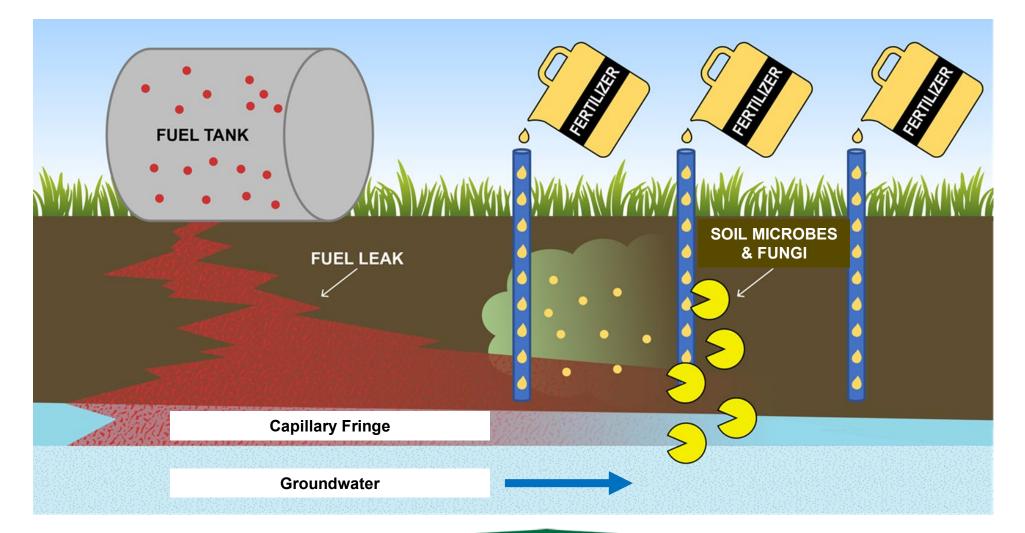


The Role of Carbonates in the Effectiveness of Biostimulatory Solutions for the Removal of Petroleum Hydrocarbon Contaminants in Cold Calcareous Soils

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	October 15	5, 2020	



Project Background





Project Background

- Data gaps surrounding effectiveness of in situ bioremediation
 - Influence of site-specific soil properties over looked in "off the shelf" remedies
- Recent studies have identified that biostimulation can work:
 - Understanding phosphorous bioavailability (Siciliano et al., 2016; Hamilton et al., 2018; Bulmer et al., 2018)
 - Sorption, complexation, precipitation
 - Addition of low molecular weight organic acids to chelate Ca and Mg ions
 increase availability of phosphorus (Siciliano et al., 2016; Chen, 2018)
- However, remediation at some sites has stalled
 - Believed to be related to buffering capacity and carbonate mineralogy



A Note on Soil Buffering Capacity

- Ability of a soil system to resist changes in pH
 - Related to CEC, base saturation, and acid neutralizing capability.
- Stalling is likely associated with a failure of the soils to buffer against the biostimulatory solutions
 - Results in formation of Ca-P complexes that are stable compounds with relatively low solubilities



Project Objective

- Understand role of site-specific properties in the effectiveness of biostimulation techniques for the remediation of petroleum impacts in cold-region calcareous soils
 - How geochemical parameters 'condition' biostimulatory solutions
- END GOAL: create a conceptual model that integrate site-specific geochemical parameters and microbial activity to estimate hydrocarbon degradation rates.



Experimental Design















Experimental Design

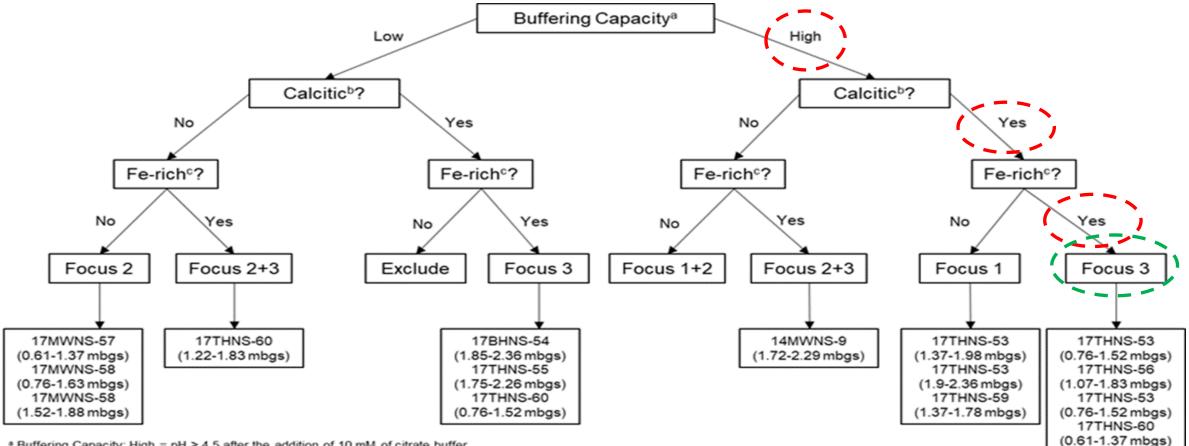
Biostimulatory Solutions

- Sodium triphosphate
- Nitric acid
- Magnesium sulphate / potassium sulphate
- Ammonium iron citrate





Experimental Design



Buffering Capacity: High = pH > 4.5 after the addition of 10 mM of citrate buffer

b Calcitic Soil: Mg:Ca Ratio < 0.12</p>

^c Fe-rich soil: Fe_{solution} > 30 mg/L

Notes:

17THNS-53 (1.9-2.36 mbgs) was not used for the final microcosm test because it did not follow the same flow as the other cores (unusual place on decision tree)

> = greater than

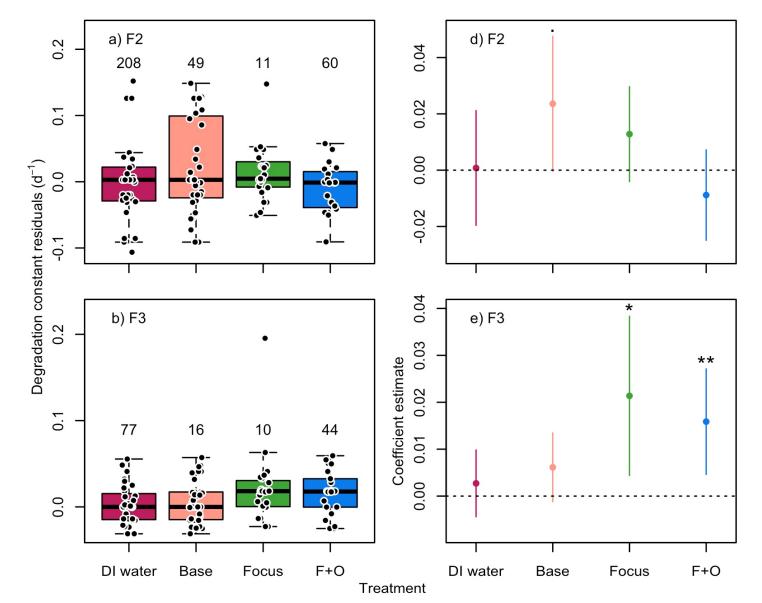
< = less than

Ca = calcium

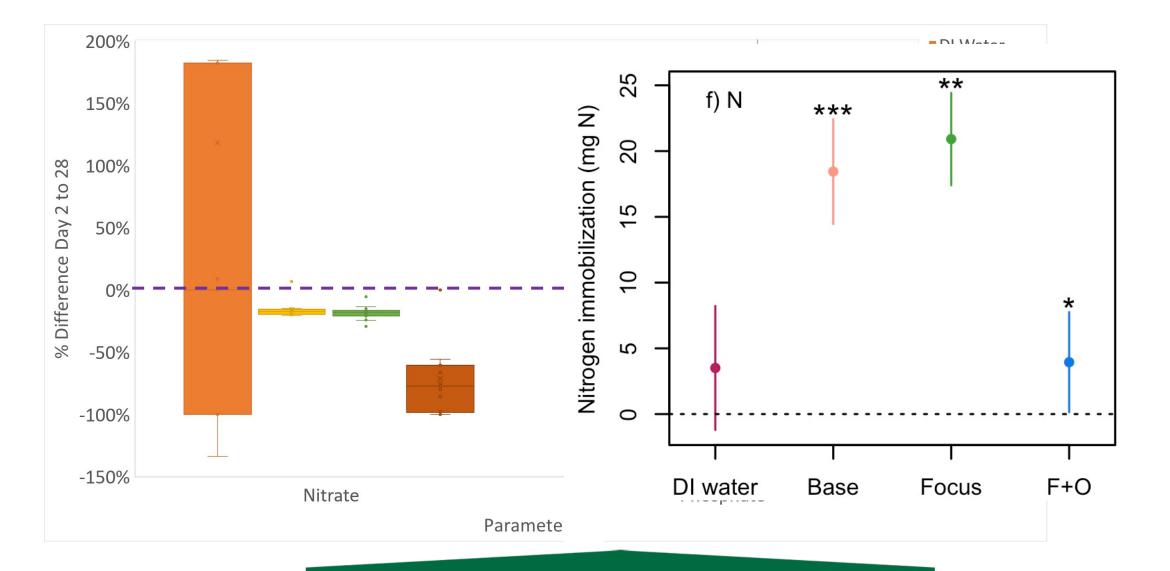
Mg = magnesium

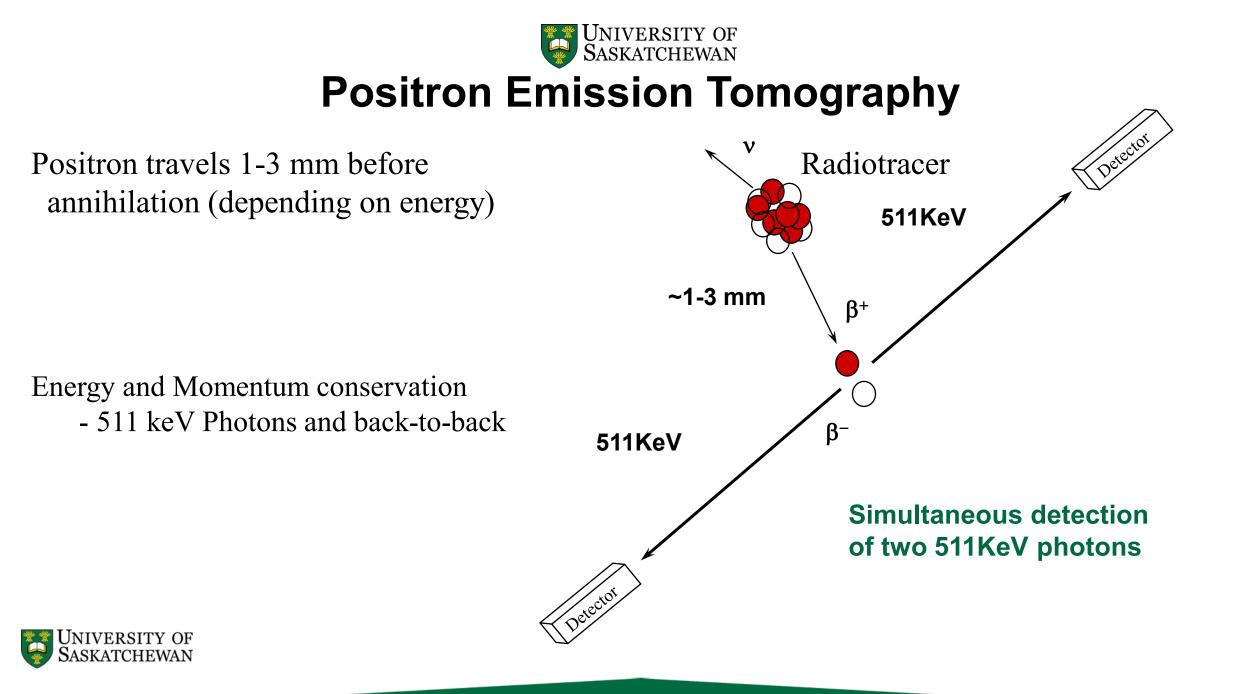


Results







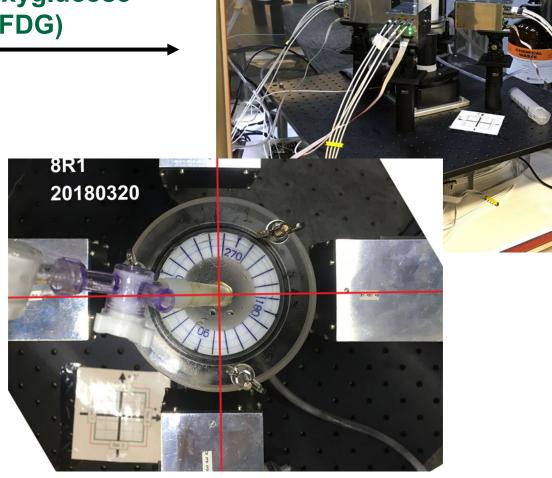






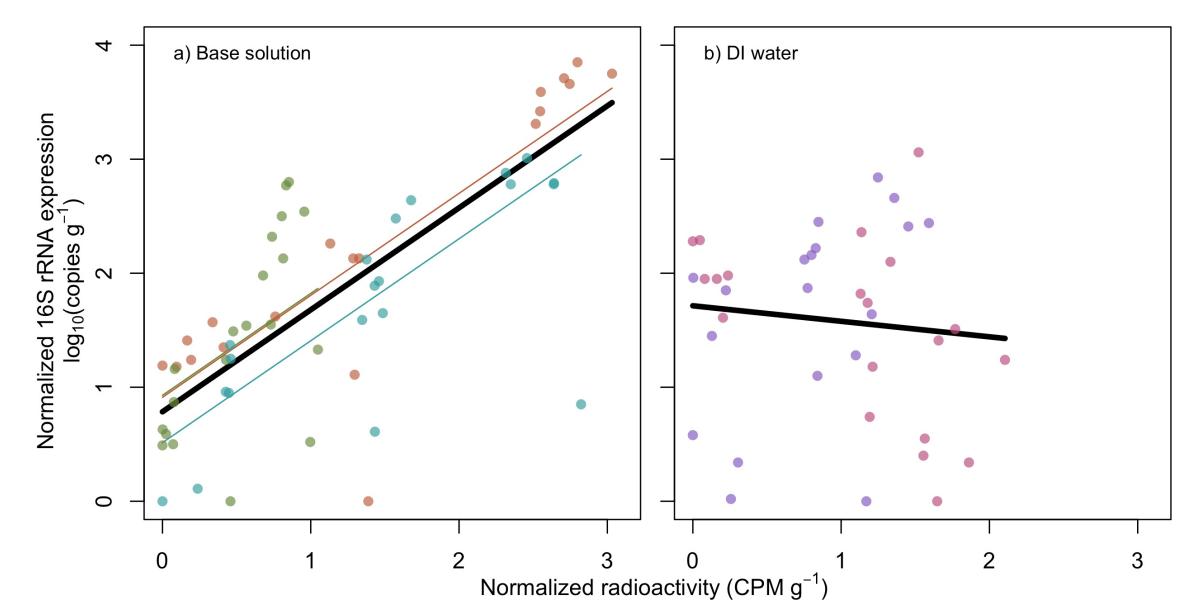
You measure what stays in the system!

¹⁸F-fludeoxyglucose (¹⁸F-FDG)





Results





Conclusions and Next Steps

- General trends were evident
 - Interference/Dilution effects due to microcosm design
- Further development of the focus solutions is warranted
- Evaluate sorption kinetics in these systems

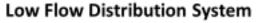


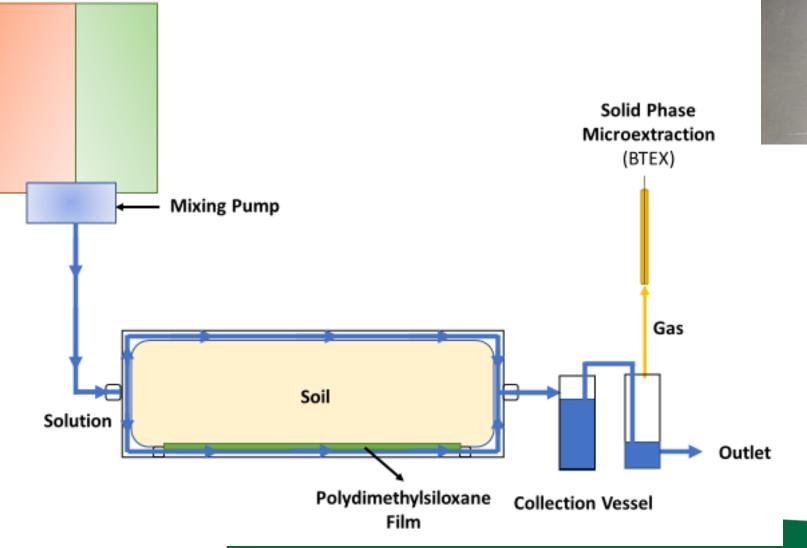
- Comparing to sorption kinetics
 experimental set-ups
- Assess sorption of PO₄



Next Steps









Summary

- Assess biostimulation holistically to develop conceptual model to estimate degradation rates
 - a) Site specific geochemistry
 - b) Microbial activity
- Goal: create a model that can predict an effective solution for all sites
 - Understand which geochemical parameters influence or condition biostimulatory solutions



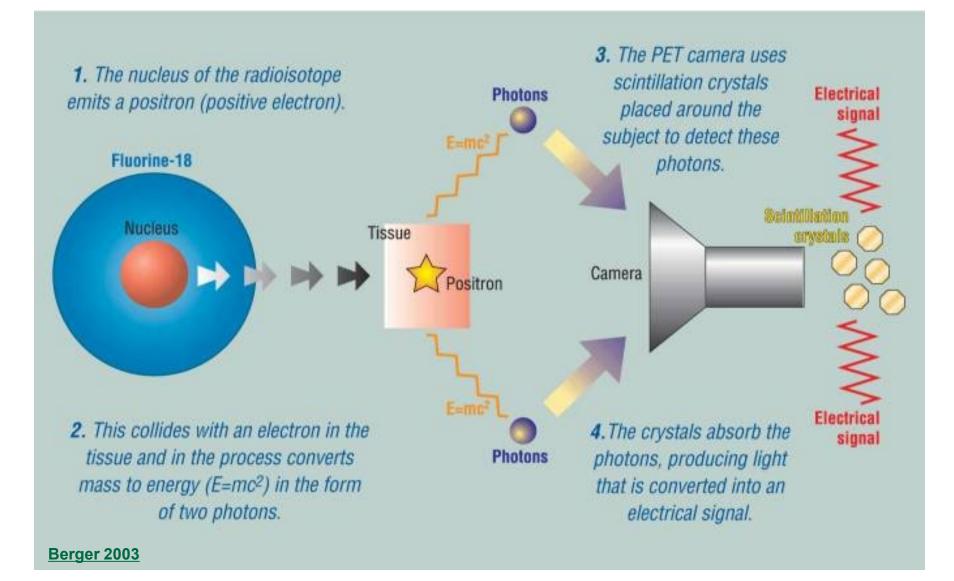
Thank you!

Acknowledgements:

Analysis: Drs. Steven Siciliano and Steven Mamet

University of Saskatchewan's Environmental Soil Toxicology Laboratory Government of Yukon's Site Assessment and Remediation Unit Jacobs Engineering

How do we trace PET isotopes?





Assessing Microbial Activity

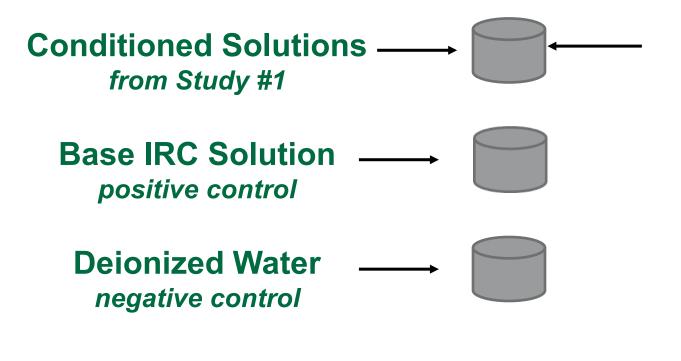
Increasing Benzene Biodegradation via Delivery of a Conditioned Biostimulatory Solution

 Objective: estimate the influence of 'conditioned' biostimulatory solutions on benzene degradation.



Assessing Microbial Activity

 Objective: estimate the influence of 'conditioned' biostimulatory solutions on benzene degradation.



20 mL Serum Bottle

- 1 mL benzene (0.4 1 ppm)
- 2 mL Ulrich culture
- 9 mL solution
- Minimum three replicates per solution
- 30 day trial
 - Benzene concentrations
 - Gene expression













Lessons Learned

