

Schulich School of Engineering

# Biodegradation of Sulfolane in Soil Using Aerobic Biopile Technology



UNIVERSITY OF  
**CALGARY**

Successful Collaboration Between Academia and Industry  
– Breaking Open Sulfolane Remediation strategies

12-Oct-2017

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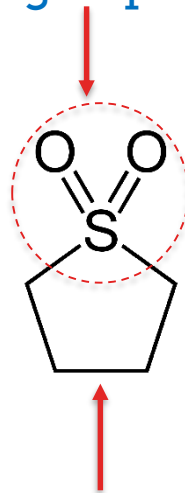
Ian Keir, Bonavista Energy Corporation

- **Sulfolane Key Properties**
- **Environmental Standards**
- **Previous Sulfolane Presentations**
- **Development of Sulfolane Treatment Technology**
- **Aerobic Biodegradation of Sulfolane in Soil**
- **Lab Degradation of Sulfolane**
- **Pilot Demonstration-Biopiles**
- **Full Remediation on Sulfolane Contaminated soil**



- Gas Sweetening
- Aromatics Extraction
- Textiles
- Production 18,000-36,000 tones<sup>[1]</sup>
- Soil and Groundwater contamination
- Ongoing Toxicity Studies (NTP)

Sulfonyl  
group

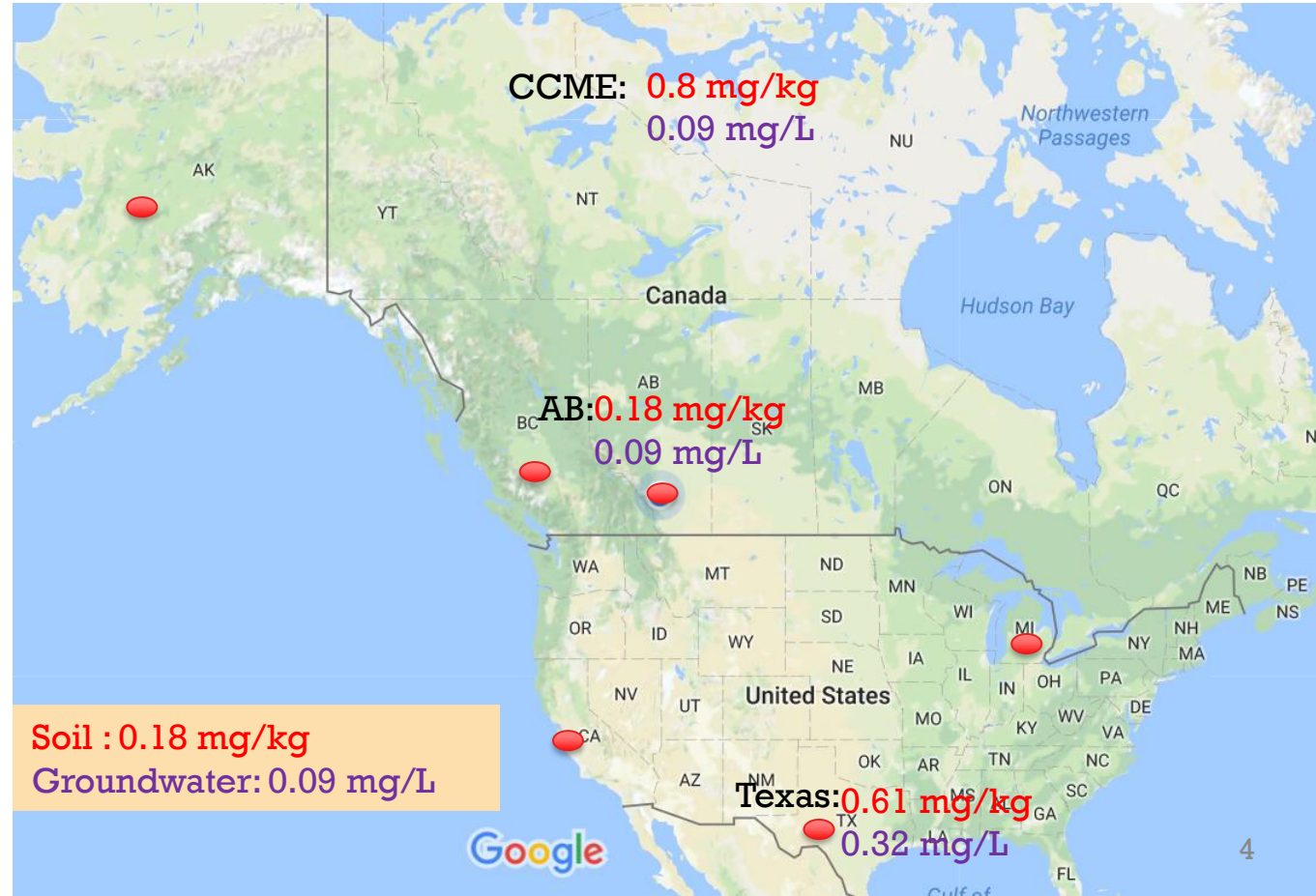


Cyclic  
structure

- ❑ Chemically stable
- ❑ Thermally stable
- ❑ Boiling Point: 287.3 °C <sup>[2]</sup>
- ❑ Vapor Pressure@20°C : 1.33 pa<sup>[3]</sup>
- ❑ Water Solubility@20°C: 1266 g/L<sup>[3]</sup>
- ❑ Soil Adsorption:
  - $K_{oc}=0.07$ <sup>[3]</sup>
  - $K_d(\text{montmorillonite})=0.94\text{L/kg}$ <sup>[4]</sup>
  - $K_d(\text{kaolinite})=0.08\text{L/kg}$  <sup>[4]</sup>

- Alberta
- BC
- Alaska
- Texas
- Louisiana
- California
- Michigan

Health Canada  
**interim** drinking  
water guideline:  
**0.04mg/L.**



# Previous Sulfolane Presentations

- EBA 2005 – Lab Scale
  - Soil: Bio-treatability
  - Groundwater: Bio-treatability; Chemical Oxidation
- Biogenie 2006 – Full Scale
  - Soil: Bio-treatability
- WorleyParsons Komex 2008 – Pilot and Full Scale
  - Groundwater: Bio-treatability
- Waterline 2016 – Pilot Scale
  - Soil: Bio-treatability; Chemical Oxidation
- Trium 2016 – Lab Scale
  - Groundwater: Chemical Oxidation
- Maxxam 2017 – Lab Scale
  - Laboratory Methods
- WorleyParsons 2017
  - Groundwater Remedial Options Review

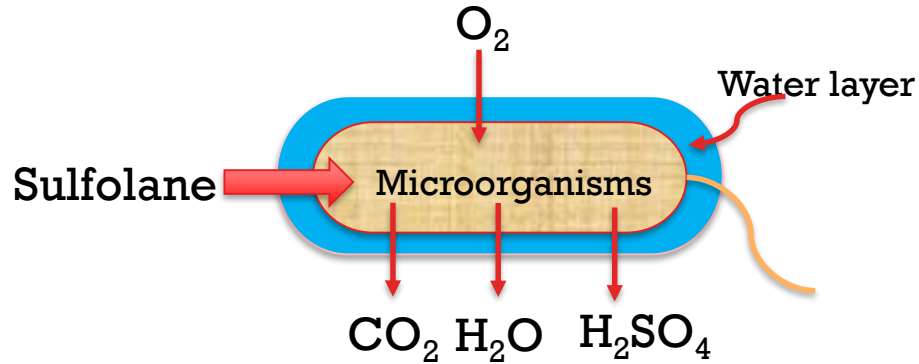


SCIENCE AND TECHNOLOGY  
**TALKS 2017**

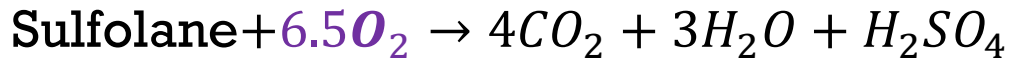


Groundwater	Soil
<ul style="list-style-type: none"><li>▪ <b>Advanced Oxidation</b> – Lab and Field Pilot</li><li>▪ <b>Bioremediation</b> – Lab and Field Pilot</li><li>▪ <b>Carbon Adsorption</b> – Lab and Field Pilot</li><li>▪ Reverse Osmosis – Lab Scale</li><li>▪ Isotope Fractionation – Lab Scale</li><li>▪ Integrated Technology-Lab scale</li></ul>	<ul style="list-style-type: none"><li>▪ <b>Bioremediation</b> – Lab Scale, Field Pilot, Full Scale</li><li>▪ Soil Flushing &amp; Washing – Lab Scale and Field Pilot</li><li>▪ In-Situ Chemical Oxidation (ISCO) – Lab Scale</li><li>▪ Oxygen Releasing Compounds (ORC) – Lab Scale and Field Pilot</li></ul>

# Aerobic Biodegradation of Sulfolane in Soil



- N, P & micronutrients
- Proper pH
- Proper temperature



## Lab Investigation:

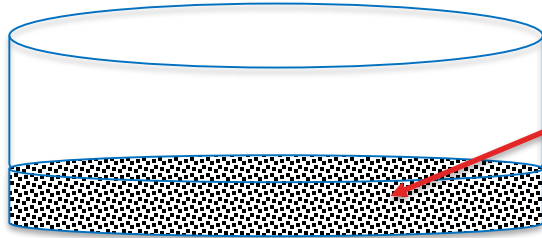
- Treatability study
- Optimization

## Pilot Demonstration:

- Evaluation
- Modification

## Full Remediation:

- Modification



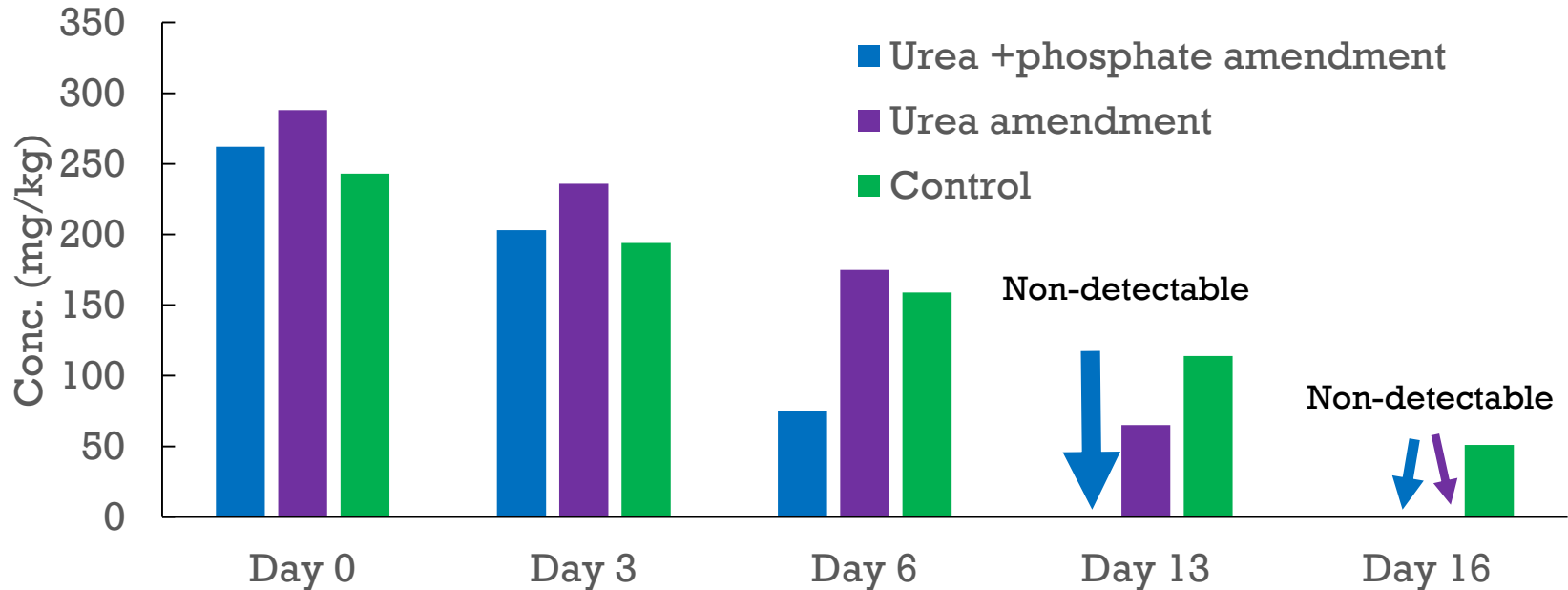
300 g of soil was  
loosely packed in a  
beaker ( $\phi = 15$  cm)

- Moisture : 18%
- Oxygen: exposed to atmosphere
- Temperature: 22 °C
- Nutrients : different conditions

<b>Soil Texture</b>	
Physical properties	Value
Sand percentage	8.3
Silt percentage	43.0
Clay percentage	48.6
Texture	<b>Silty Clay</b>

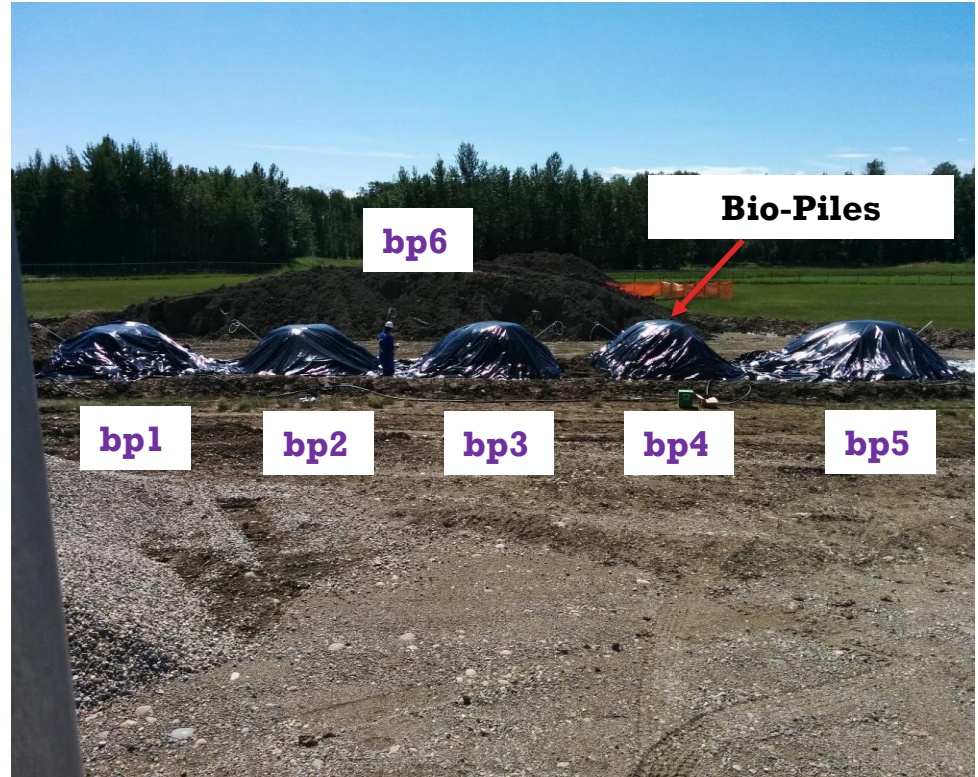
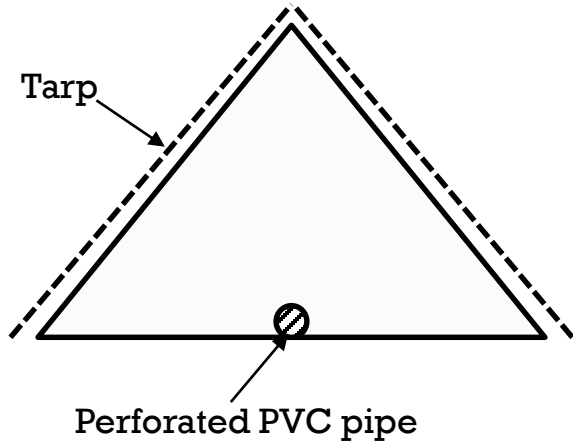


# Lab Study: Different Lab Conditions



- **Sulfolane metabolized microorganisms** were present in the contaminated soil.
- **N-P** amendment samples yield the best degradation results.

# Pilot Study: Setup of Soil Bio-Piles



Soil pile	Size of pile (m <sup>3</sup> )	Covered with Tarps	Nutrient Amendment			Aeration
			Nitrogen nutrient	Phosphate nutrient	Alfalfa green	
bp 1	25	Yes	Yes	NO	NO	Yes
bp 2	25	Yes	Yes*	Yes	NO	Yes
bp 3	25	Yes	Yes	Yes	NO	Yes
bp 4	25	Yes	NO	NO	Yes	Yes
bp 5	50	Yes	NO	NO	NO	Yes
bp 6	500	No	NO	NO	NO	NO

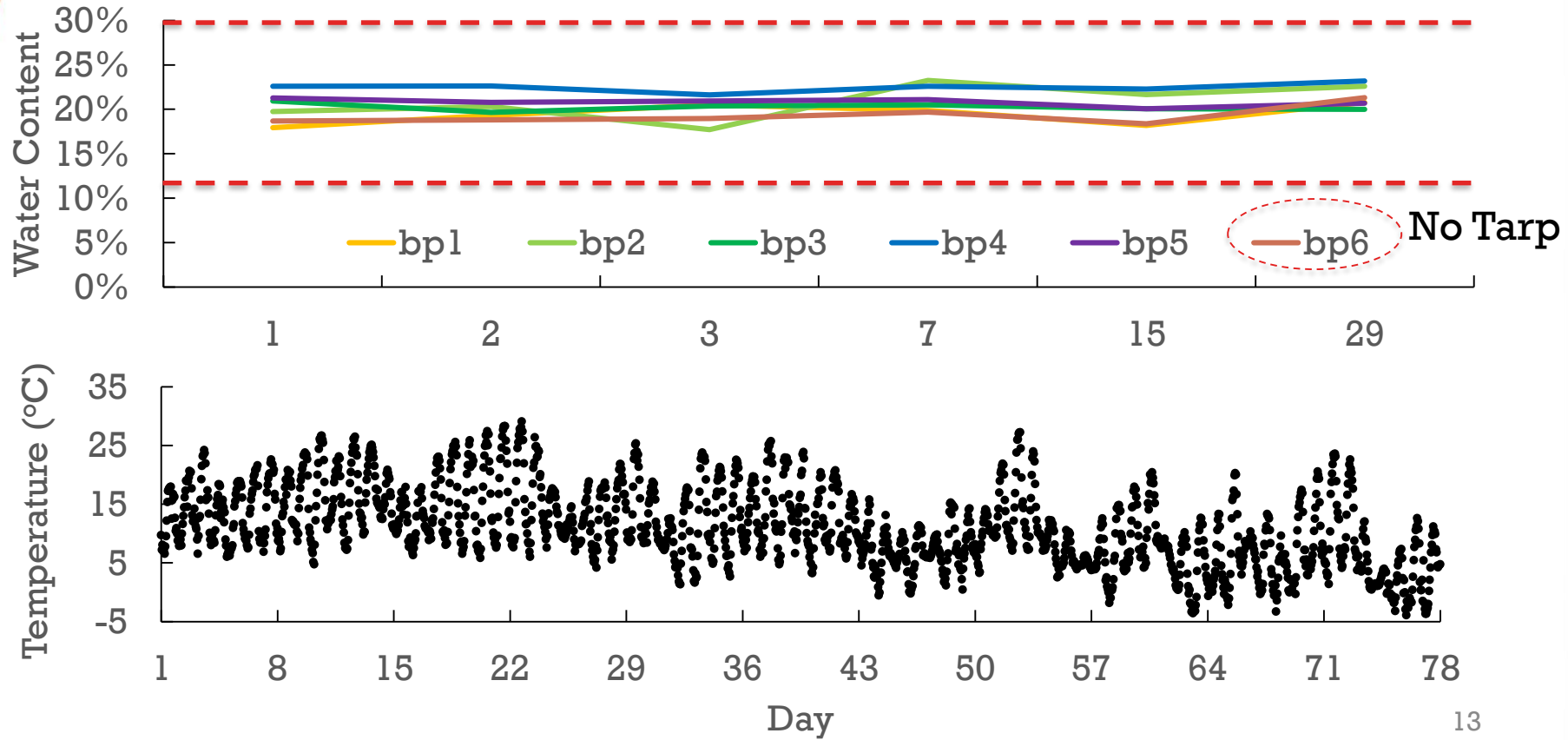
\* The amount of nitrogen added in bp 2 was only 1/10 of that in bp1

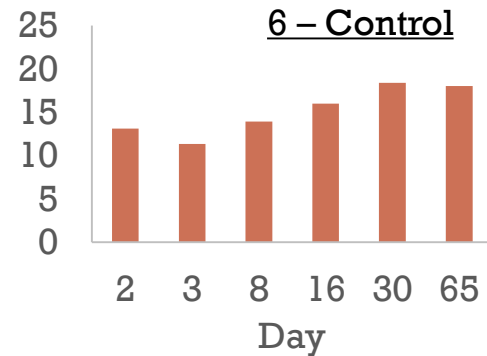
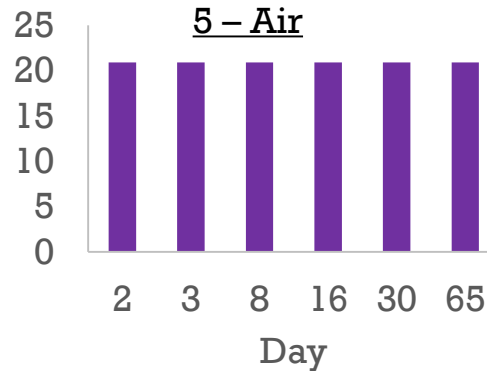
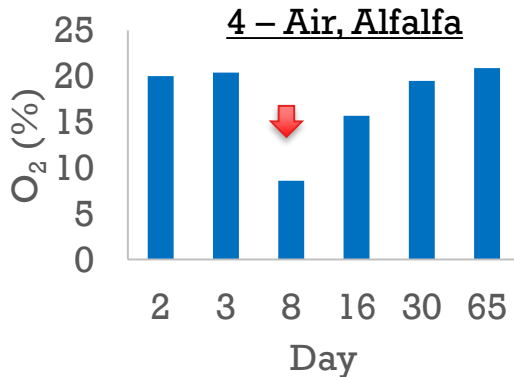
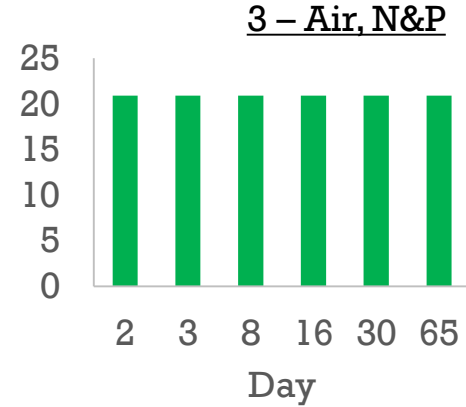
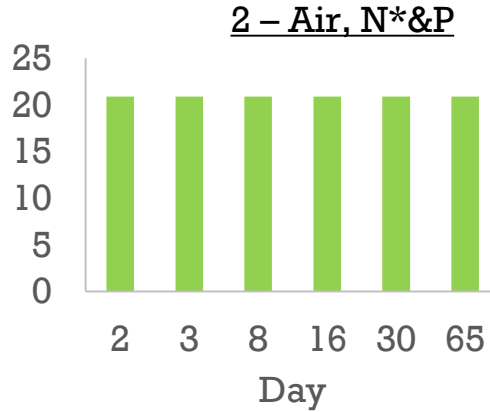
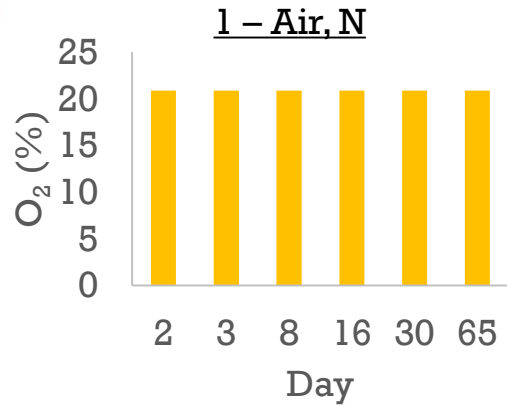


- Six random samples were collected from each soil pile
- $\text{CO}_2$ ,  $\text{O}_2$  and water moisture content were monitored.
- Temperature data was obtained from Alberta Climate Information service

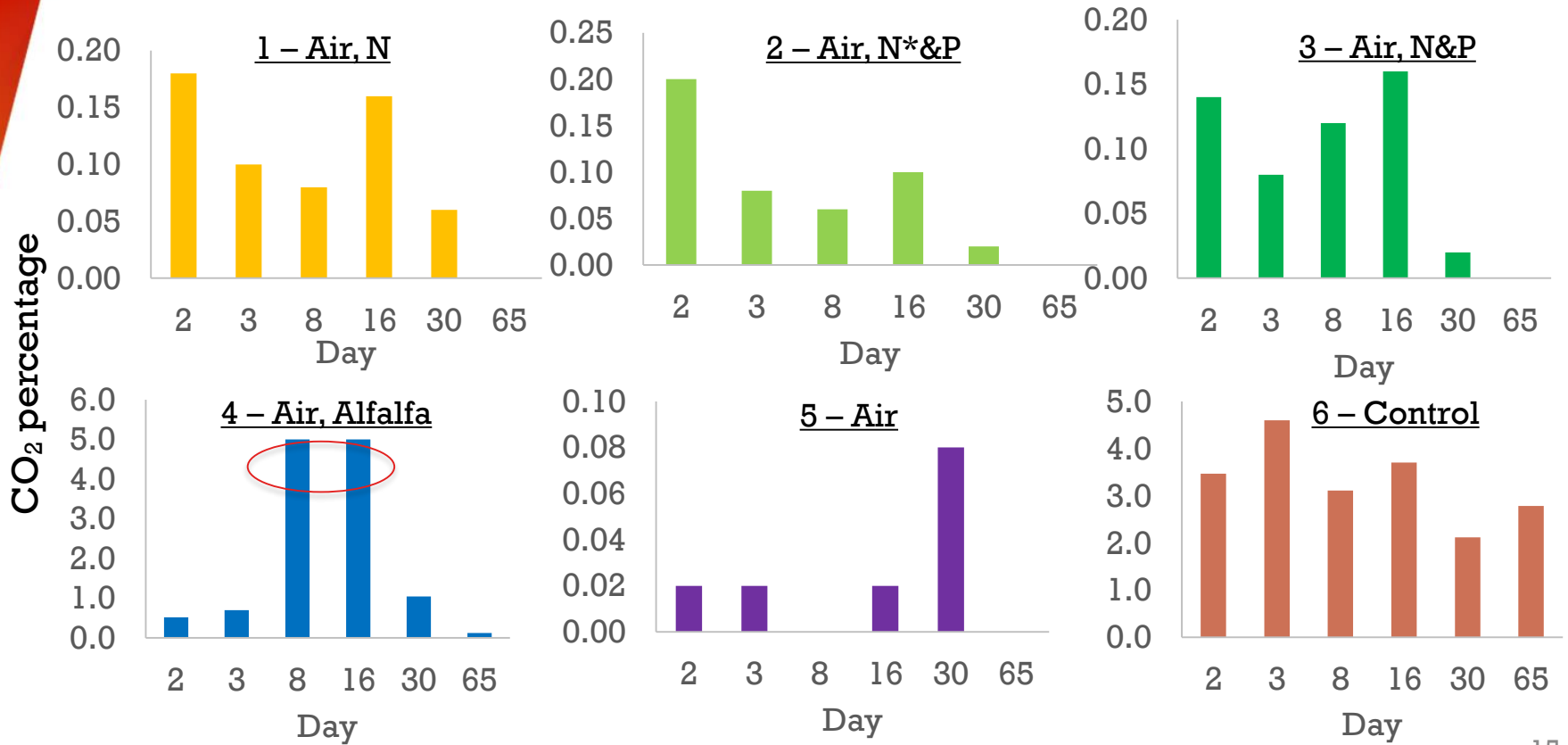


# Water Content & Temperature

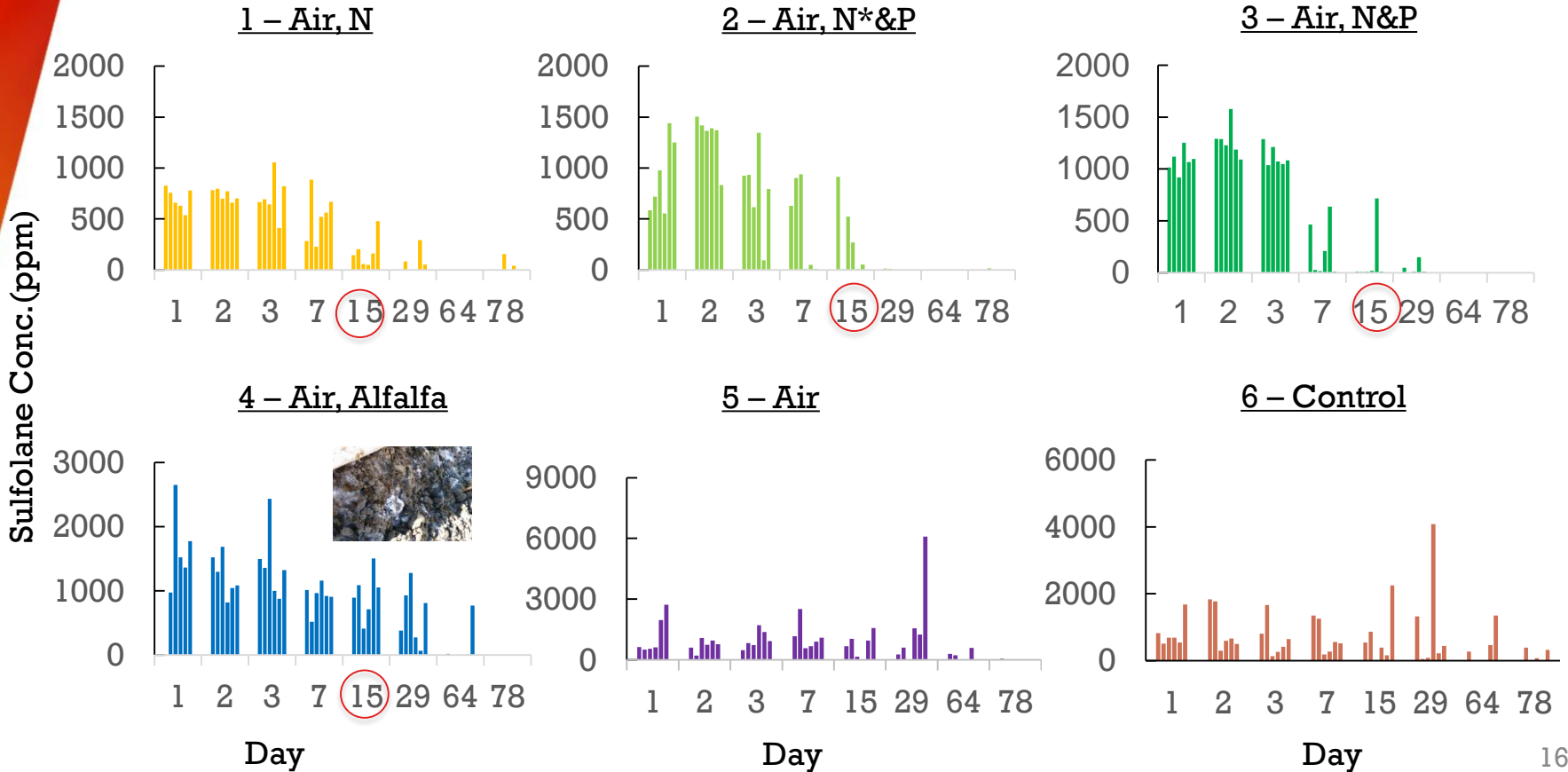








# Treatment Comparison



Soil Piles		BP1	BP2	BP3	BP4	BP5	BP6
		<u>Air, N</u>	<u>Air, N*&amp;P</u>	<u>Air, N&amp;P</u>	<u>Air, Alfalfa</u>	<u>Air</u>	<u>Control</u>
First Order Kinetics	K (Day <sup>-1</sup> )	0.09	0.09	<b>0.17</b>	0.03	NA	NA
	Half life (Day)	5.3	5.4	<b>3.0</b>	17.9	NA	NA
Zero Order Kinetics	Rate (mg/kg/Day)	24	26	<b>42</b>	17	NA	NA

The highest zero order degradation rate observed in lab was **220** mg/kg/day.

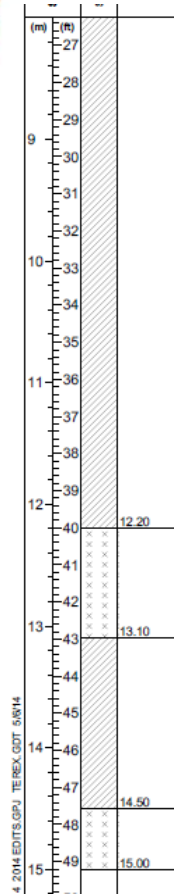


~26,000 m<sup>3</sup>

~8,000 m<sup>3</sup>

~20,000 m<sup>3</sup>





- Former Flare Pit
  - Sulfolane, DIPA, PHCs
  - Soil Texture
    - 23% Sand
    - 40% Silt
    - 37% Clay
  - Sulfolane [ 0.42 - 8170 mg/kg]
    - Average 364 mg/kg
  - Impacts 2 – 9 mbgs



- **Former Flare Pit**
  - Excavated June & July 2016





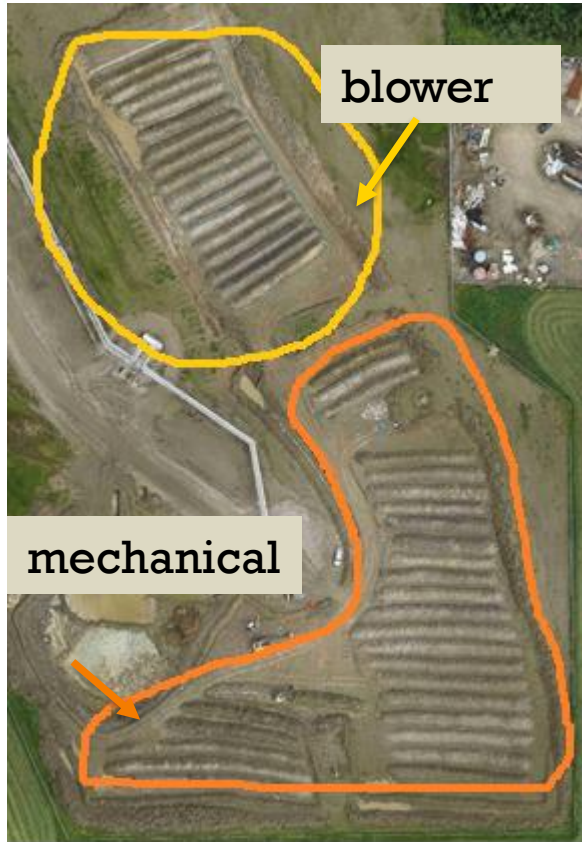


- ~8000 m<sup>3</sup> soil placed in windrows
  - volume assessed with drone
- Per m<sup>3</sup> of soil: 0.1 kg MAP and 0.29 kg urea
  - based on TOC and 100:5:1- C:N:P
- Oxygen
  - blower aeration
  - mechanical aeration





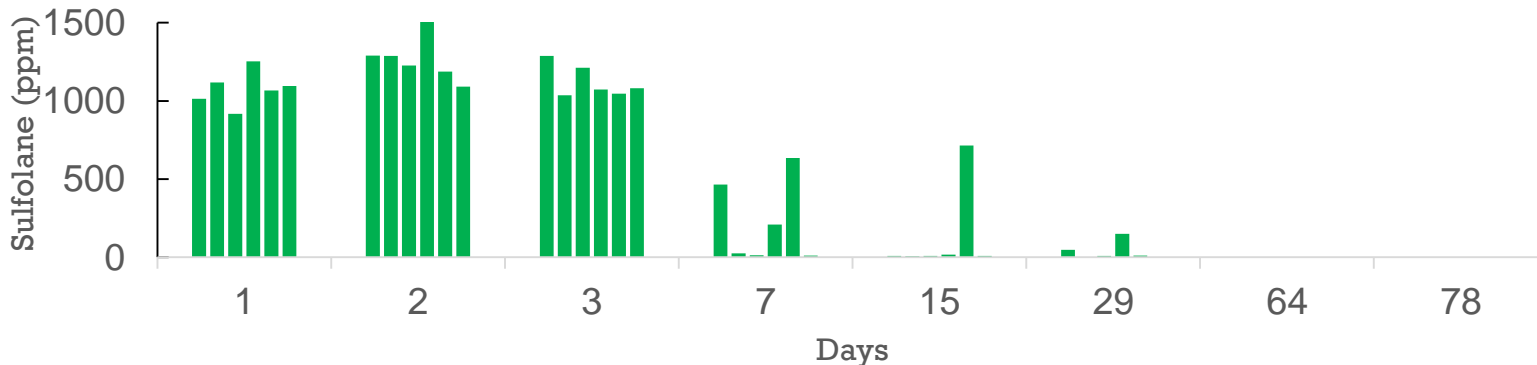




- Mechanical aeration
  - July and August
- Blower aeration
  - 24/7
- Based on half-life from pilot: ~35 days to clean soil with 346 mg/kg of sulfolane.
- 81 days between excavation and confirmatory samples
- 2 of 44 windrows exceeded sulfolane guideline



- Aerobic biodegradation of sulfolane was observed both in the **lab** and in the **field**.
- The addition of nutrients and forcing aeration enhanced sulfolane degradation (Pilot).
- Supplemented with both “N” and “P” nutrient resulted the best sulfolane degradation rate, the half-life is **3 days** (optimal pilot conditions)
- **Mechanical** and **forced aeration** were both successful in full scale



- [1] CCME. 2006. Canadian Environmental Quality Guidelines for Sulfolane:Water and Soil.
- [2] Kirk-Othmer. 1999. Encyclopedia of Chemical Technology. Fourth Edition, 1999. John Wiley & Sons.
- [3] Shell Chemicals Europe Limited. 1994. Sulfolane Data Sheet, Shell Chemicals UK Ltd.
- [4] Luther, S.M., Dudas, M.J. and Fedorak, P.M. 1998. Sorption of sulfolane and diisopropanolamine by soils, clays and aquifer materials. *Journal of contaminant hydrology*, 32(1), pp.159-176.



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- Collin Hennel, Bonavista Energy Corporation

**Thank you!**



**Questions?**