



# Table of Contents

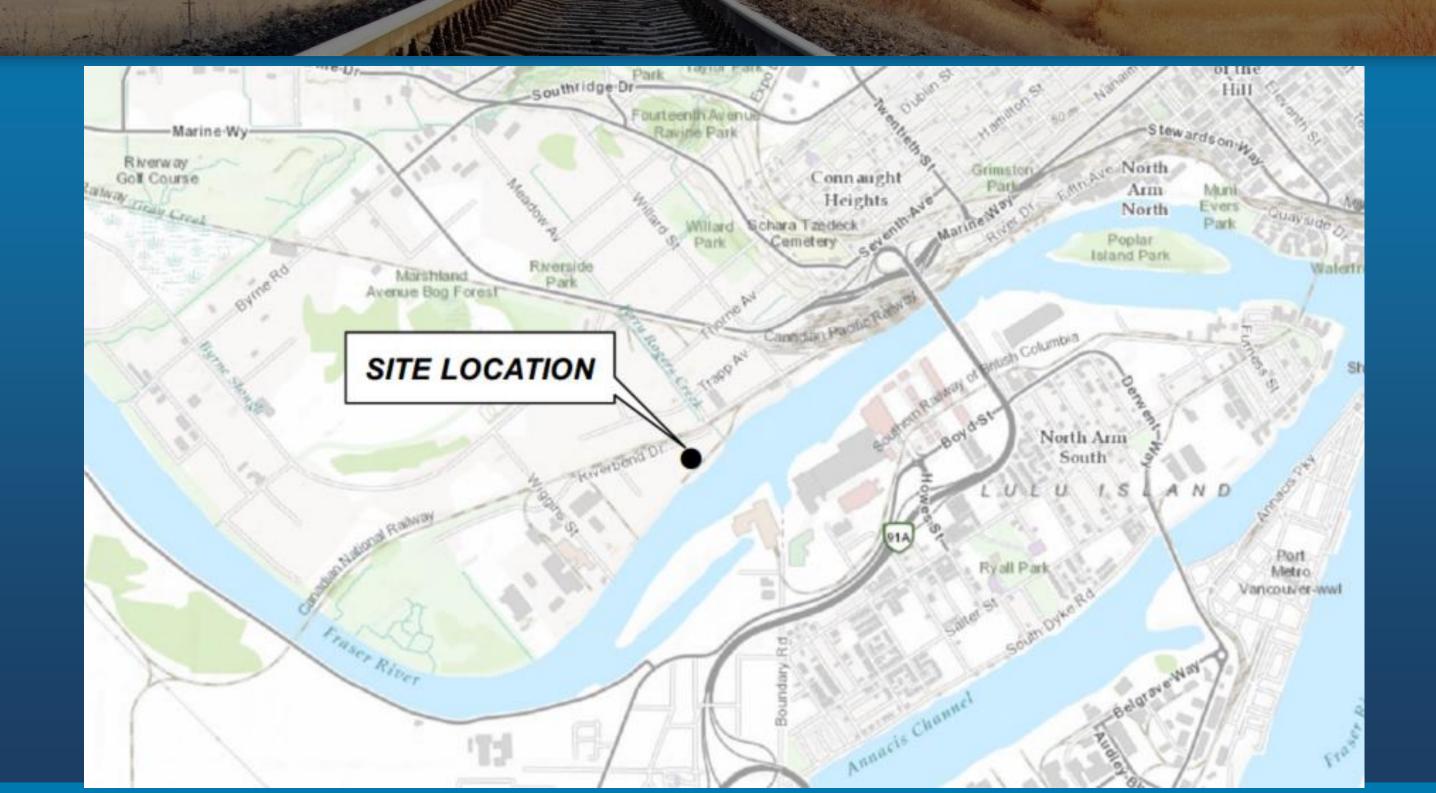


- 1. Introduction
- 2. Site Background
- 3. Conceptual Site Model
- 4. Regulatory Environment
- 5. Arsenic Geochemistry
- 6. Chemical Injection Program
- 7. Results





# BACKGROUND: SITE LOCATION

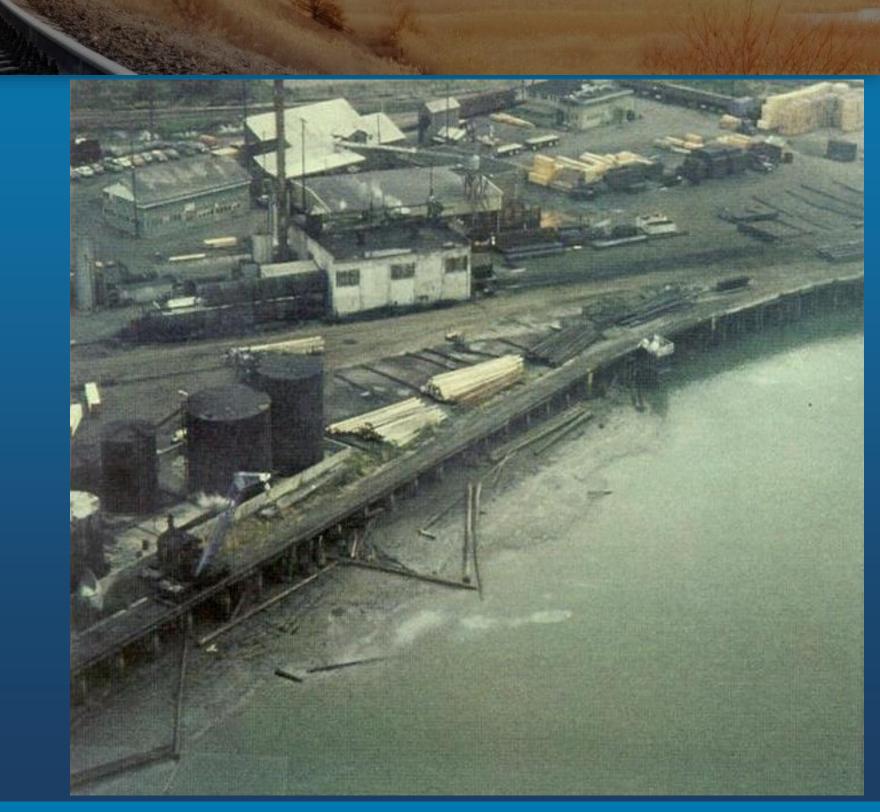




#### BACKGROUND: SITE HISTORY

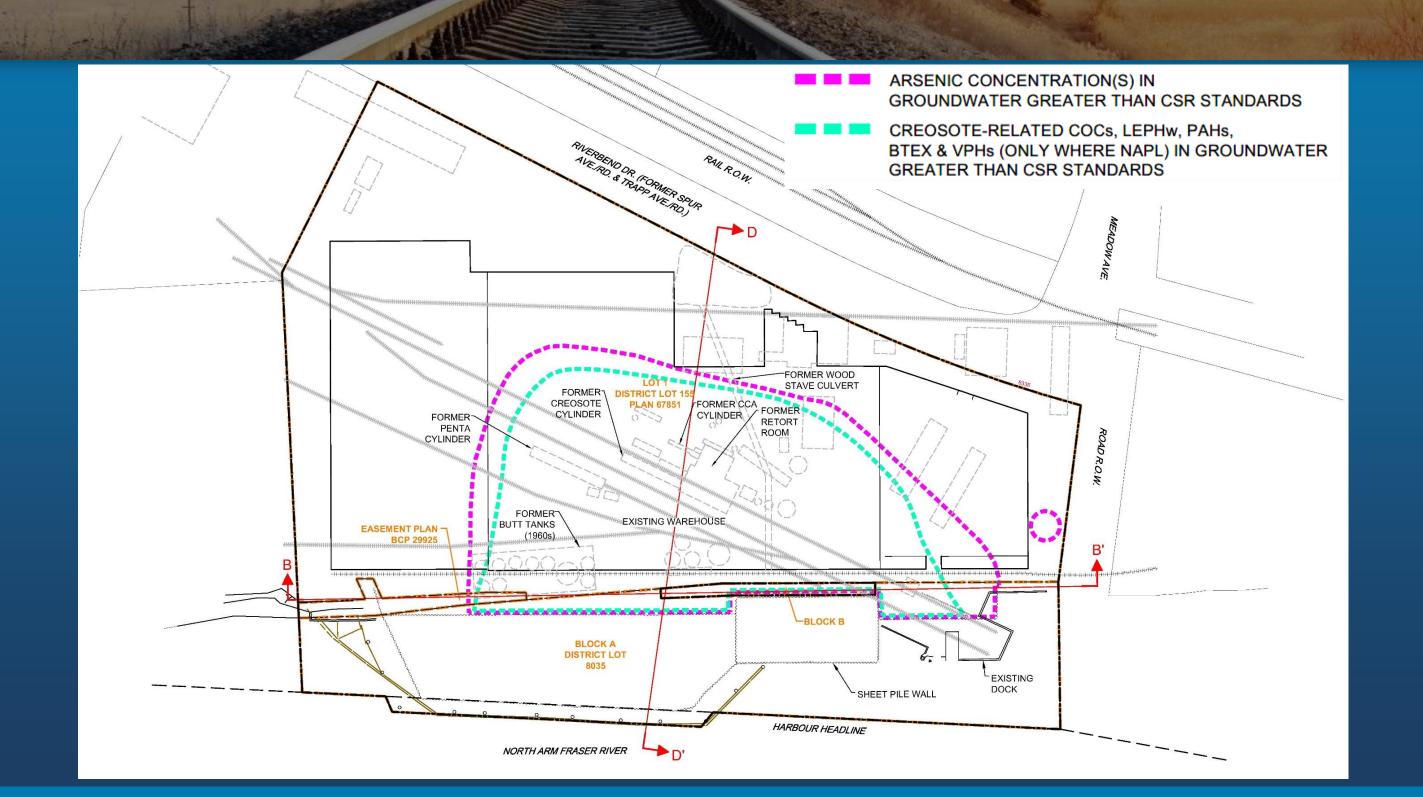
#### **Wood Preservation Activities**

- Creosote
   DNAPL and dissolved phase constituents (PAHs)
- Copper-chromium-arsenate (CCA)
   Dissolved arsenic
- Pentachlorophenol (PCP)



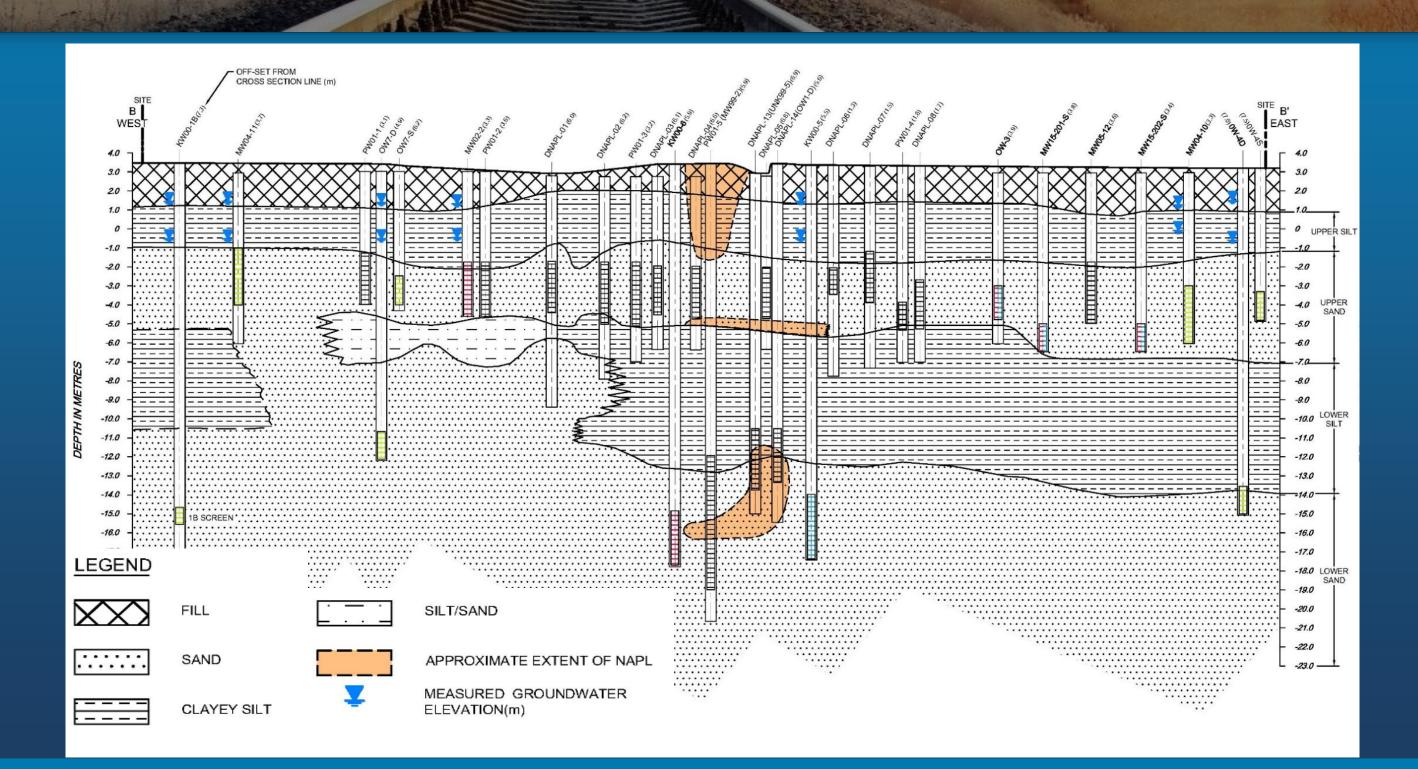


### BACKGROUND: SITE HISTORY





# CONCEPTUAL SITE MODEL SITE GEOLOGY





#### Remedial works completed in 2004:

- Dredging
- Installation of cap and sheet pile walls
- Construction of new industrial wharf
- Backfilling and construction of a new marsh



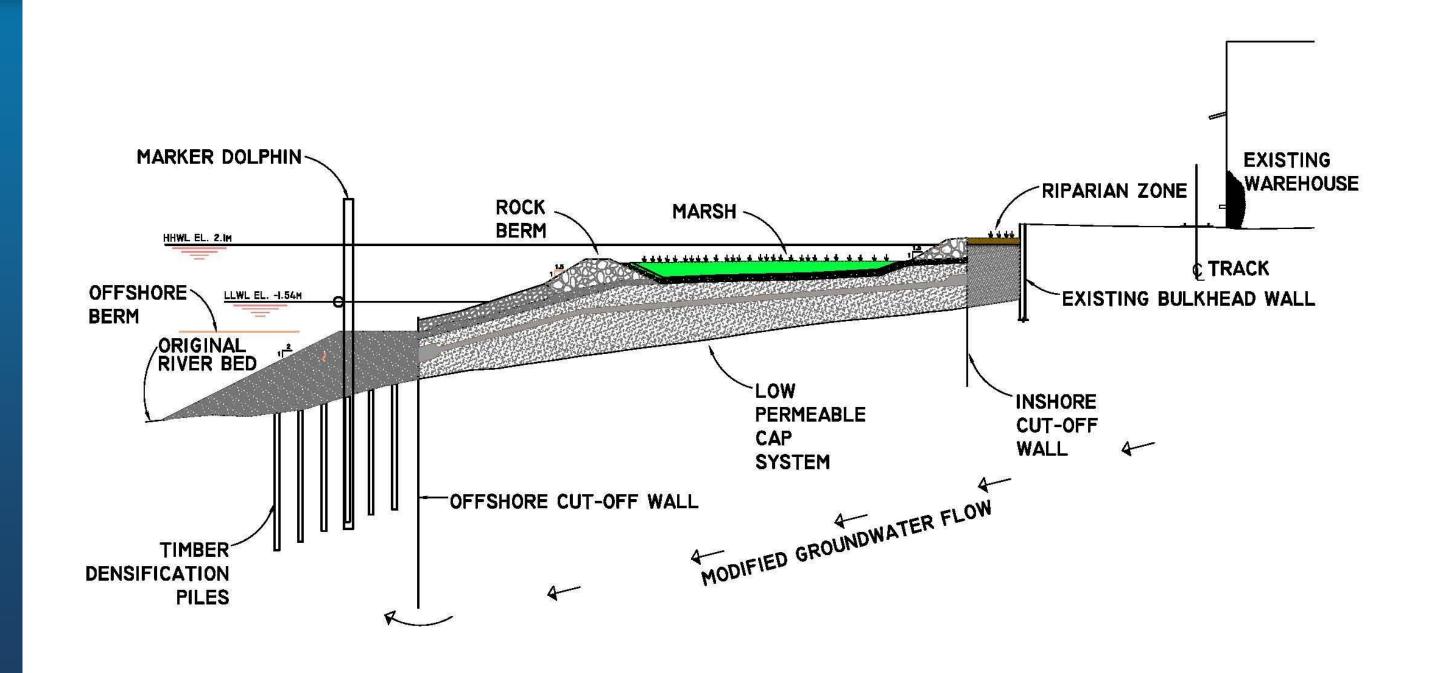














### REGULATORY ENVIRONMENT

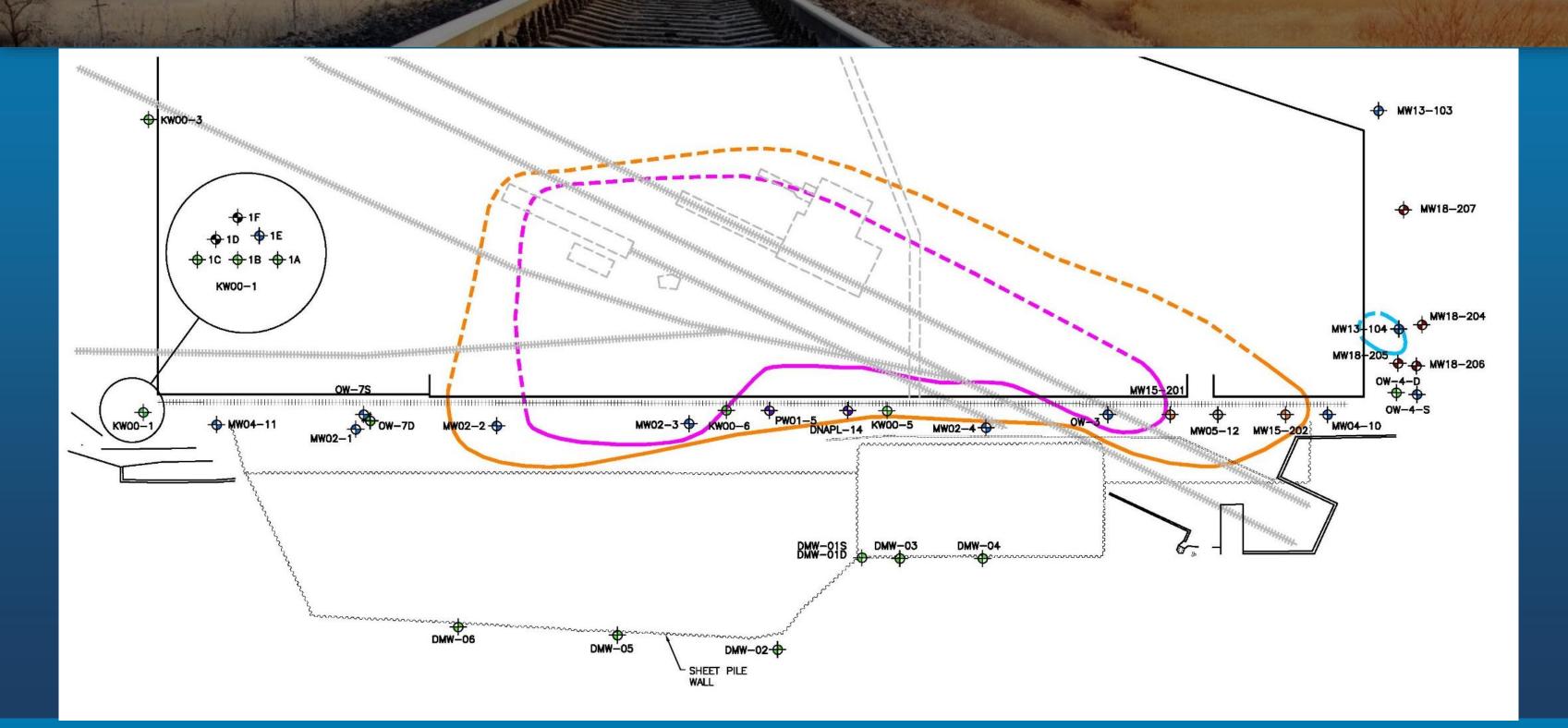


#### Performance Monitoring Plan (PMP)

- Defines ongoing sampling, monitoring, and inspections
- Used BC CSR framework as basis to establish site-specific toxicity reference values (TRVs)
- Establishes trigger criteria to increase or reduce frequency of sampling
- Initiates further actions if thresholds are exceeded



### DISSOLVED ARSENIC





### SITE CONDITIONS



- Warehouse occupies more than 90% of the upland footprint.
- Spur line runs between warehouse and river.
- Active warehouse operations 24/7.



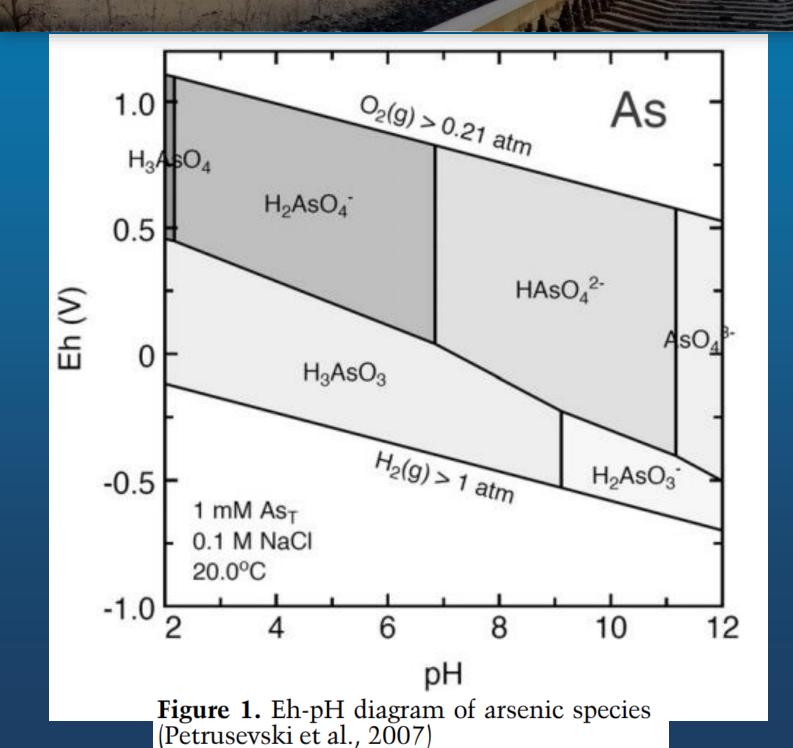
#### SITE GEOCHEMISTRY



- The site fluvial sediments at the site contain organics which contribute to reducing conditions at the site
- PAH contamination and active anaerobes consuming electron acceptors at the site influences the site geochemistry to have strong reducing conditions



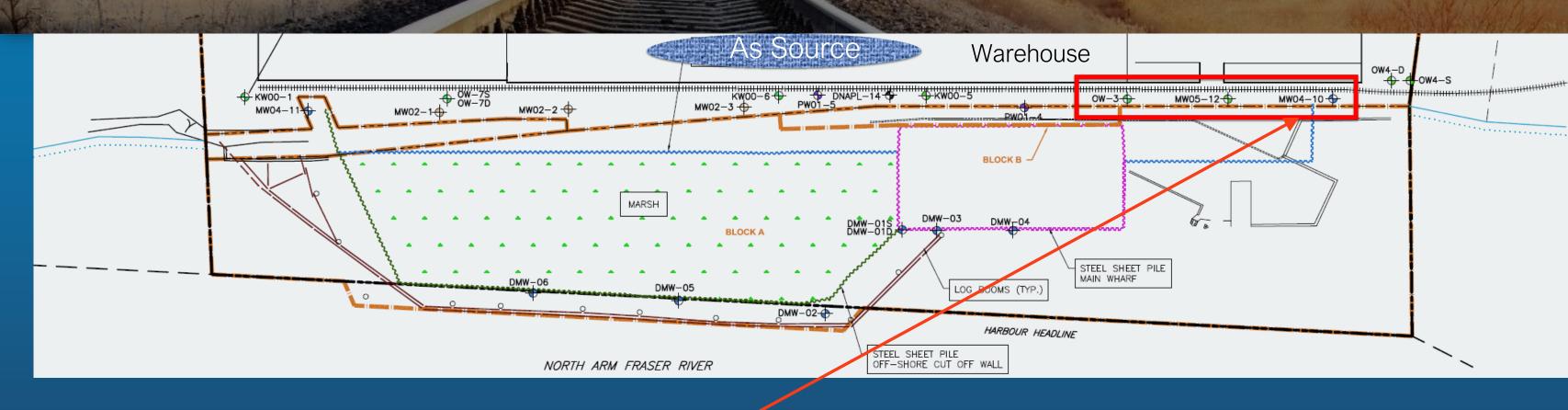
#### ARSENIC GEOCHEMISTRY



- Arsenic mobility is greatly affected by its speciation.
- In groundwater, inorganic Arsenic
   predominantly exists as As (V) arsenate, and As
   (III) arsenite.
- Arsenic in groundwater is predominantly in the form of arsenite at the site due to the reducing conditions present
- Arsenite is more mobile than arsenate, and typically remediation of arsenic in groundwater focusses on oxidizing arsenic to arsenate.



### DISSOLVED ARSENIC EXCEEDANCES



Sample ID	RDL				MW04-10				TRV	CSR
Date Sampled		02-Jun-09	25-May-10	24-May-11	24-May-12	04-Dec-12	11-Jun-13	02-Jun-14		(AW <sub>FW</sub> )
Arsenic	0.1	3.3	10	12.4	21.9	<u>93</u>	<u>104</u>	<u>88.5</u>	55	50

Target []

Dissolved As concentration increasing over time



# Pre-Treatment Dissolved Arsenic Concentrations 2008-2015 OW-3 (near the source) and MW04-10 (near the discharge zone)





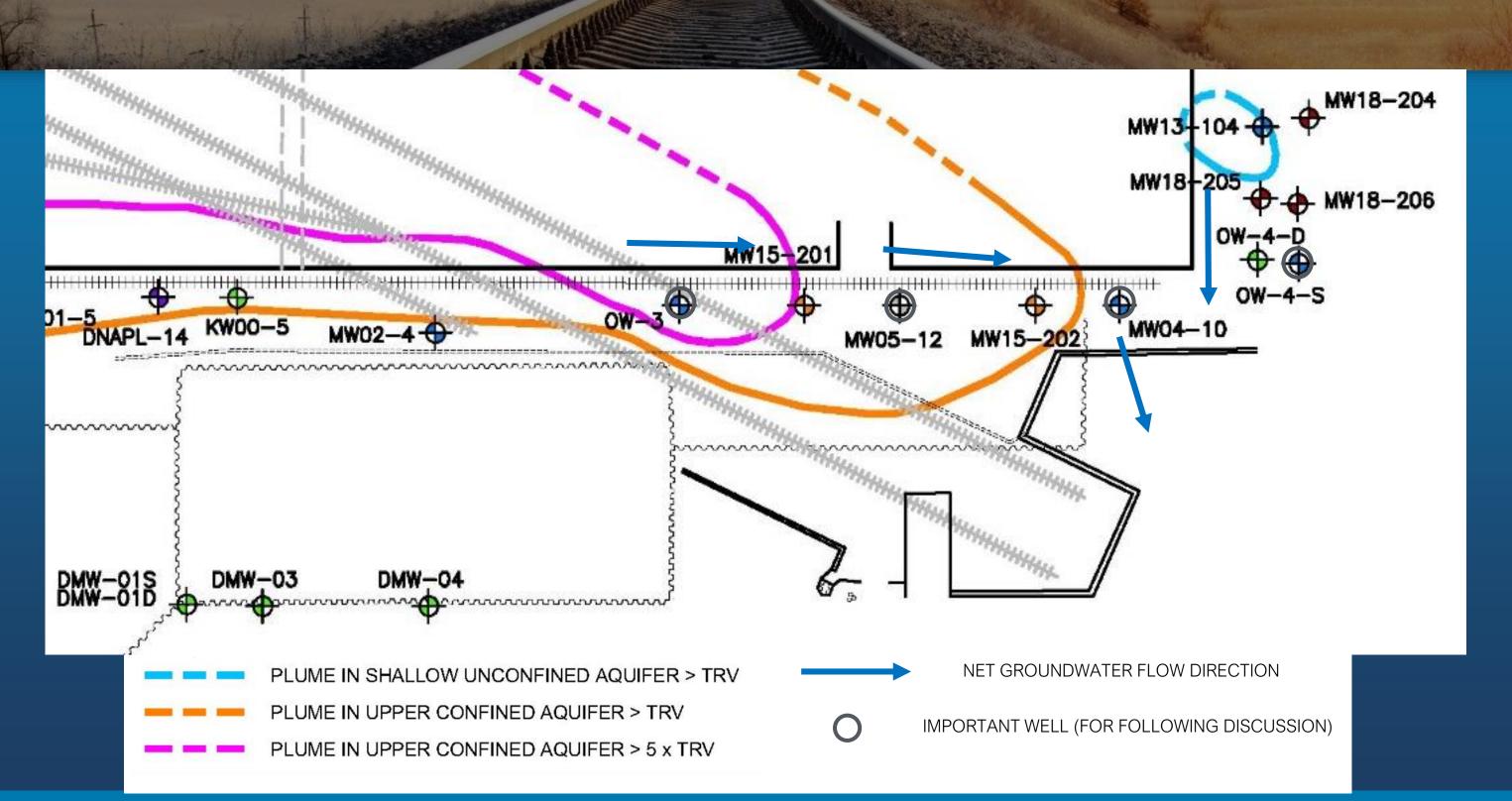
### INJECTION PROGRAM OBJECTIVES



- Investigate the rate of arsenic immobilization, the injection radius of influence, and the effects, benefits and limitations of different injection methods on the site geochemistry
- Reduce dissolved arsenic concentrations in groundwater below applicable standards (CSR and TRV)
- Reach stable / decreasing concentrations



### ARSENIC REMEDIAL STRATEGY





# CHEMICAL INJECTION AREA





# PILOT CHEMICAL INJECTION PROGRAM CHEMICAL SELECTION



Arsenic reacts with iron and sulphide from reduced sulphate to form arsenopyrite (FeAsS) precipitate

- Organic amendment
- Zero-valent iron
- Sulphate source



#### FORMATION OF STABLE ARSENIC PRECIPITATE



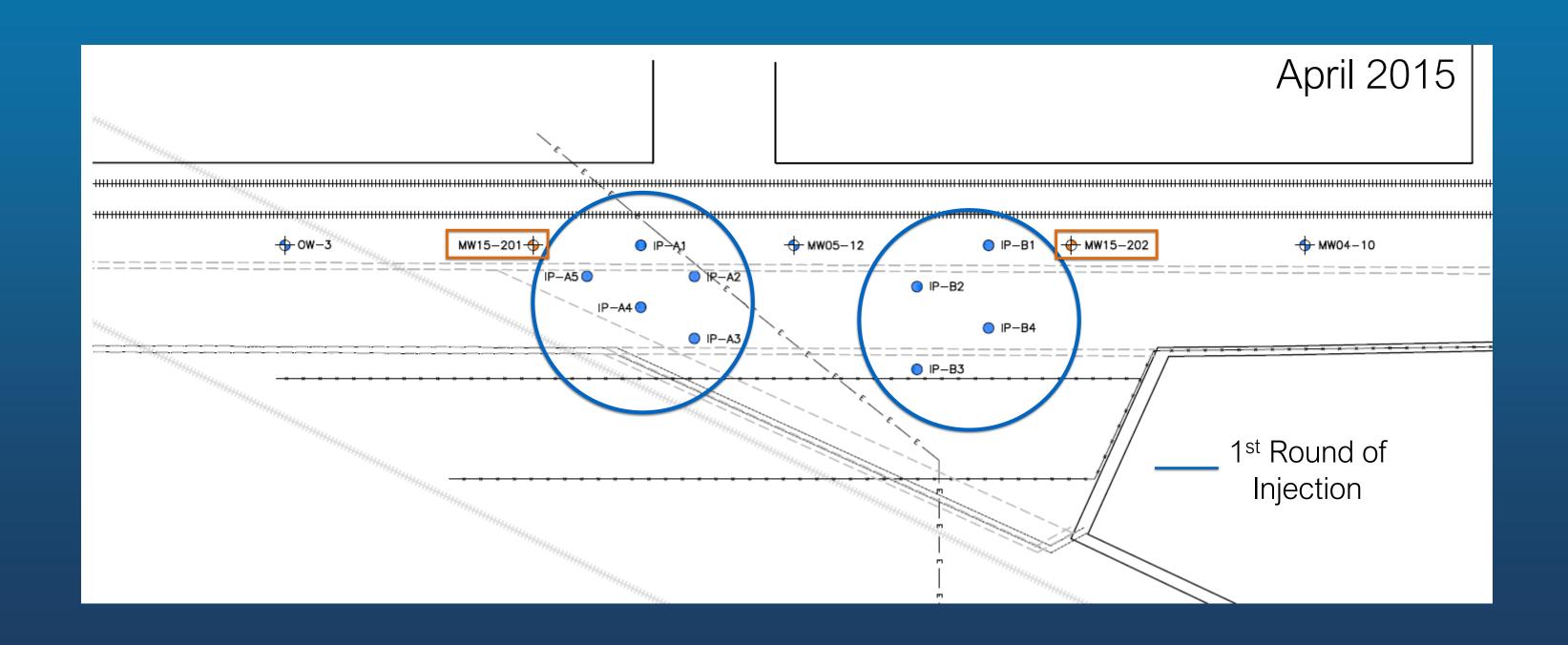
$$2 \text{ Fe}^0 + \text{O}_2 + 2\text{H}_2\text{O} = 2 \text{ Fe}^{2+} + 4 \text{ OH}^-$$

$$C\text{H}_4 + S\text{O}_4^{2-} = \text{HCO}_3^- + \text{HS}^- + \text{H}_2\text{O}$$

$$2\text{Fe}(\text{OH})_3 + 2\text{HS}^- + 2\text{H}_3\text{AsO}_3 = 2\text{FeAsS} + 4\text{OH}^- + 3\text{O}_2 + 5\text{H}_2\text{O}$$



### PILOT CHEMICAL INJECTION PROGRAM





### FIRST ROUND OF CHEMICAL INJECTION

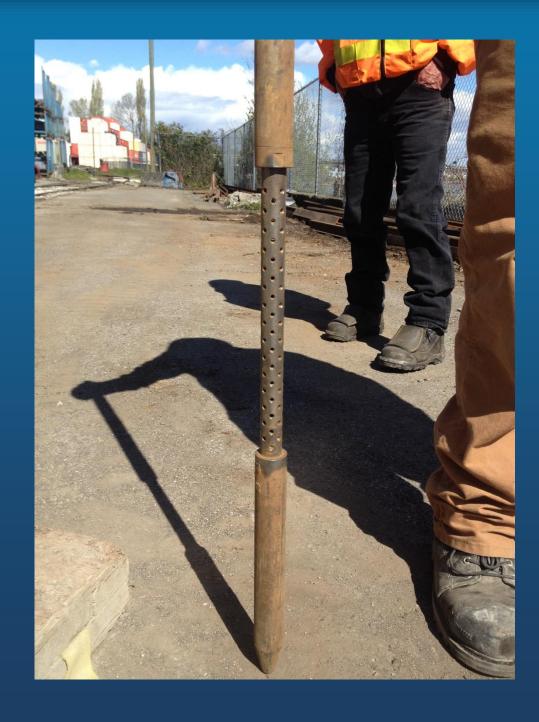


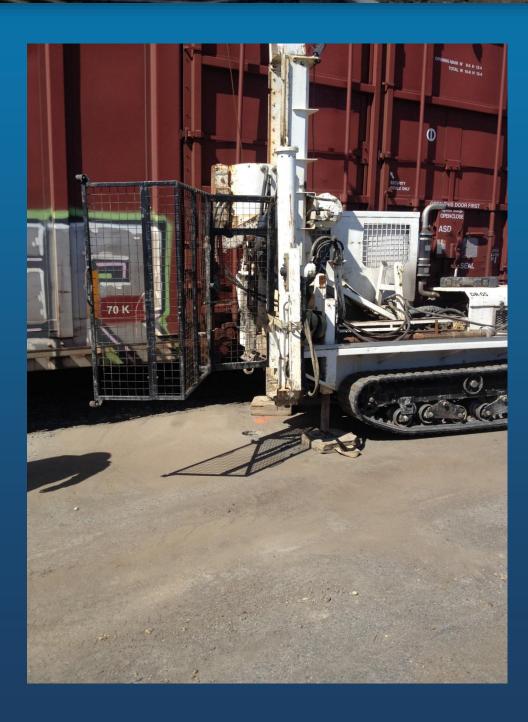
- Direct push injection selected
- Targeted slurry mixture of 25%-35% solids content
- Delivered chemical through soil fracture to 3 m radius, at depth from 10 mbg to 5 mbg
- Delivered half of the calculated chemical through this method
- Problem = short-circuiting from soil fracture into adjacent MW





# FIRST ROUND OF CHEMICAL INJECTION

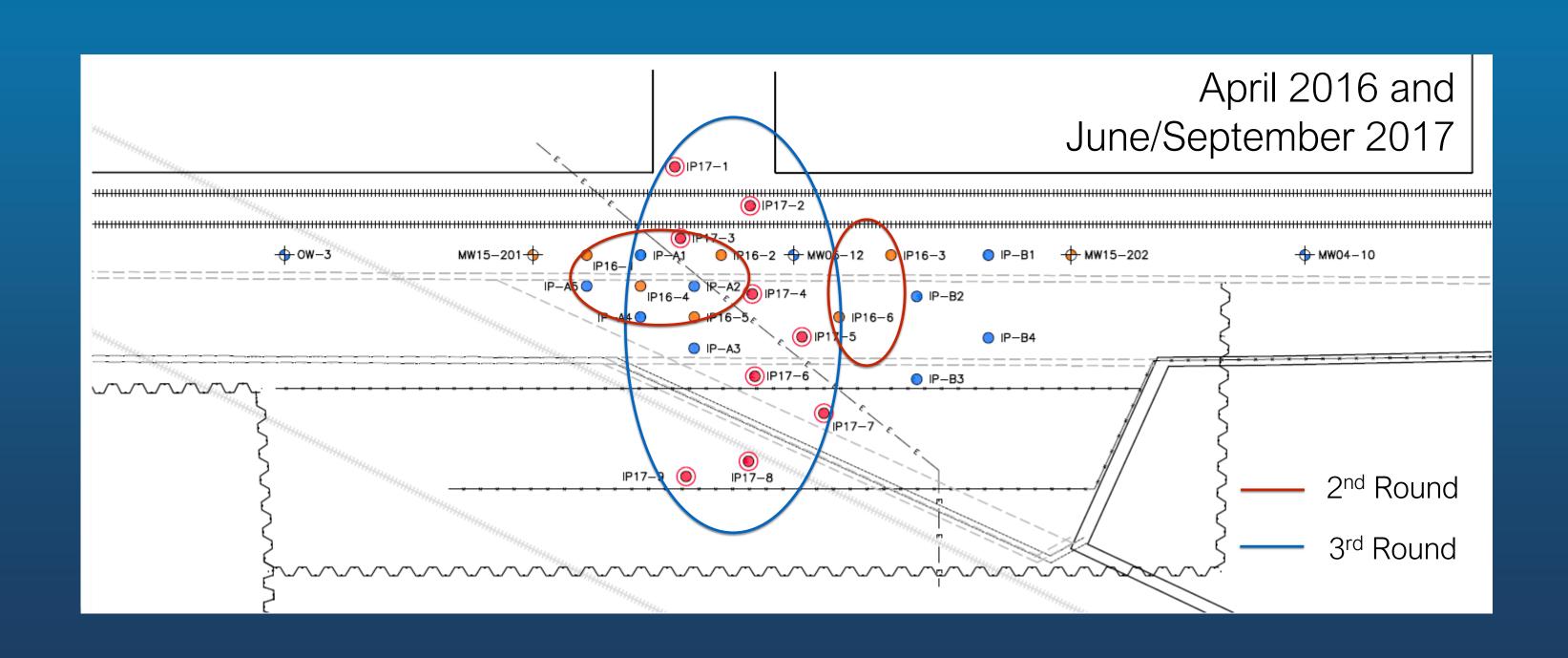








### SECOND & THIRD ROUND OF CHEMICAL INJECTION

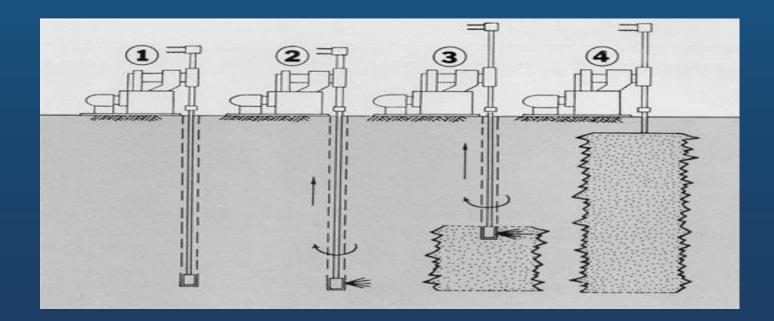




# SECOND & THIRD ROUND OF CHEMICAL INJECTION INJECTION METHODOLOGY



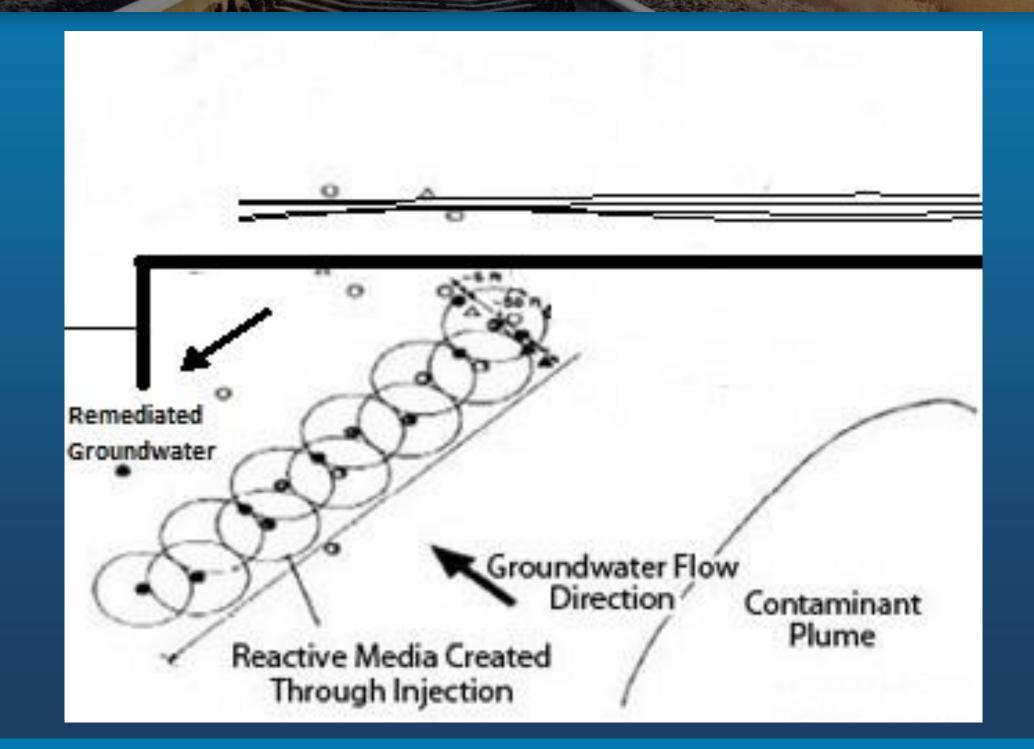
- Chemical injected at 16 locations
- Pressure grout method selected
- Delivered chemical through high-pressure rotating tip to form chemical / soil mix columns
- Displaced media is released to surface through the drill pipe to reduce potential for short-circuiting to other release points





# INJECTION LAYOUT FOR THIRD AND FOLLOWING ROUNDS







#### CHEMICAL SELECTION AMENDMENT



#### **Slurry matrix Round 2:**

- EHC-M®
- Calcium Lactate carbon source
- Epsom salt (MgSO4) fast Rx SO<sub>4</sub>

#### **Slurry matrix Round 3/4:**

- EHC-M®
- Calcium Lactate Gluconate carbon source
- Epsom salt (MgSO4) fast Rx SO<sub>4</sub>
- Gypsum slow Rx SO<sub>4</sub>
- Ferrous Sulphate additional iron source and medium Rx SO<sub>4</sub>







# CHEMICAL INJECTION EQUIPMENT







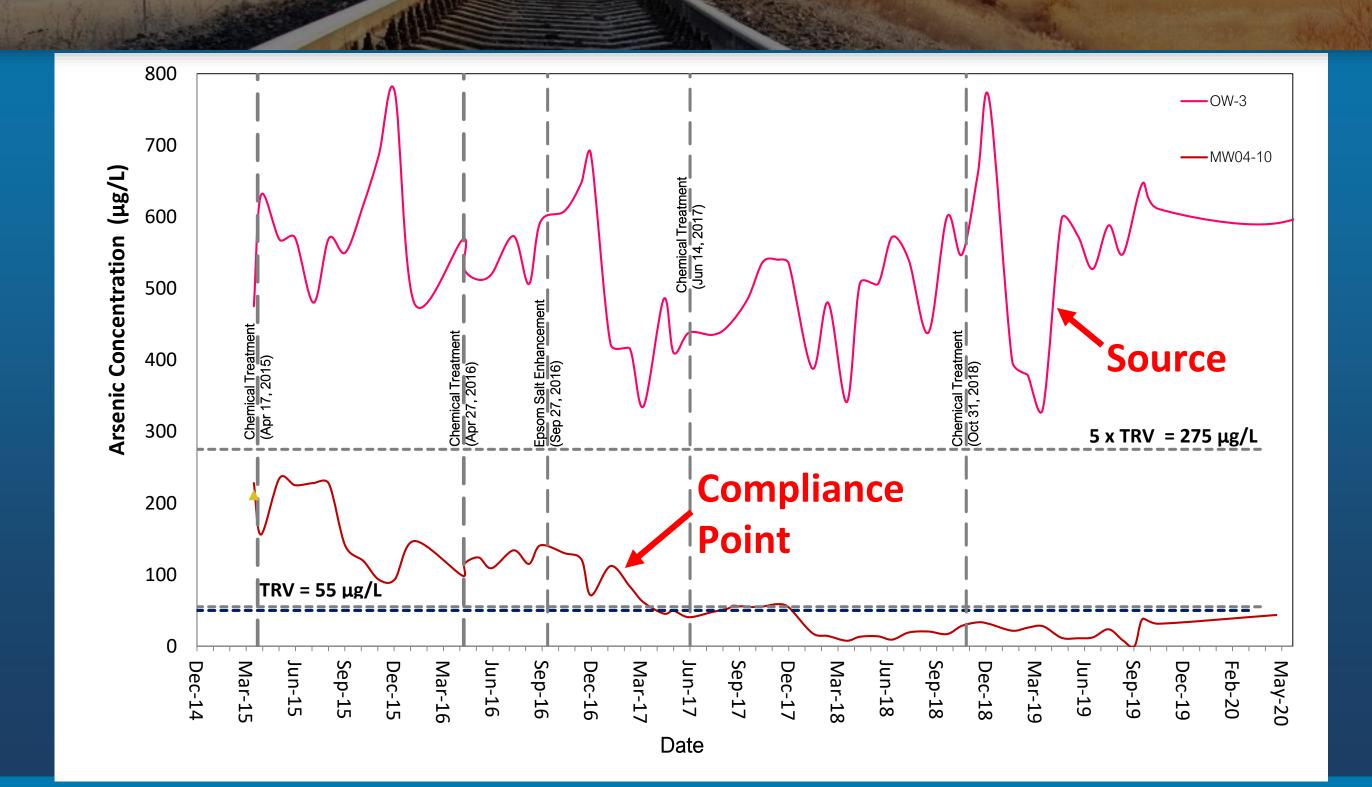
# CHEMICAL INJECTION EQUIPMENT







#### ARSENIC RESULTS





#### SUMMARY OF INITIAL PILOT PROGRAM



- Reduced concentrations below standard
- Preference to achieve lower ORP following injection
- Sulphate was moving past the injection area soon after the injection event
- Post-treatment monitoring shows slow increasing concentration trend due to ongoing source which requires further monitoring and treatment



### INJECTION CHEMISTRY CHANGE



#### Injection was modified to include:

- Ferric Sulphide
- Zero Valent Iron
- Iron oxides
- Carbonates







#### **THANK YOU**

