## Bioremediation of a Dissolved Ammonia and Nitrate Plume Through In-Situ Denitrification

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# Background

- Small unincorporated community in central Alberta, Canada
- Petroleum facilities including gas bar and bulk fuel facility in operation since the 1930's
- Multiple ESAs since 2001 identifying PHC impacts in soil and groundwater
  - On-site
  - Off-site
- USTs were removed in 2006 and replaced with ASTs



# Background



#### **Site Characteristics**

Lithology

 Soil stratigraphy generally consists of clay till underlain by saturated gravel/sand followed by bedrock

Groundwater

- GW depth approximately 1.5 mbg
- Flow is toward the north

Gradient

• GW gradient 0.0086 m/m



#### **Previous Remedial Efforts**

**Bio-Stimulation** 

- Addition of nitrogen-based liquid fertilizer into the nearsurface GW
- 500-L of 10-34-0 mixed with 500-L of 28-0-0 liquid fertilizer
  - Direct injected into nine monitoring wells and injection header





#### **Previous Remedial Efforts**

- As a result of limited carbon (low PHC impacts, coarsegrained geology), the added nutrients were not completely consumed
- GW nutrient concentrations (NO<sub>3</sub>, NO<sub>2</sub>, NH<sub>4</sub>) exceed applicable regulatory guidelines
- Elevated concentrations of orthophosphate also present

#### **Plume Characteristics**

• Nutrient impact area estimated to be 16,850 m<sup>2</sup>



- Thickness of the impacted aquifer is approximately 2.5 m
- Porosity approximately 35%
- Estimated 14,745 m<sup>3</sup> of nitrate impacted groundwater requiring treatment

#### **Remedial Approach**

- Average nitrate concentration of 27 mg/L
- Average ammonia concentration 10.63 mg/L
- Estimated mass:
  - Nitrate estimated to be 398 kg (880 lbs)
  - Ammonia estimated to be 157 kg (350 lbs)



#### **Remedial Approach**

- Site has limited carbon
- Elected to use ethanol



- Often used as a treatment for fertilizer impacts
- Environmentally inert
- Reasonable choice because of **low PHC concentrations**
- Ethanol is very labile and would have low toxicity

#### **Balanced Stoichiometric Equation**

 $NO_3^{-}+ 5/12 CH_3 CH_2 OH = 1/2 N_2 + 5/6 CO_2 + 3/4 H_2 O + OH^{-}$ 

#### **Remedial Approach**

- Based on literature review, 1.4 mg of ethanol is required to denitrify 1 mg of nitrate
  - =1,088 kg of ethanol required
- Consideration also given to additional carbon consumption
  - Presence of iron (Fe<sup>3+</sup>)
  - Oxygen
- Recommendation to double the mass
  - =2,175 kg

#### **Injection Summary**

- Ethanol added to the same wells which received fertilizer amendment
  - Approximately 20,600 litres of 15% ethanol injected
- GW monitoring and sampling programs completed pre- (-30 days) and post-amendment (48, 139, 244, 365 days)
  - Wells were developed by purging 100L per well





#### **Field Measured Oxidation Reduction Potential**

Amended Wells

Non-amended Wells



- ORP decreased significantly following amendment
  - Reducing environment, returning close to baseline 240 days post amendment, but remains negative +1yr
- Larger decrease noted in wells receiving ethanol amendment versus non-amended wells

#### Petroleum Hydrocarbons (Total BTEX)



- PHC concentrations remained low between pre- and post-amendment
- EtOH is a simple carbon source, preferential consumption
- TOC (bioavailable carbon) increased



Ammonia

- Ammonia showed initial nitrification after 48 days (~96%)
- ~64% increase from baseline after 365 days (potentially diffused back into the area once carbon source was consumed)
- If we take out one well, ammonia concentrations were static



Nitrate

#### Amended Wells

- Stable decreasing trend
  - ~90% reduction from baseline to final sampling • event



- Initial increase in nitrite concentrations a good indication of ammonia nitrification
- Decreasing concentrations of nitrite may indicate that the rate of nitrification is slowing

![](_page_16_Figure_1.jpeg)

- Dissolved iron concentrations increased ~620% at third monitoring event and remains ~260% after 1 year
- The significant decrease in ORP resulted in the mobilization of metals
  - Evaluation of which metals were mobilized is still ongoing

#### **Other Results**

- Orthophosphate concentrations were below method detection limits
  - Orthophosphate may act as a rate limiting step in nitrogen compound degradation
- Bacteria concentrations
  - No discernable pattern for bacterial communities (SRB, IR, HA and SF)

# Conclusions

- Nitrification of ammonia would decrease DO and produce anaerobic conditions
  - Site was near anaerobic to begin. DO concentrations
    <1 ppm both pre and post amendment</li>
  - Ethanol amendment forced redox negative environment
- Denitrifying bacteria would use nitrate as an electron acceptor during anaerobic respiration producing nitrogen gas
  - Nitrogen gas was not measured during investigation
  - Nitrogen gas likely increased based on reduction of nitrate concentrations throughout investigation area

## Conclusions

- Nitrite concentrations would also increase
  - Initial increase in nitrite concentrations was a good indication of ammonia degradation – a 37% increase in nitrite concentration was noted
- Reduction of ammonia and nitrate via anammox process
  - Ammonia initially reduced, but rebounded (potentially the result of diffusion)
  - Average nitrate concentration went from 31 ppm in the baseline event to 1.3 ppm after third post amendment event
  - Anammox processes likely not taking place at the site

#### **Questions?**

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