

Can Autonomous, High Data Density Sensors Save Time & Costs in Adaptive Management of Hydrocarbon Impacted Sites?

Steven Mamet

Nicholas Higgs, Bipinal Unni, Amy Jimmo, Curtis Senger, & Steven Siciliano

Environmental Material Science, Inc.

Saskatoon, SK



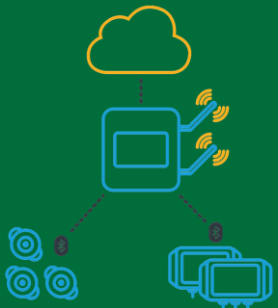
13 October 2023

EMS: Who we are and what we do

Environmental Material Science (EMS) has **10 years of R&D of real time monitoring** to improve understanding: **1) contaminated sites**, then **2) soil health and VRT optimization**, now **3) GHG emissions monitoring**.

- Technology can be utilized across all industry sectors

Real-Time Measurement



Quantify mass reduction with fringe placement

Accuracy



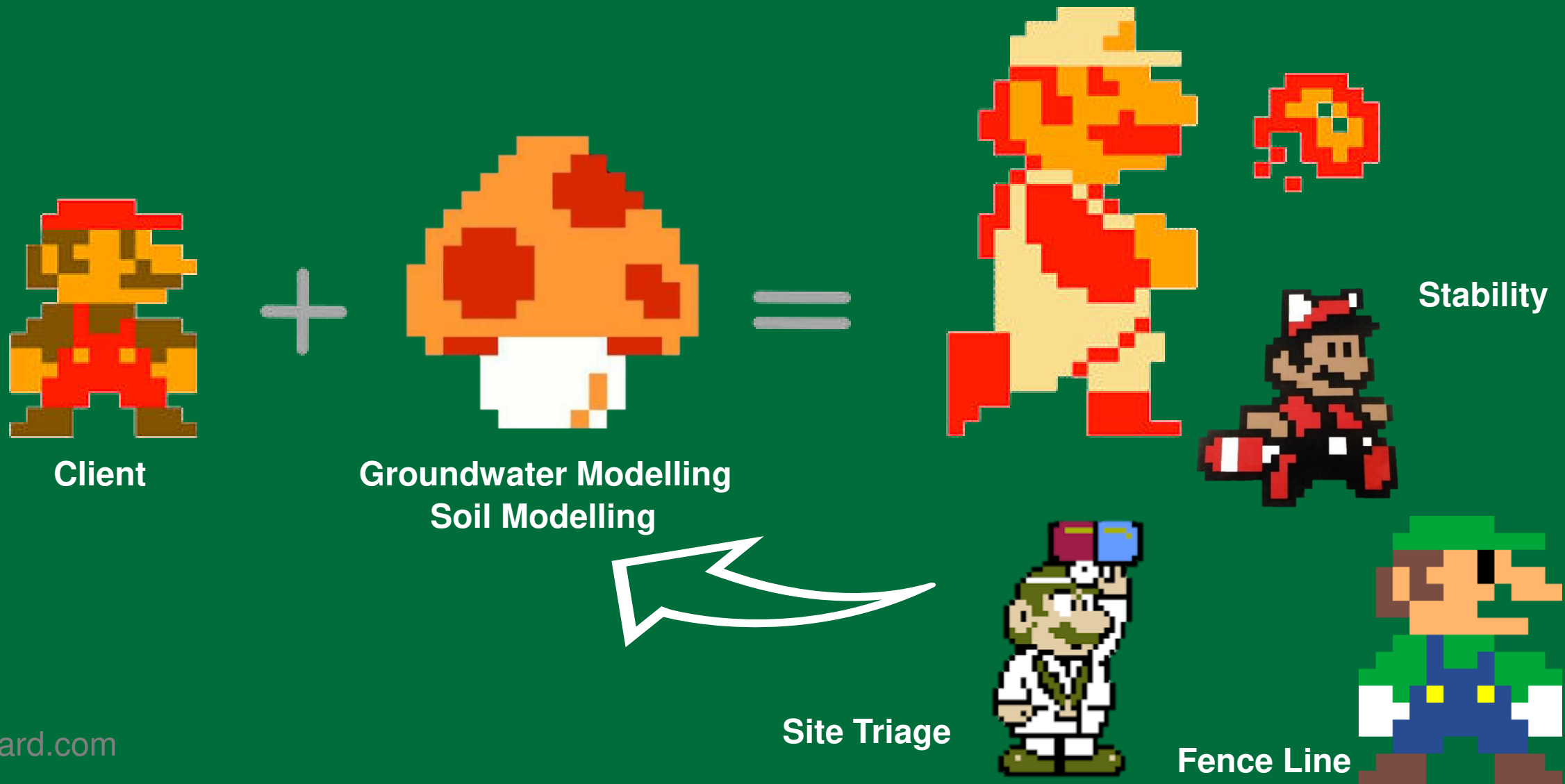
Year-round tracking = more robust site picture than PIT

Engagement

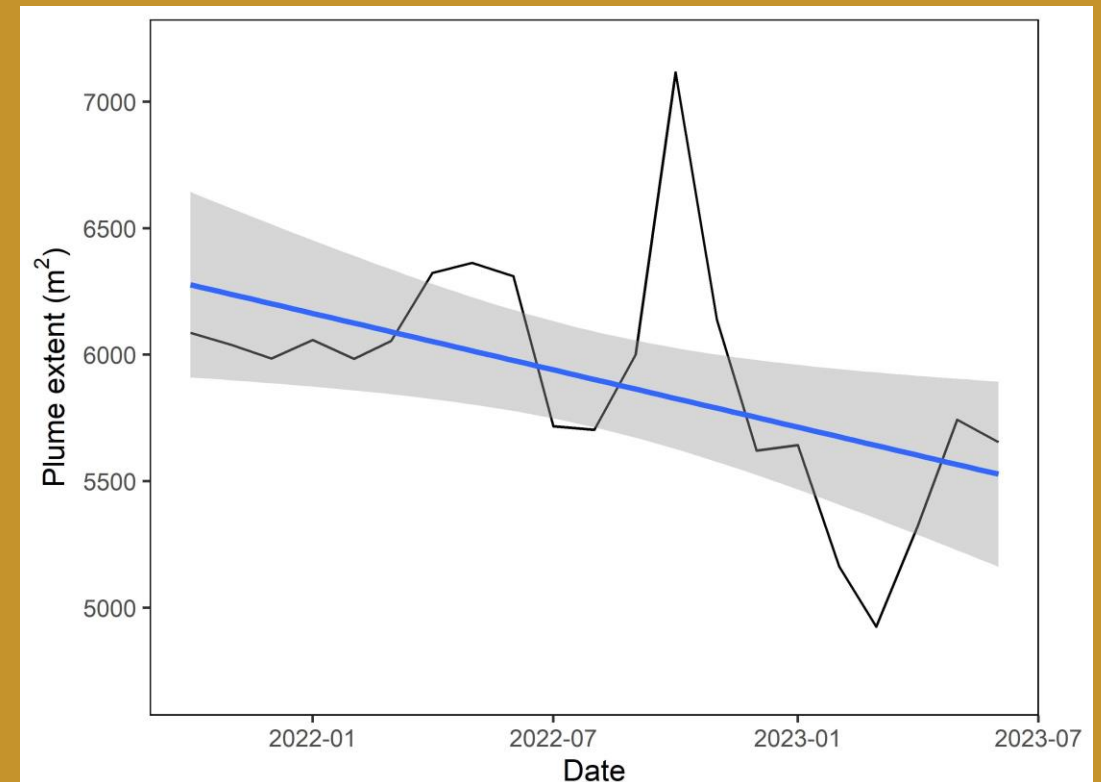


Real-time data facilitates better engagement with regulators

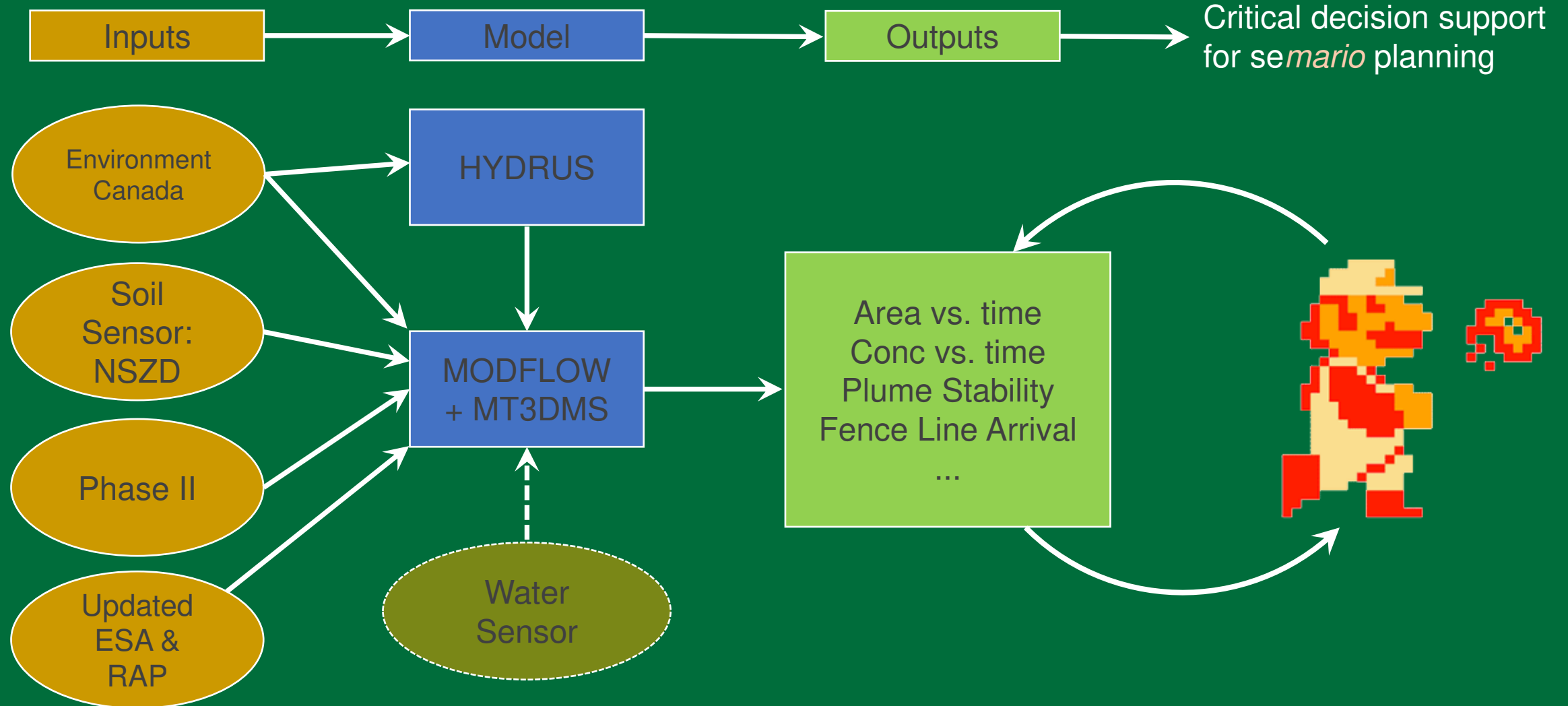
Adaptive Management Strategies



Groundwater and Soils tell us two different facets of a contaminated site



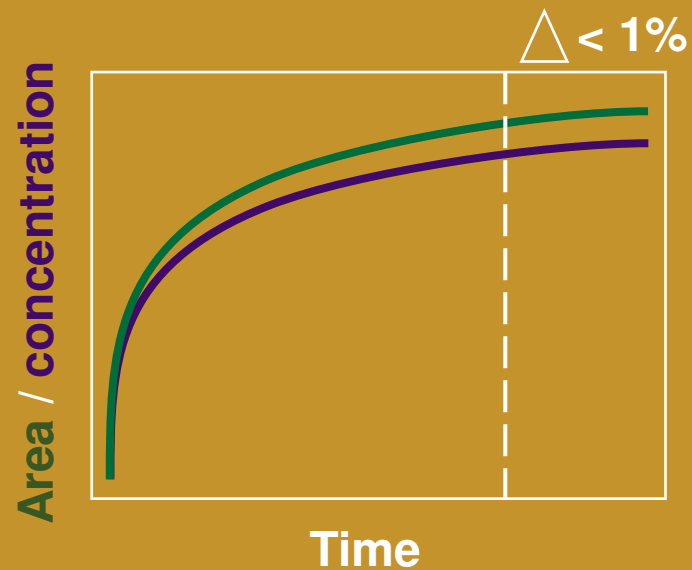
Groundwater Modeling (Benzene)



Groundwater Modeling (Benzene)

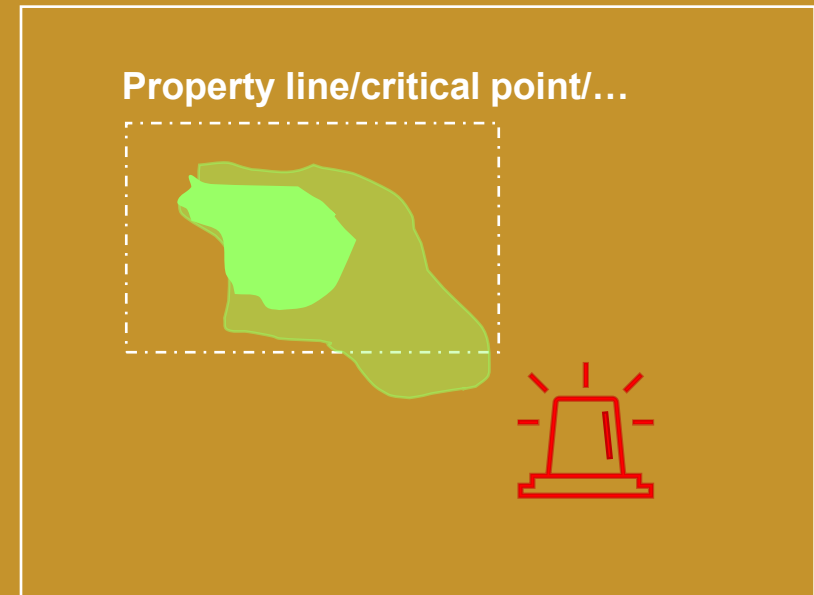
Plume Stability

When does the plume stop expanding and/or moving?



Fence Line Arrival

Does the plume hit a critical receptor?



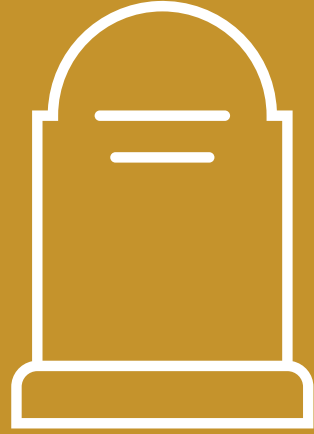
Groundwater Modeling (Benzene)

THREE scenarios

- Do nothing
- NSZD
- Biostimulation

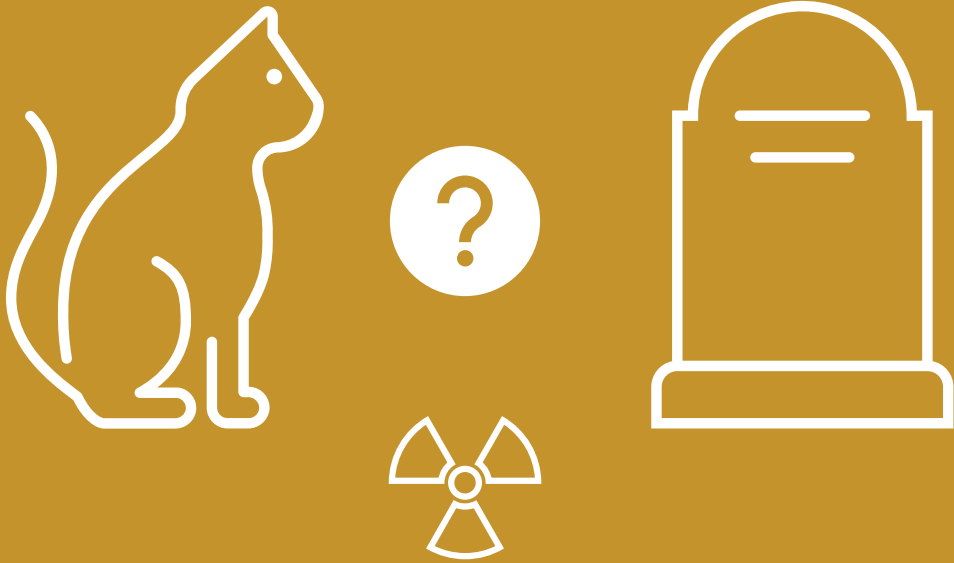
Schrödinger's Cat

Physics



Schrödinger's ~~Cat~~ Goomba (NSZD)

Physics

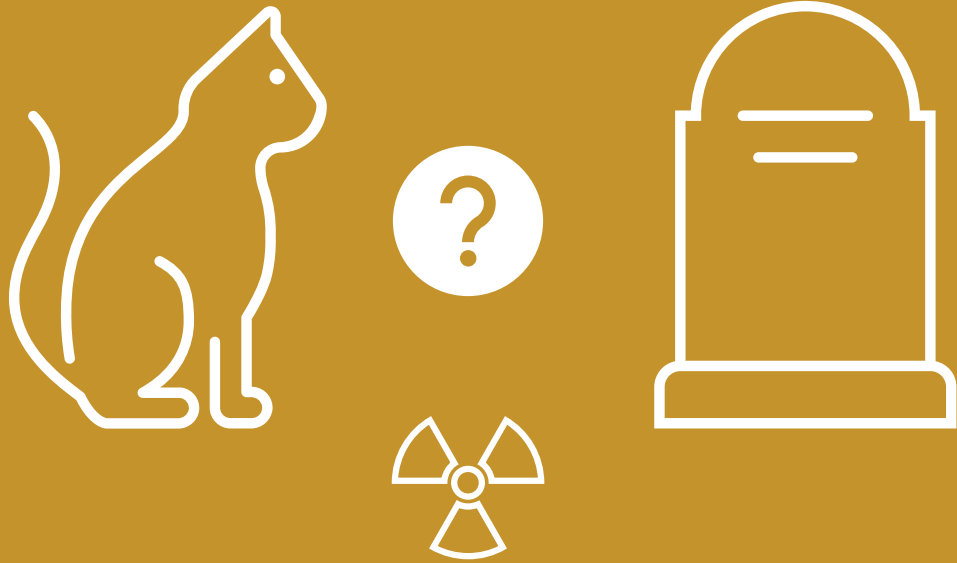


Groundwater
Risk



Schrödinger's Cat Goomba (NSZD)

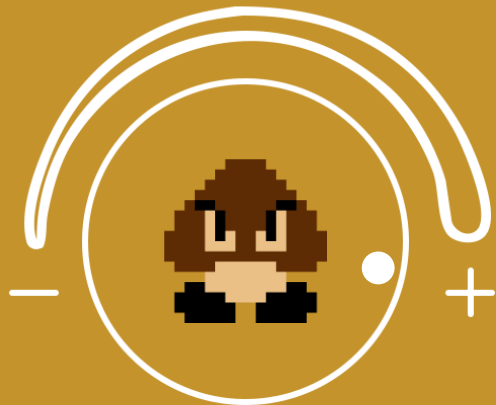
Physics



Plume Size
w/ no NSZD



Groundwater
Risk



NSZD



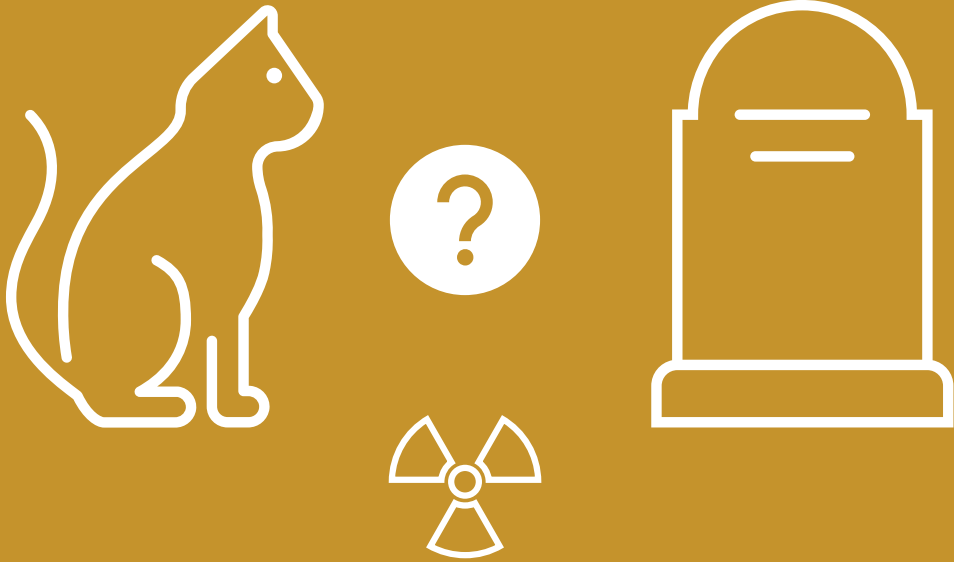
modflow



Risk

Schrödinger's Cat Goomba (NSZD)

Physics



Plume Size
w/ no NSZD



Groundwater
Risk



NSZD



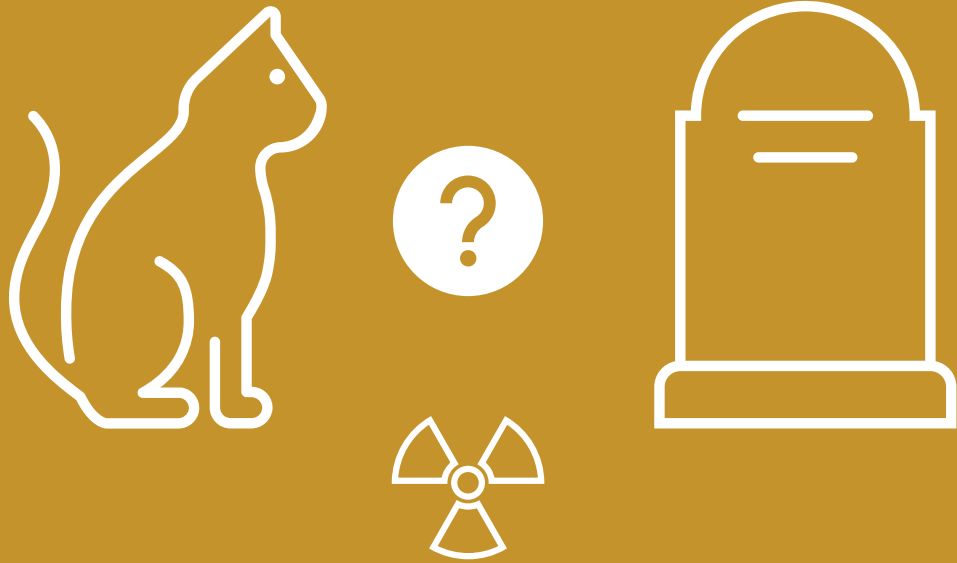
modflow



Risk

Schrödinger's Cat Goomba (NSZD)

Physics



Plume Size
w/ NSZD



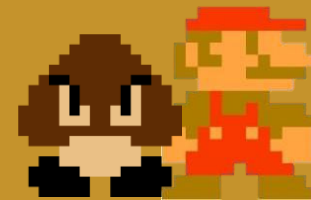
Groundwater
Risk



modflow



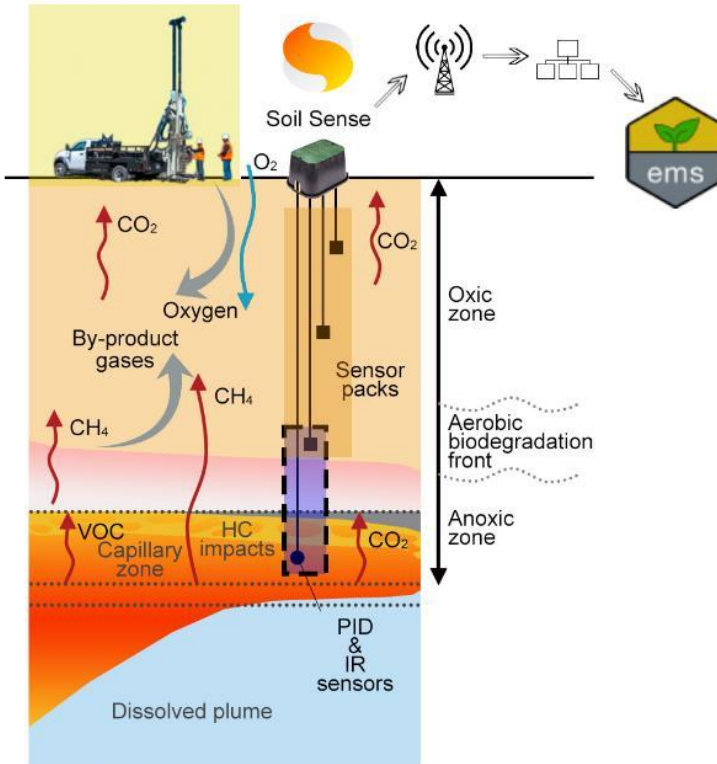
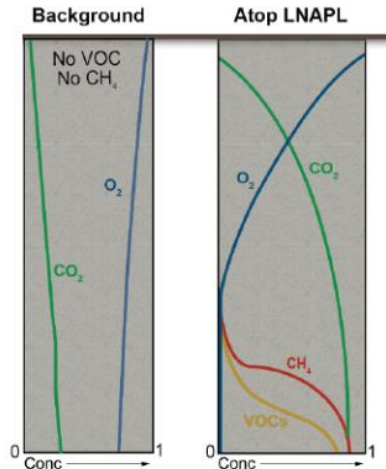
↓ Risk



NSZD: Concentration Gradient Method

Site-Specific

Empirical stoichiometric conversion: benzene-CO₂



CRC Care 2018

$$1. J = D_v^{eff} \left(\frac{dC}{dz} \right)$$

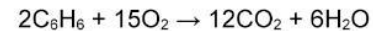
J = steady state diffusive flux (g/m²-soil/s)

D_v^{eff} = effective vapour diffusion coefficient (m²/s)

$\frac{dC}{dz}$ = soil gas concentration gradient (g/m³-m)

$$2. J_{impacted} - J_{background} = J_{corrected}$$

3. Theoretical stoichiometric conversion: benzene-CO₂



$$2C_6H_6: 2 \times (12.011 \text{ g/mol} \times 6 + 1.008 \text{ g/mol} \times 6) = 156.223 \text{ g/mol}$$

$$12CO_2: 12 \times (12.011 \text{ g/mol} + 15.999 \text{ g/mol} \times 2) = 528.096 \text{ g/mol}$$

When 156 g of C₆H₆ are consumed, 528 g CO₂ are produced

$$Stoich_{CO_2} = 156 / 528$$

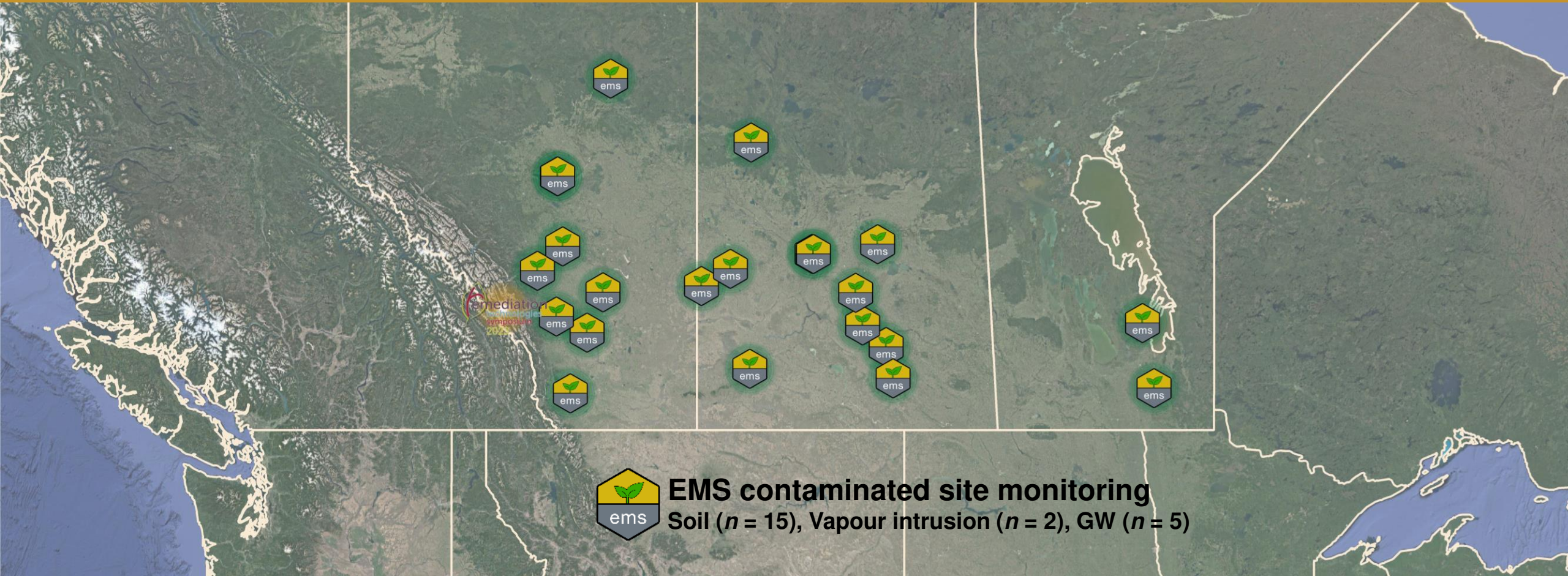
4. Natural Source Zone Depletion

$$NSZD = J_{corrected} \times Stoich_{CO_2}$$

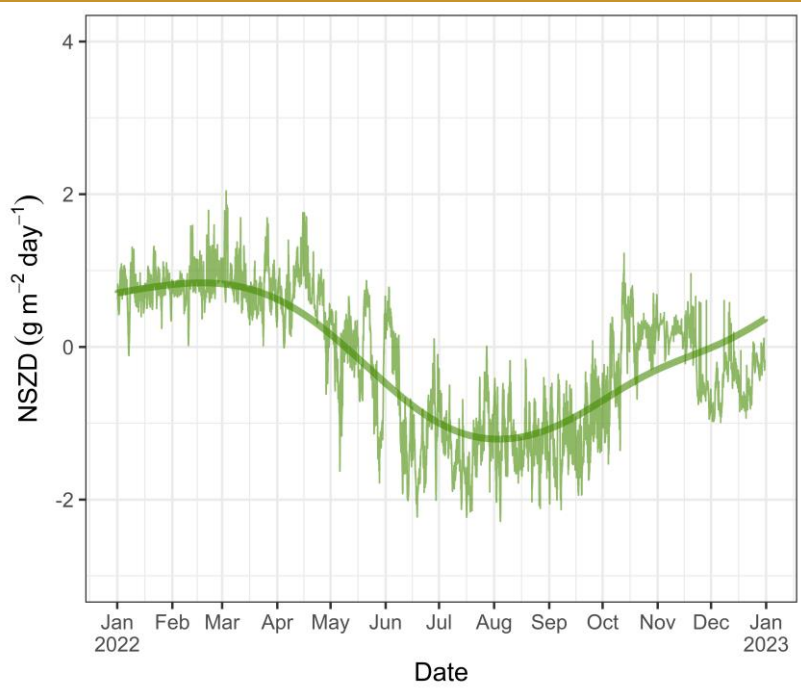
Persistent Measurements

Hydrocarbons (source) and CO₂ efflux (products)

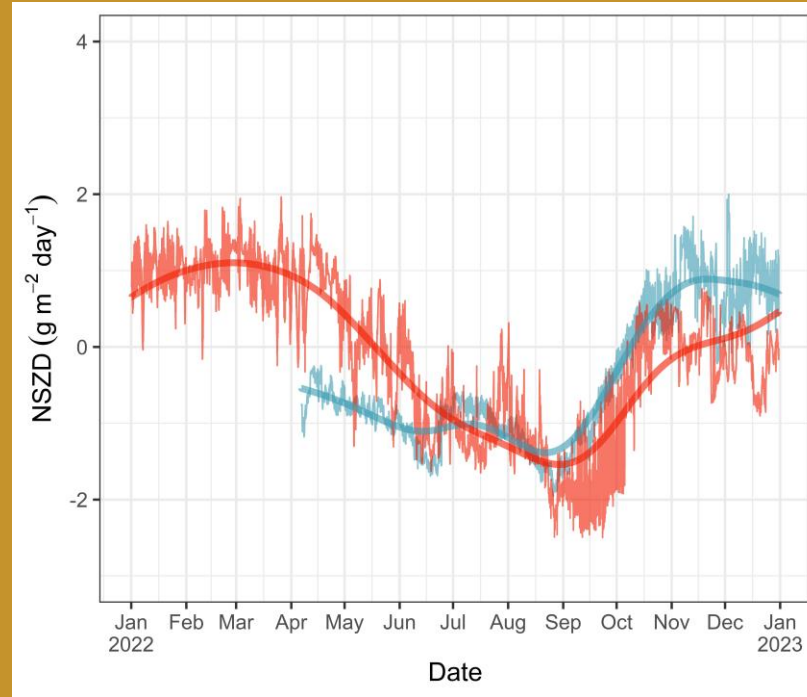
NSZD: Concentration Gradient Method



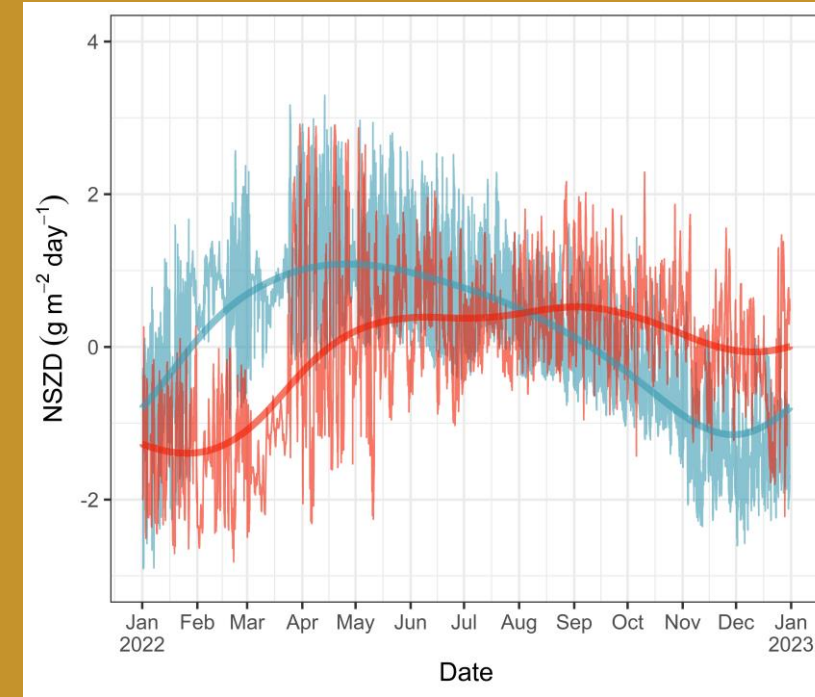
NSZD: Seasonal Patterns



**Regional mean
(normalized)**



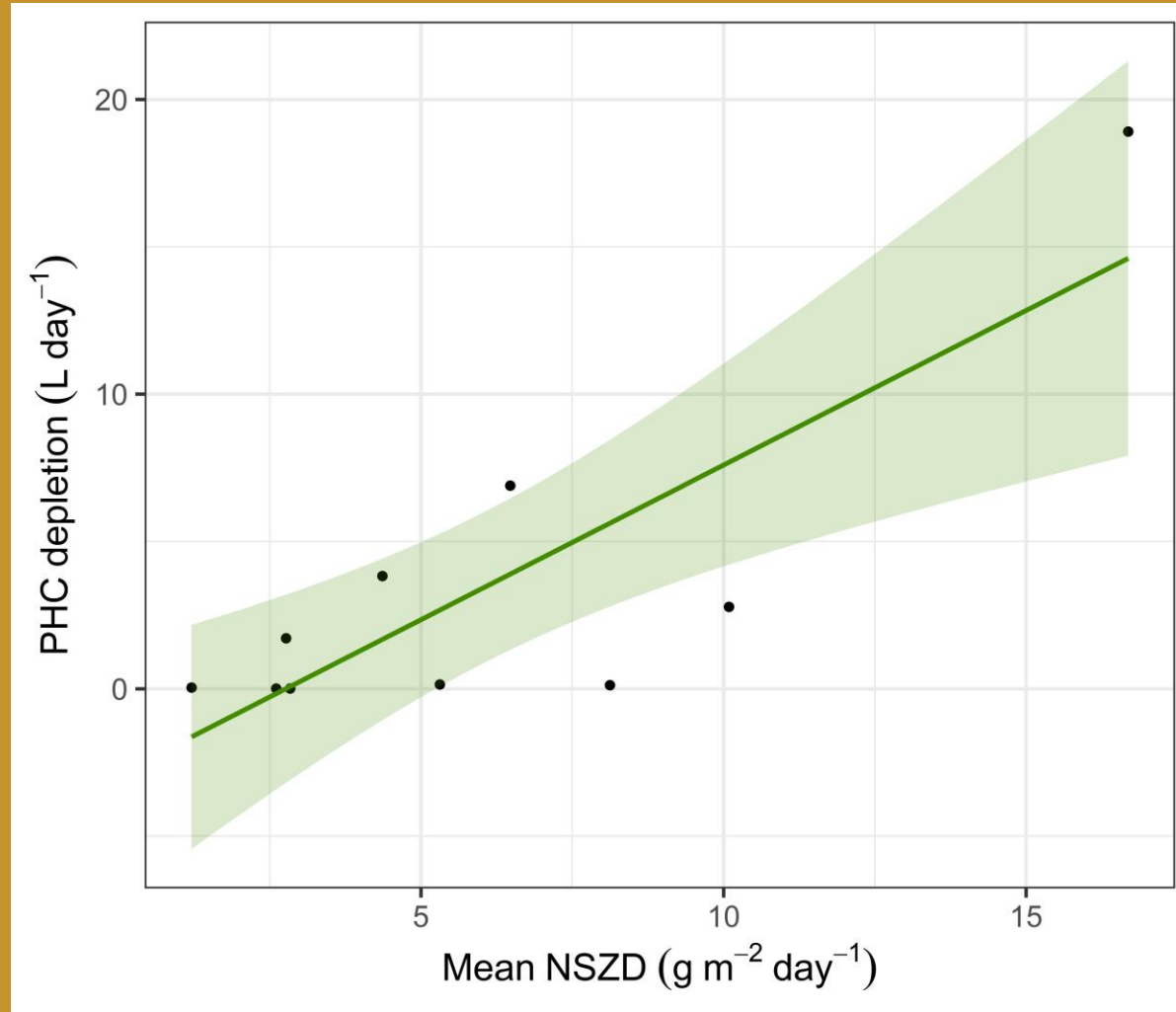
**Some sites follow
regional mean**



Others, not so much

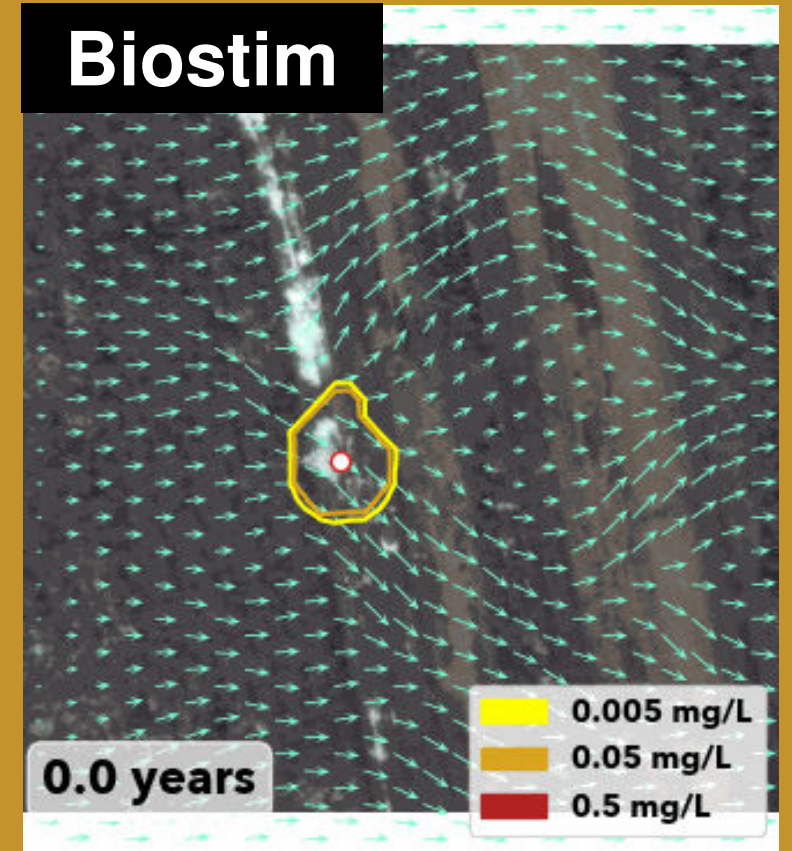
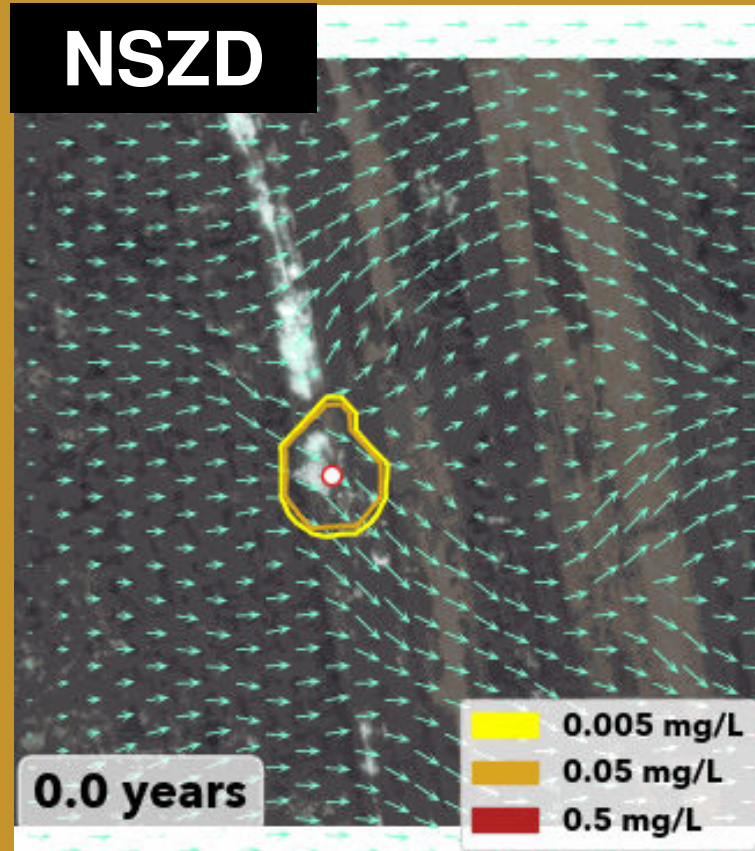
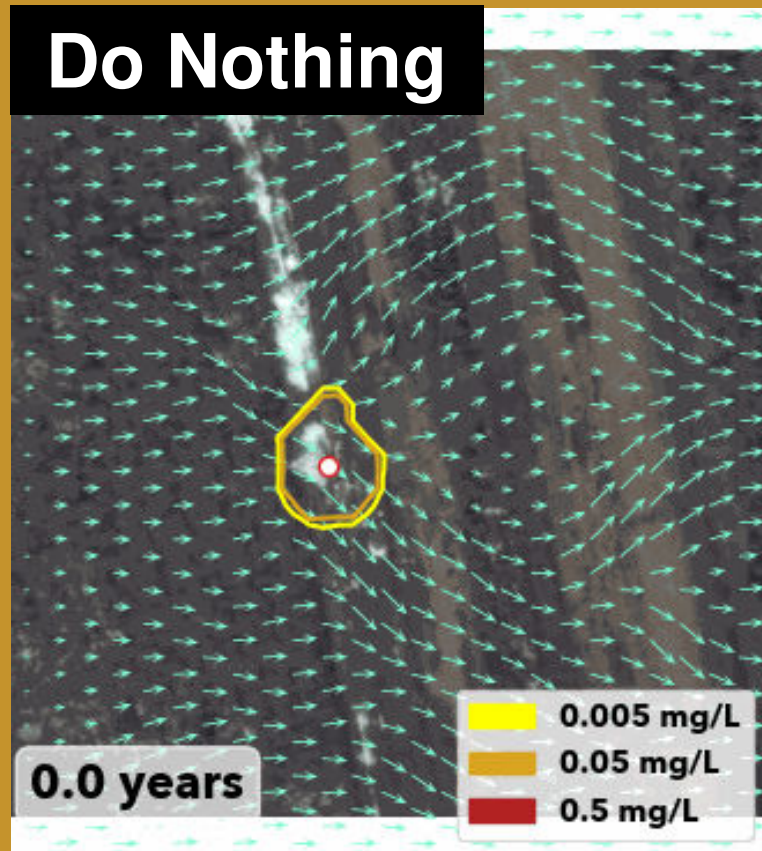
NSZD meets PHC depletion

Estimated measuring hydrocarbons directly

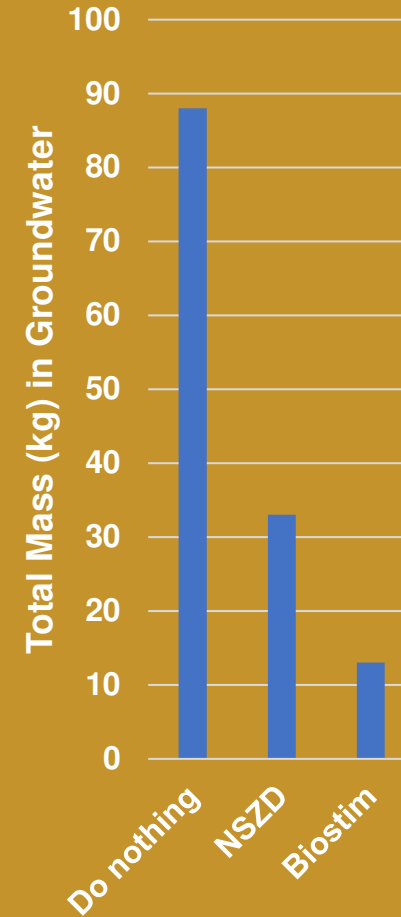
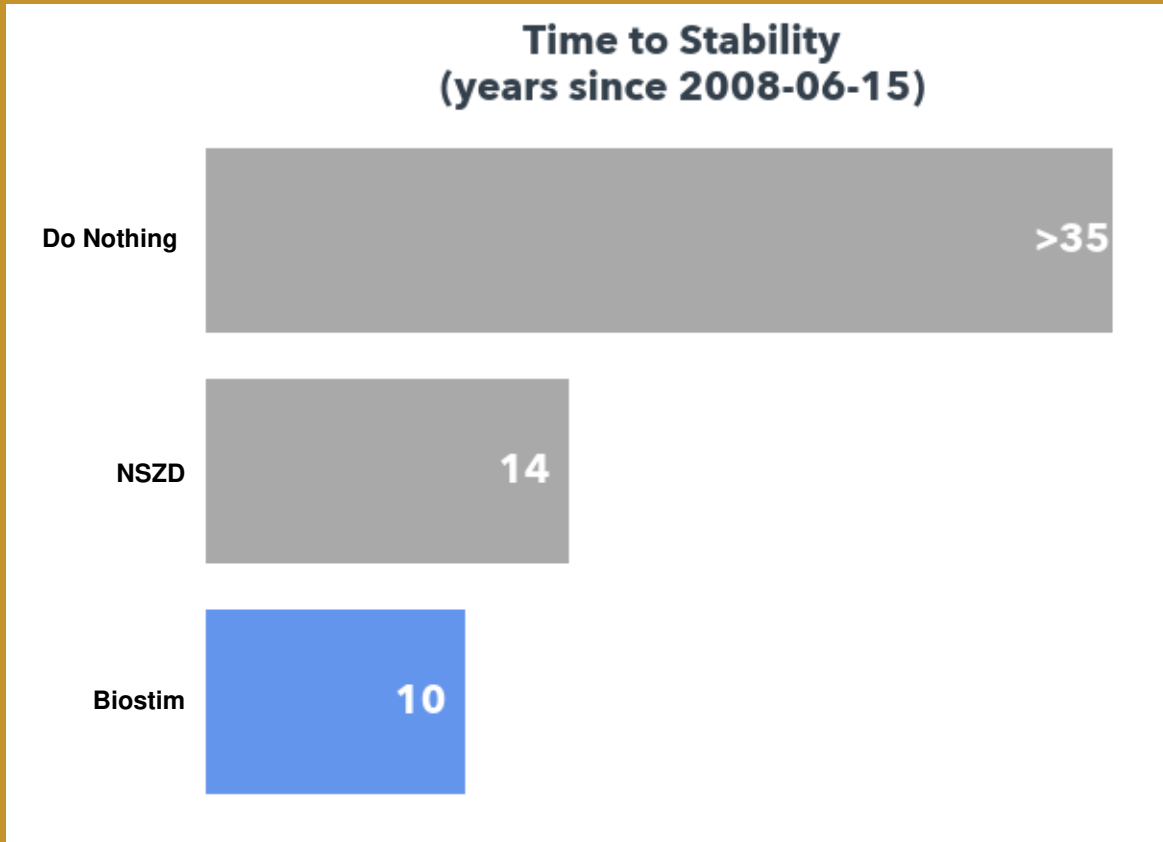


Estimated using indicator gases

Groundwater Modeling (Benzene)

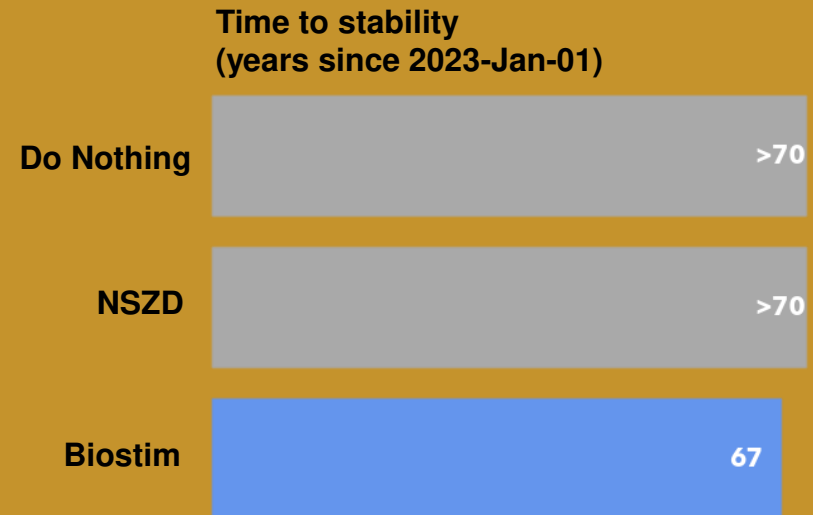
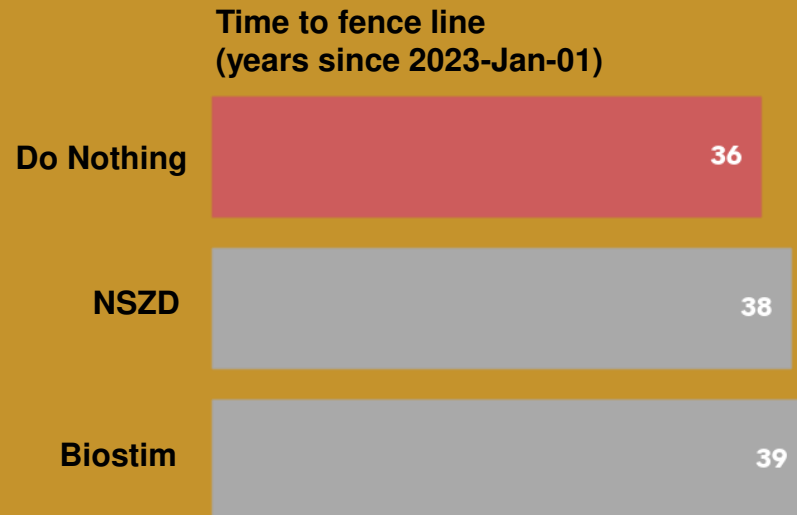


Groundwater Modeling (Benzene)

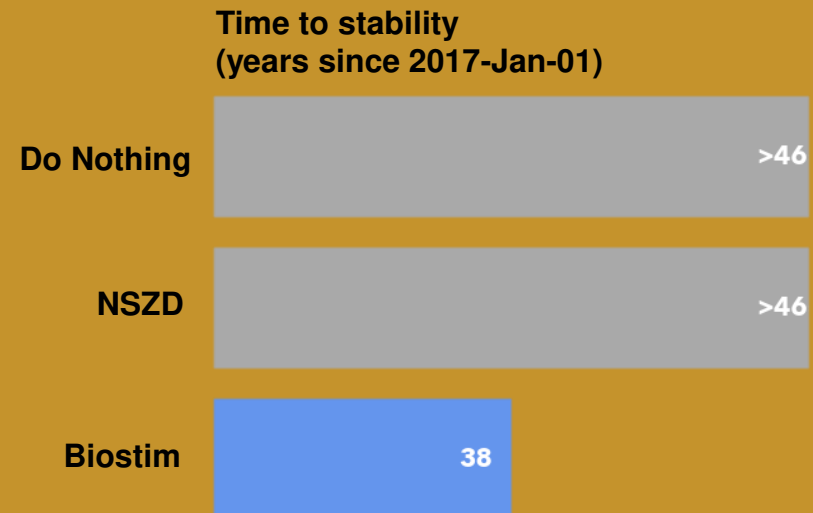
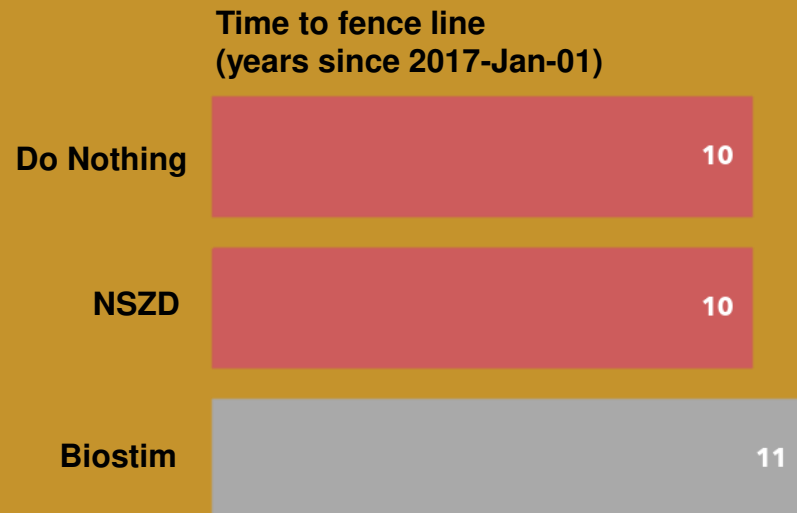
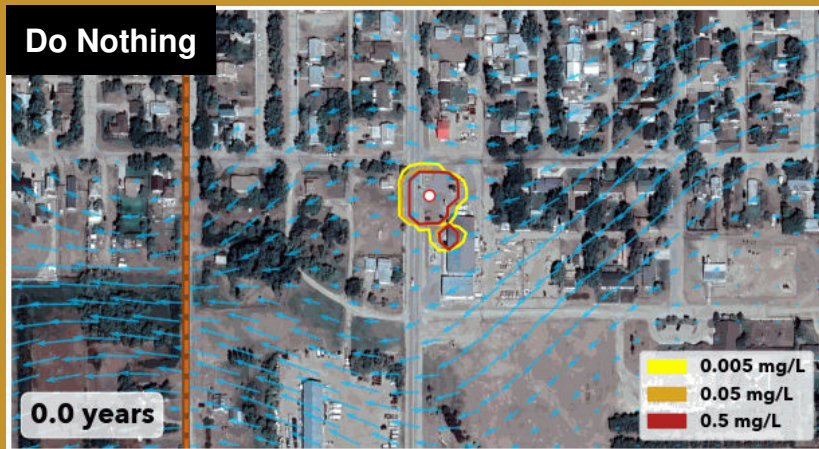


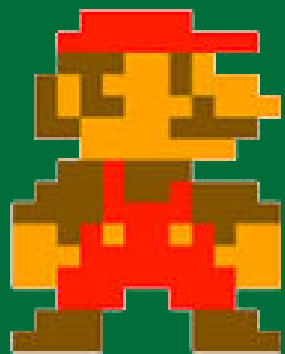
Time to stability based on area and concentration

Groundwater Modeling (Benzene)



Groundwater Modeling (Benzene)





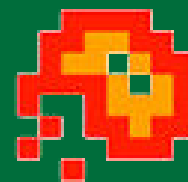
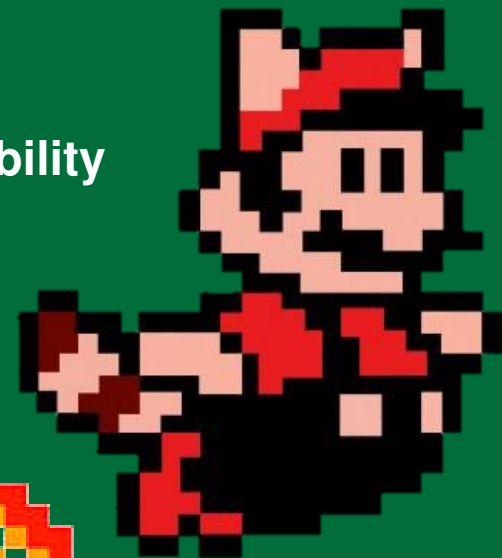
+



=



Stability



Site Triage



Highlights: Groundwater Risk and NSZD

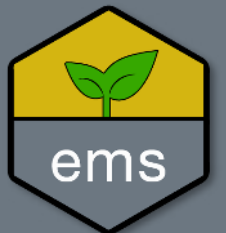
- Groundwater modeling is about changing risk vector over time – soil may be PHC source to GW
- Soil modeling is about remediation - **source** removal
- Comprehensive groundwater (risk) model should incorporate stimulated and/or natural depletion
- Continuous, real-time data streams empower managers to:
 - Make evidence-based decisions
 - Adapt management strategies to achieve site objectives in a more **timely** and **cost-effective** fashion.

Can Autonomous, High Data Density Sensors Save Time & Costs in Adaptive Management of Hydrocarbon Impacted Sites?

Thank you!

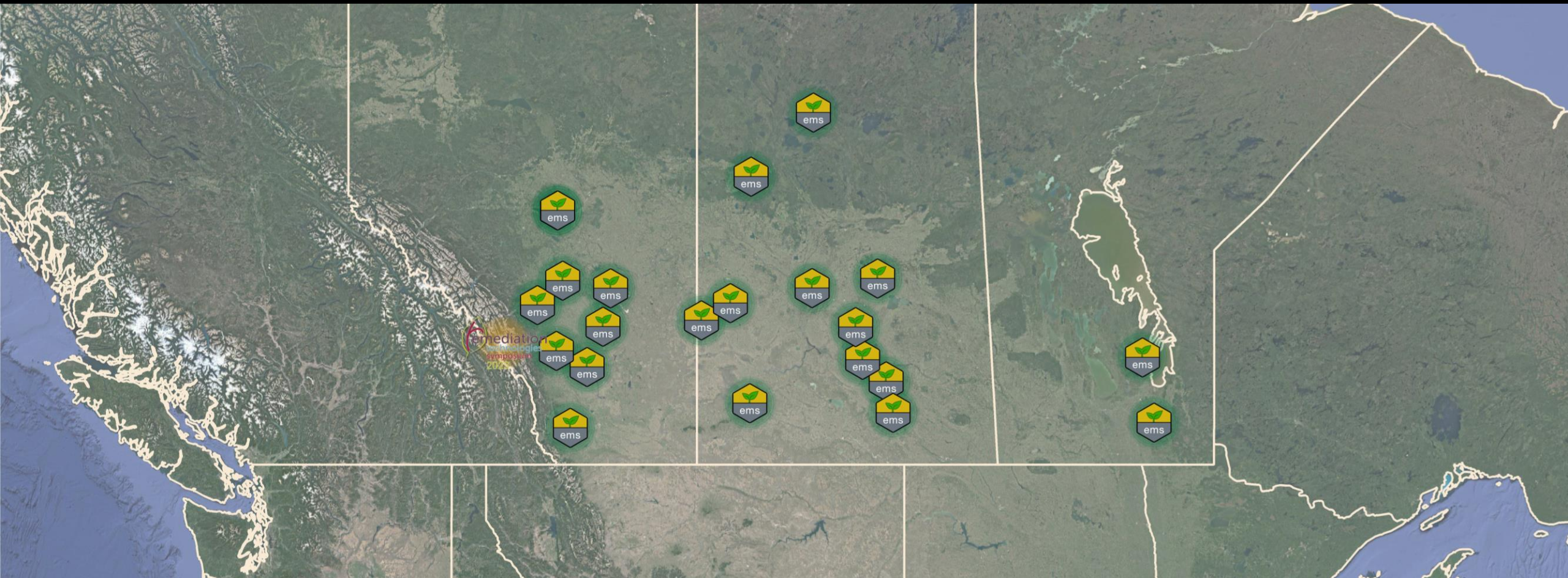
1(306)952-0403
stevemamet@ems-inc.ca
stevesiciliano@ems-inc.ca

<https://ems-inc.ca/>



Your roadmap for today





Mediation
Technology
Symposium
2022



Natural Source Zone Depletion Explained

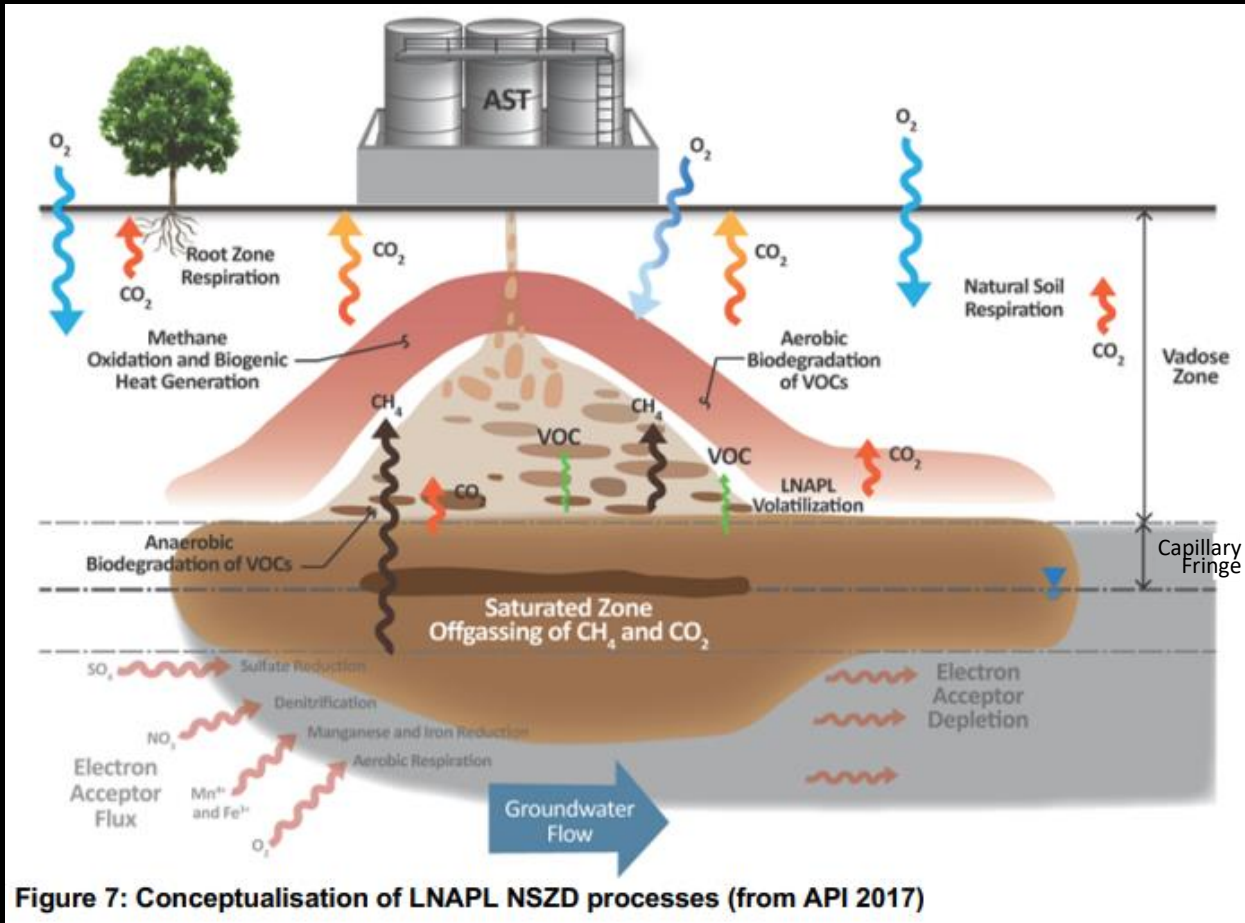
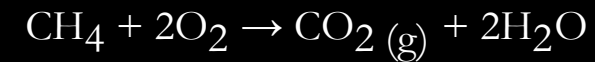


Figure 7: Conceptualisation of LNAPL NSZD processes (from API 2017)

← **Aerobic Transport**

← **Methane Oxidation**



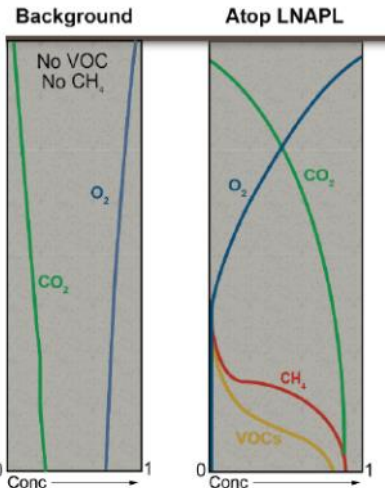
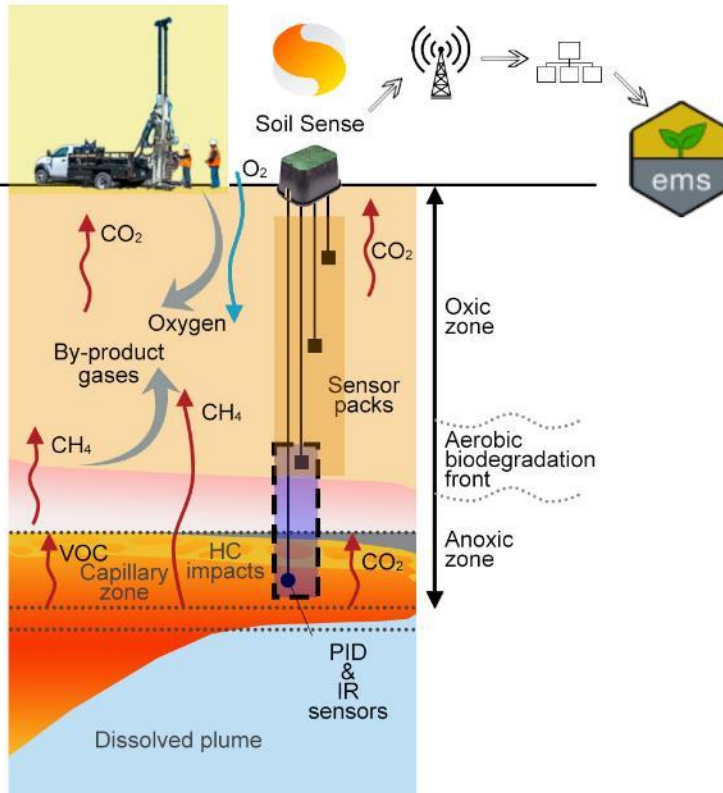
← **Methane Generation**



Concentration Gradient Method

Site-Specific

Empirical stoichiometric conversion: benzene-CO₂



Fick's First Law of Diffusion - rate of diffusion is proportional to the concentration and surface area

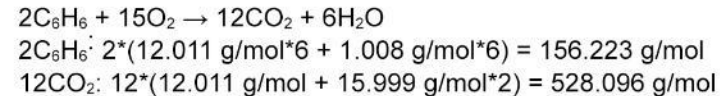
CRC Care 2018

$$1. J = D_v^{eff} \left(\frac{dC}{dz} \right)$$

J = steady state diffusive flux (g/m²-soil/s)
 D_v^{eff} = effective vapour diffusion coefficient (m²/s)
 $\frac{dC}{dz}$ = soil gas concentration gradient (g/m³-m)

$$2. J_{impacted} - J_{background} = J_{corrected}$$

3. Theoretical stoichiometric conversion: benzene-CO₂



When 156 g of C₆H₆ are consumed, 528 g CO₂ are produced
 $Stoich_{CO_2} = 156 / 528$

4. Natural Source Zone Depletion

$$NSZD = J_{corrected} * Stoich_{CO_2}$$

Persistent Measurements

Hydrocarbons (source) and CO₂ efflux (products)

Why CH₄ and why high data density?

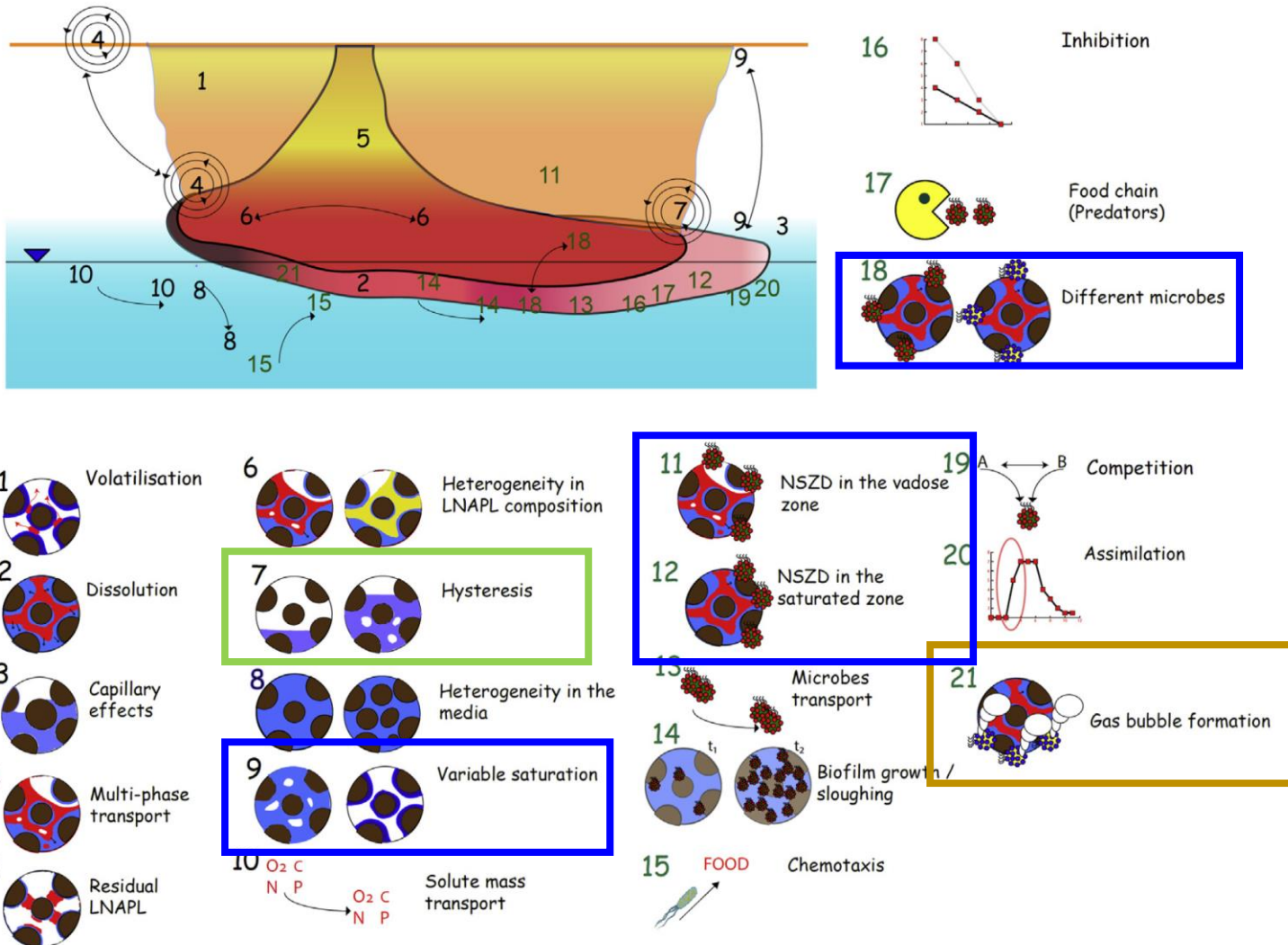
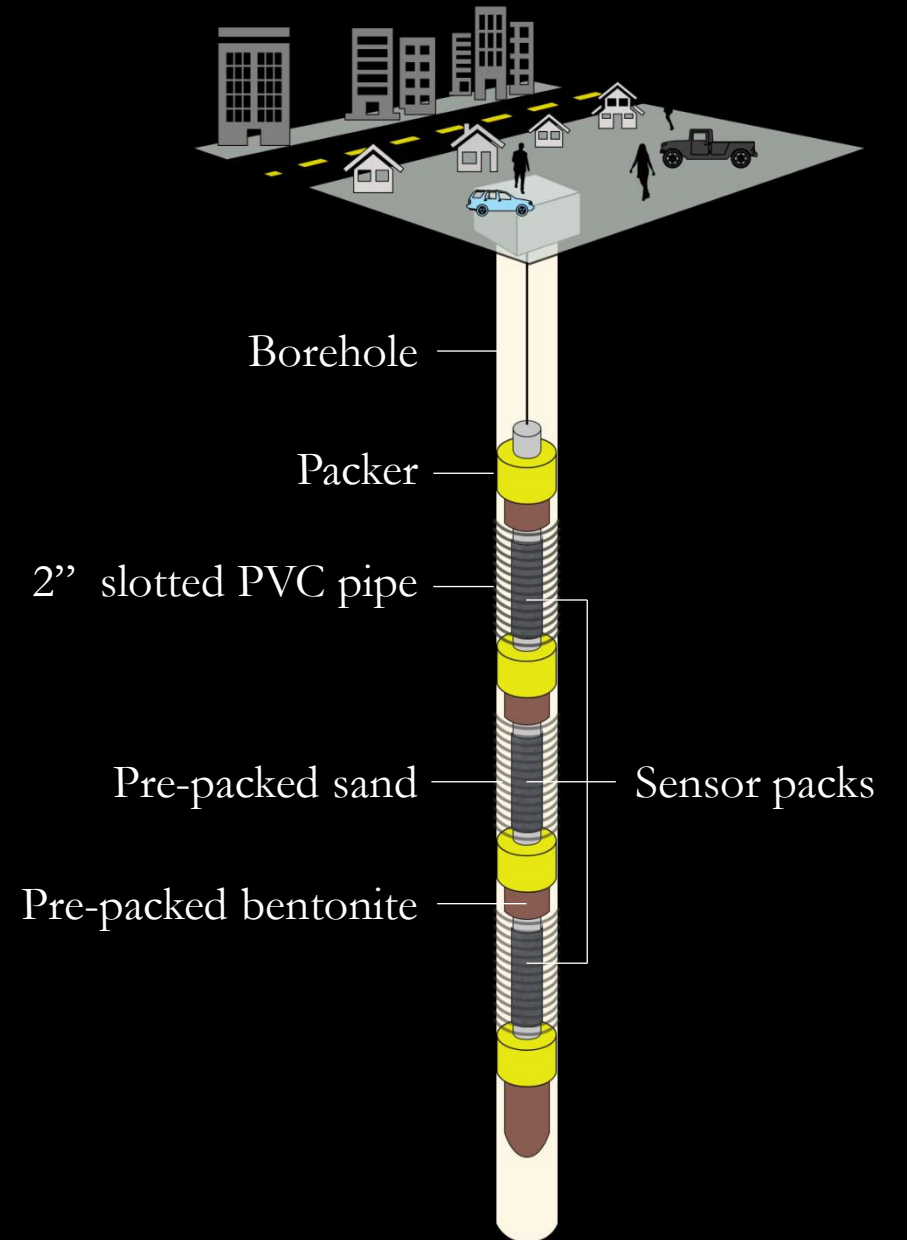
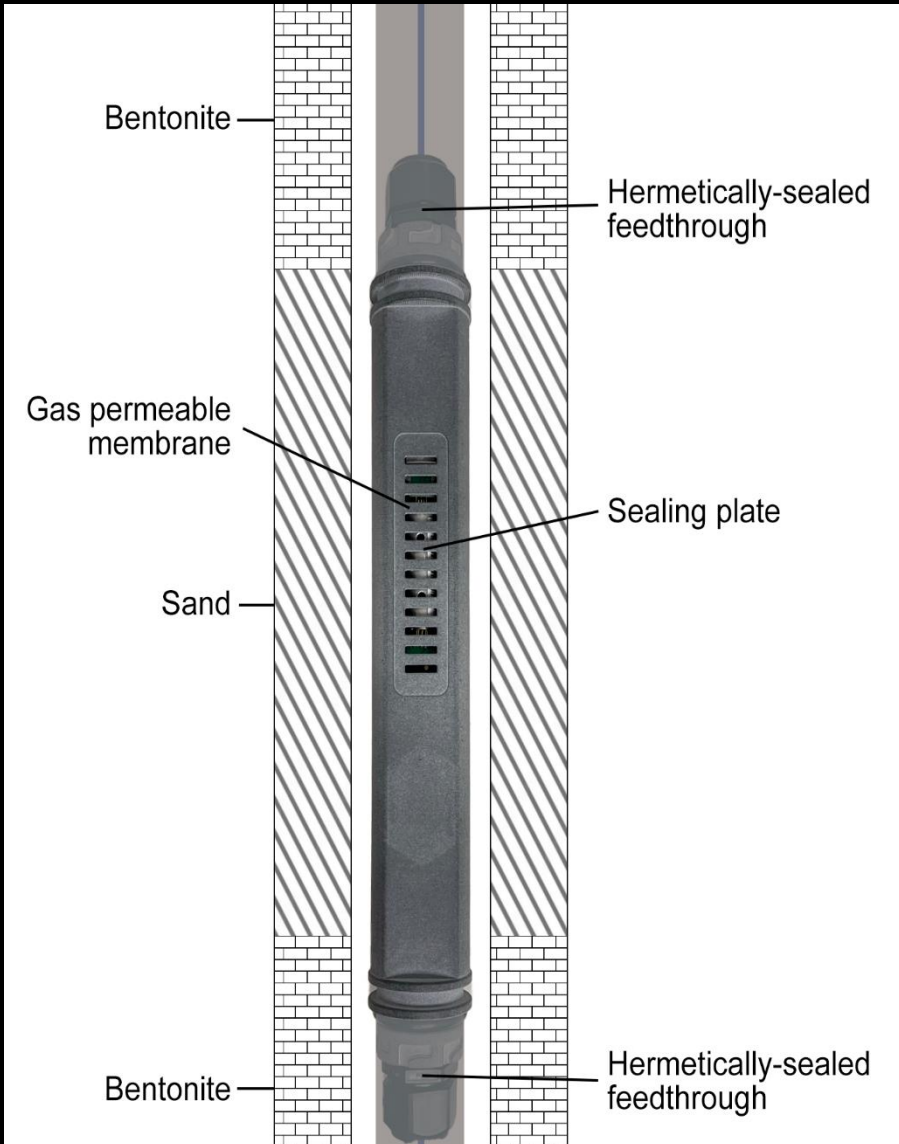


Fig. 1. Some of the major processes and parameters in NSZD.

- Early attenuation processes dominated by methanogenic degradation
- “Signal shredding”
- Lag between peak subsurface gas concentrations and peak surface efflux.

Soil Sense



Soil Sense

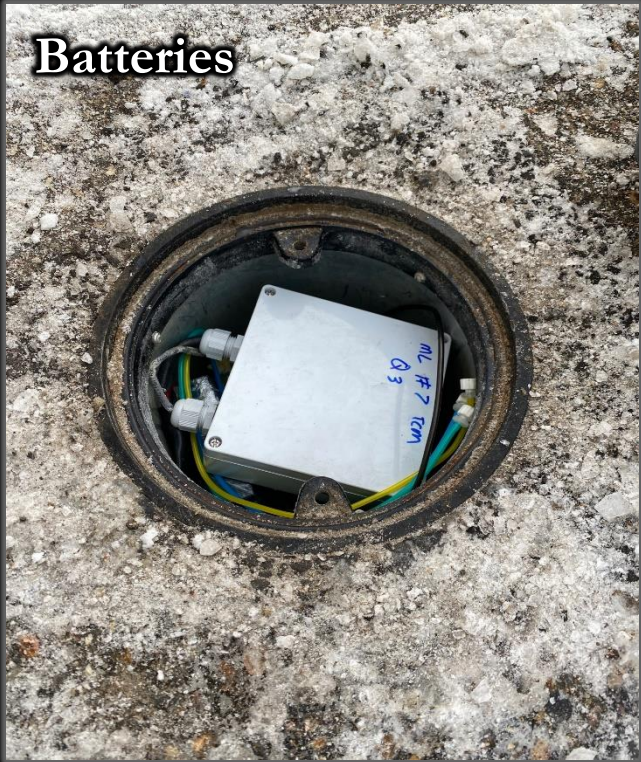
Hardwired



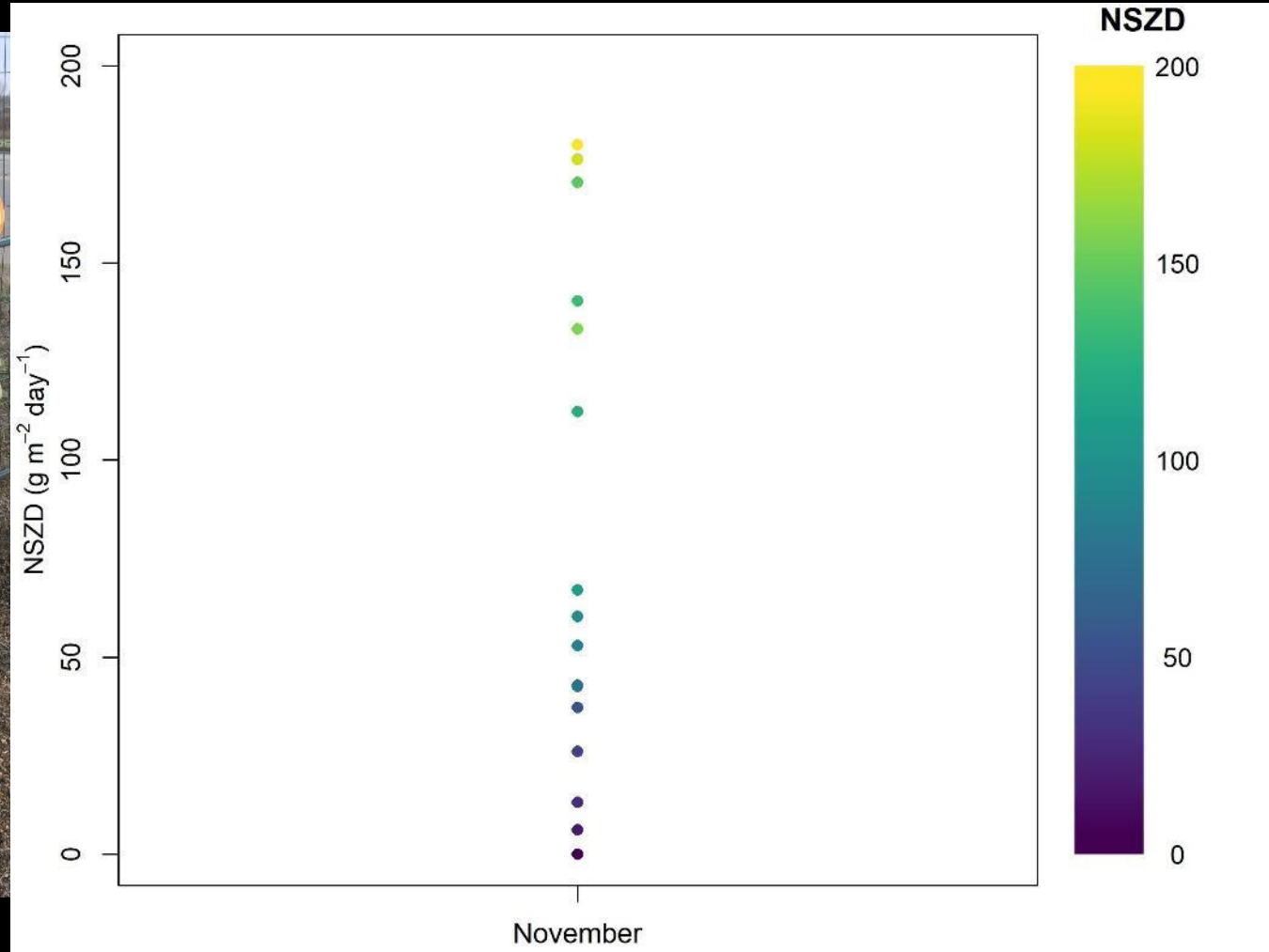
Solar



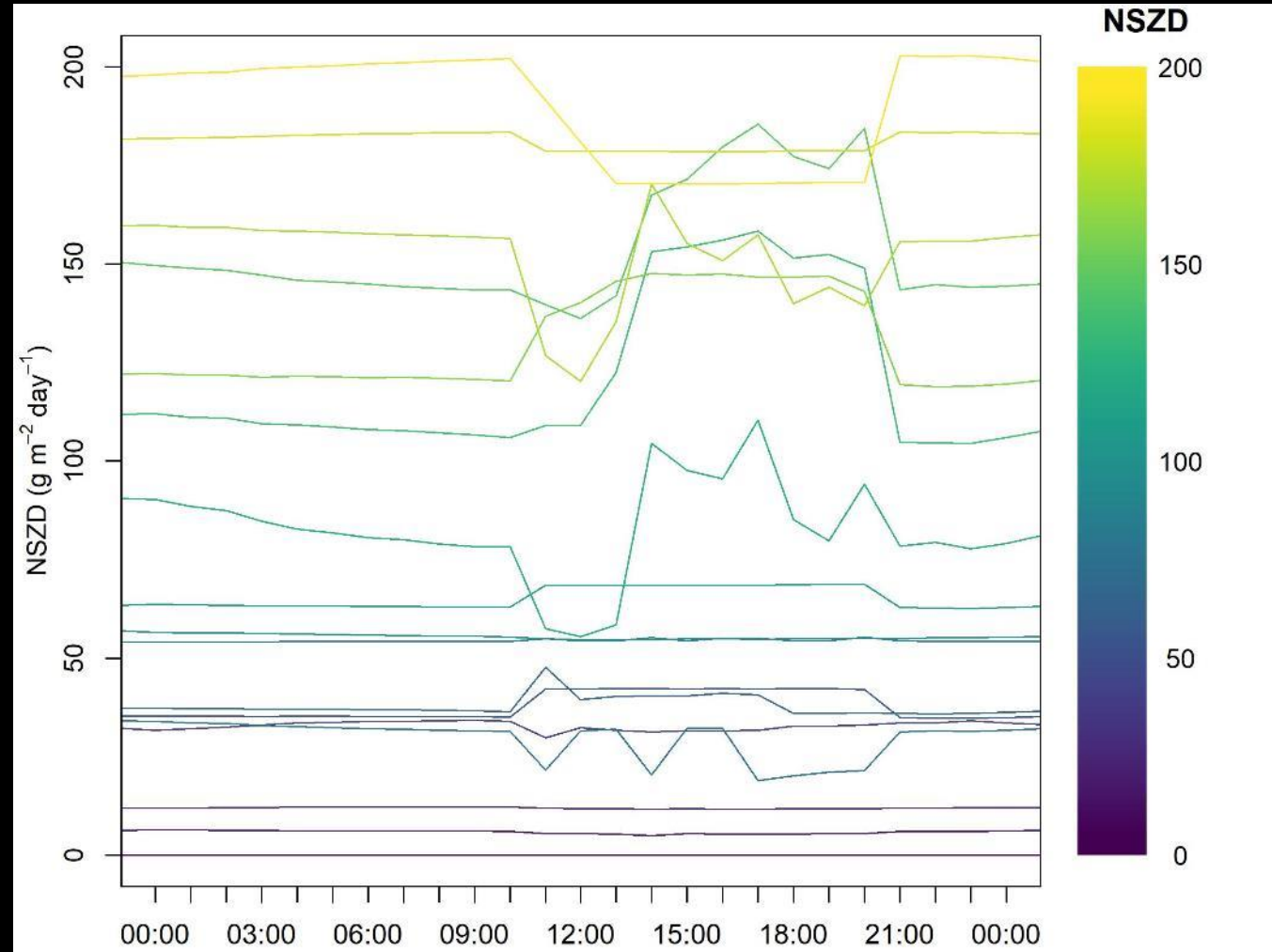
Batteries



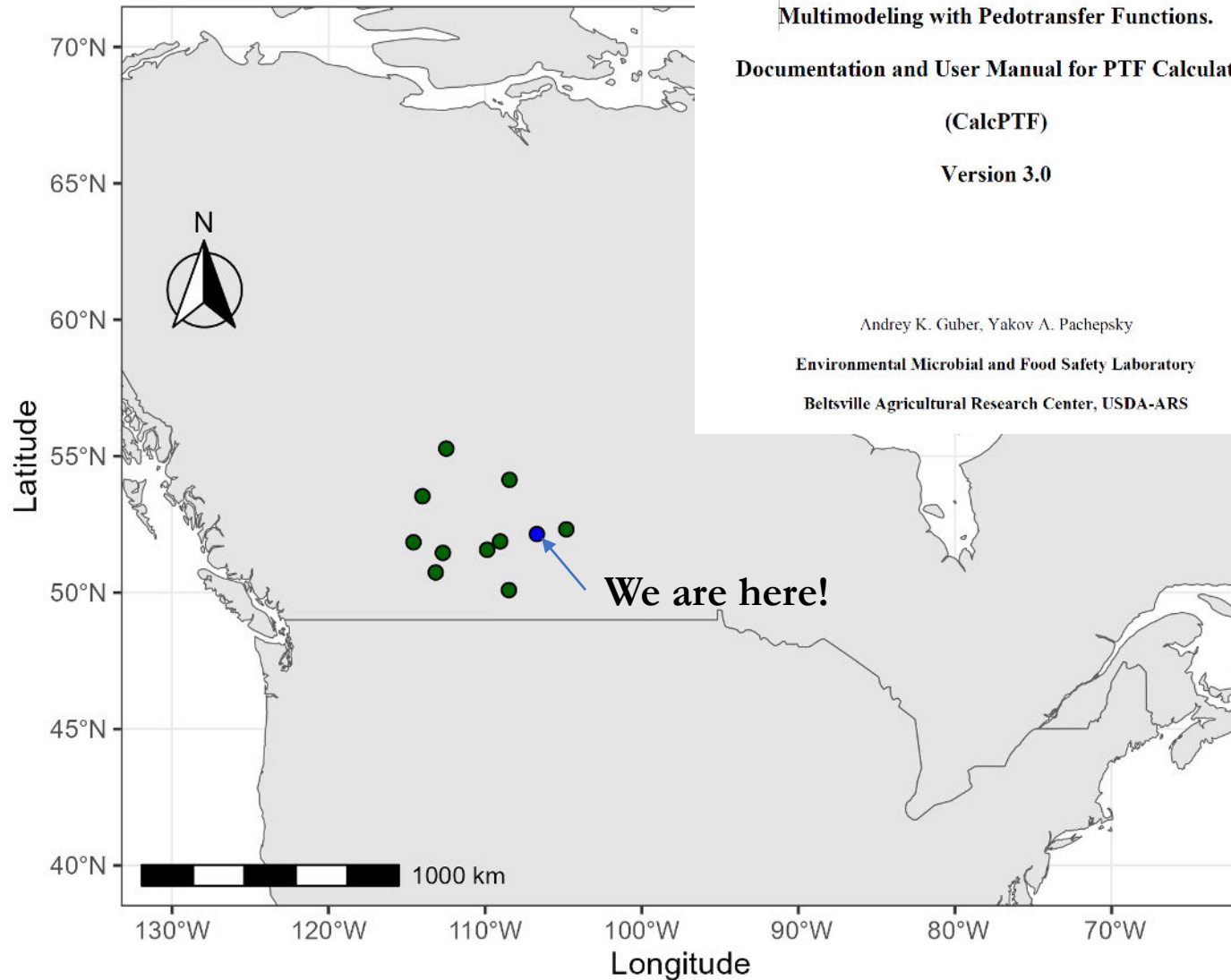
Low sampling frequency (point-in-time is the “**pit_s**”)



Soil Sense collects 48 measurements per day

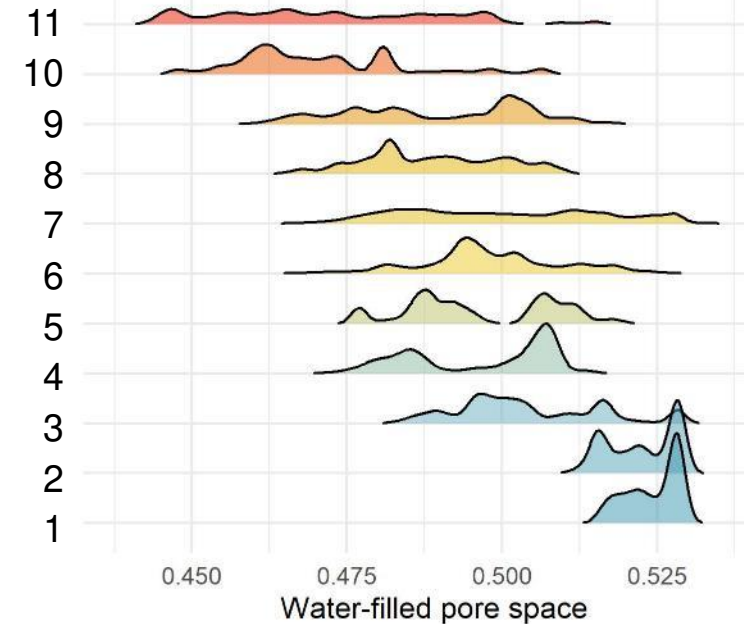
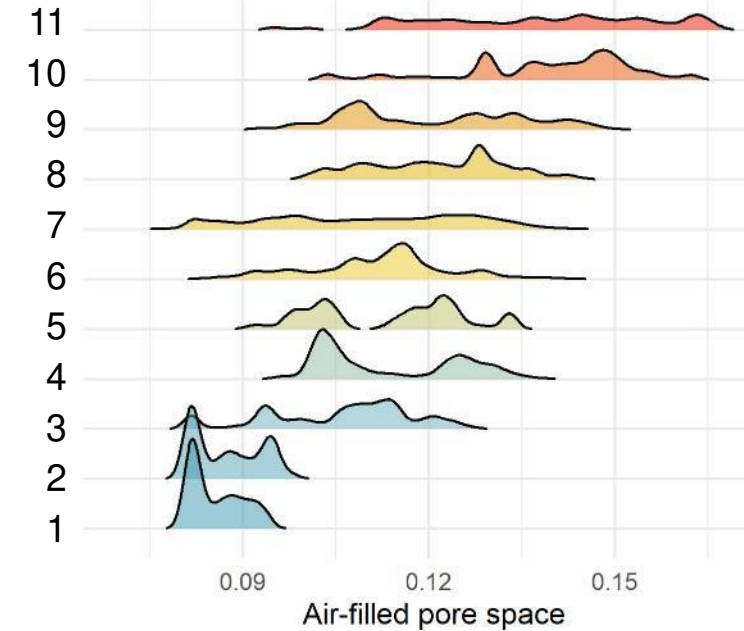


The Sites



Multimodeling with Pedotransfer Functions.
Documentation and User Manual for PTF Calculator
(CalcPTF)
Version 3.0

Andrey K. Guber, Yakov A. Pachepsky
Environmental Microbial and Food Safety Laboratory
Beltsville Agricultural Research Center, USDA-ARS

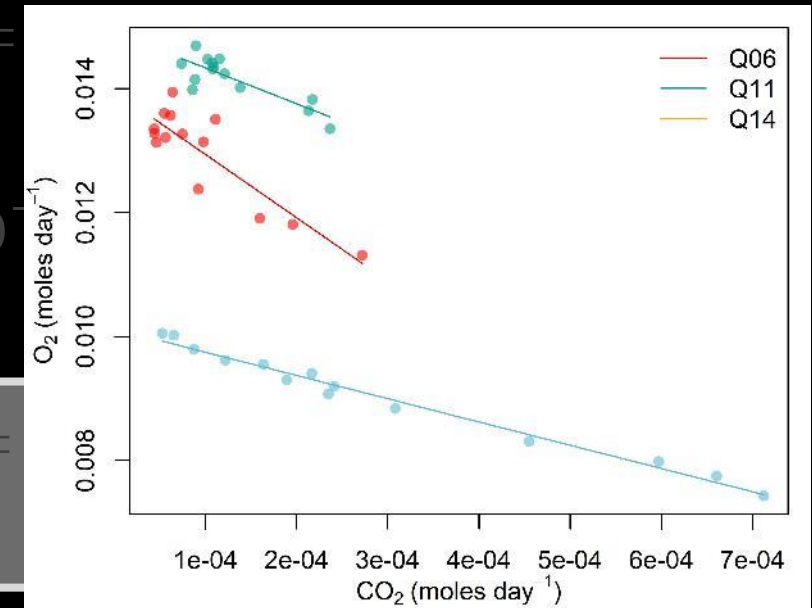
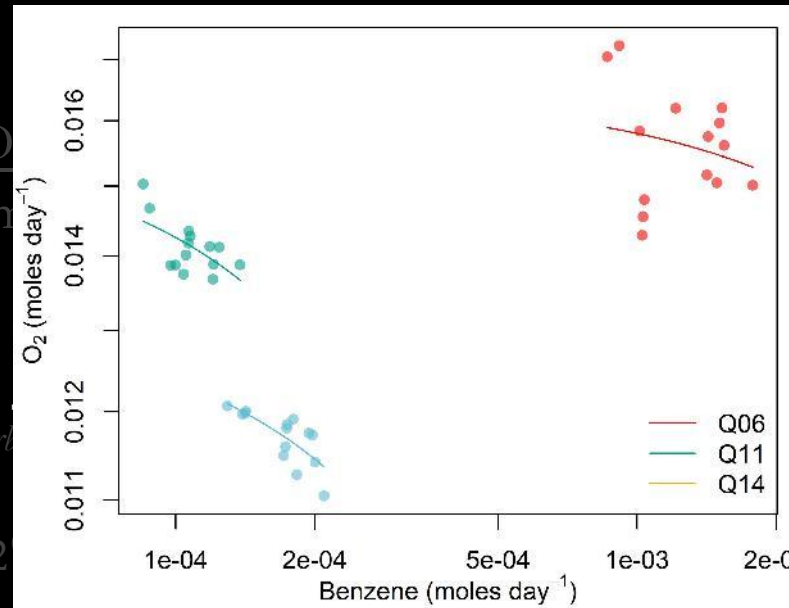
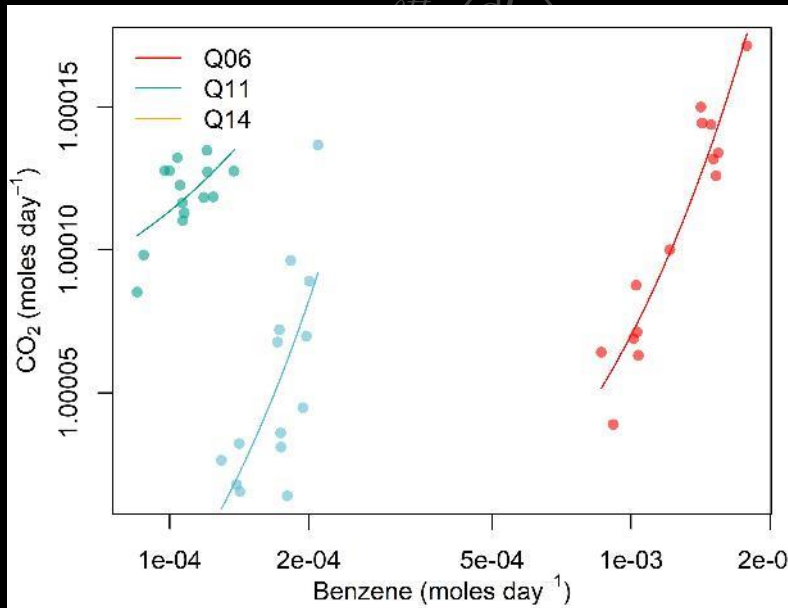


Guidance values



Convert Flux to Hydrocarbon Depletion (GUIDANCE)

CONVERT FLUX TO HYDROCARBON DEPLETION (SITE-SPECIFIC)



$$\text{Molar Ratio} = \frac{78.11 \frac{\text{gC}_6\text{H}_6}{\text{mol}}}{44.11 \frac{\text{gCO}_2}{\text{mol}}} \cdot 2 \text{ molC}_6\text{H}_6$$

Ratio	Guidance (mass)	Site-specific (mass)
C ₆ H ₆ :CO ₂	0.296	0.329
C ₆ H ₆ :O ₂	0.325	0.179
C ₆ H ₆ :CH ₄	1.038	1.038

Convert Flux to Hydrocarbon Depletion (GUIDANCE)

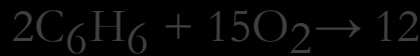
CONVERT FLUX TO HYDROCARBON DEPLETION (SITE-SPECIFIC)

$$J = D_v^{eff} \left(\frac{dC}{dZ} \right)$$

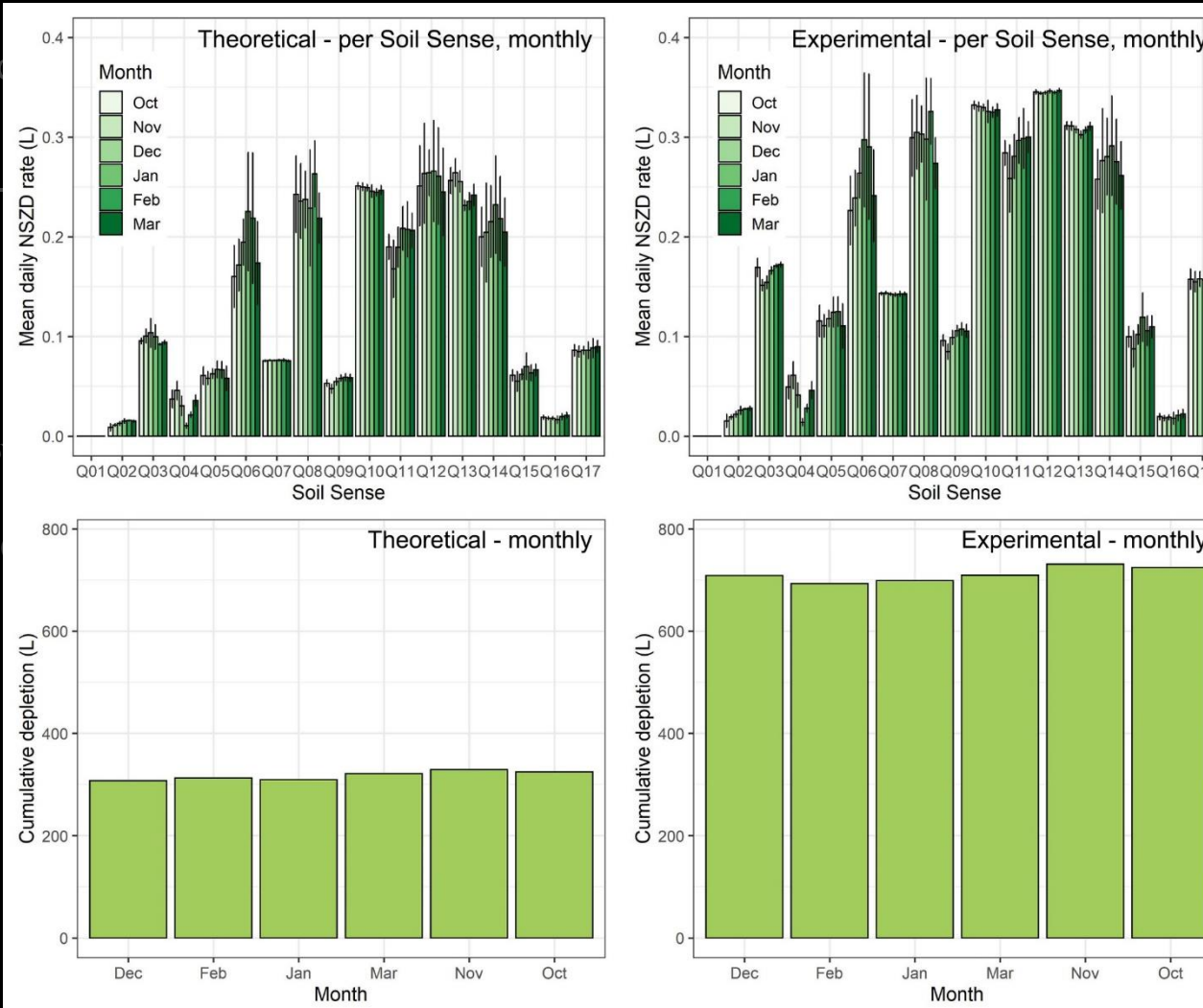
$$J = 3.8 \times 10^{-7} \frac{m^2}{s}$$

$$J = 2.43 \times 10^{-7}$$

Convert to g - Represent



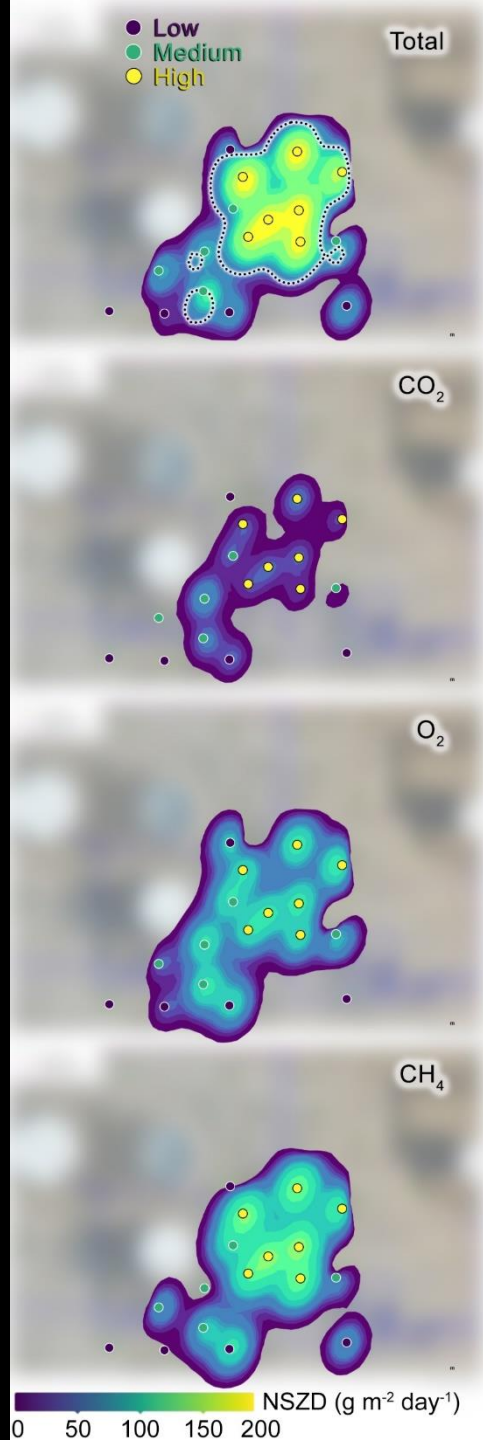
$$\text{Molar Ratio} = \frac{78.11 \frac{gC_6H_6}{mol}}{44.11 \frac{gCO_2}{mol}}$$



ed * Molar Ratio

* 0.3 * 86400

$\frac{H_6}{d}$



NSZD through space

- NSZD estimates incorporate methane and paired CO₂ production/O₂ consumption
- More representative estimates of biological activity.

NSZD Site Variability

Localized Activity

12 sites

76 Soil Senses

228 Sensor packs

>100,000,000 data points!

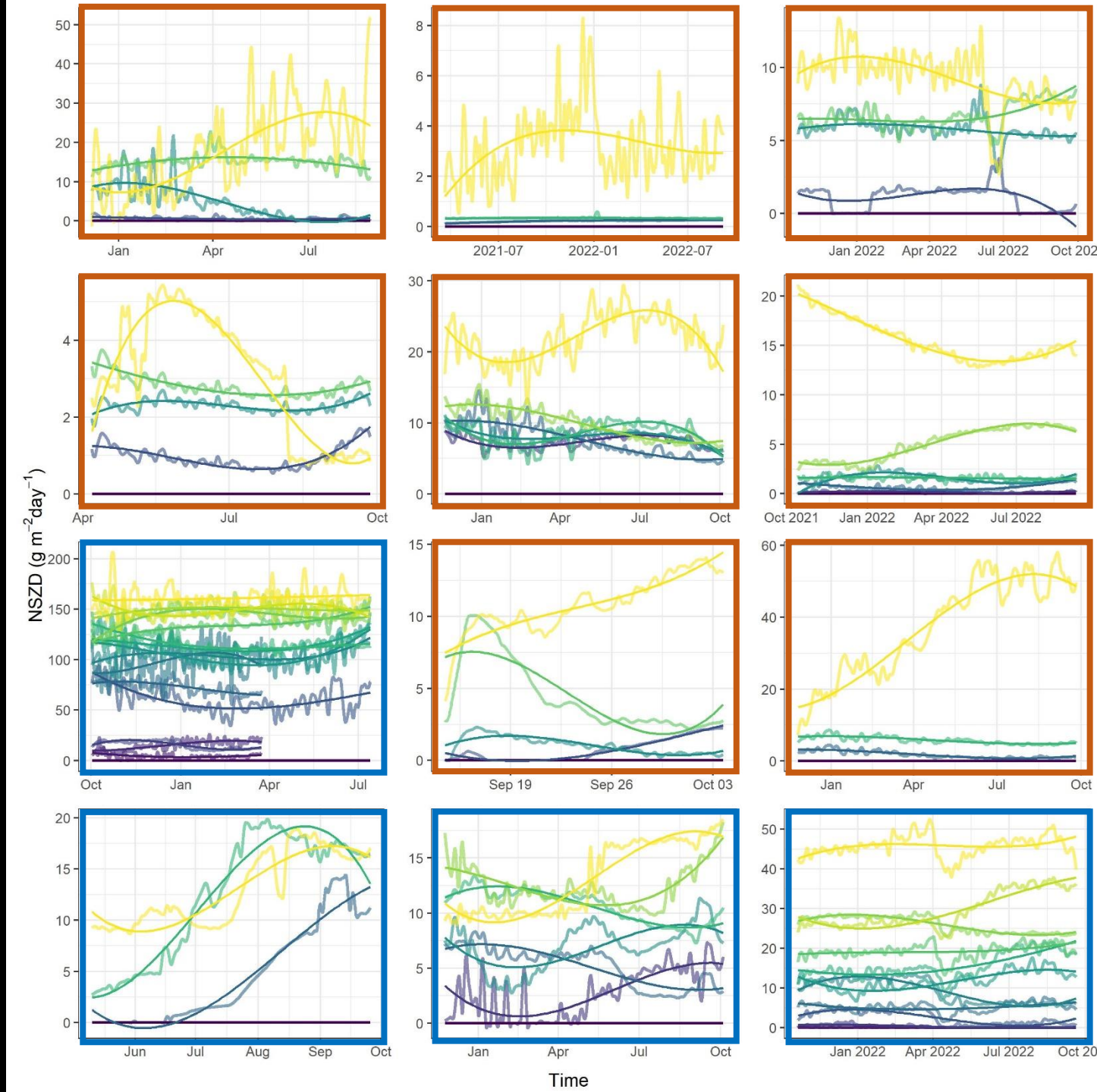
Site Wide Activity

Low

High



NSZD



NSZD

Site Variability

Localized Activity

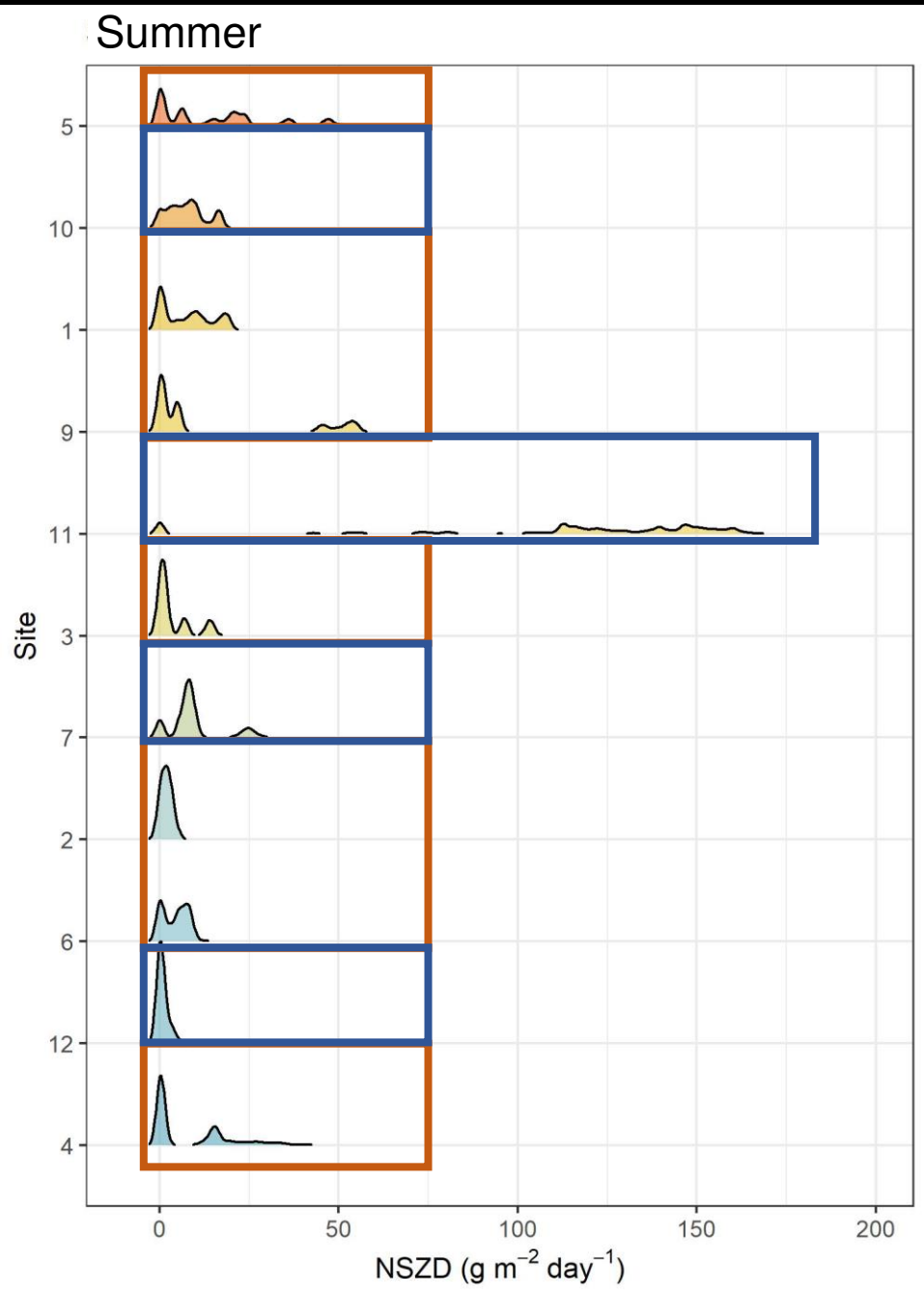
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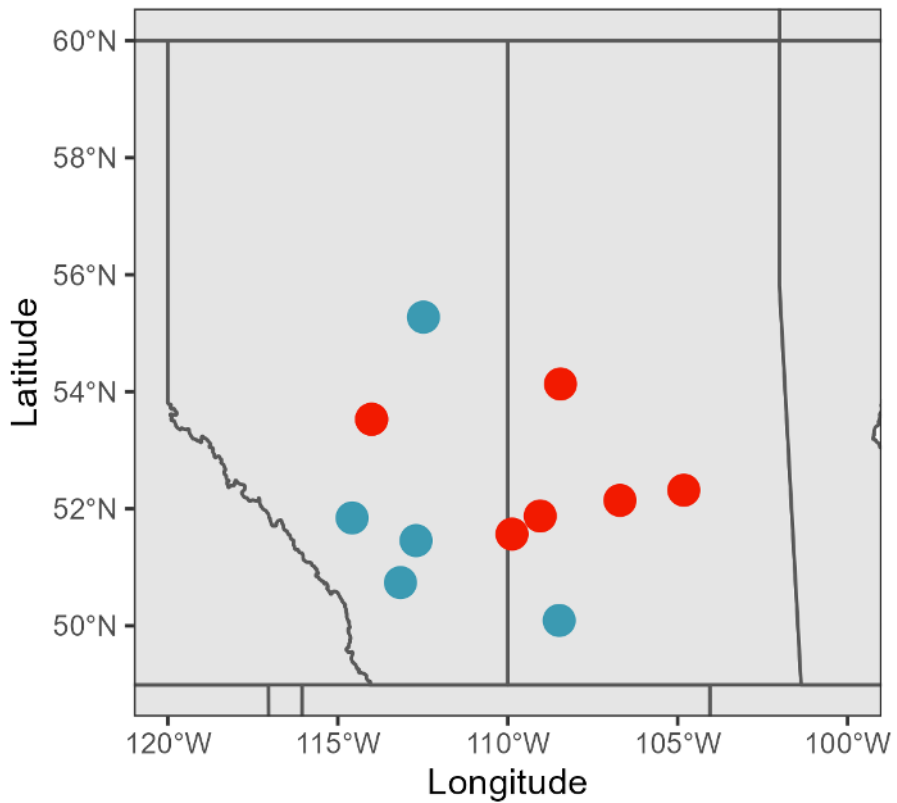
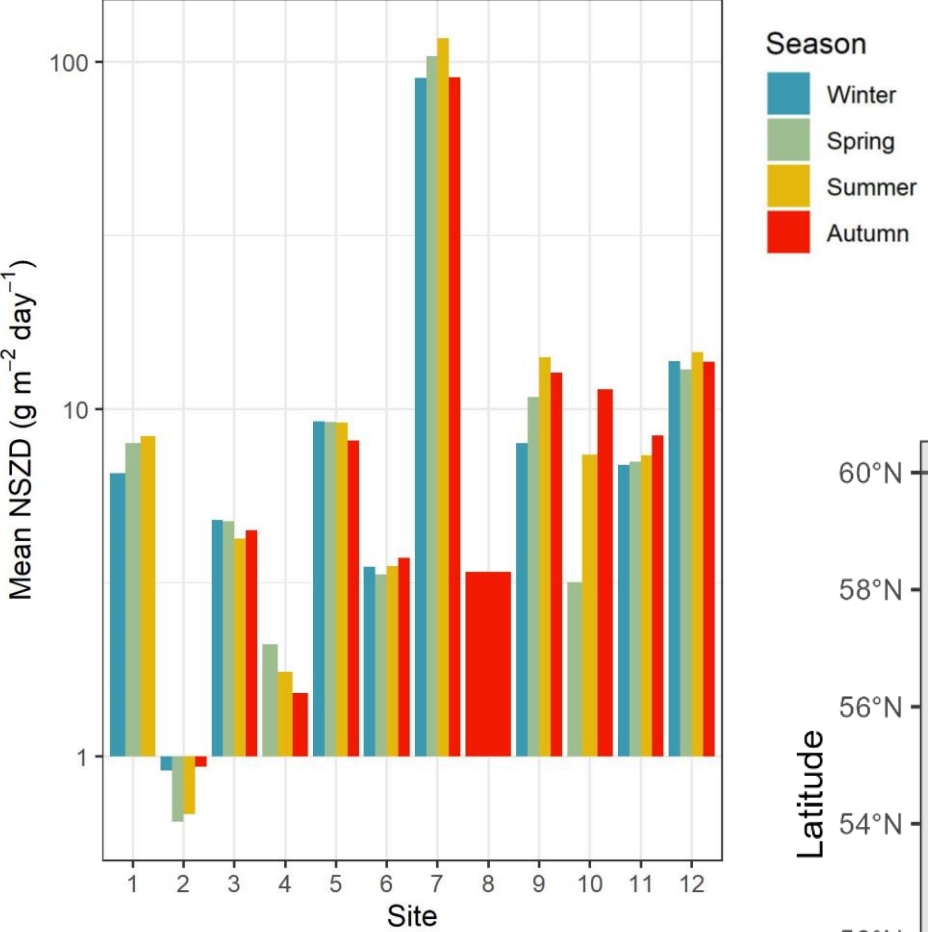
>100,000,000 data points!

Site Wide Activity



NSZD Seasonality

NSZD peaks in either winter/spring or summer/fall

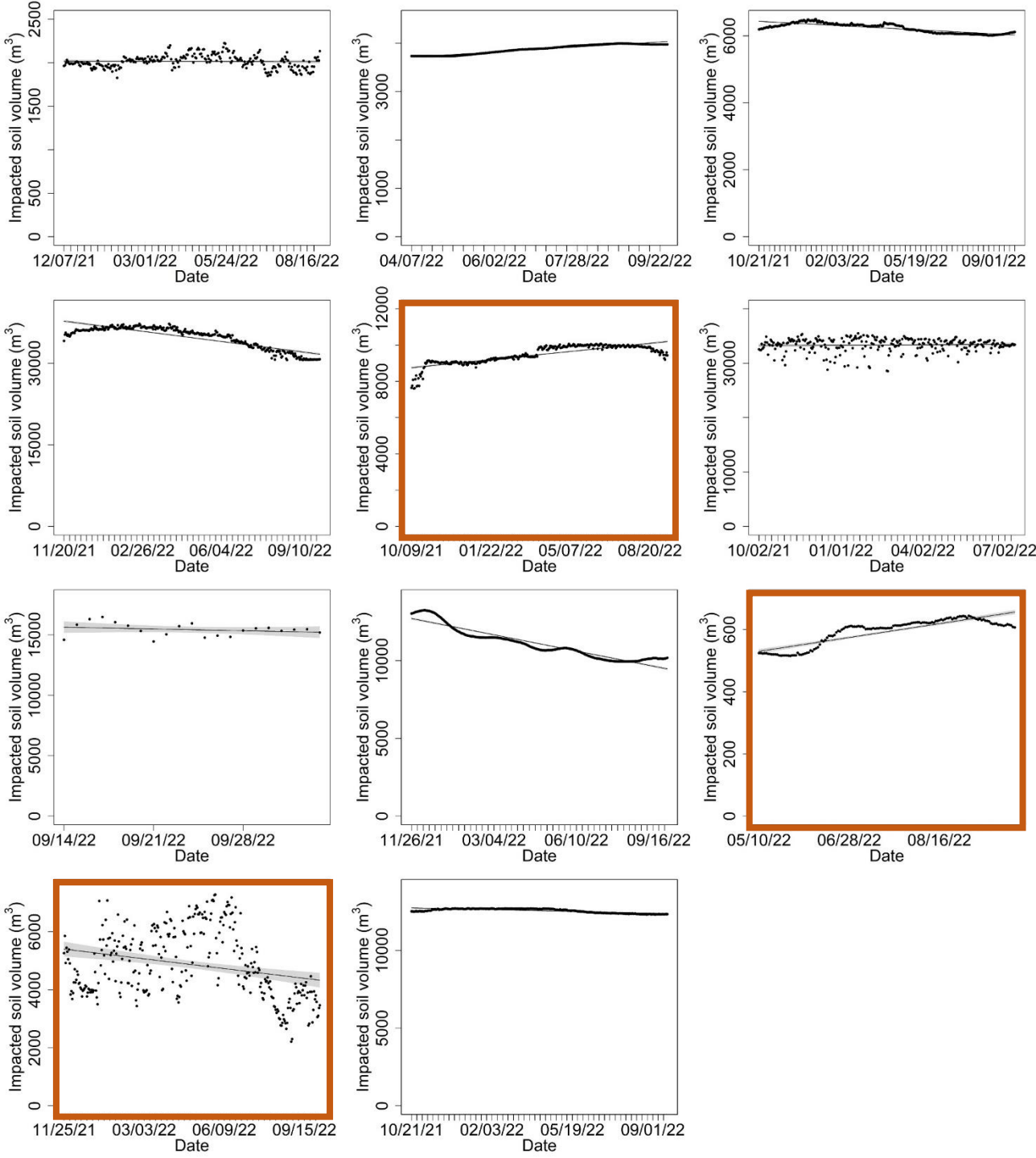


● Winter/Spring ● Summer/Autumn

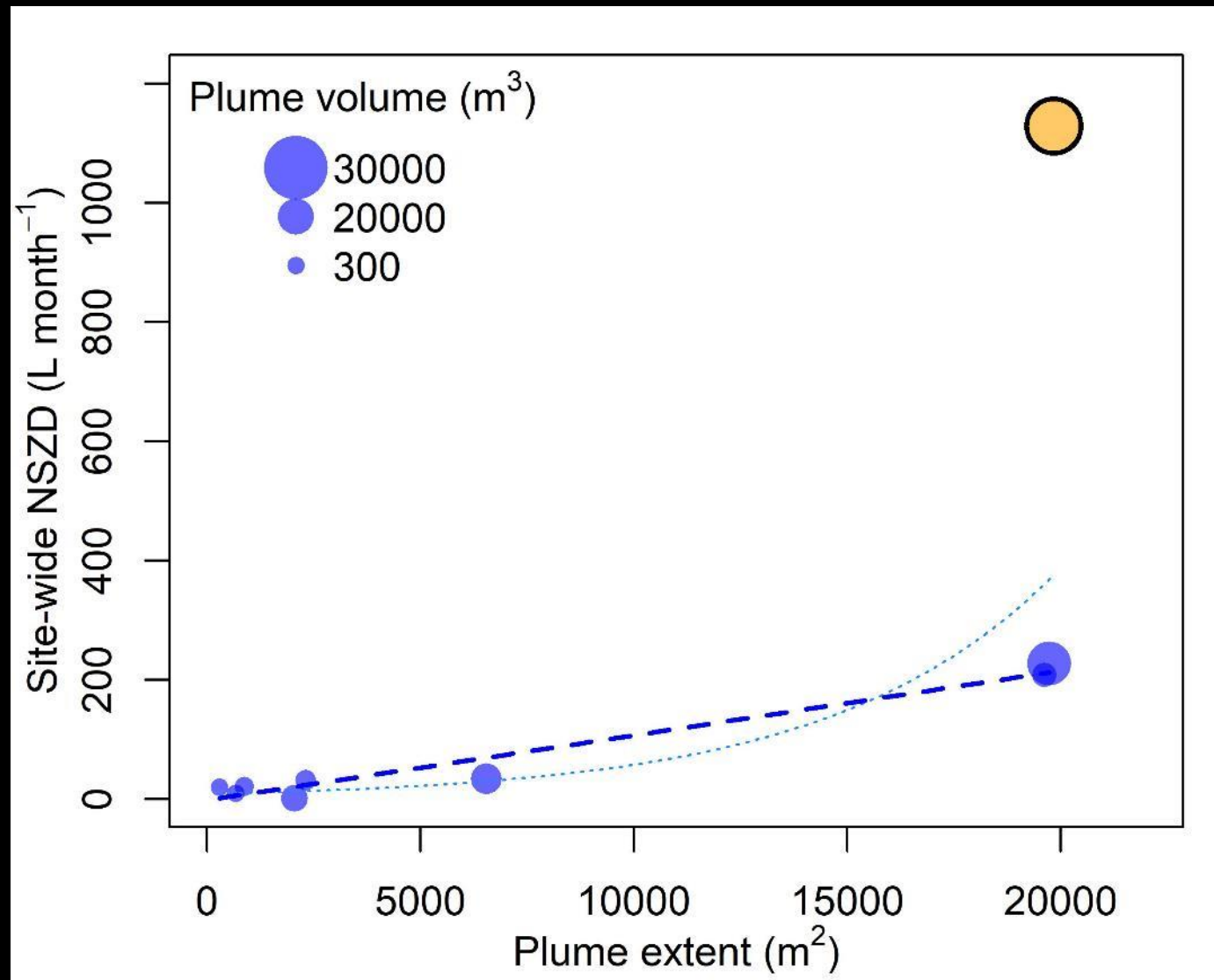
Plume volume

Typically stable over seasons

- Active site
- Additional sensor install and equilibration
- Site with high-water table and fall ice-up.



How do sites compare?



Continuous Site Monitoring

Money saved

Water saved

Updated for: Q2 2022

- Benzene Plume Border
- NSZD Plume Overlay
- Overlay Opacity

Money Saved
219747 CAD\$

Water Saved
348 kg



Memorandum

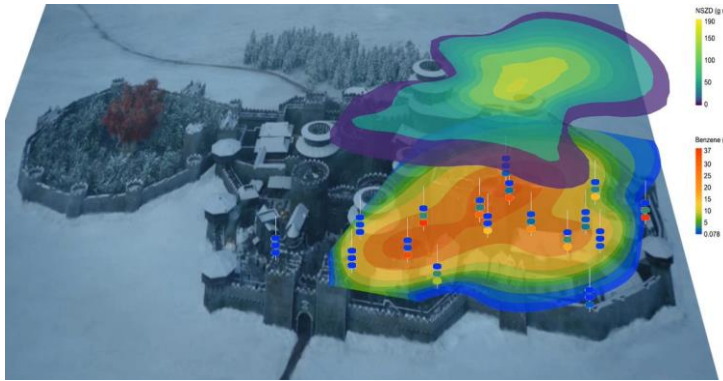
Environmental Material Science Inc.
#114, 116 Research Drive
Saskatoon, SK S7N 3R3

To: [Redacted]
From: Steven Siciliano, Environmental Material Science, Inc.
CC: Amy Jimmo, Environmental Material Science, Inc.
Date: [Redacted]
Subject: [Redacted]

Executive Summary

A Soil Sense network was installed at [Redacted] and around an Light Non-Aqueous Phase Liquid (LNAPL) plume. We operationally define a plume as a discrete spatial interval with hydrocarbon signals or surrogates ¹ above SK Tier 2 EQGs (e.g., [Redacted] kg⁻¹ for benzene here). The network collected 6,467,400 data points since [Redacted] providing the following estimates:

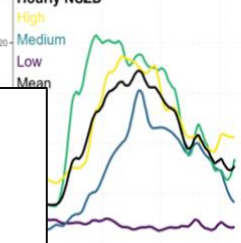
- Plume areal extent based on soil vapour estimates was [Redacted] with peak plume volume (maximum concentration and area) at 5 m below ground. Plume volume increased slightly since [Redacted]. Mean plume volume = 26,924 m³.
- Mean Natural Source Zone Depletion (NSZD¹) was estimated to be 56.9 g m⁻² day⁻¹ of benzene. Mean air temperatures cooled from 7.2 to 11.8°C [Redacted] while NSZD rates decreased from 66.6 g m⁻² day⁻¹. Site wide activity was estimated at 887 kg day⁻¹.



Mean NSZD



Hourly NSZD



liquids is equal to the vapor pressure of the pure component multiplied by its mole fraction in the mixture. Mathematically, this combination works out to:

$$C_{T,i} = \frac{C_{a,i} (\theta_w + K_{OC,i} f_{OC,i} \rho_s + H_i \theta_a)}{H_i \rho_b} \quad [1]$$

Where:

- $C_{T,i}$ = total soil concentration of component i (mg L⁻¹)
- $C_{a,i}$ = soil vapour concentration (from IR; mg L⁻¹)
- H_i = Henry's constant for each component i at the temperature of interest (0.230 for benzene at 25°C; however, see equations 2 and 3 below, dimensionless)
- ρ_b = dry soil bulk density (1.85 g m⁻³)
- θ_w = volumetric water content in vadose zone (see equation 9; dimensionless)
- θ_a = volumetric air content (see equation 10; dimensionless)
- $K_{OC,i}$ = organic carbon water partitioning coefficient for each component (79.4 L kg⁻¹ for benzene)
- $f_{OC,i}$ = organic carbon fraction (0.001; dimensionless)

However, mean subsurface soil temperatures are typically less than standard state (i.e., 25°C) and use of Henry's law constant under these conditions may overpredict contaminant volatility resulting in artificially low soil concentrations. Henry's law constant may be corrected for mean soil temperature using the Clausius-Clapeyron relationship⁷. First, we approximate the enthalpy of vaporization of the contaminant at mean soil temperature ($\Delta H_{v,TS}$) from the enthalpy of vaporization at the normal boiling point:

$$\Delta H_{v,TS} = \Delta H_{v,b} \left[\frac{(1 - \frac{T_S}{T_C})}{(1 - \frac{T_b}{T_C})} \right]^n \quad [2]$$

Where:

- $\Delta H_{v,TS}$ = enthalpy of vaporization at the mean soil temperature (kJ mol⁻¹)
- $\Delta H_{v,b}$ = enthalpy of vaporization at the normal boiling point (kJ mol⁻¹)
- T_S = mean soil temperature (K)
- T_C = critical temperature (K)
- T_b = normal boiling point (K)
- n = exponent (dimensionless)

$\Delta H_{v,TS}$ is then substituted into equation 3 to derive Henry's law constant corrected for mean soil temperature:

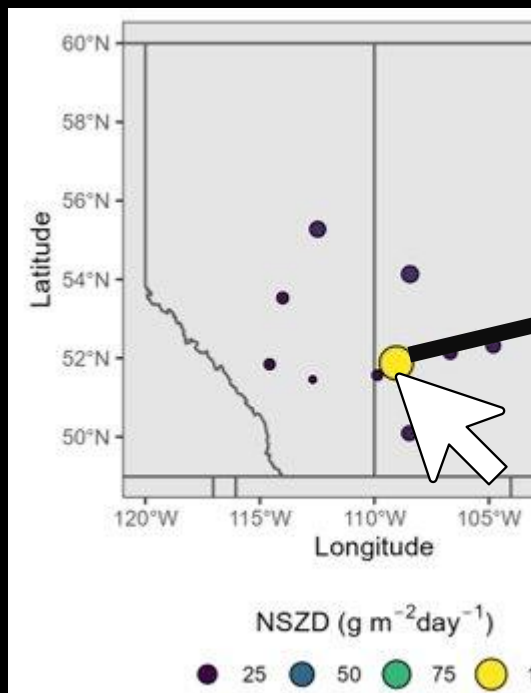
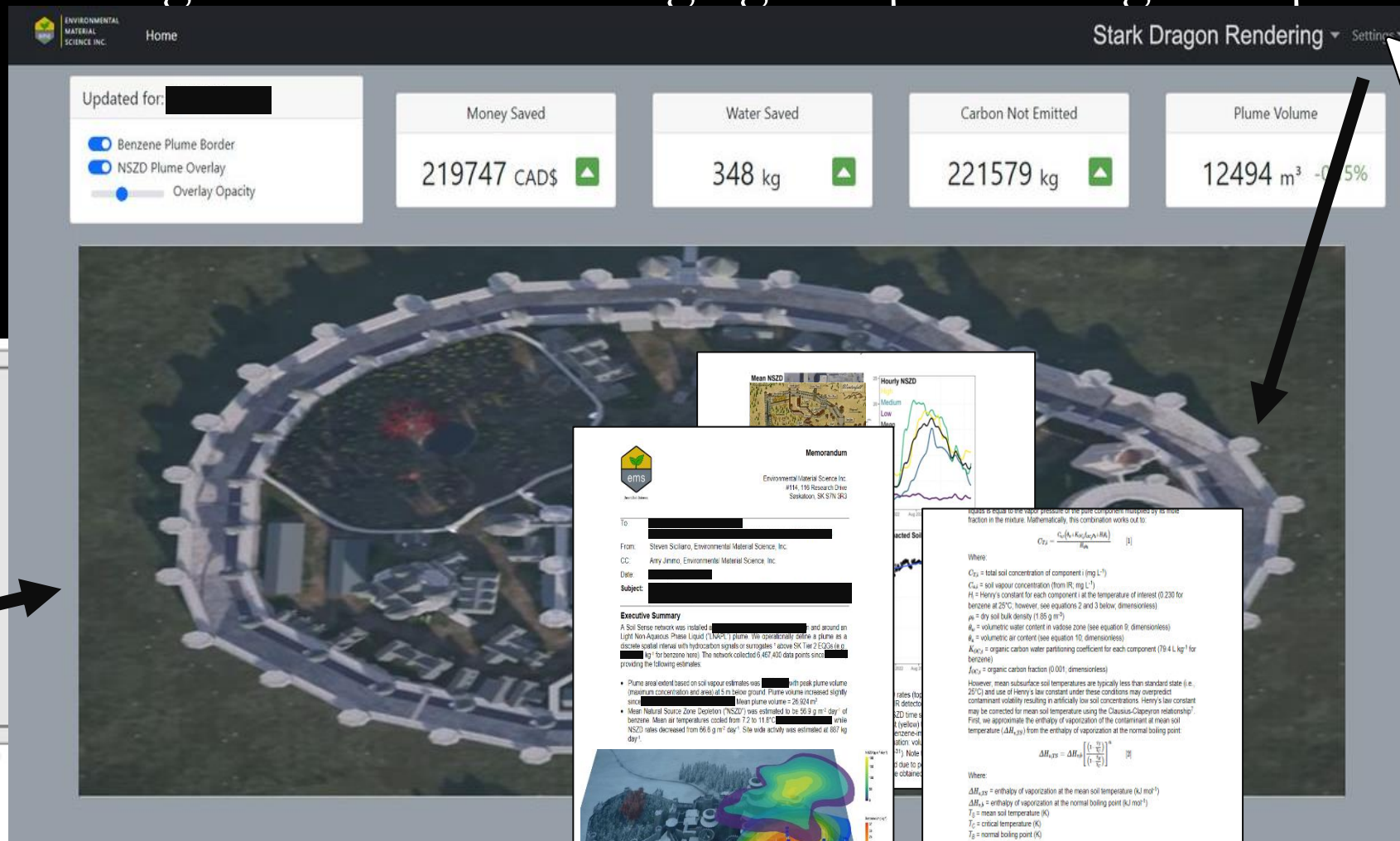
$$H_{TS} = \frac{\exp \left[\frac{\Delta H_{v,TS}}{R C} \left(\frac{1}{T_S} - \frac{1}{T_b} \right) \right] H_R}{R T_S} \quad [3]$$

Where:

- H_{TS} = Henry's law constant at the mean soil temperature (dimensionless)

Managing Multiple Sites

- The goal is to make managing and prioritizing multiple



Memorandum

Environmental Material Science Inc.
 #16, 116 Research Drive
 Sudbury, ON S7N 3R3

To: [REDACTED]

From: Steven Sciarro, Environmental Material Science, Inc.

CC: Amy Jimco, Environmental Material Science, Inc.

Date: [REDACTED]

Subject: [REDACTED]

Executive Summary

A Soil Sense network was installed at [REDACTED] and around an Light Non-Aqueous Phase Liquid (LNAPL) plume. The spatially defined plume as a discrete spatial interval with hydrocarbon signals is roughly active S/N 2, 520 (in [REDACTED] kg) for benzene here). The network collected 9,487,400 data points since [REDACTED] providing the following estimates:

- Plume area extent based on soil vapour estimates was [REDACTED] with peak plume volume (maximum concentration and area) at 5 m below ground. Plume volume increased slightly since [REDACTED] when plume volume = 26,924 m³.
- Mean Natural Source Zone Occurrence (NSZD) was estimated to be 69.8 g m⁻² day⁻¹ of benzene. Mean air temperatures cooled from 7.2 to 11.8°C [REDACTED] while NSZD rates decreased from 66.6 g m⁻² day⁻¹. Site wide activity was estimated at 867 kg day⁻¹.

$$C_{T2} = \frac{C_u (K_{oc} K_{ow} K_{oc} K_{oc})}{R_{so}} \quad (1)$$

Where:

- C_{T2} = total soil concentration of component i (mg L⁻¹)
- C_u = soil vapour concentration (from IR, mg L⁻¹)
- R = Henry's constant for each component i at the temperature of interest (0.230 for benzene at 25°C, however, see equations 2 and 3 below, dimensionless)
- ρ_b = dry soil bulk density (1.85 g m⁻³)
- θ_v = volumetric water content in vadose zone (see equation 9, dimensionless)
- θ_w = volumetric air content (see equation 10, dimensionless)
- K_{oc} = organic carbon water partitioning coefficient for each component (79.4 L kg⁻¹ for benzene)
- K_{ow} = organic carbon fraction (0.001, dimensionless)

However, mean subsurface soil temperatures are typically less than standard state (i.e. 25°C) and use of Henry's law constant under these conditions may overpredict contaminant volatility resulting in artificially low soil concentrations. Henry's law constant may be corrected for mean soil temperature using the Clausius-Clapeyron relationship⁷. First, we approximate the enthalpy of vaporization of the contaminant at mean soil temperature (ΔH_{T2}) from the enthalpy of vaporization at the normal boiling point:

$$\Delta H_{T2} = \Delta H_{Tb} \left[\frac{(T - T_b)}{(T - T_c)} \right]^n \quad (2)$$

Where:

- ΔH_{T2} = enthalpy of vaporization at the mean soil temperature (kJ mol⁻¹)
- ΔH_{Tb} = enthalpy of vaporization at the normal boiling point (kJ mol⁻¹)
- T = mean soil temperature (K)
- T_c = critical temperature (K)
- T_b = normal boiling point (K)
- n = exponent (dimensionless)

ΔH_{T2} is then substituted into equation 3 to derive Henry's law constant corrected for mean soil temperature:

$$H_{T2} = \frac{\exp\left(\frac{\Delta H_{T2}}{R_c} \left(\frac{1}{T} - \frac{1}{T_b}\right)\right) H_{Tb}}{RT_2} \quad (3)$$

Where:

R_c = Henry's law constant of the pure component at standard state

Where are we now?

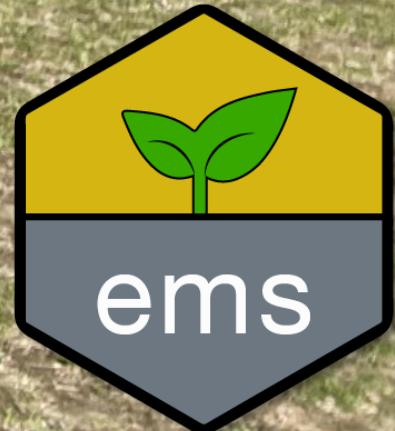
- High density continuous measurements
 - Diurnal and seasonal nuances, site-driven variation
 - CH₄ may make up the bulk of NSZD in some sites
- Soil moisture
 - Implications for fluid transfer and PHC distributions
- Site-specific estimates
 - Theoretical stoichiometry may not represent site processes
- Stay-tuned for next year!
 - modflow
 - Data assimilation
 - AgTech.

Thank you!

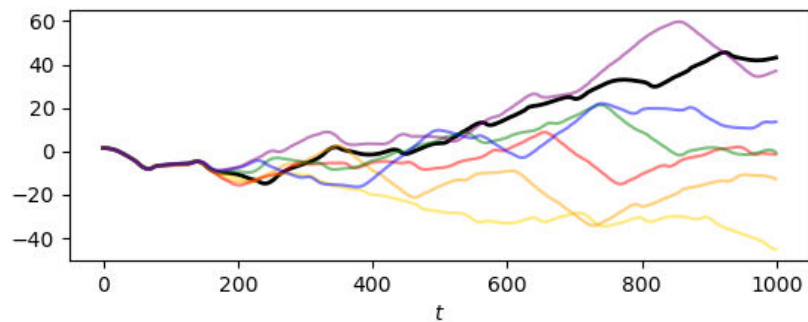
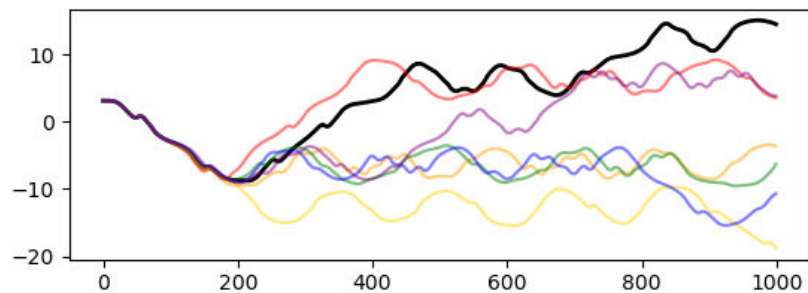
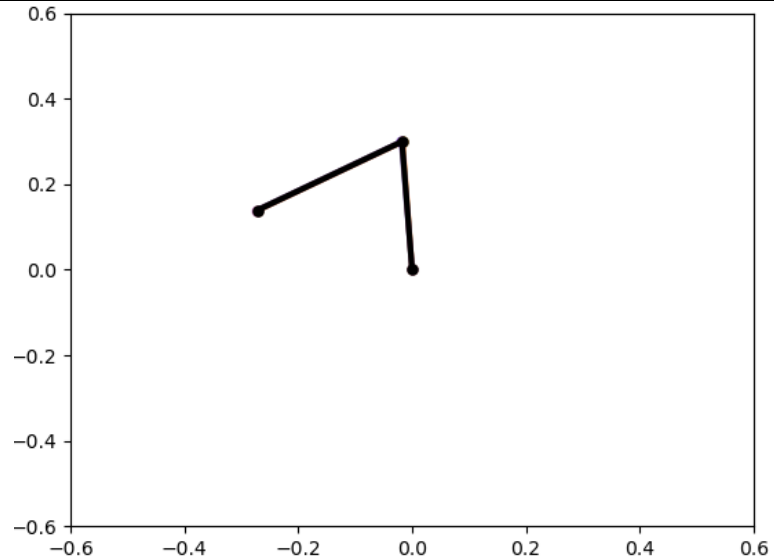
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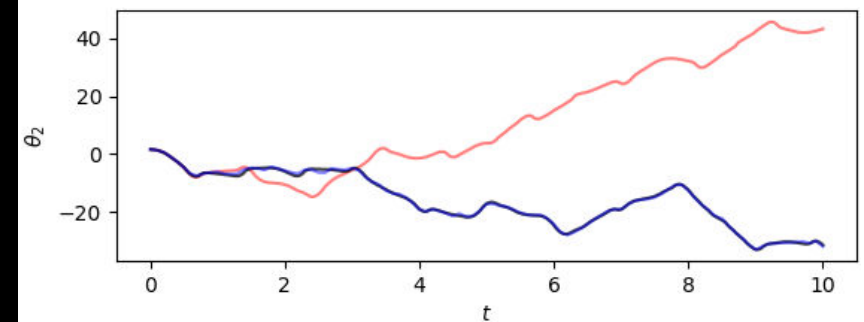
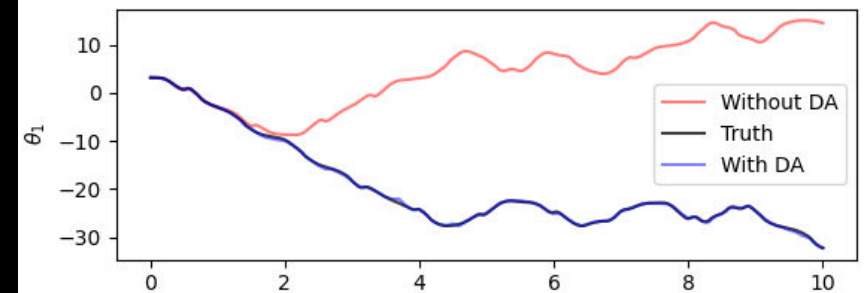
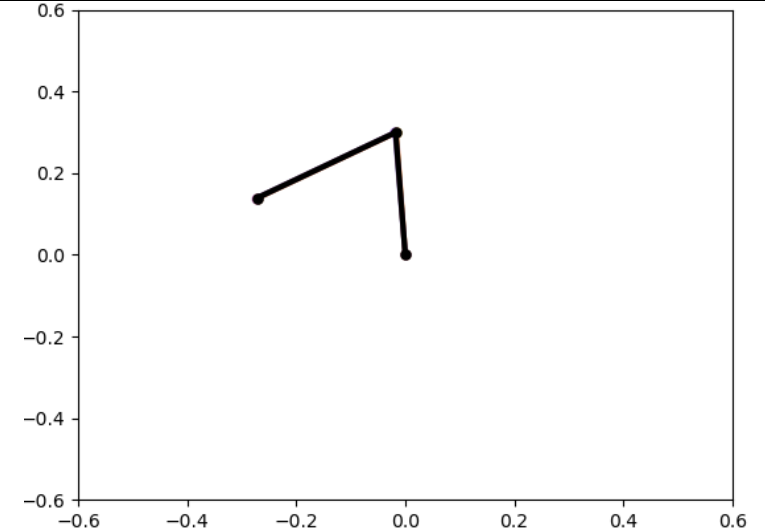
stevemamet@ems-inc.ca
<https://www.ems-inc.ca/>



Data Assimilation – Double Pendulum Example

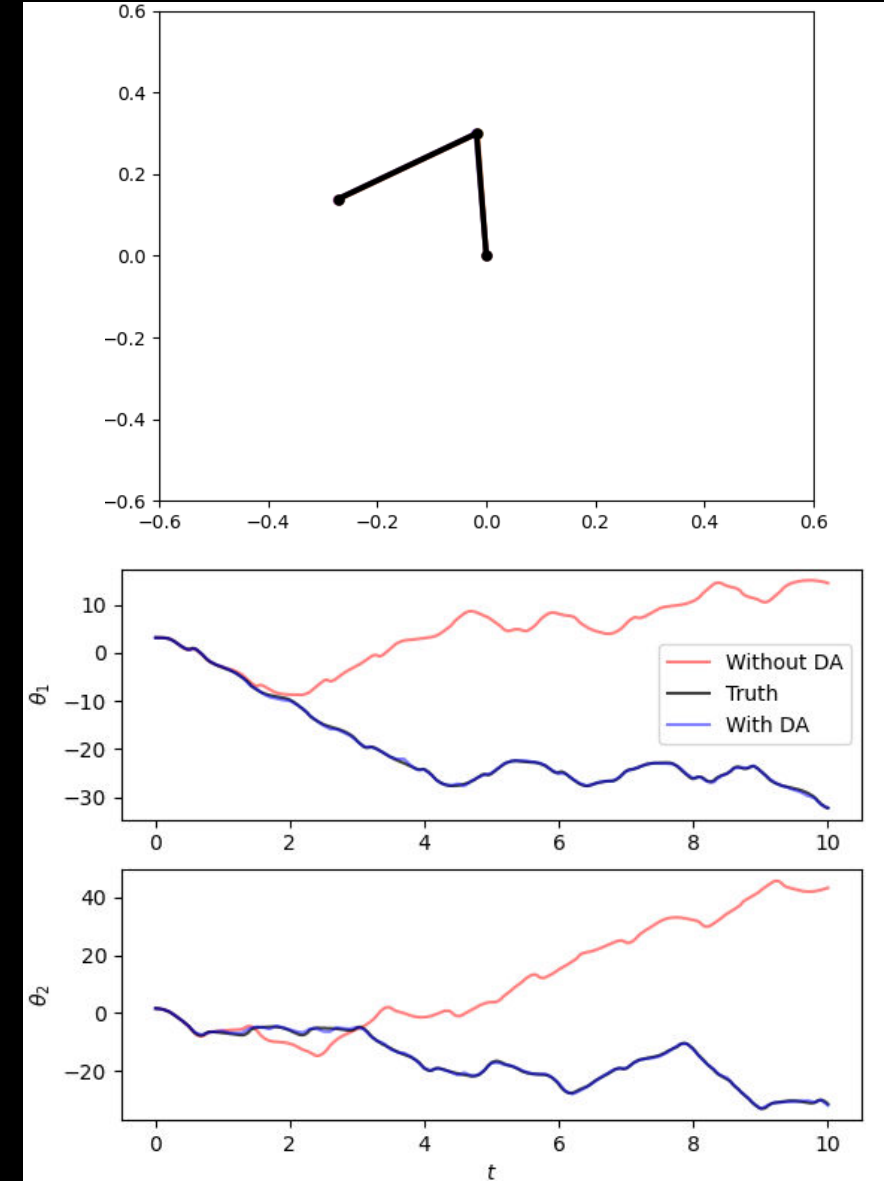
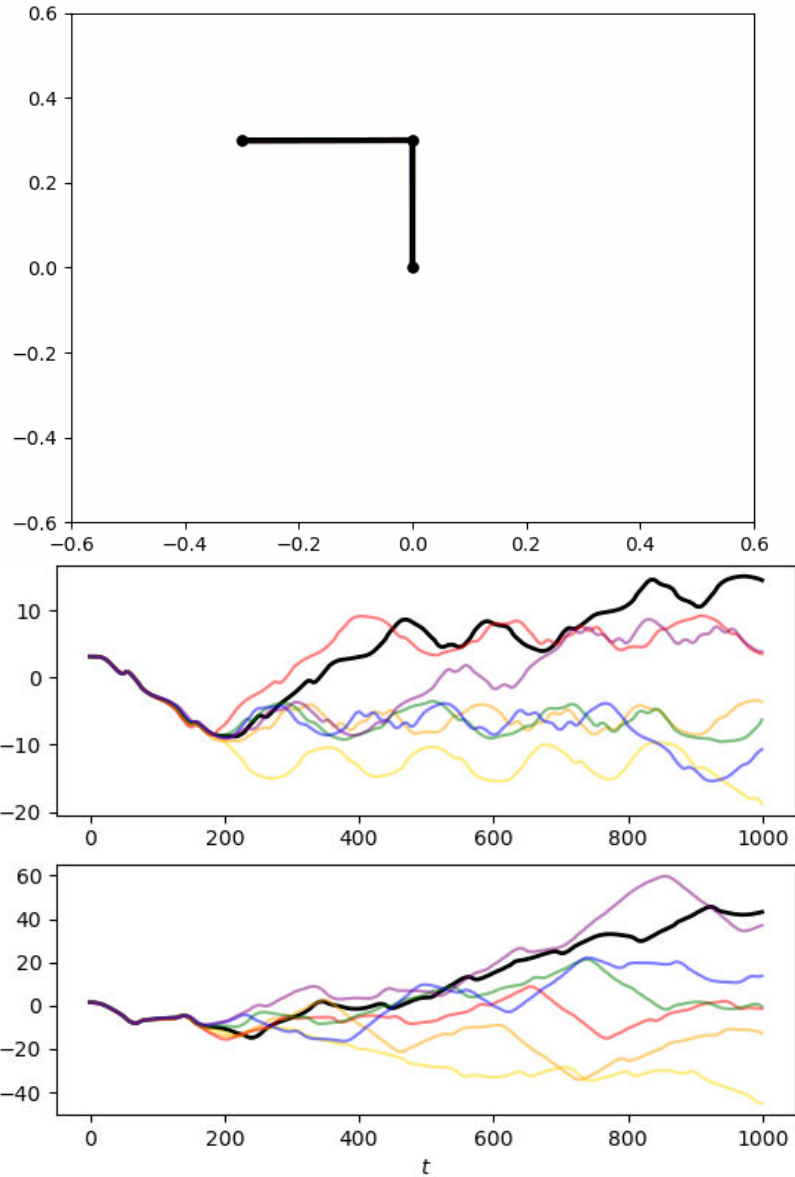


- Small differences in parameters or initial conditions propagate into large intractable errors in model output
- Data Assimilation uses observations/measurements to update models to prevent them from diverging.

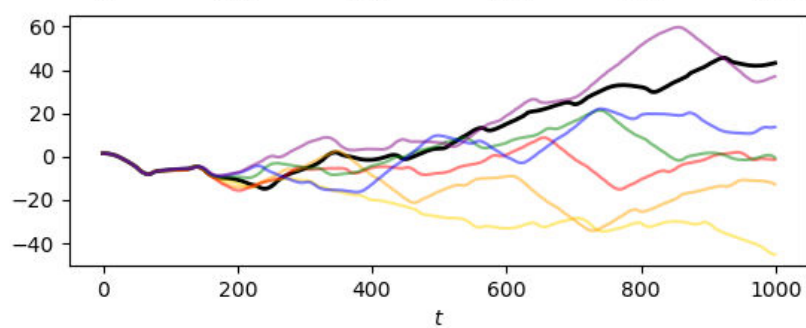
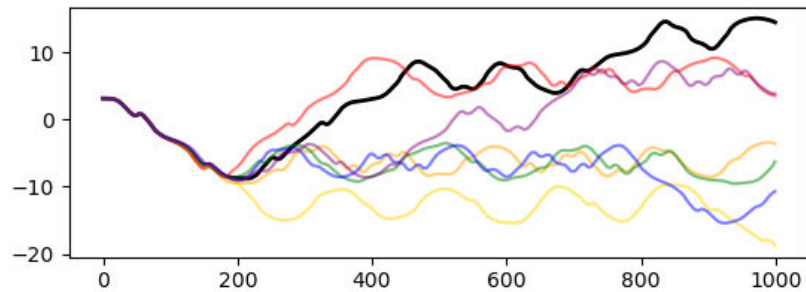
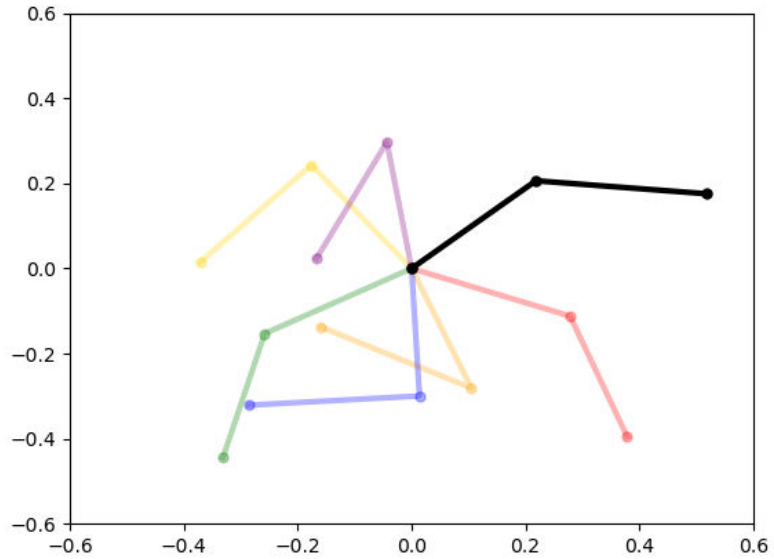


Data Assimilation – Double Pendulum Example

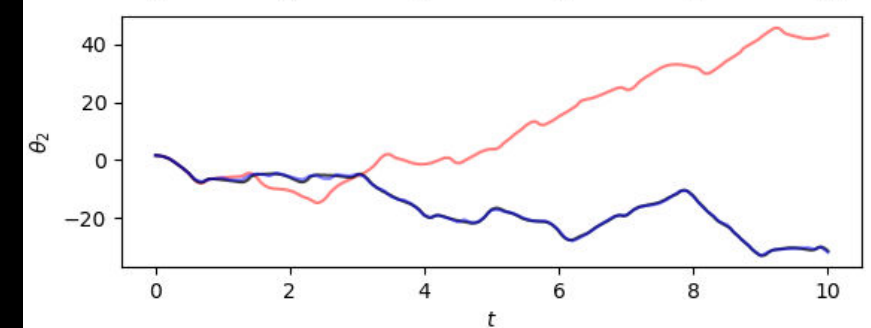
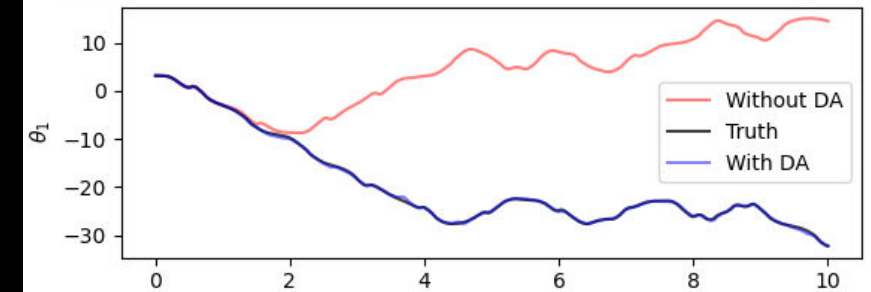
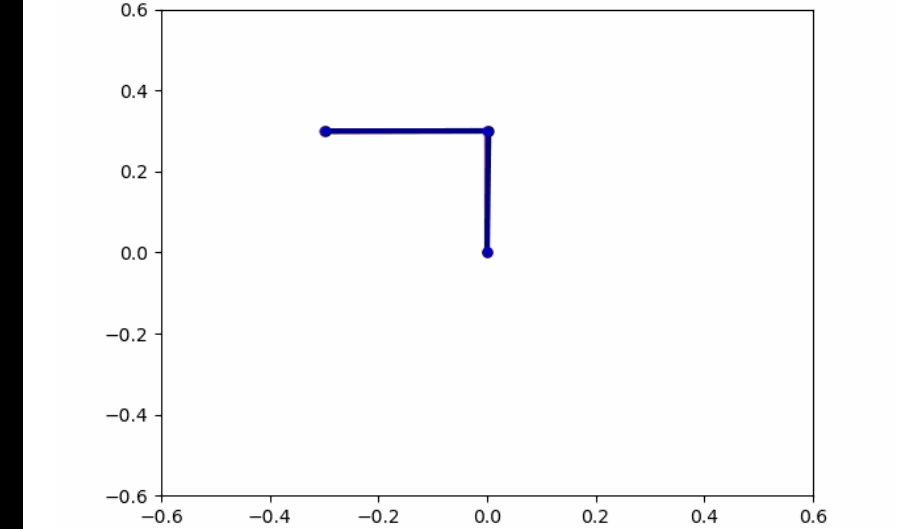
- Small differences in parameters or initial conditions propagate into large intractable errors in model output
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Data Assimilation – Double Pendulum Example

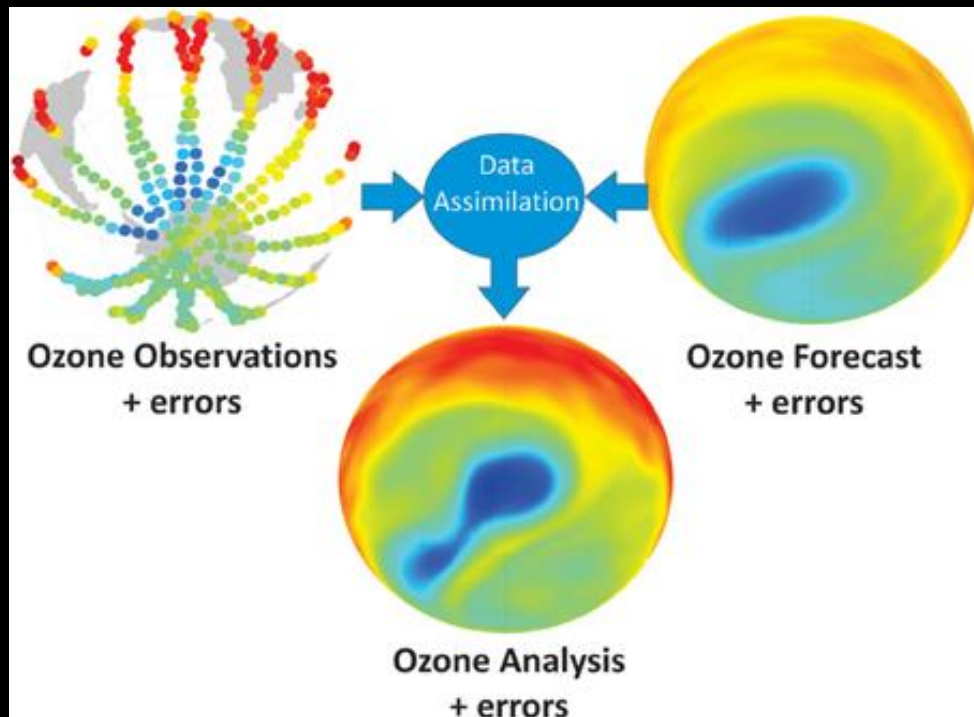


- Small differences in parameters or initial conditions propagate into large intractable errors in model output
- Data Assimilation uses observations/measurements to update models to prevent them from diverging.



Data Assimilation

- Data Assimilation is utilized in a large number of fields including weather prediction, self-driving cars, missile tracking, nuclear power, and GPS



- EMS' sensor data leverages Data Assimilation to improve estimates of plume-maps, biodegradation, and soil gas profiles.

