



# PFAS Impacts on Solid Waste Landfills

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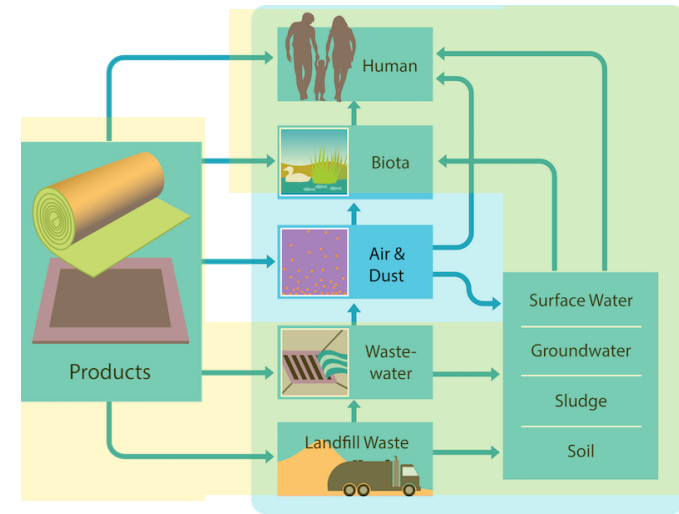


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RemTech, Banff, AB – October 10-12, 2018

# Outline

1. Background
2. PFAS concentrations in landfill leachate
3. Oxidizable precursors
4. Fate of PFAS in biological treatment systems
5. Potential treatment technologies



Source: [www.buildinggreen.com](http://www.buildinggreen.com)

# PFAS Uses & Products

Heat, oil, stain, and grease resistant coatings

Clothing

Furniture

Carpet stain protection

Food packaging

Paper coating

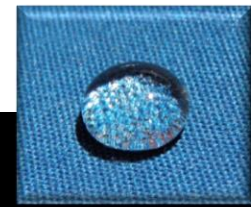
Non-stick cooking surfaces

Electrical wire insulation

Chromium plating mist suppressants

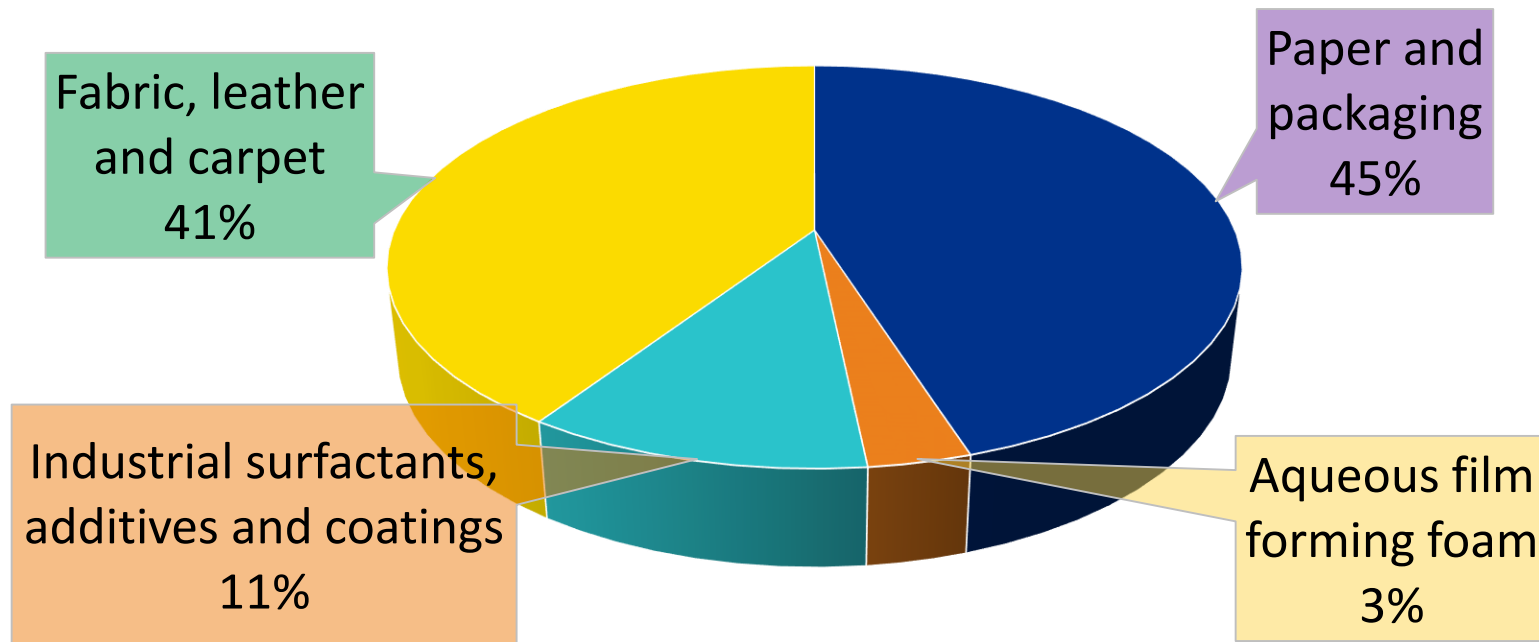
Photolithographic chemicals

Many other uses



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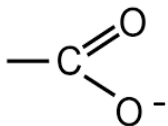
# Pre-2003 North American PFAS Production



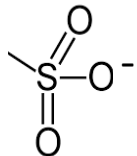
(Schultz, et al, 2003)

# PFAS

- Naming convention by
  - Number of carbons
  - Functional group
- Perfluoro Carboxylic Acids (PFCA)



- Perfluoro Sulfonic Acids (PFSA)



C4 **B**utane  
C5 **P**entane  
C6 **H**exane  
C7 **H**eptane  
C8 **O**ctane  
C9 **N**onane  
C10 **D**ecane  
C11 **U**nodecane  
C12 **D**odecane  
C13 **T**ridecane  
C14 **T**etradecane

## Perfluorooctanoic acid (PFOA)



## Perfluorooctane sulfonate (PFOS)



## Perfluorobutane sulfonate (PFBS)

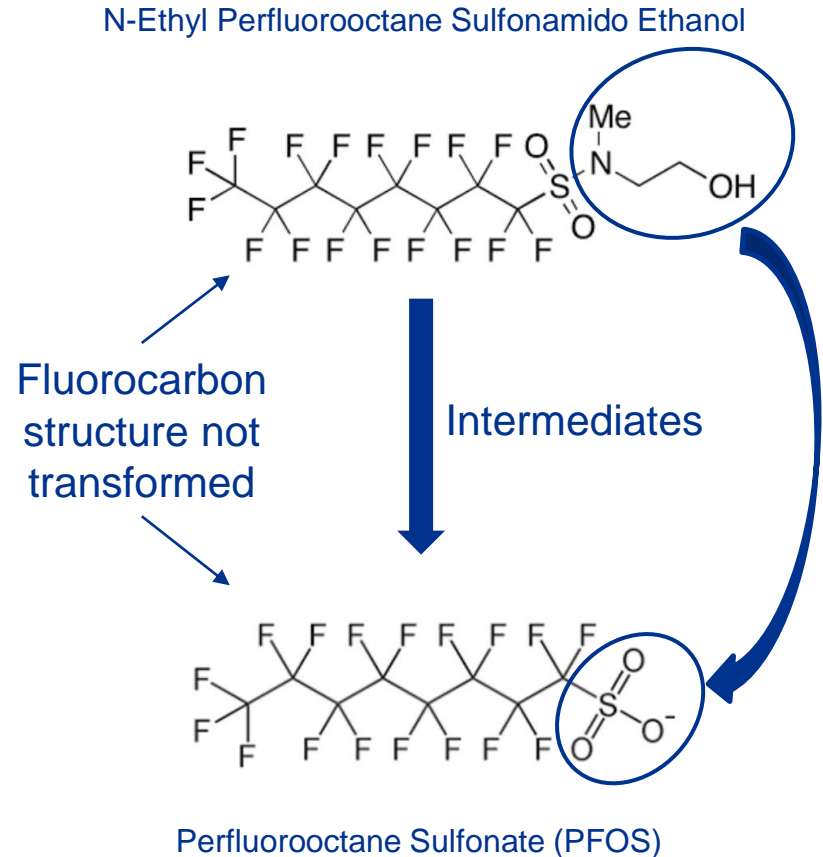


# Fluorotelomers

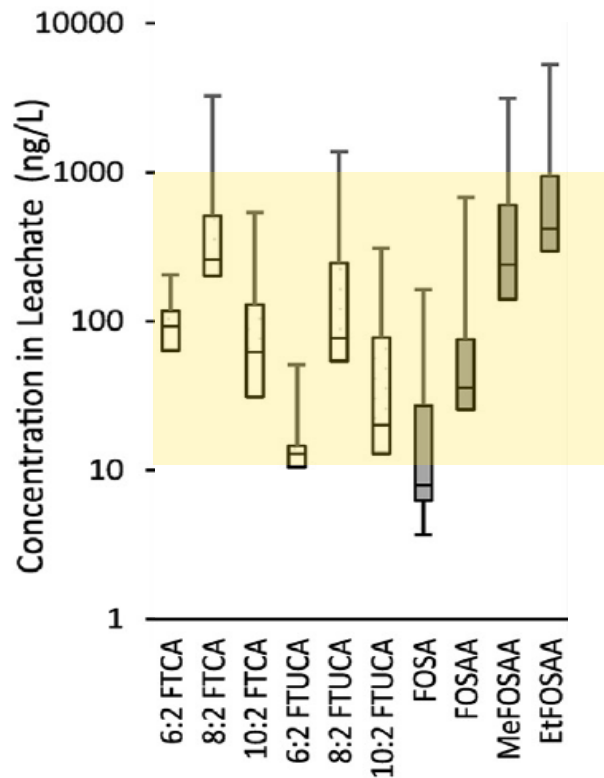
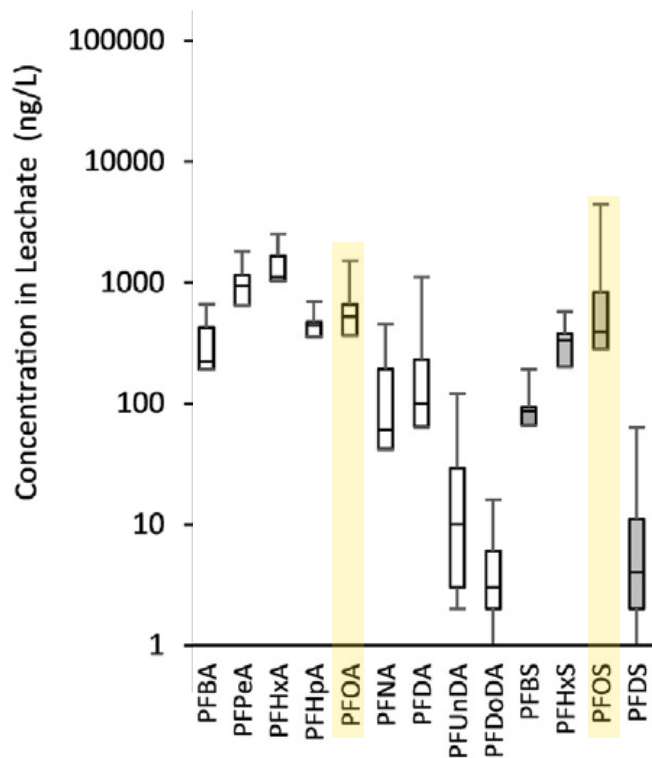
- “Telomer” small polymer (usually 2 to 5 monomer units)
- Fluorotelomer acrylate (FTA) polymers are used to coat a wide range of textiles, paper, and carpet
  - water, oil, and stain repellency
- FTA products expected to end up in landfills leading to long-term presence in the environment of a potential source of PFOA

# Precursors

- 1000's of PFAS compounds created – very few quantified analytically
- Precursors may be transformed through biological treatment
- Generates stable end-products of PFOA or PFOS
- Treatment of PFOA & PFOS not feasible with current observed biology



# PFAS Concentrations – Canadian Leachate Study

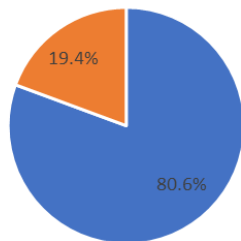


Source: Benskin et. al., 2012 ES&T, as presented in Humid et. al., 2018 Environmental Pollution.

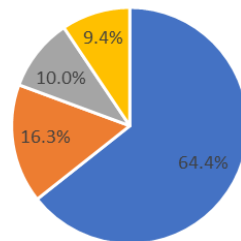


# PFAS Distribution - Landfill Leachate

United States

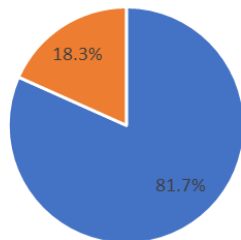


■ % PFCA ■ % PFSA

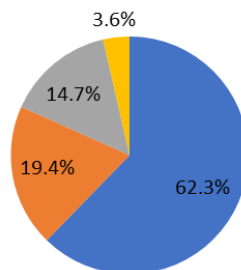


■ % SC CA ■ % LC CA ■ % SC SA ■ % LC SA

Canada



■ % PFCA ■ % PFSA

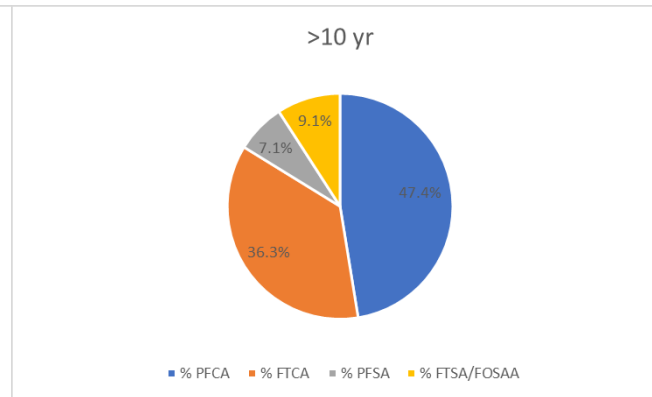
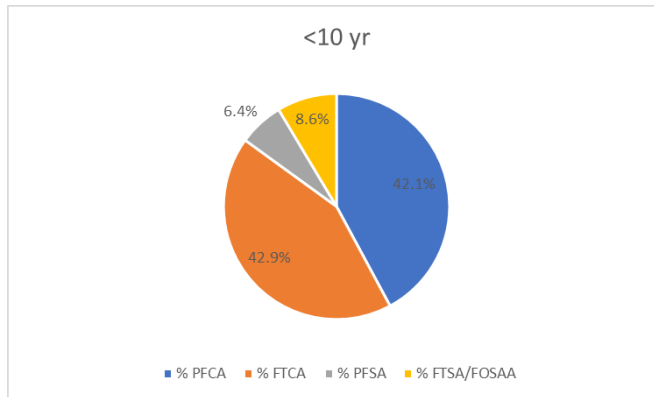


■ % SC CA ■ % LC CA ■ % SC SA ■ % LC SA

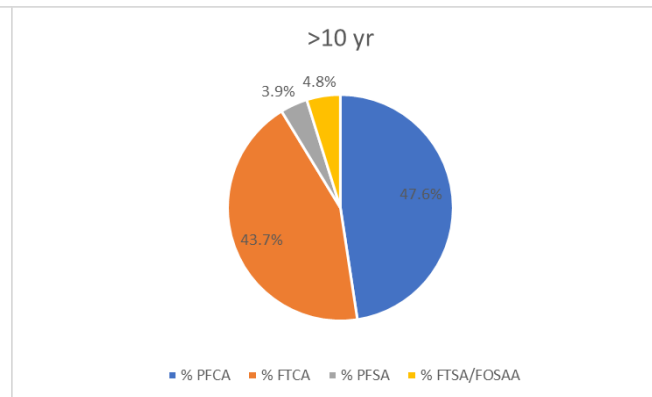
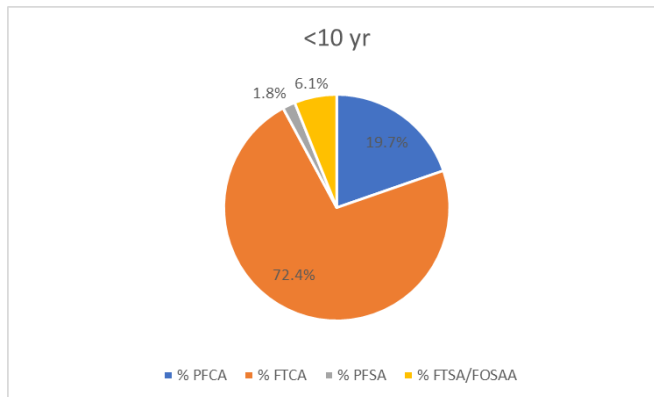
Source: Huset et. al., 2011 Chemosphere; Benskin et. al., 2012 ES&T

# PFAS Distribution - Landfill Leachate

Wet Climate



Temperate  
Climate



Source: Lang et. al., 2017 ES&T

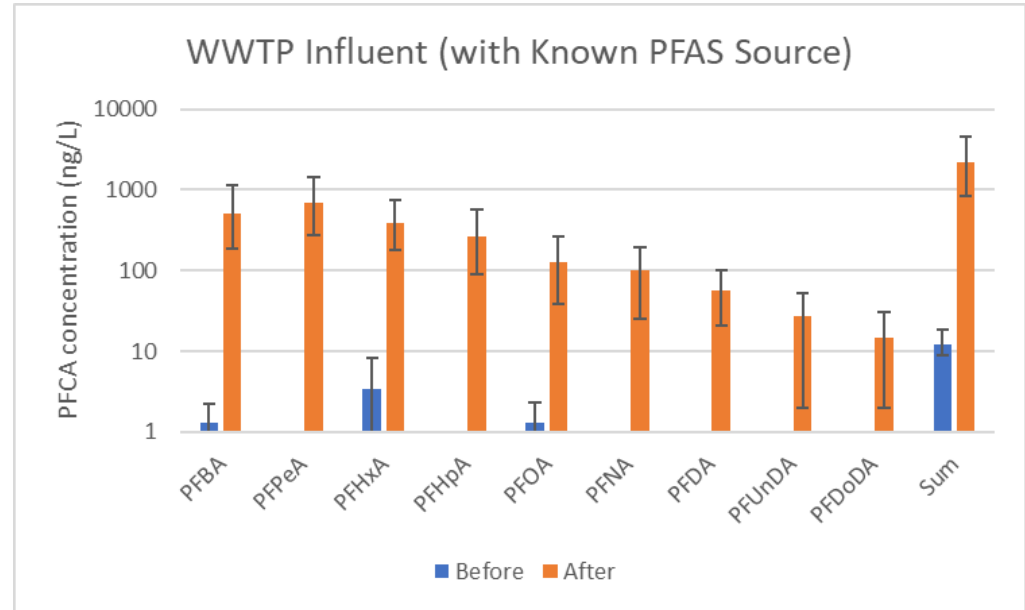
# Total Oxidizable Precursor (TOP) Assay



- Undefined potential precursors to PFOA/PFOS
- Best offered technique to towards “total PFAS”
- Allows for analysis utilizing traditional PFAS methodologies

# TOP Analysis Results

- Leachate TOP analysis conducted
  - 100,000 ng/L of detectable PFOA
  - 1,300,000 ng/L of potential precursor material from TOP



# PFAS Distribution & Fate – Municipal WWTP

Location	PFHxA	PFHpA	PFOA	PFHxS	PFOS	Total
Influent	59	13	206	24	134	444
Effluent	60	13	200	28	240	560
Influent	9.7	2.2	3.1	6.6	12	35
Effluent	31	3.7	14	48	22	120

- Precursors discharged to WWTP cause PFAS increase across aeration
- Measured PFAS pass through WWTP with limited/no reduction
- PFAS also leave plant through biosolids

# PFAS in Biosolids

- Annual biosolids generation
  - 660,000 tons in Canada (land application varies by province)
- PFAS concentrations in biosolids have remain relatively constant <sup>1</sup>
- Biosolids increased PFAS in soil and bioaccumulate in earthworms <sup>2</sup>
  - PFAS concentrations 9.88 to 99.9 ng/g (d.w.) compared to control 1.76 ng/g (d.w.)
- Limited data on crop uptake groundwater
  - Some data show translation to crops <sup>3</sup>
- Rainfall can cause leaching to groundwater
  - One study showed GW concentrations < Health Canada Screening Values <sup>4</sup>

# Adsorption

- Granular Activated Carbon (GAC) well demonstrated
  - Bituminous-based GAC: increasing number of full-scale installations
  - Regenerated agglomerated shows better performance
- Ion Exchange Resin is becoming more demonstrated
  - IX generally outperforms GAC -- becomes cost-benefit exercise
  - Not effectively regenerable (ECT2 is focused on on-site regen)
- Generally more effective for
  - Long chain versus shorter chain
  - Sulfonates versus carboxylic acids of same length

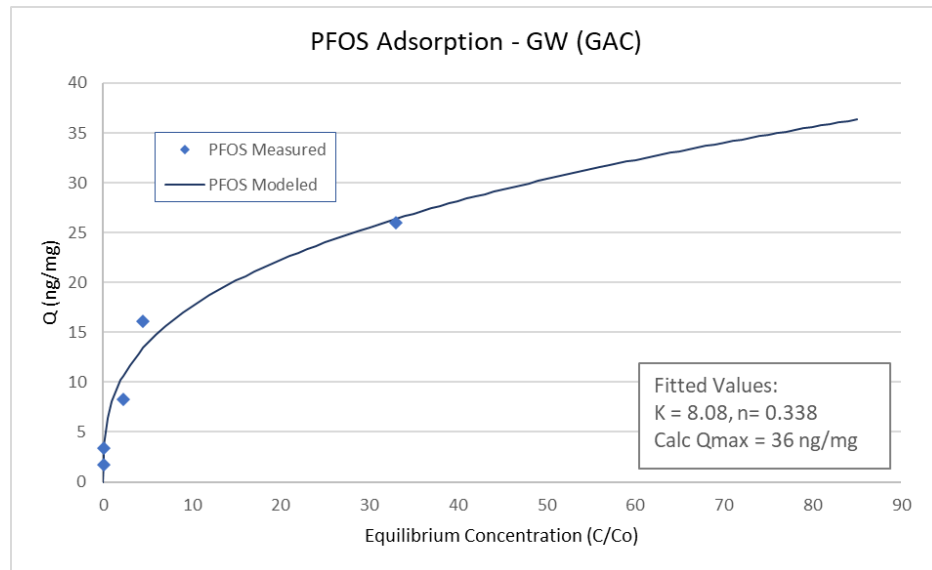
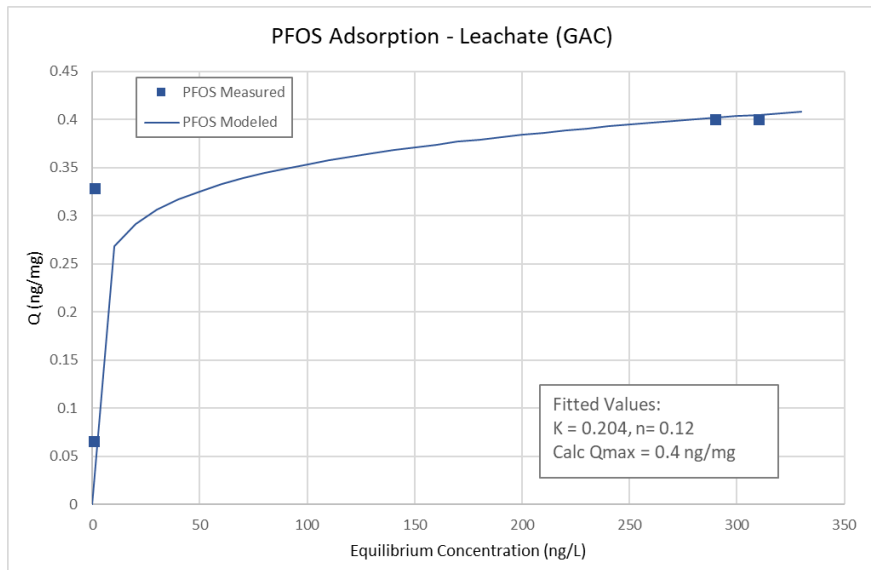


# Leachate-specific considerations for adsorption

- Many competing constituents
  - Slows transport kinetics (speed that constituents adsorb)
  - Limits adsorption capacity (how much PFAS can be adsorbed)
- GAC
  - Background organics
- IX
  - Background organics
  - Anions (chlorides, sulfates)
- Batch Application
  - Standalone
  - PACT



# Adsorption – Leachate vs DW



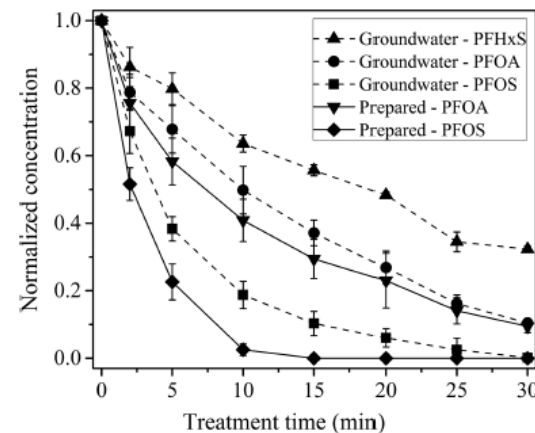
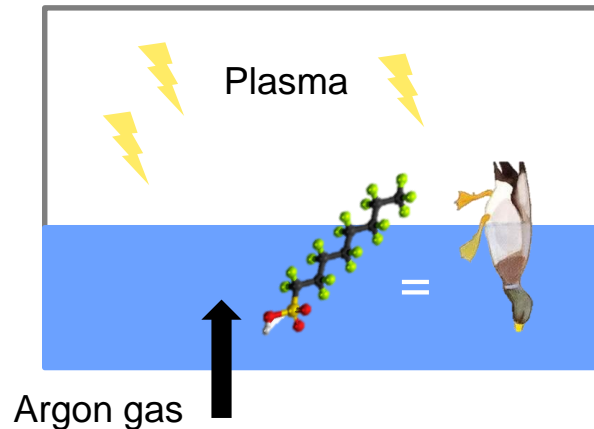
# Reverse Osmosis

- Effective for removal
  - Both long and short-chain PFAS
- Concerns
  - Managing liquid concentrates
  - Application in leachate
    - Usually require pretreatment (i.e., MBR)
- Areas of development
  - Effectiveness of precursor removal
  - Lower pressure NF applications



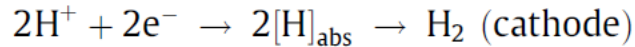
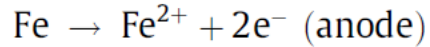
# Advanced Reduction Processes

- Non-Thermal Plasma
  - Free electron
  - In water unstable, short lifetime
  - Research shows liquid penetration of ~2.5 nm
- “Solvated electron” ( $e^-_{aq}$ )
  - Free electron
  - In water unstable, short lifetime
  - Research shows liquid penetration of ~2.5 nm
- Highly reductive species
  - Reduction is gain of electrons
    - Net increase in C-H bonds
    - Net decrease in C-F bonds

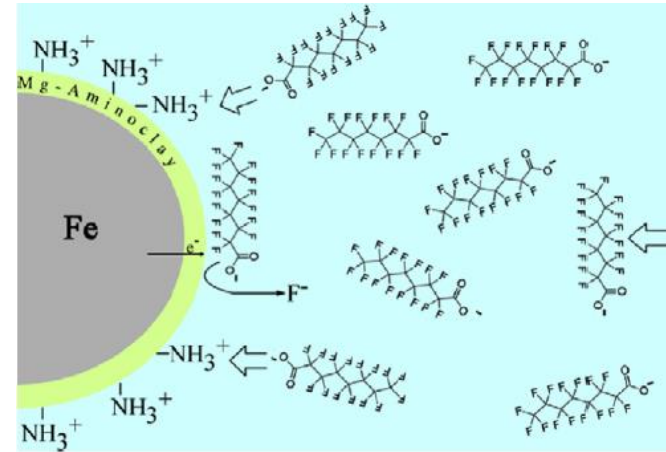


# Zero Valent Iron

- Another means of chemical reduction – generation of electrons

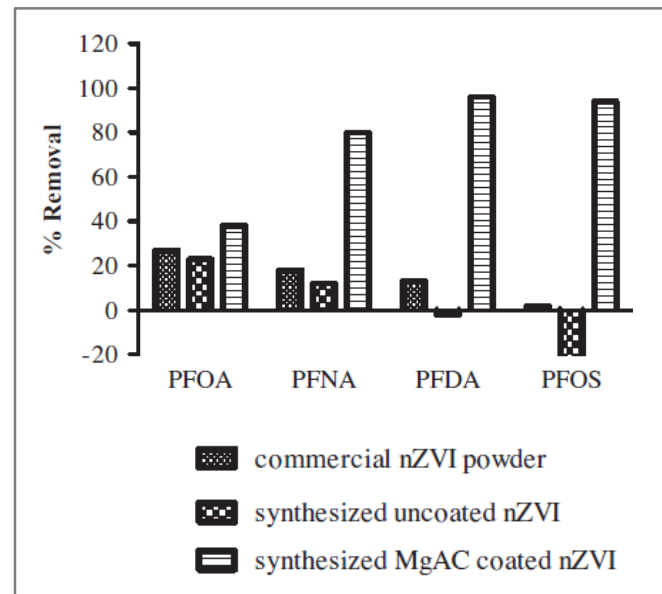


- Cathode indicates low pH is preferred
- PFAS attracted to surface-liquid boundaries
- pKa of PFAS  $\ll 7$  (anion in solution)
  - Positive surface charge



# Zero Valent Iron

- Nano or colloidal ZVI much more reactive (higher surface area)
- Surface modifications
  - Improve adsorption of PFAS to surface
  - Limits passivation of reactive iron surface
    - Amino-clay
    - Sulfidation
    - Noble metals
    - Manganese



# Thank you for your time

## Questions?

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