



ARSENIC SPECIATION, MOBILITY AND TREATABILITY AT A FORMER INDUSTRIAL SITE IN EDGEWATER, NEW JERSEY

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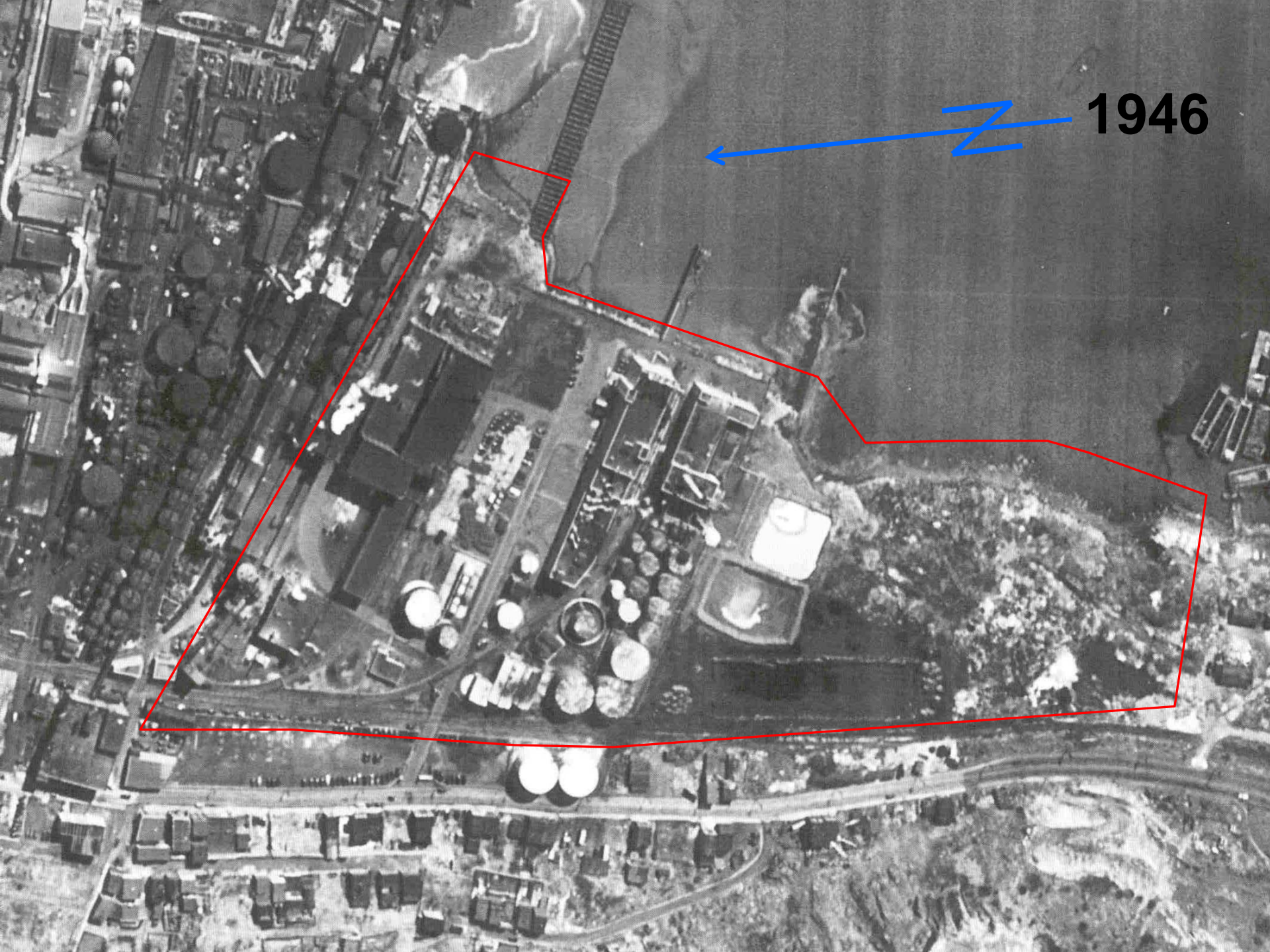




Project Overview

- 15-Acre Riverfront Property
- Former Industrial Usage: chemical mfg, roofing pitch storage, hydrogen gas plant, edible oil, soap and detergents
- Proposed Redevelopment as a Mixed Use Residential and Commercial Property
- Contaminated with Arsenic (Metals), Roofing Tar/Pitch Material, Benzene
- Northern Portion of Site Impacted by Adjacent Superfund Site

1946



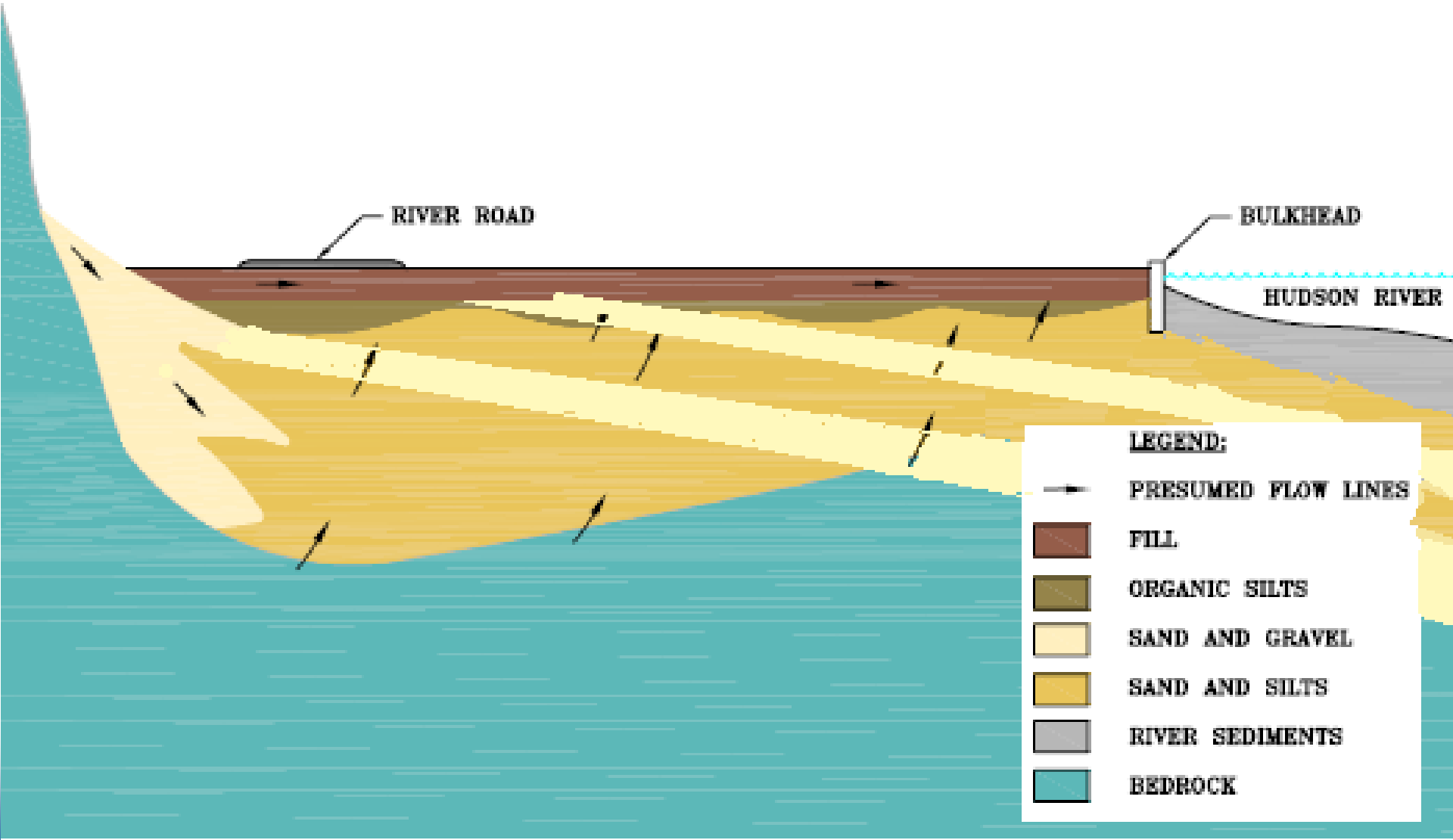


Site Geology/Hydrogeology

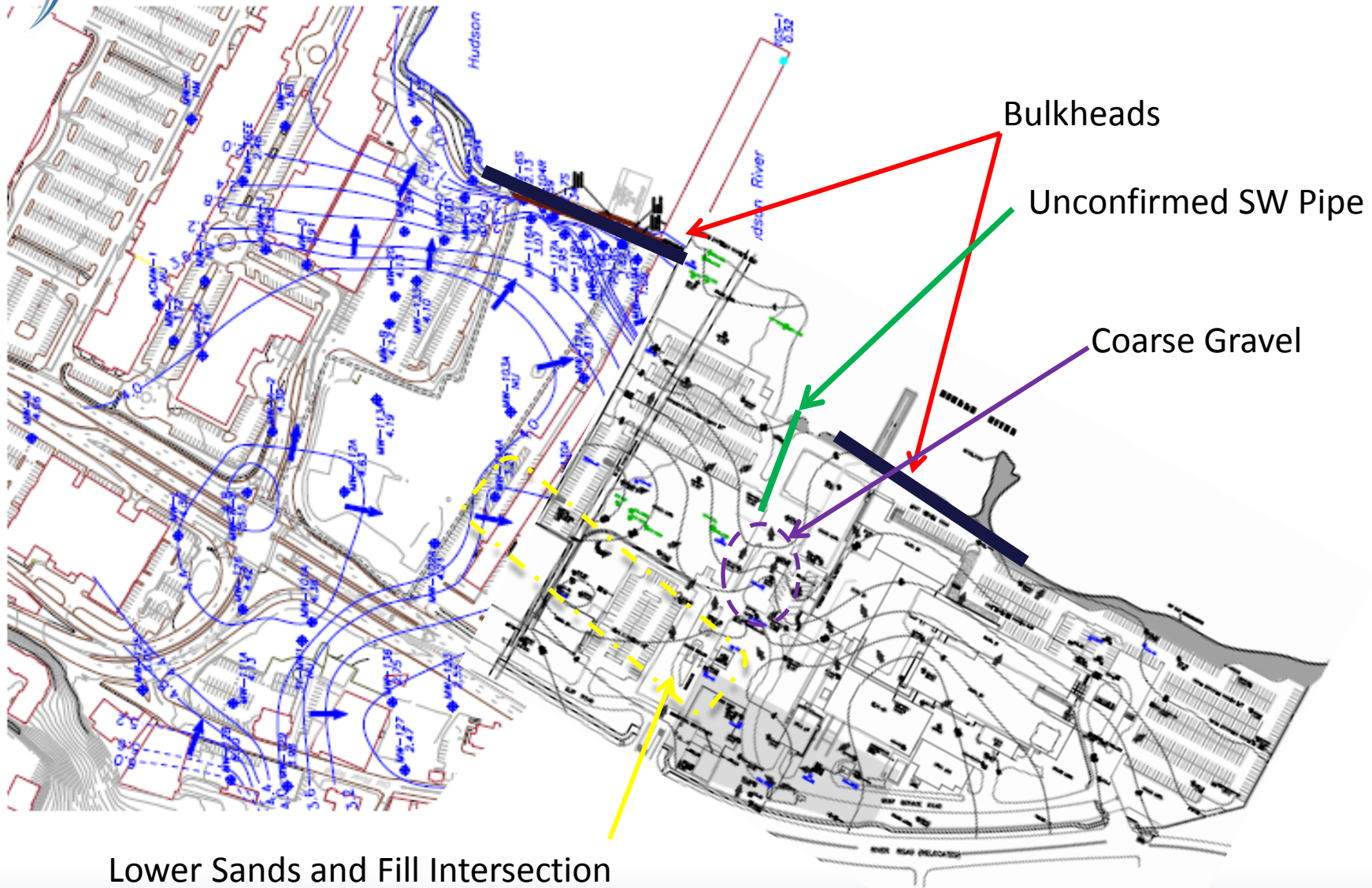
- Fill – Silts, Sands, Debris. 5-25' Thick.
 - 20 to >100 ft/day
- Upper Sands. F/M Grain. Lacustrine Fan
 - 2 to 20 ft/day
- Organic Silts and Clays. Swamp/Marsh
 - 0.001 to 2 feet/day
- Lower Sands. F/M w Gravel. Lacustrine Fan
 - 5 to 20 ft/day
- Stockton Formation. 50 to >90 feet bgs



Site Geology/Hydrogeology



Shallow Groundwater Flow





Background/Concerns

- Extent and Magnitude of Arsenic Initially Under Reported
- Substantial amounts of fill placed between 1880 and 1930
- High Concentrations of Arsenic in Soils and Groundwater
- Sulfuric Acid Plant located North of Site
- USEPA and DEP believed Arsenopyrite-Rich Slag was Source of Arsenic
- Capping and Institutional Controls Not Sufficiently Protective of Groundwater or Hudson River



Procedure

- Conduct Petrographic and XRD Study
- Evaluated Site Geochemistry and Speciation
- Develop ARS for Arsenic per NJDEP Guidance
- Evaluated Solidification/Stabilization Technologies to Address Arsenic Impacted Soils



Petrographic and X-Ray Diffraction Investigation

- Petrography on Cinder/Slag
Samples indicated:
 - Amorphous glass
 - Mullite,
 - Iron Oxide
 - Spherulitic Chalcedony
 - Arsenopyrite (found precipitated
gabbro and quartzite
 - Pyrite

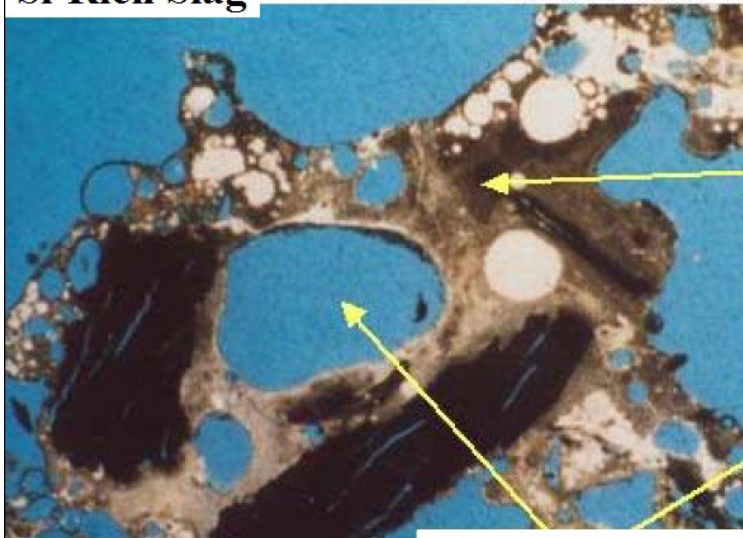




THIN SECTION PHOTOMICROSCOPY

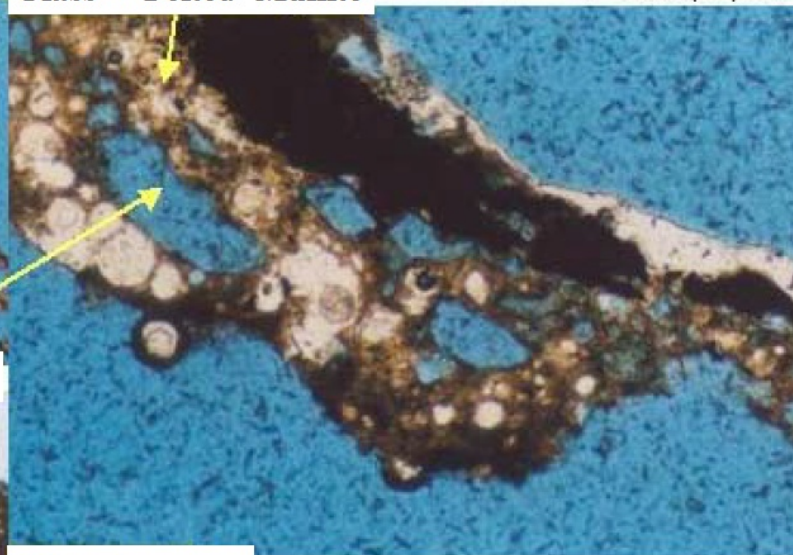
Si-Rich Slag

400 μ M (1.6X) UXN



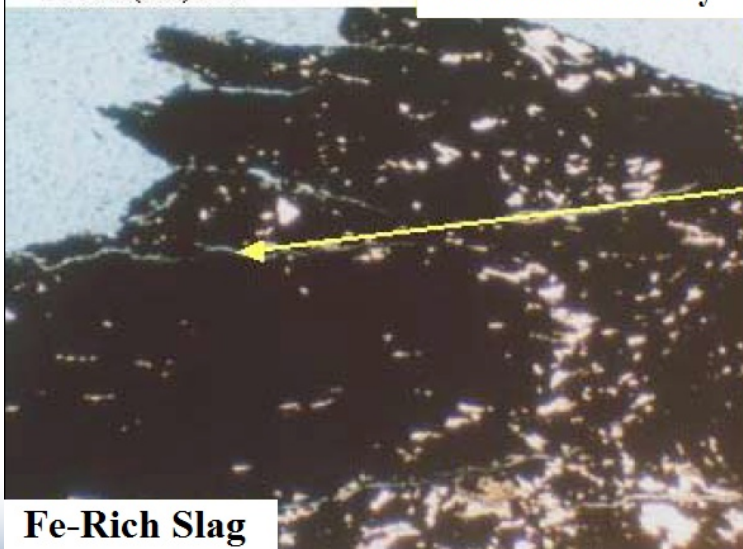
Glass + 'Felted' Mullite

125 μ M (10X) UXN



200 μ M (3.2X) UXN

Vesicular Porosity



Microfractures

GZA-143 (7.5-8.5')

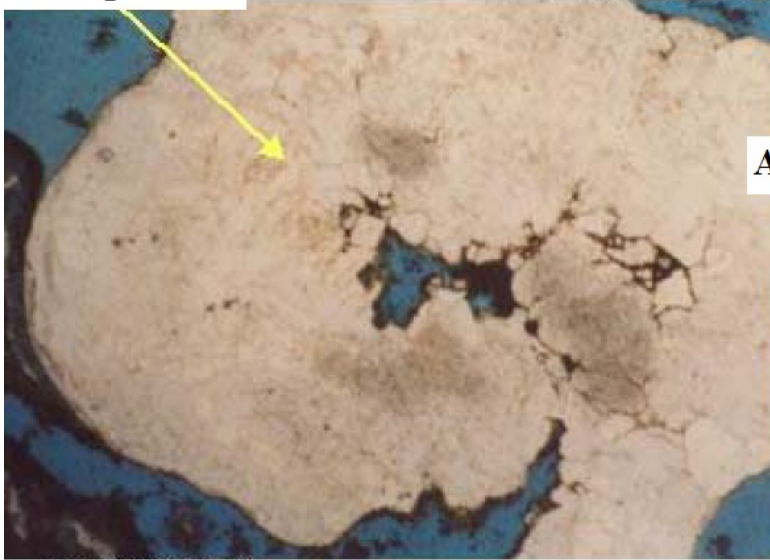
Fe-Rich Slag



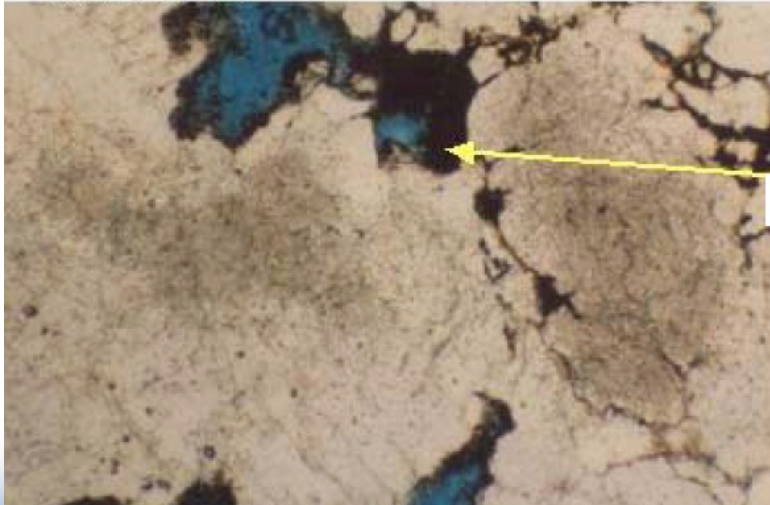
THIN SECTION PHOTOMICROSCOPY

Metaquartzite

400 μM (1.63 \times) USXN



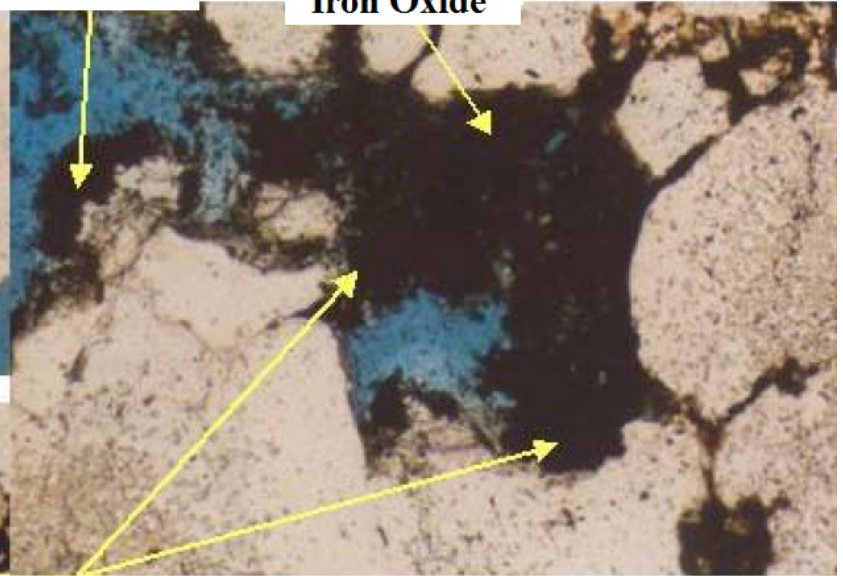
200 μM (3.23 \times) USXN



Arsenopyrite

Iron Oxide

125 μM (103 \times) USXN

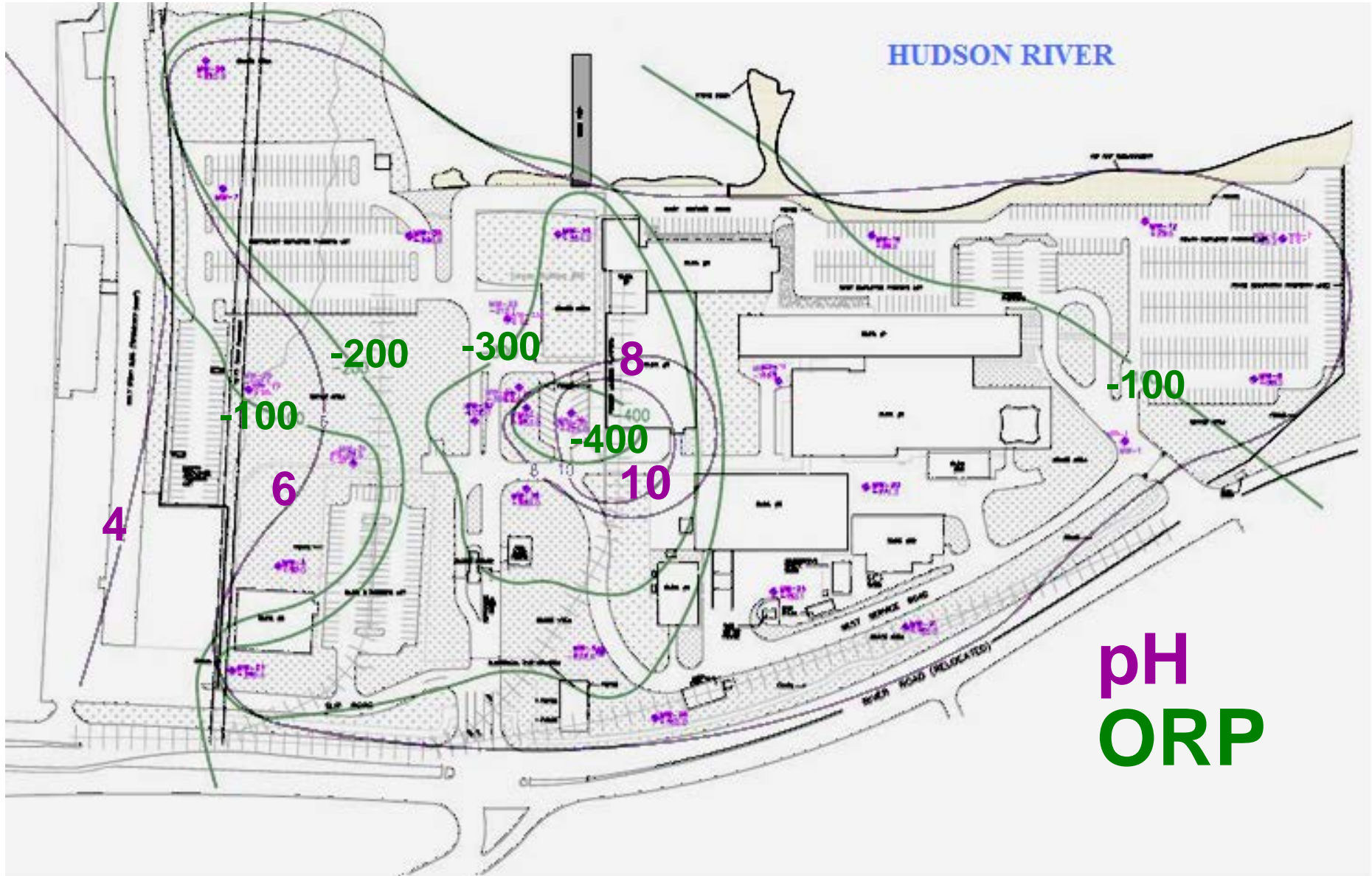


Pyrite

GZA-163 (1.6-1.7')



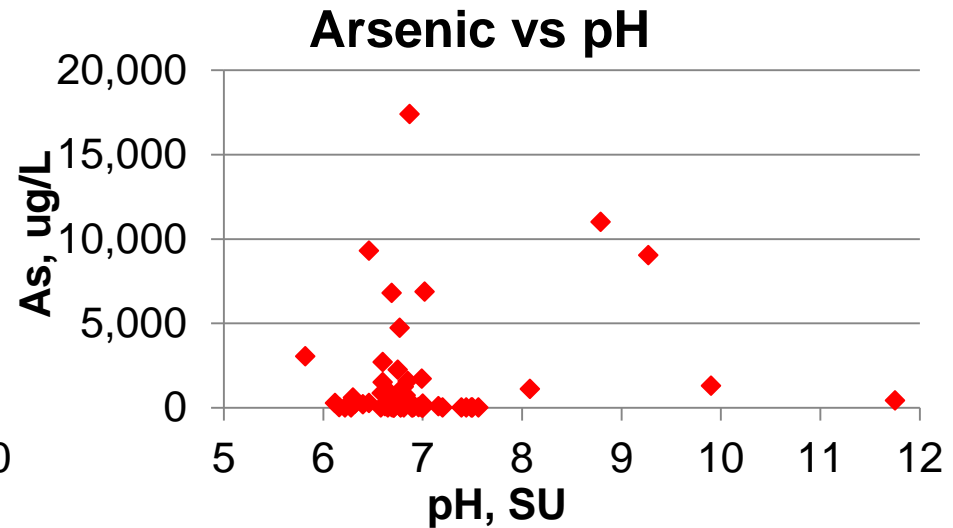
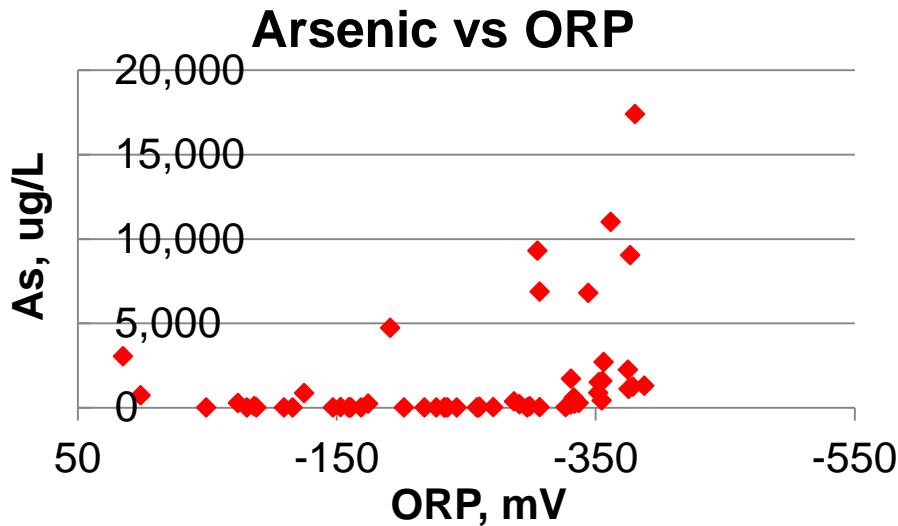
pH and ORP Distribution





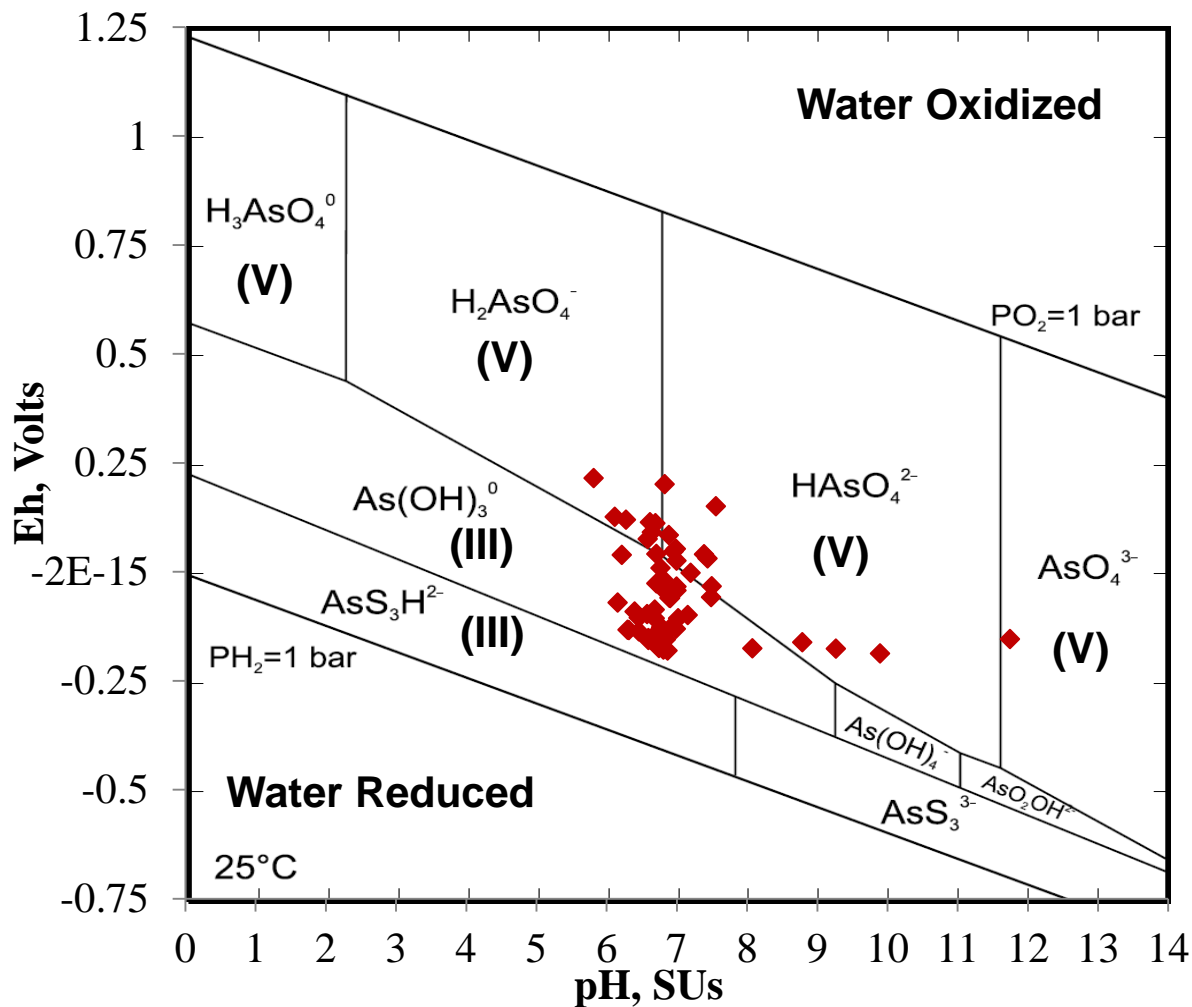
Arsenic and Groundwater Geochemistry

- Found Different Geochemical Zones Corresponding with Dissolved Arsenic
 - Low Eh / High pH (Zone 1)
 - High Eh / Low pH (Zone 2)



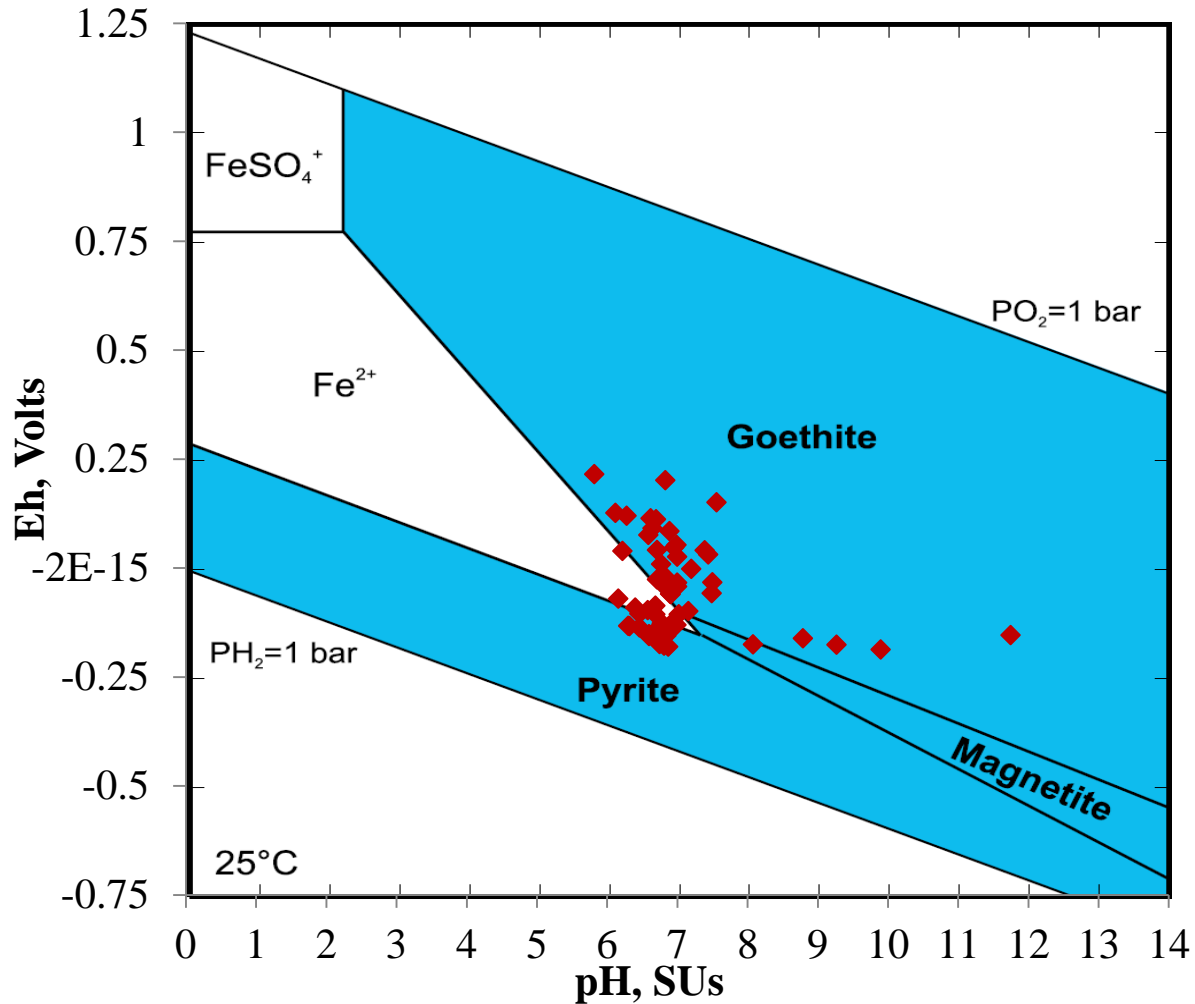


- Eh/pH range indicated groundwater zones straddled the Arsenite (As[III]) stability field





- Eh/pH range indicated groundwater in two zones fell outside iron oxyhydroxide stability field





Groundwater Piper – Arsenic

CATIONS

Ca = 6,900. mg/l

Mg = 390. mg/l

Na = 3,300. mg/l

K = 340. mg/l

ANIONS

HCO₃ = 2,100. mg/l

CO₃ = 1,500. mg/l

Cl = 9,400. mg/l

SO₄ = 3,500. mg/l

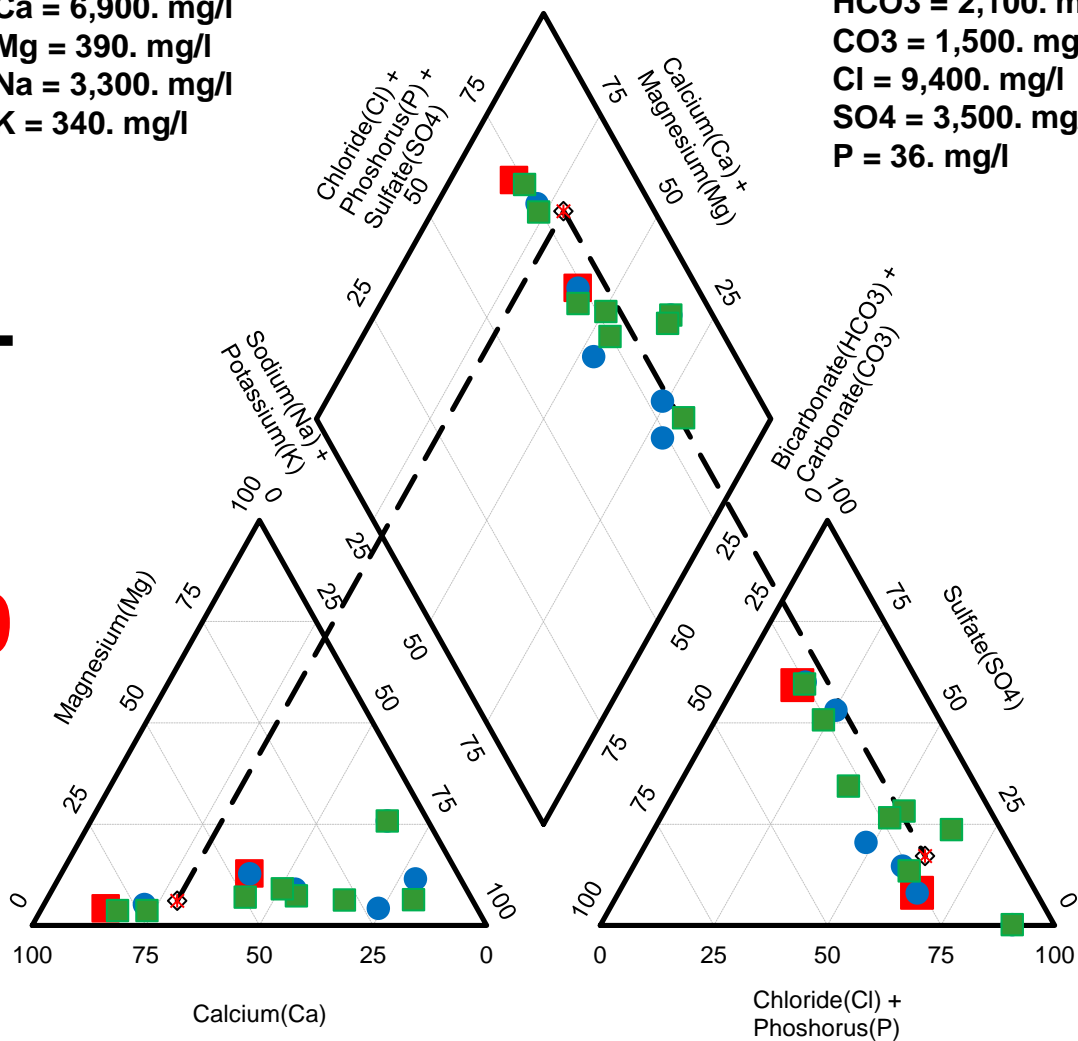
P = 36. mg/l

As, ug/L

> 1,000

> 5,000

> 10,000





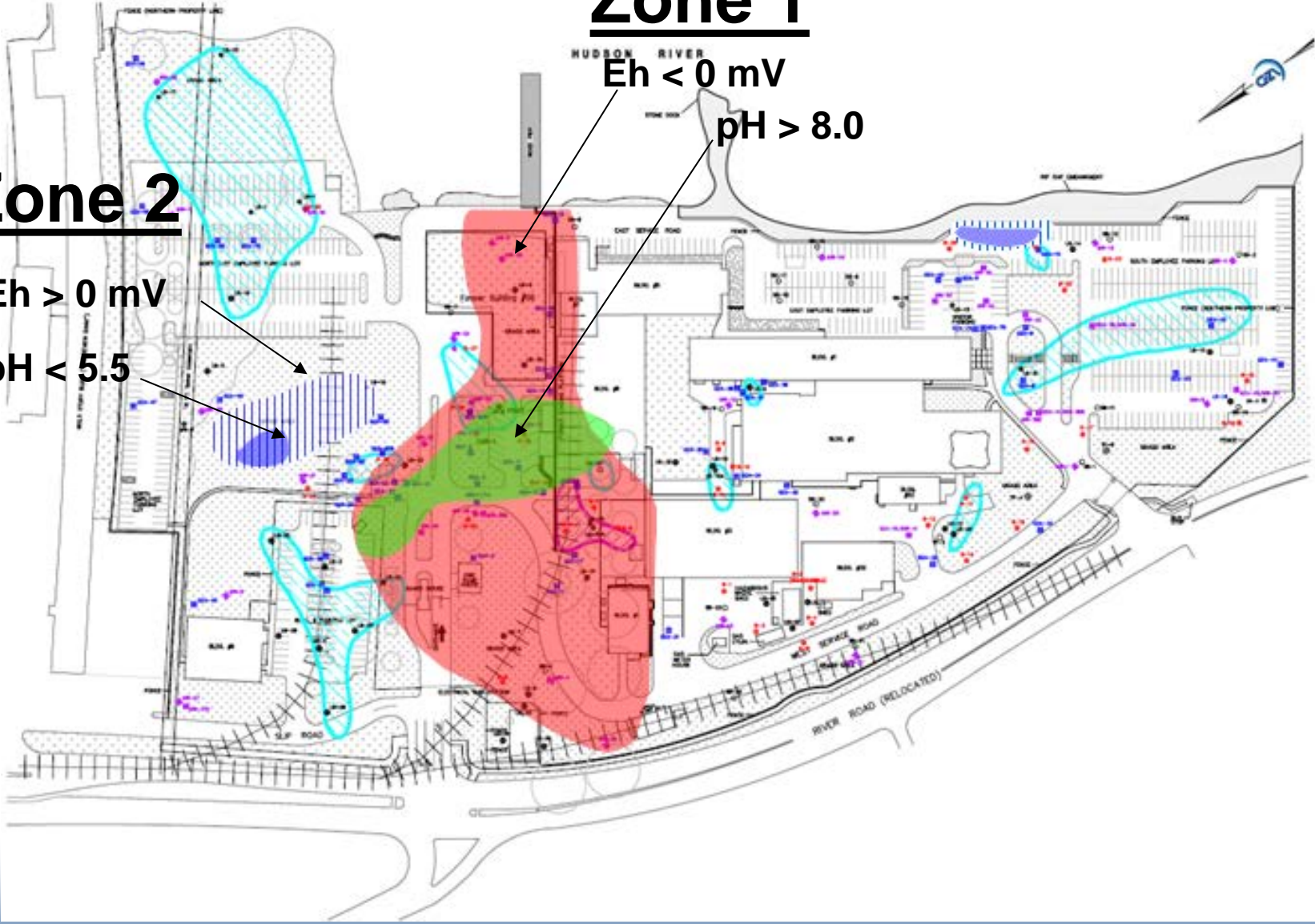
Groundwater Arsenic Distribution

Zone 1

$Eh < 0 \text{ mV}$
 $pH > 8.0$

Zone 2

$Eh > 0 \text{ mV}$
 $pH < 5.5$





Arsenic Cleanup Standard

- NJDEP set direct contact SCC
- Issued guidance for calculating Impact to Groundwater ARS
 - Analyze soils for SPLP compare to LS (3 ppb)
 - ARS = Highest C_T for which C_L d LS = 22 ppm
 - ARS Using site Specific K_d

$$K_d = \frac{(C_T M_s - C_L V_L) / M_s}{C_L} \quad ARS = C_{gw} \left\{ K_d + \frac{\theta_w + \theta_a H'}{\rho_b} \right\} DAF$$

- K_d ranged from 22 to 17,000 L/kg. ARS using 22 = 0.8 ppm
- Regression Analysis of C_T vs C_L = Failed

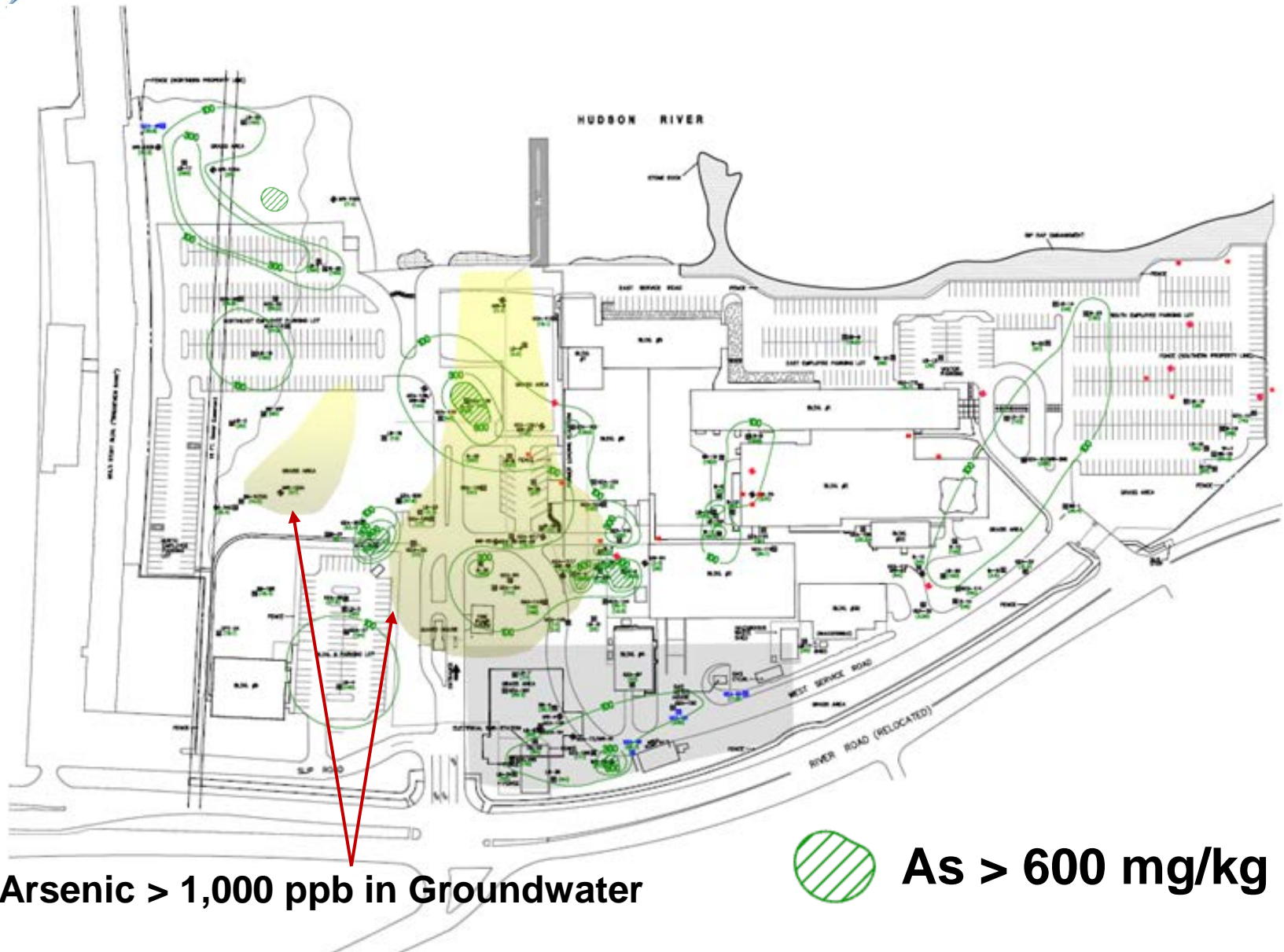


Arsenic Cleanup Standard

- Arsenic cleanup Standard Very Dependent on Site Geochemistry
 - No clear correlations with SPLP results or Kd
- Argued that Arsenic solubility was dependent on Eh, pH and Ironoxyhydroxide stability. We would excavate soils above HFM Maximum of 1,096 ppm
 - NJDEP willing to compromise
 - Look at correlations between soil and groundwater hot spots

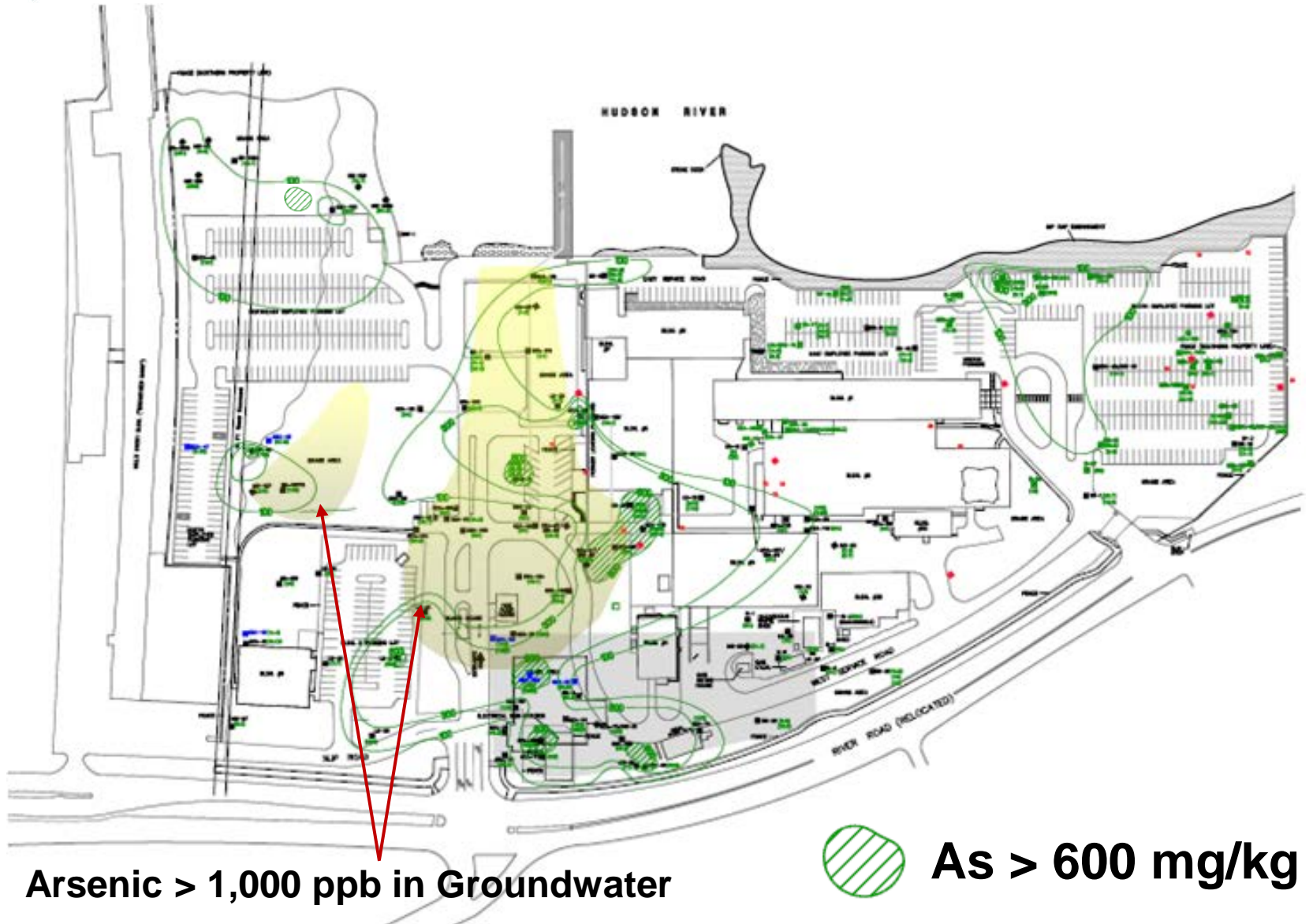


Arsenic in Unsaturated Soil





Arsenic in Saturated Soil





Treatability Study

- Prepared Standard Operating Procedures
- Mixture Designs Applied
 - Type I Portland Cement
 - Addition rates of 10%, 7.5%
- Bench scale Hobart type mixer
- 4-inch diameter plastic cylindrical curing molds
- Cured for 28 days





Treatability Study Results

- PAHs and VOCs Passed NJDEP Default Leachate Criteria
- Compressive Strength Strong
- Low Permeability
- Leachate Results
 - Lead Present in One Sample
 - Arsenic Present In Leachate



Treatability Study Results

➤ Lead

- Lowering Cement Content to 7.5%
Reduced Lead Leachability

➤ Arsenic

- Arsenic Present as Arsenite, Soluble Under Reducing Conditions
- Add Ferric Sulfate to Increase Oxidation State of Arsenic Yielding Arsenate
- Ferric Sulfate Resulted in Formation of Iron Oxyhydroxides Promoting Arsenic Bonding



Conclusions

- Arsenic Source Cinder-Ash Coal, not Arsenopyrite-Rich Slag
- Arsenic Solubility Due to Site Specific Geochemical Parameters
- Established Reasonable ARS for Arsenic
- NJDEP Required Groundwater Cut-off Wall to Control Future Arsenic Discharge to the Hudson River
- Demonstrated that ISS was a Treatment Technology for Arsenic Impacted Soils