



Pilot scale *in situ* treatment of chlorinated solvent source zone – use of compound-specific isotope analysis (CSIA) to assess treatment performance and injection design

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Niagara Falls, Ont
May 30th to June 1st

Context

Remediation technologies aim at stimulating a specific contaminant removal process:

- i. Destructive (chemical and biological processes)
- ii. non-destructive (physical process)

How do we know if the intended removal process is actually occurring ?

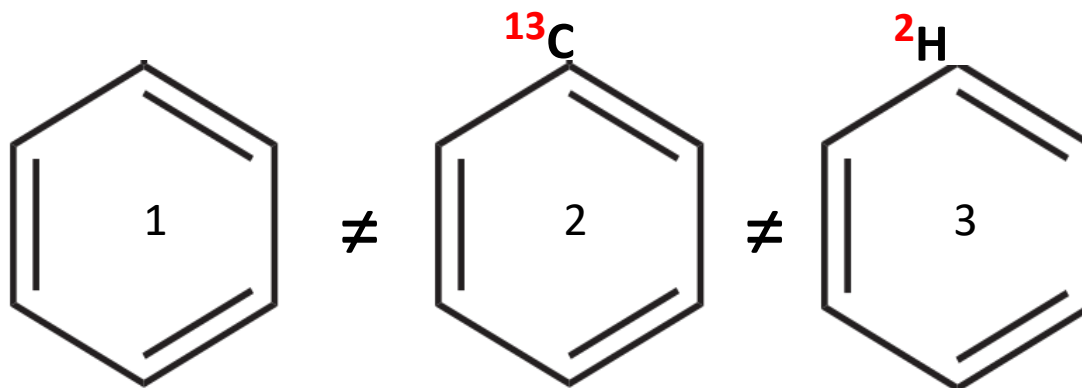
- i. Presence of co-occurring processes decreasing contaminant concentrations:
 - o Dilution vs destruction
 - o Dilution vs non-destructive removal
 - o Destruction vs non-destructive removal

What is your monitoring program?

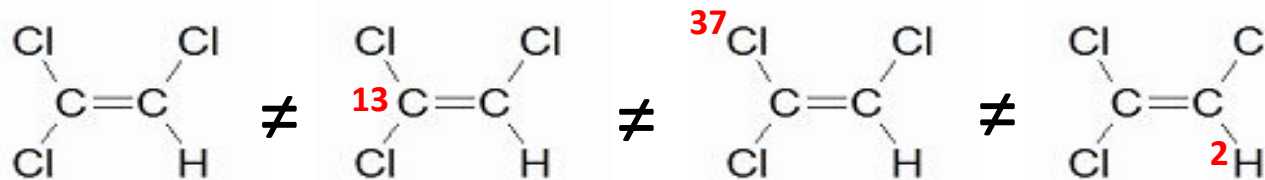
- i. Targeted VOC
 - a. Concentration
 - b. Mass flux
- ii. Hydrogeochemical parameters:
 - a. DO, ORP, pH
 - b. NO_3 , Fe, SO_4 ,
 - c. H_2S , ethene/ethane/methane
- iii. Microbial population
 - a. Change over time
 - b. Right population present
- iv. Are the compounds of concern degraded ?
 - a. Compound-specific isotope analysis (CSIA)

Implication of heavy isotopes

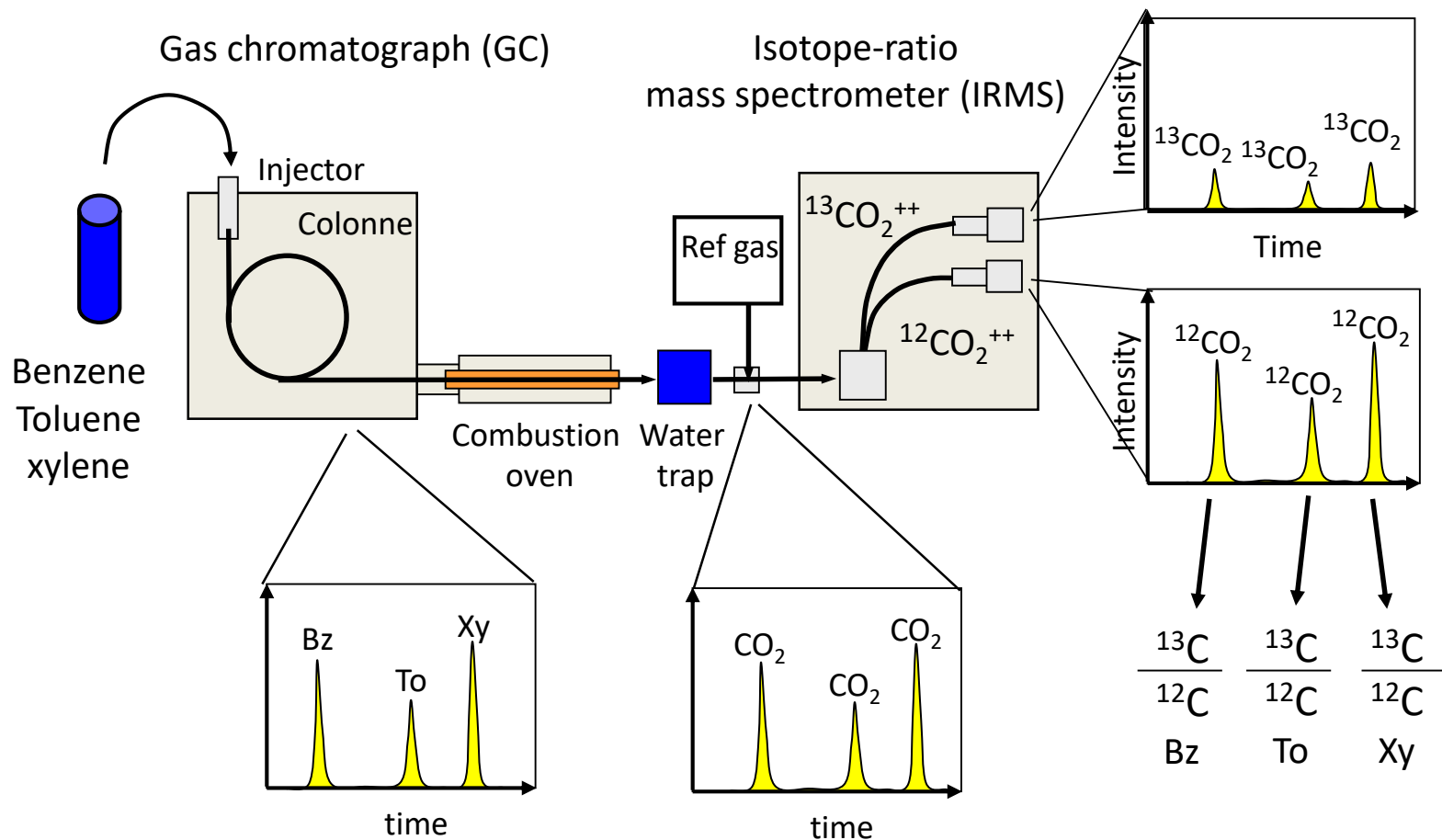
benzene



TCE



Sample analysis



Delta (δ) notation

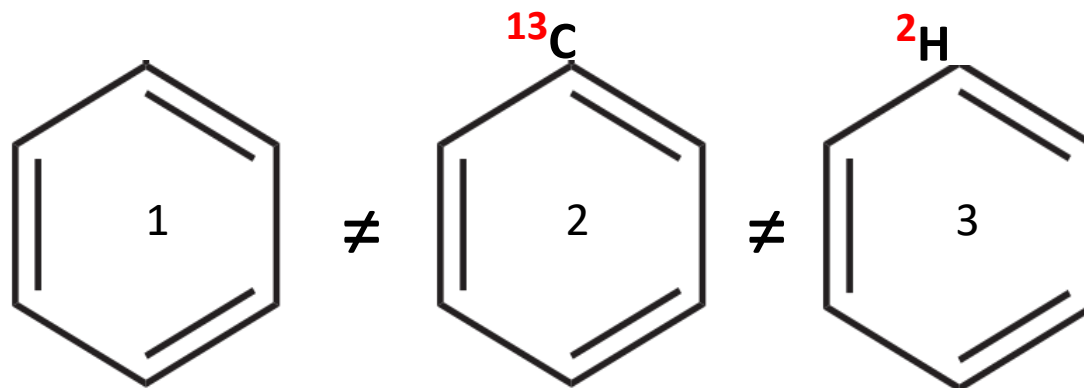
$$\delta^{13}\text{C} (\text{‰}) = \frac{{}^{13}\text{C}/{}^{12}\text{C}_{\text{sample}} - {}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}}{{}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}} \times 1000$$

International standard

- Carbon: V-PDB
- Hydrogen: V-SMOW
- Chlorine : V-SMOC

Implication of heavy isotopes

benzene



Degradation rate	normal	slower	slower
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Method aim:

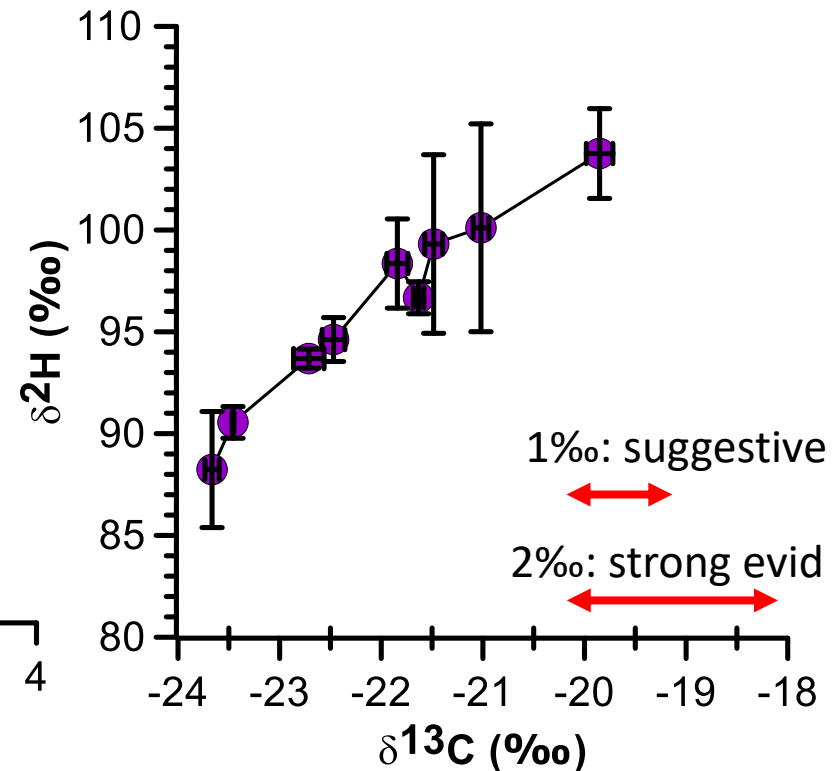
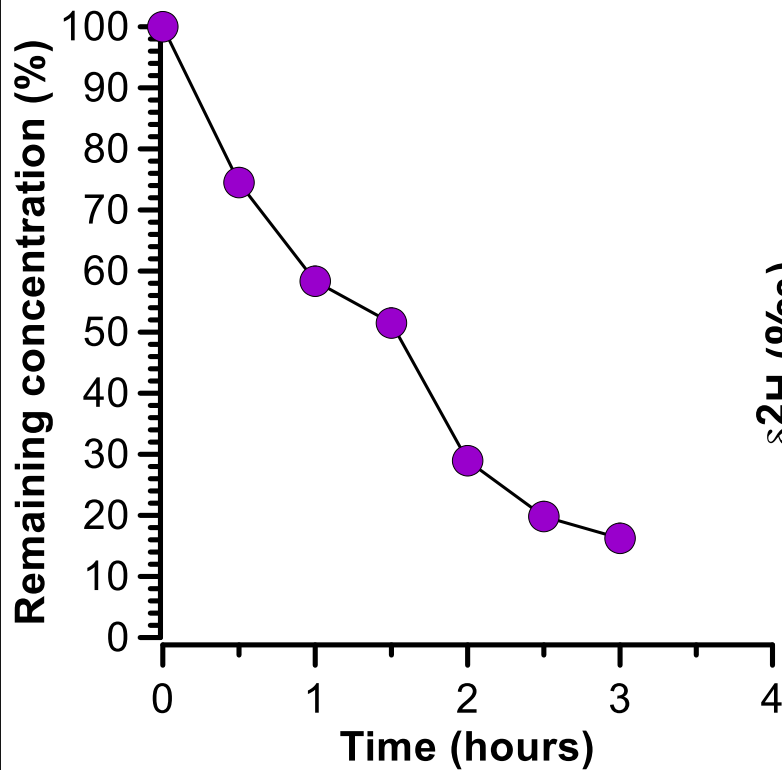
tracking the $^{13}\text{C}/^{12}\text{C}$ ratio evolution over time

Introduction

Fundamentals

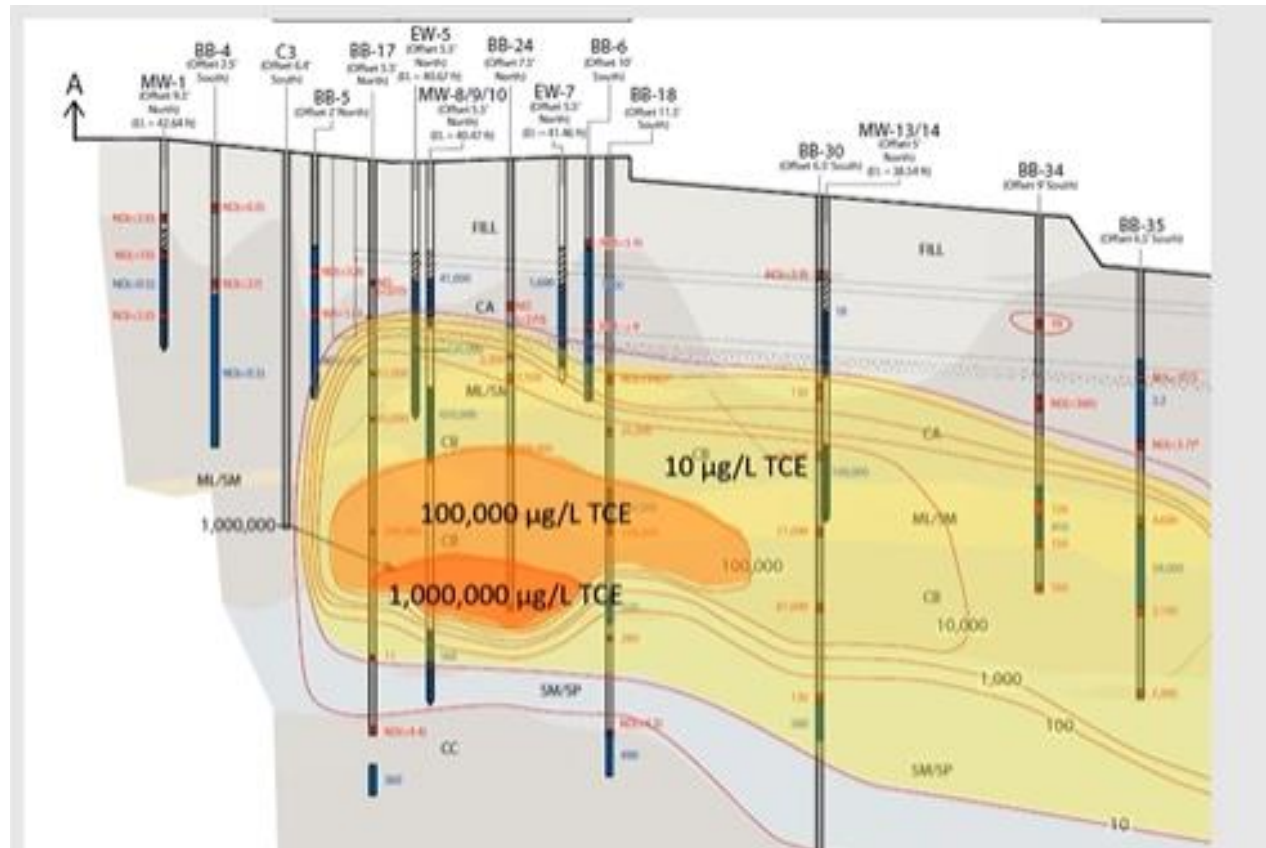
Implication of heavy isotopes

- Biodegradation of benzene



Site Description

- Contaminant: TCE (source zone)
- Sandy silt layer above a sandy layer

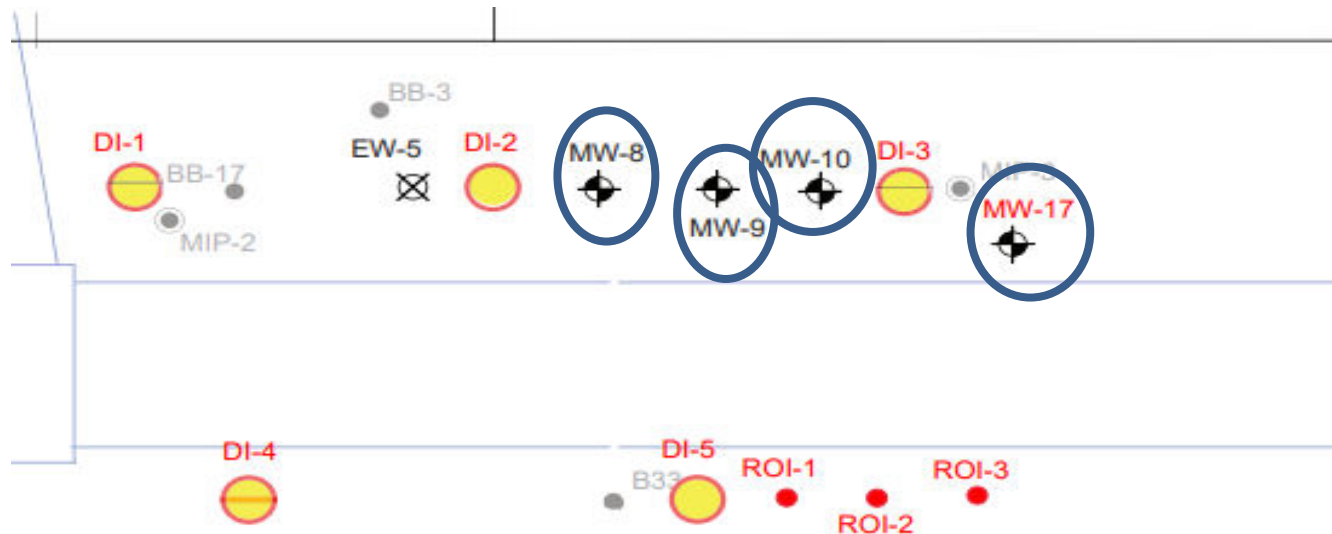


CSIA
Application
during In situ
Remediation
Treatment

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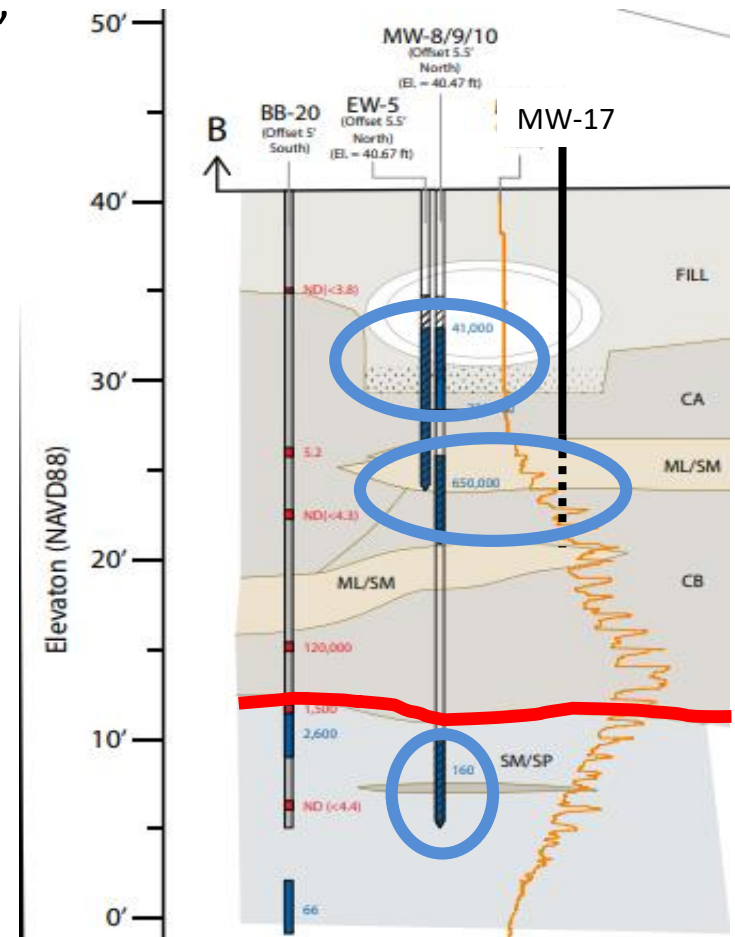
- Enhanced anaerobic biodegradation
 - EVO (EDS-ER™) + dhc (KB-1™) + ZVI
- in situ bioremediation - pilot test
 - 5 injection points
 - 2500 gallons (500 gal / point)
 - larger volume at the bottom of silt layer
 - performance monitoring
 - 4 MWs
 - baseline + 3 post injection sampling events



Site Description

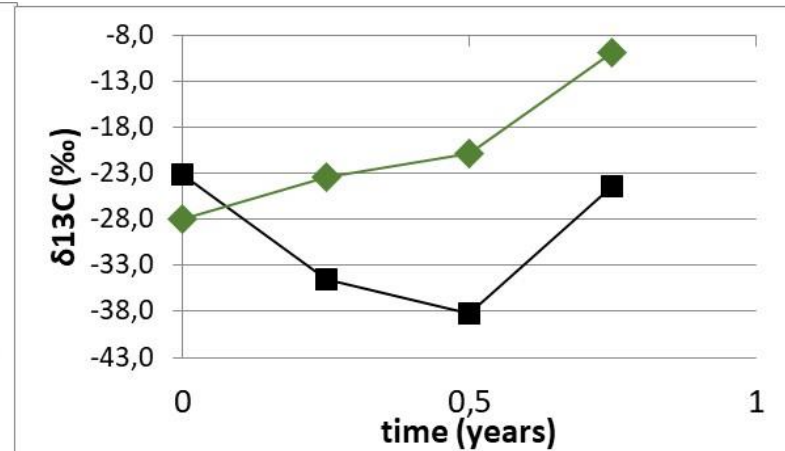
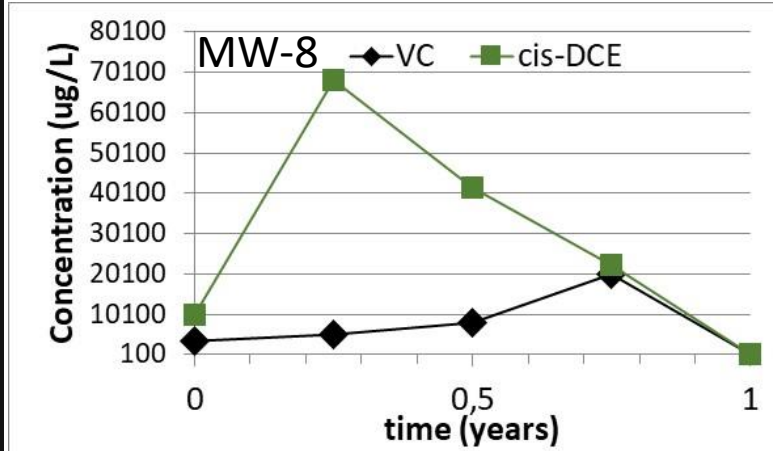
CSIA Application during In situ Remediation Treatment

- Questions for CSIA:
 - Is the treatment occurring as intended ?
 - Dilution or degradation?
 - Was TCE pushed downward, into the sandy layer ?



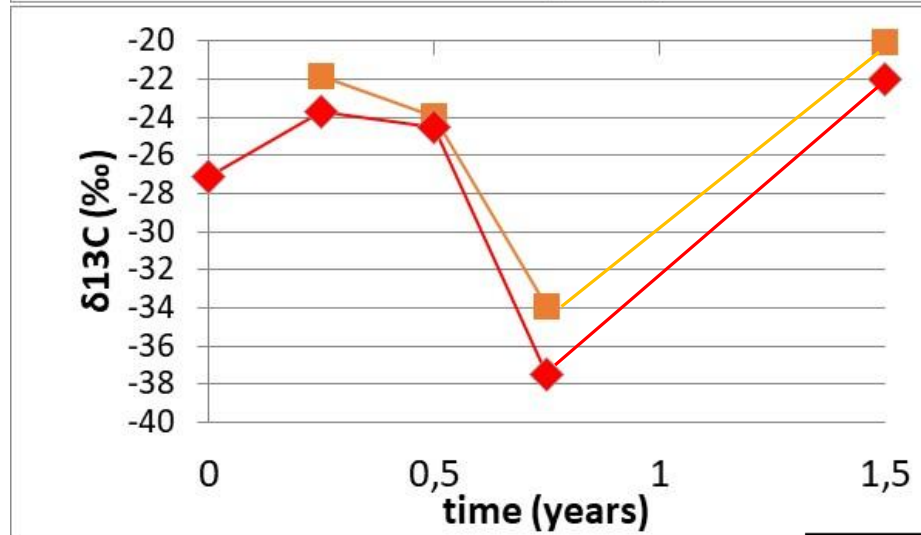
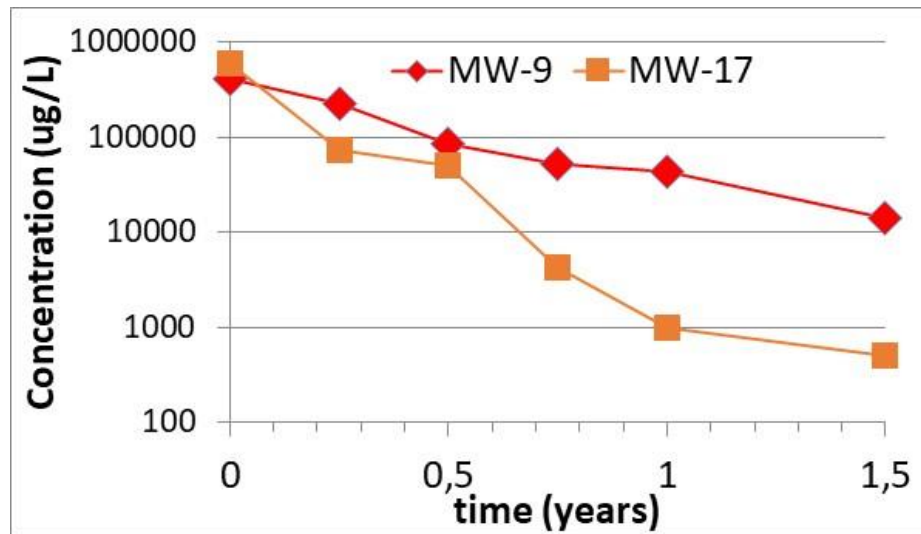
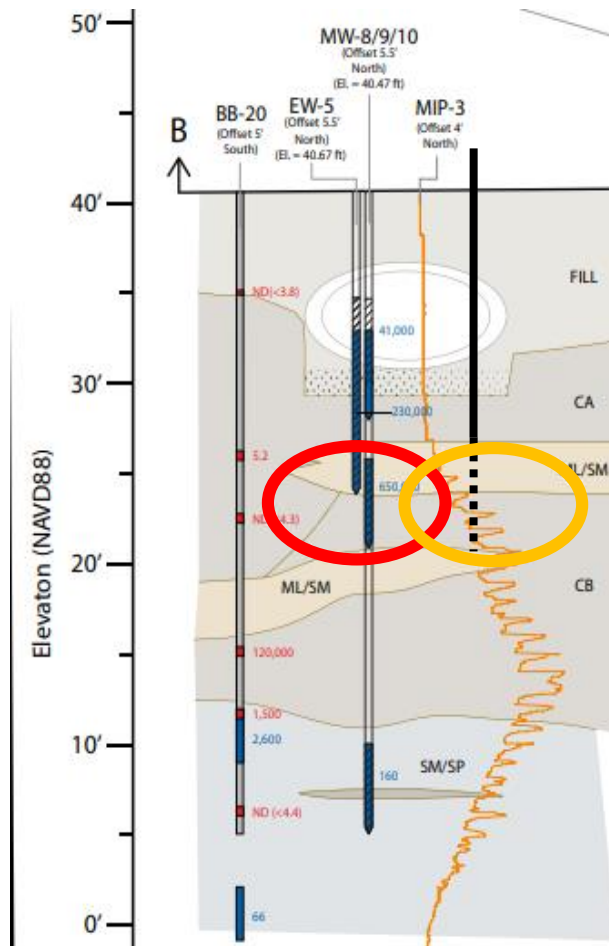
Results

CSIA
Application
during In situ
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Results

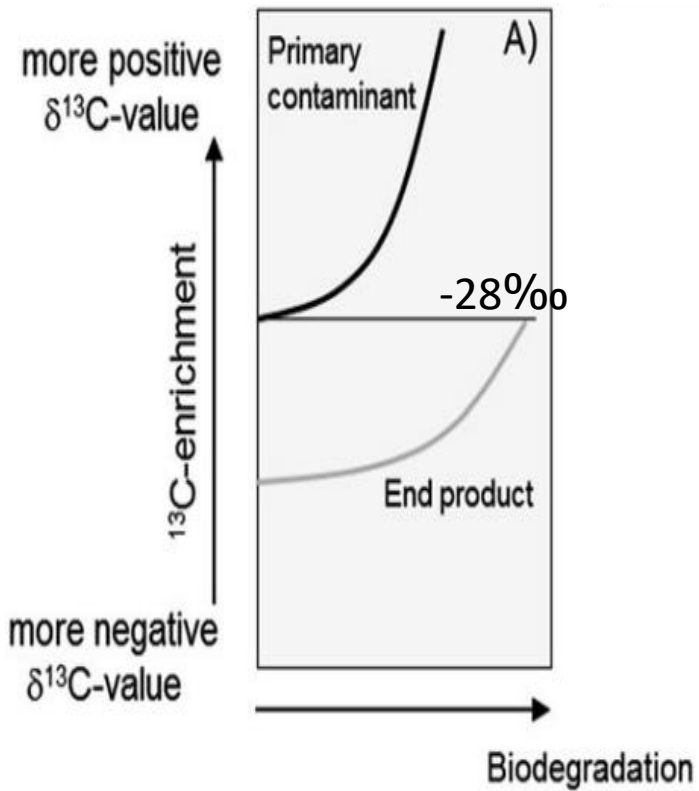
CSIA
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Results - Isotopic balance

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MW	event	TCE		cis-DCE		VC		Total $\delta^{13}\text{C}$ (‰)
		Conc (ug/L)	$\delta^{13}\text{C}$ (‰) mean	Conc (ug/L)	$\delta^{13}\text{C}$ (‰) mean	Conc (ug/L)	$\delta^{13}\text{C}$ (‰) mean	
MW-8	Baseline	1960	-25,0	10100	-28,0	3580	-23,1	-26,1
	Q 1	<1000	BDL	68200	-23,5	5210	-34,6	-24,7
	Q 2	<250	BDL	41400	-20,8	7870	-38,2	-24,8
	Q 3	57	-12,5	22200	-9,9	19900	-24,5	-18,4
	Q 4	24,5	--	118	--	19,9	--	--



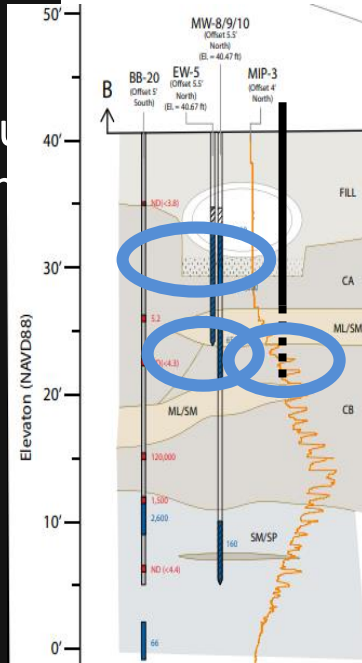
Isotopic balance :

$$\delta^{13}C_{sum} = \underbrace{x_{PCE} \cdot \delta^{13}C_{PCE}}_{\text{No full dechlor}} + x_{TCE} \cdot \delta^{13}C_{TCE} + \underbrace{x_{DCE} \cdot \delta^{13}C_{DCE}}_{\text{No full dechlor}} + x_{VC} \cdot \delta^{13}C_{VC} = -28\text{‰}$$

= > -28‰ full dechlor
 = -28‰ No full dechlor

Results - Isotopic balance

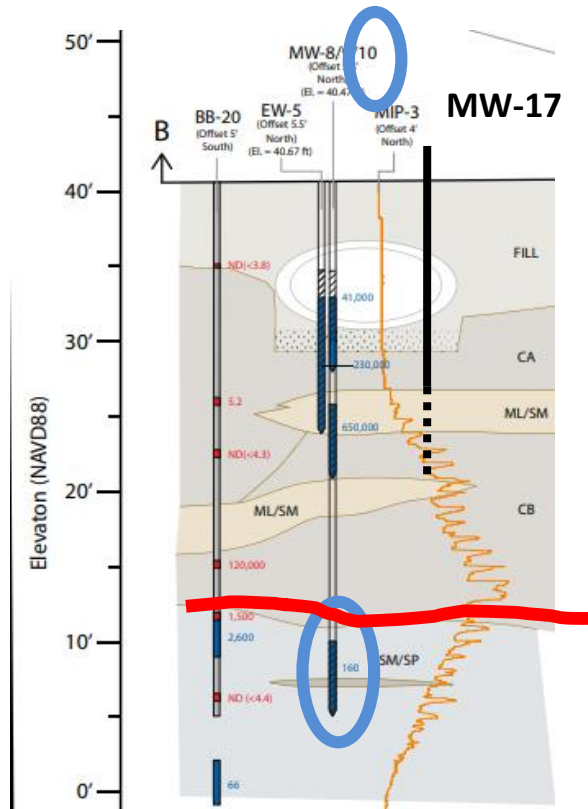
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	Q 4	24,5	--	118	--	19,9	--	--
MW-9	Baseline	409000	-27,1	42700	-32,0	<5000	-26,3	-27,7
	Q 1	229000	-23,7	307000	-29,9	<2500	-16,4	-27,6
	Q 2	86700	-24,5	357000	-24,7	<5000	-37,2	-24,9
	Q 3	52100	-37,5	420000	-31,2	2690	-30,0	-31,7
	Q 4	44100	--	430000	--	7850	--	--
	Q-5	13900	-22,0	619000	-23,7	47800	-39,8	-25,4
MW-17	Baseline	598000	na	48900	na	<1000	na	--
	Q 1	74000	-21,8	561000	-25,8	2170	-27,7	-25,4
	Q 2	50500	-24,0	258000	-28,2	3660	BDL	-27,1
	Q 3	4230	-34,0	557000	-23,6	68700	-35,3	-25,5
	Q 4	<10000	--	321000	--	80800	--	--
	Q-5	<5000	-20,1	290000	-21,8	55100	-26,1	-22,8

Site 4 –Results

CSIA
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during In situ
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MW	event	TCE	
		Conc (ug/L)	$\delta^{13}\text{C}$ (‰) mean
MW-10	Baseline	100	-21,4
	Q 1	14,4	-16,0
	Q 2	3,71	na
	Q 3	2,74	-11,8
	Q 4	8,74	--

Benefits of CSIA

1. Confirms COC degradation has occurred as intended (large shift)
2. No DCE stall, but periodic VC stall
3. Supports full dechlorination occurrence (due to abiotic and biotic processes)
4. Supports the selected dosage
5. Direct push did not displace TCE downward
6. Mass degraded:
 - I. Rayleigh equation – not fully respected
 - II. Enrichment factor – mixt of biotic and abiotic process

CSIA is a reliable tool to assess performance of remediation treatments

- i. Gain information independently from concentration data
- ii. For various in situ approaches (biological, chemical, physical)

CSIA application possibilities

- i. Using 1 isotope:
 - destruction versus dilution
- ii. Using 2 isotopes – to target a specific mass removal process among co-occurring mass removal process
 - Aerobic vs anaerobic biodegradation
 - Chemical oxidation vs biodegradation
 - Biodegradation vs volatilization

To document treatment efficiency

- i. Pilot scale
 - i. Proof of concept
 - ii. Dimensioning full scale
- ii. Full scale
 - i. Document the progress
 - ii. Support decision to pursuit, to optimize the treatment
 - iii. Provide evidences of treatment limitations have been reached
 - iv. Support transitioning to next phase

Acknowledgments

Field work

- Tyson Fulmer – CE solution
- David Alden - Tersus Environmental

CSIA

- Isotope Tracer Technologies - Orfan Shouakar Stash



*** Thank You**

Site 4 –Results

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MW-10	Baseline	100	-21,4	400	-27,8	<10	na	-26,8
	Q 1	14,4	-16,0	413	-24,6	23,7	BDL	-22,4
	Q 2	3,71	na	272	na	225	na	--
	Q 3	2,74	-11,8	118	-21,3	164	-20,8	-20,9
	Q 4	8,74	--	133	--	137	--	--

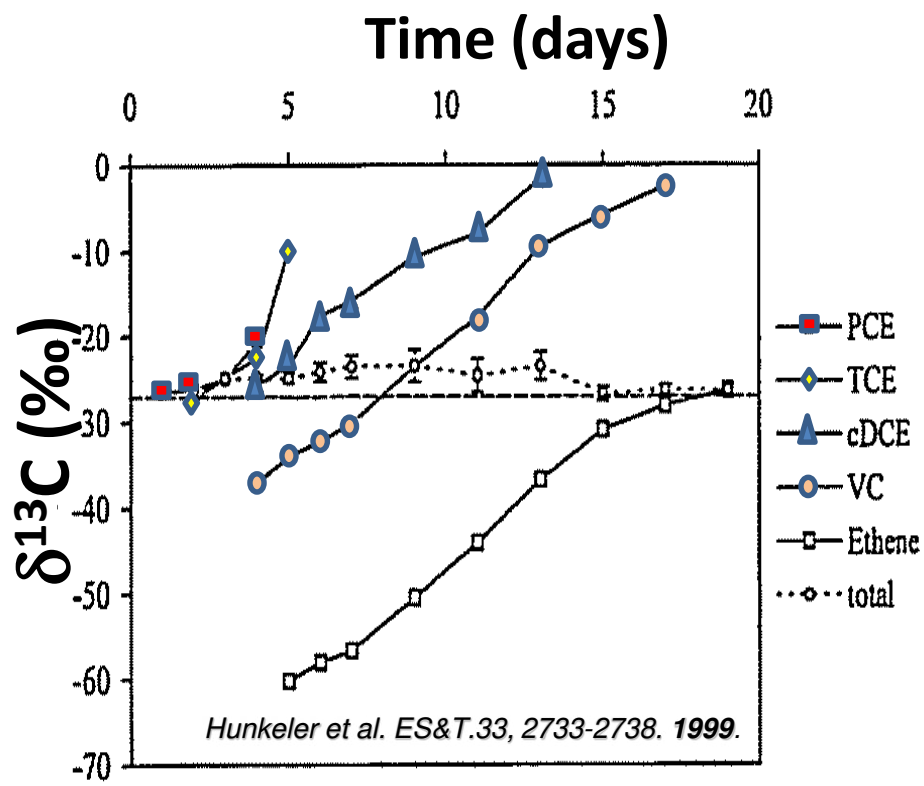
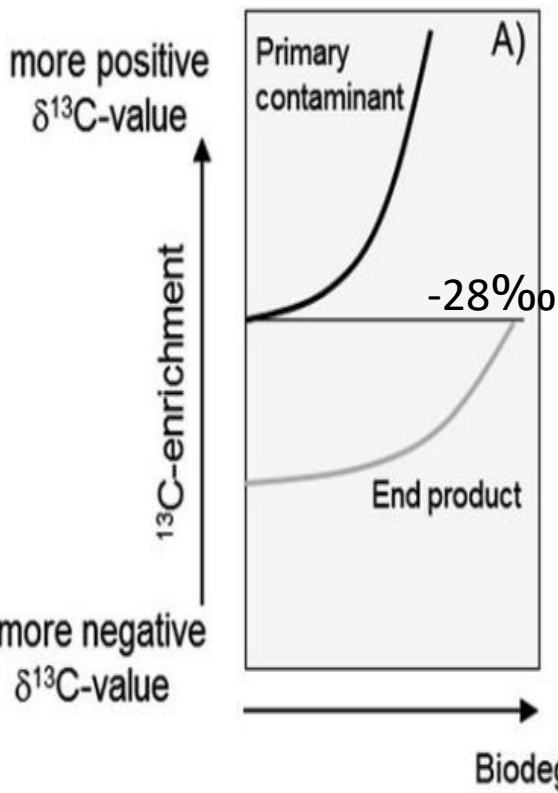
Fully dechlorination is occurring, thanks to abiotic process

Sampling procedures

- i. Groundwater Sampling
 - Similar field procedure as for VOC concentrations
 - Additional 40 ml vials
 - Preservative, shipping and storage conditions
 - Longer holding time (1-3 months)
 - Detection limits (BTEX)
 - For $\delta^{13}\text{C}$: 5-25 ug/L
 - For $\delta^2\text{H}$: 50-100 ug/L
- ii. Gas-phase sampling
 - Sorbant tubes
 - Solvent-based method

Costs (approx.)

- $\delta^{13}\text{C}$: 140 \$ / compound
- $\delta^2\text{H}$: 150 \$ / compound
- $\delta^{37}\text{Cl}$: 450 \$ first cVOC, then 200 \$ (PCE + TCE = 650\$/sample)

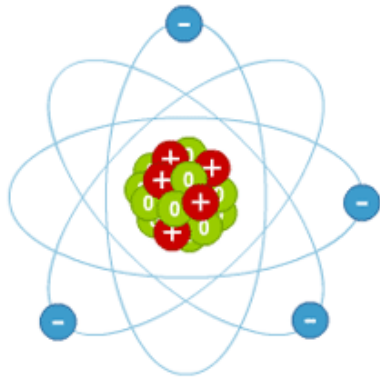


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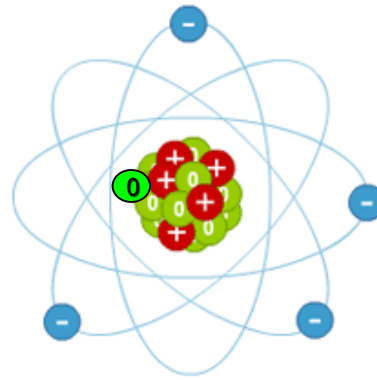
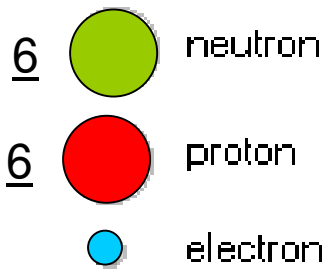
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Carbon atom



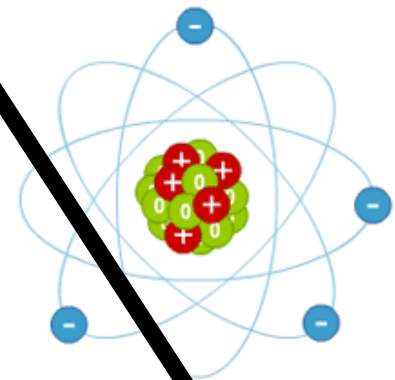
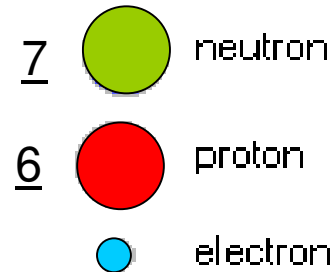
Carbone 12

Atome stable



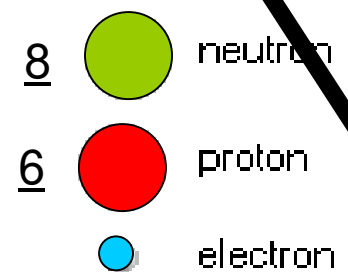
Carbone 13

Atome stable

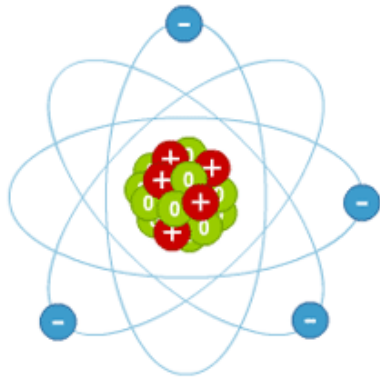


Carbone 14

Atome instable

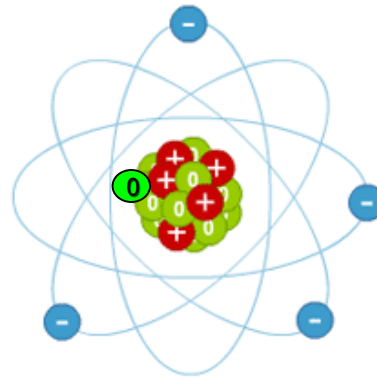
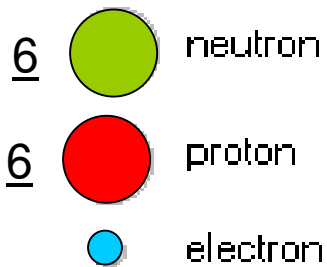


Carbon atom



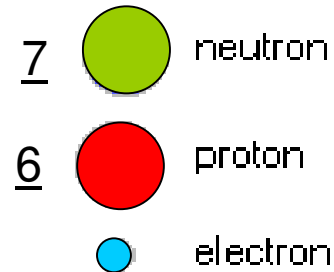
Carbone 12

Atome stable



Carbone 13

Atome stable



Abondance naturelle

1 : 88

^{12}C : 98,89 %

^{13}C : 1,11 %

^1H : 99,985 %

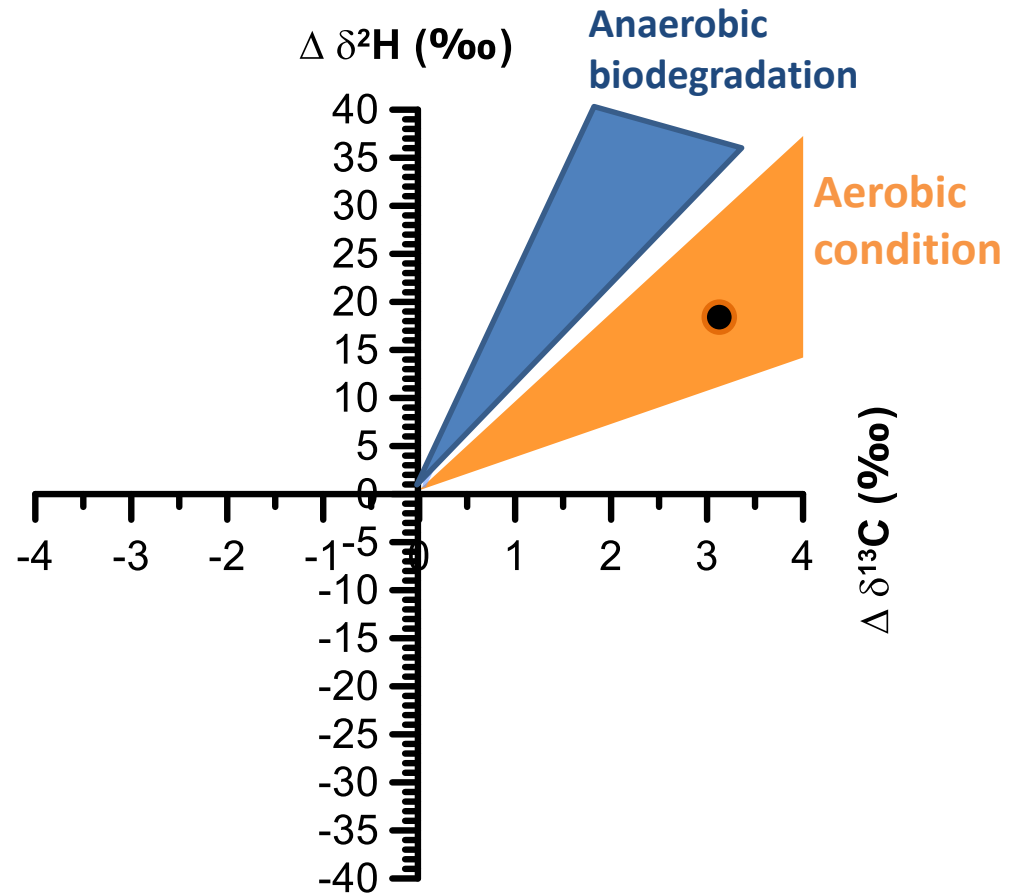
^{35}Cl : 75,7 %

^2H : 0,0155 %

^{37}Cl : 24,3 %

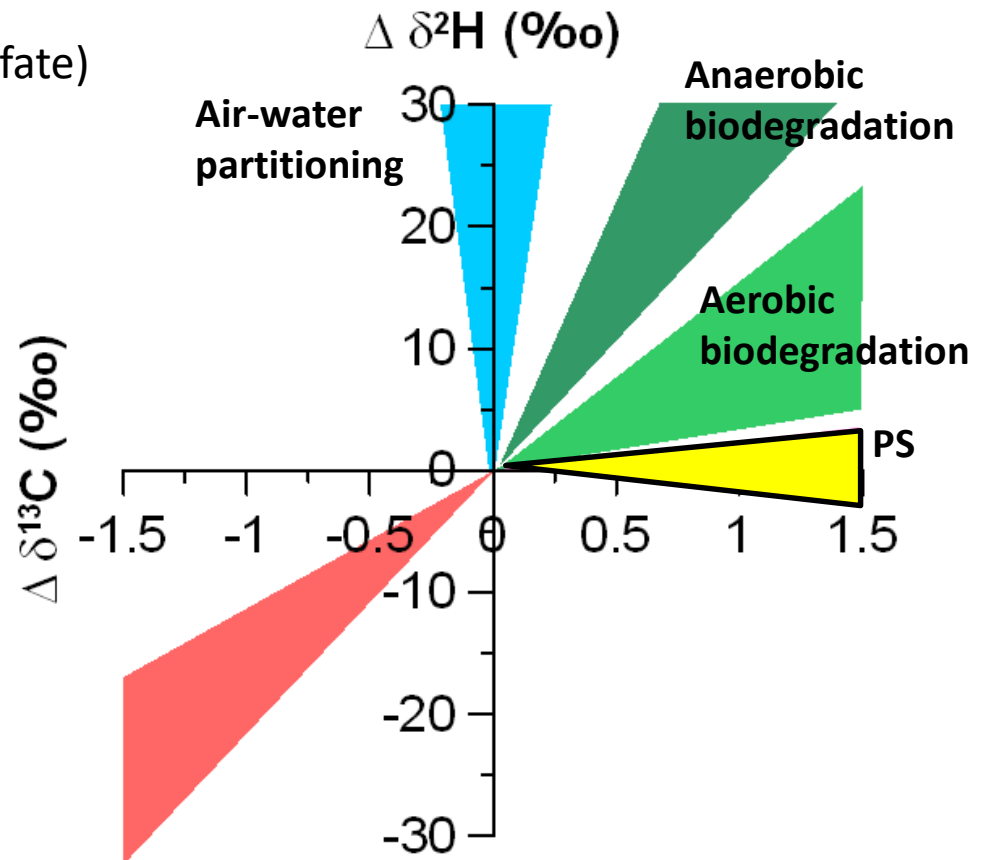
Implication of heavy isotopes

- Biodegradation of benzene
- Different enrichment pattern
 - Aerobic
 - Anaerobic



Implication of heavy isotopes

- Biodegradation of benzene
- Different enrichment pattern
 - Aerobic
 - Anaerobic
 - ISCO (unact Persulfate)
 - Air-NAPL parti
 - Air-Water parti



Dual isotope plot concept: to distinguish destructive and non-destructive processes

How to plot field results:

- Calculate Δ (‰) for C and H
- δ Sample - $\delta^{13}\text{C}$ Source
- $(\Delta \delta^{13}\text{C}, \Delta \delta^2\text{H}) = (1.5, 20)$

