

Vertex Environmental Inc.



Optimizing Oxidant Delivery for In-Situ Remediation

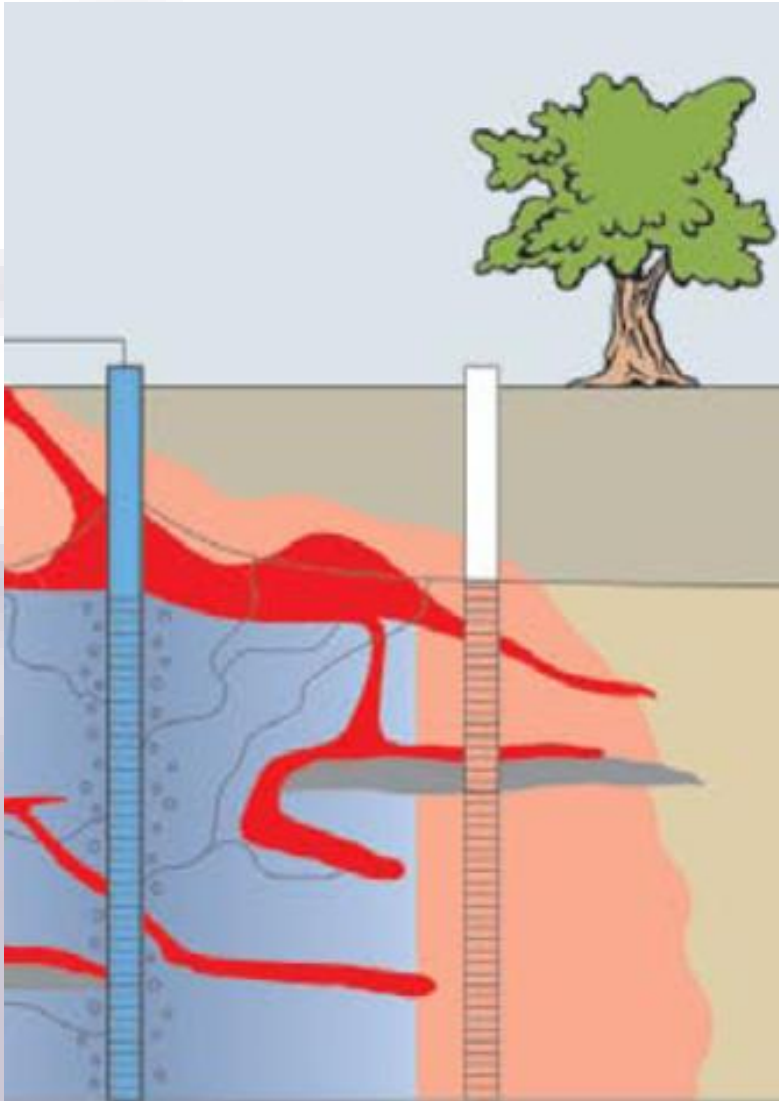
Remediation Technologies Symposium 2012

Banff, Alberta

October 19, 2012

Bruce Tunncliffe

Overview



- Background
 - Purpose of Talk
 - Vertex
 - What is In-Situ? Key Factors
- Contaminant Distribution
 - Free Phase Product
 - Dissolved Phase Plume
- Oxidant Delivery
 - Delivery Method
 - Delivery Experiment
- Questions



Purpose of Talk

- In-situ remediation is now commonly considered and applied
- Currently: pre-remediation data & design for an in-situ project is similar to excavation
- Purpose: Explore what we need to consider to better deliver a remediation liquid (i.e oxidant) to the subsurface to improve remedial success.

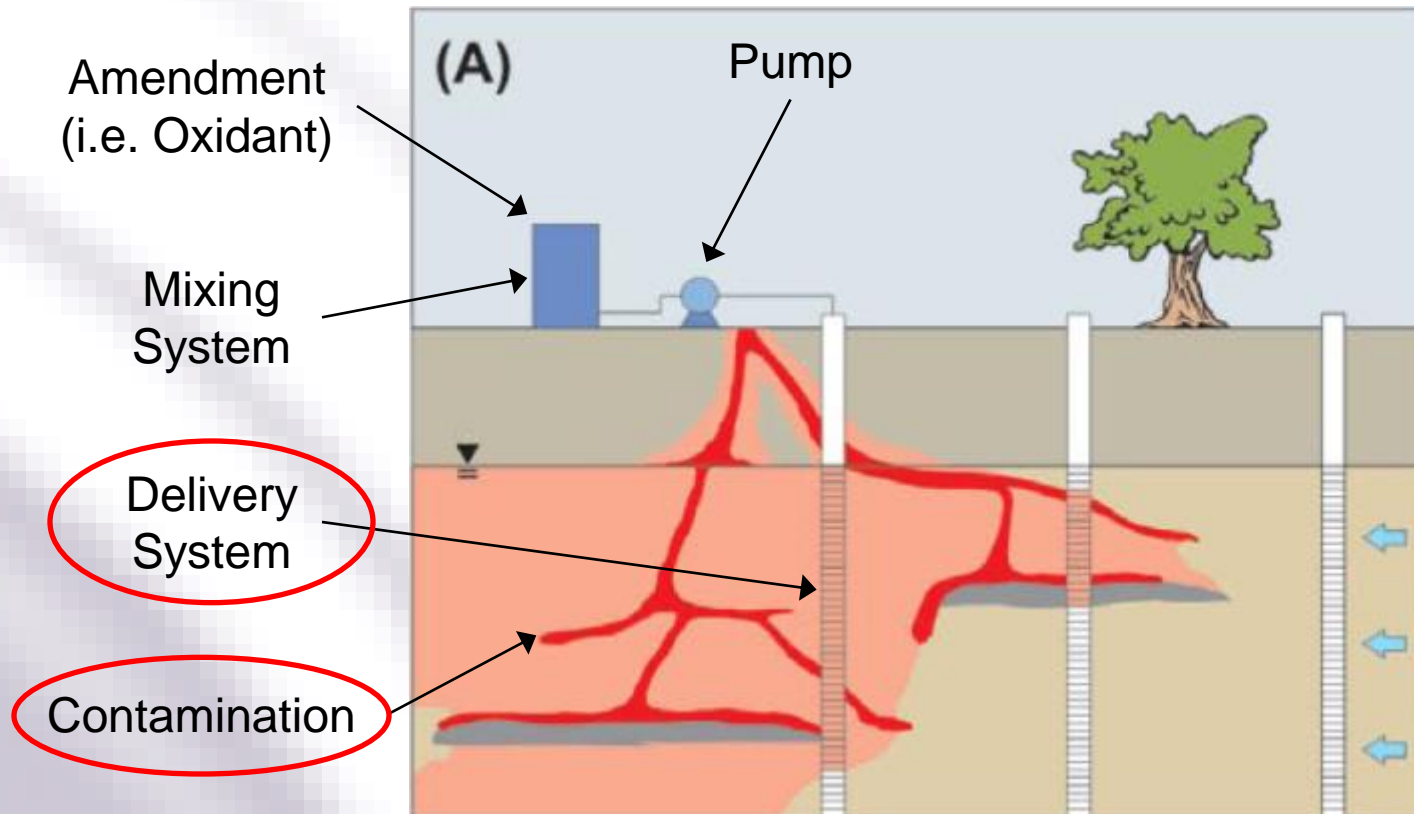


Vertex Background

- Environmental Contracting
- Remediation and injection services
- Clients are consultants
 - Consultant: Phase II ESA
 - Vertex: Remedial Design, Remediation (bench, pilot, full-scale)

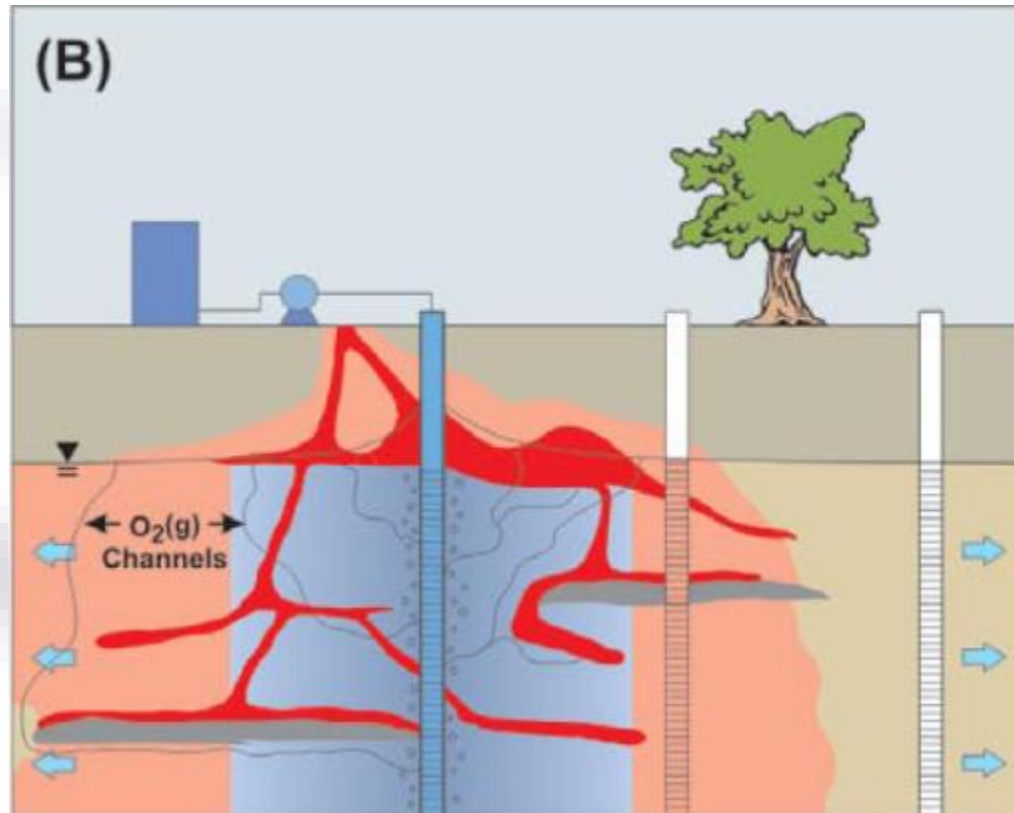


What Is In-Situ?

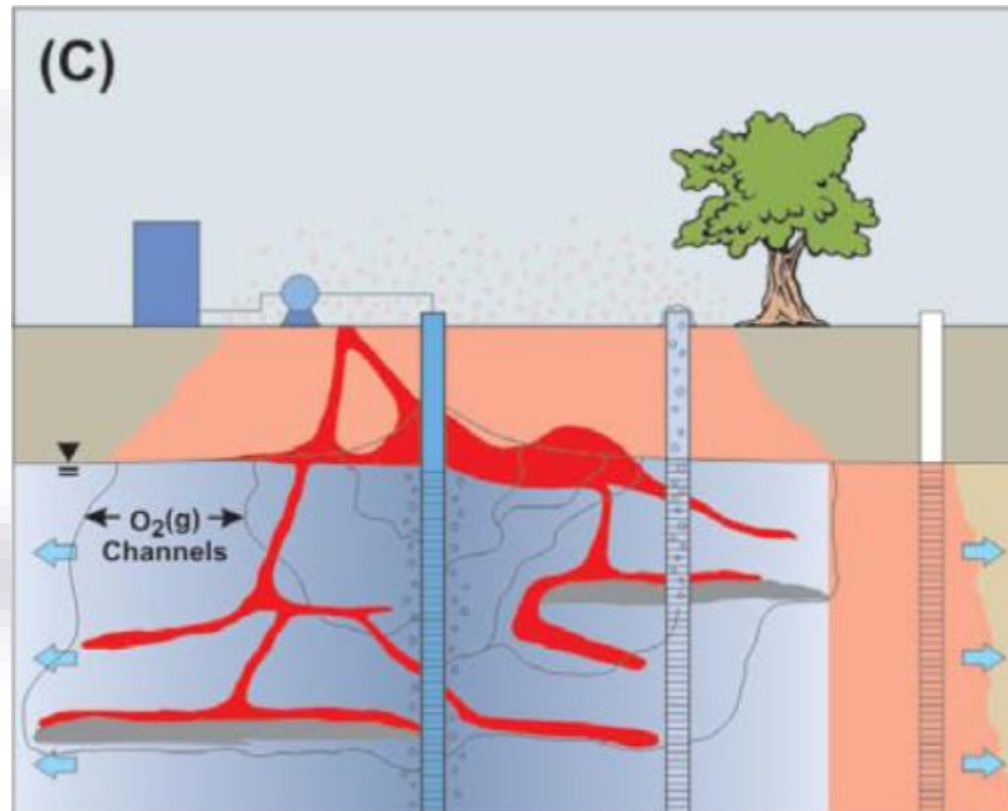


- In-Situ is a Latin phrase meaning “in its place”
- Remediation completed in the ground

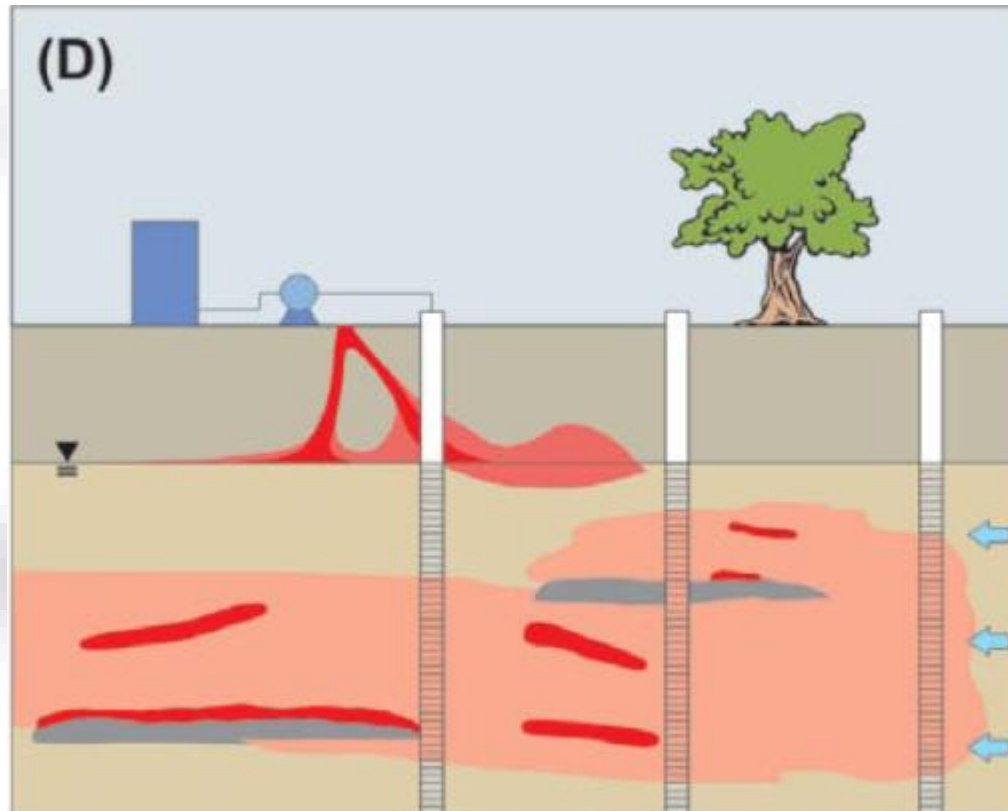
What Is In-Situ?



What Is In-Situ?



What Is In-Situ?



Contaminant Characterization



Advanced Characterization Tools:

- Laser Induced Fluorescence
 - LIF
 - Free Product (LNAPL)
- Membrane Interface Probe
 - MIP
 - Dissolved Phase (Plume)



Contaminant Characterization

- Laser Induced Fluorescence



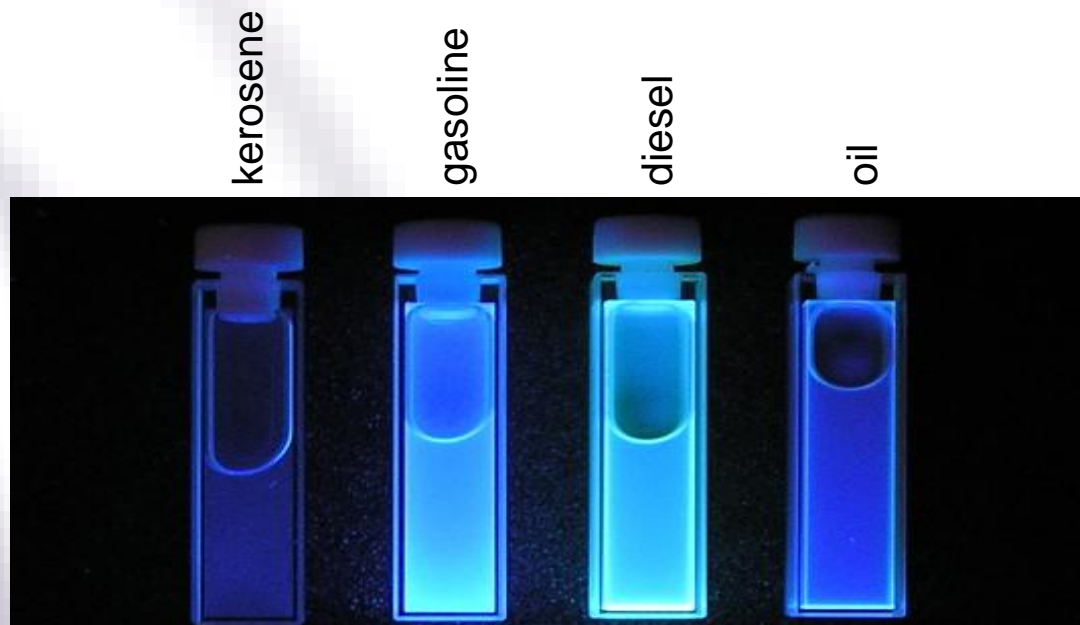
Laser Induced Fluorescence

- Developed in early 1990s
 - US Army Corps of Engineers
- UVOST (Ultra Violet Optical Screening Tool)
- Light-based
- Equipment
 - Geoprobe (direct push)
 - Fibre optic cable
 - Sapphire window



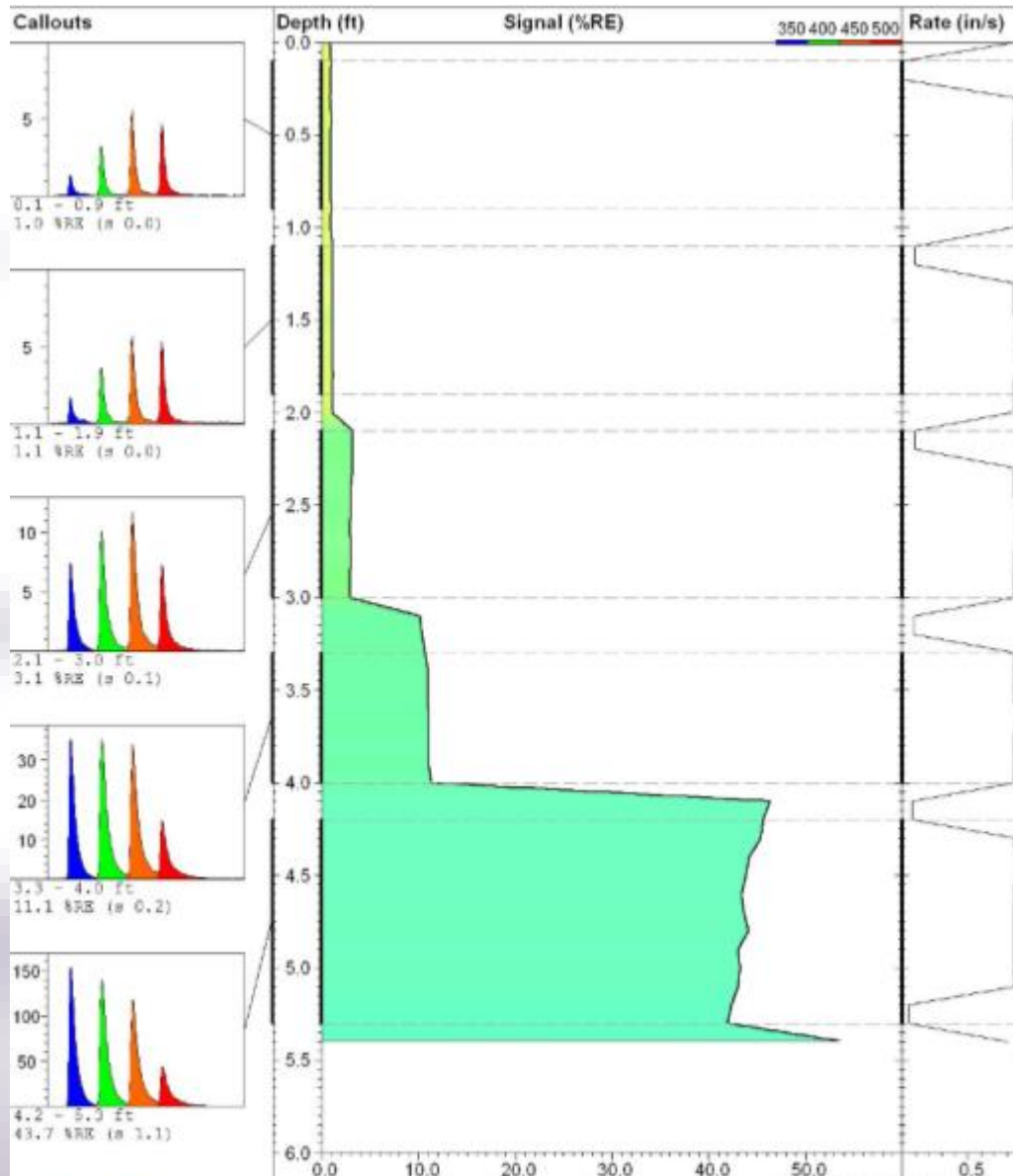
Laser Induced Fluorescence

- Spectroscopy – molecules absorb light (gain energy) and then emit light (lose energy = fluoresce)
- Aromatic molecules (PAHs) readily absorb and emit light



Laser Induced Fluorescence





Clean Sand

350 ppm

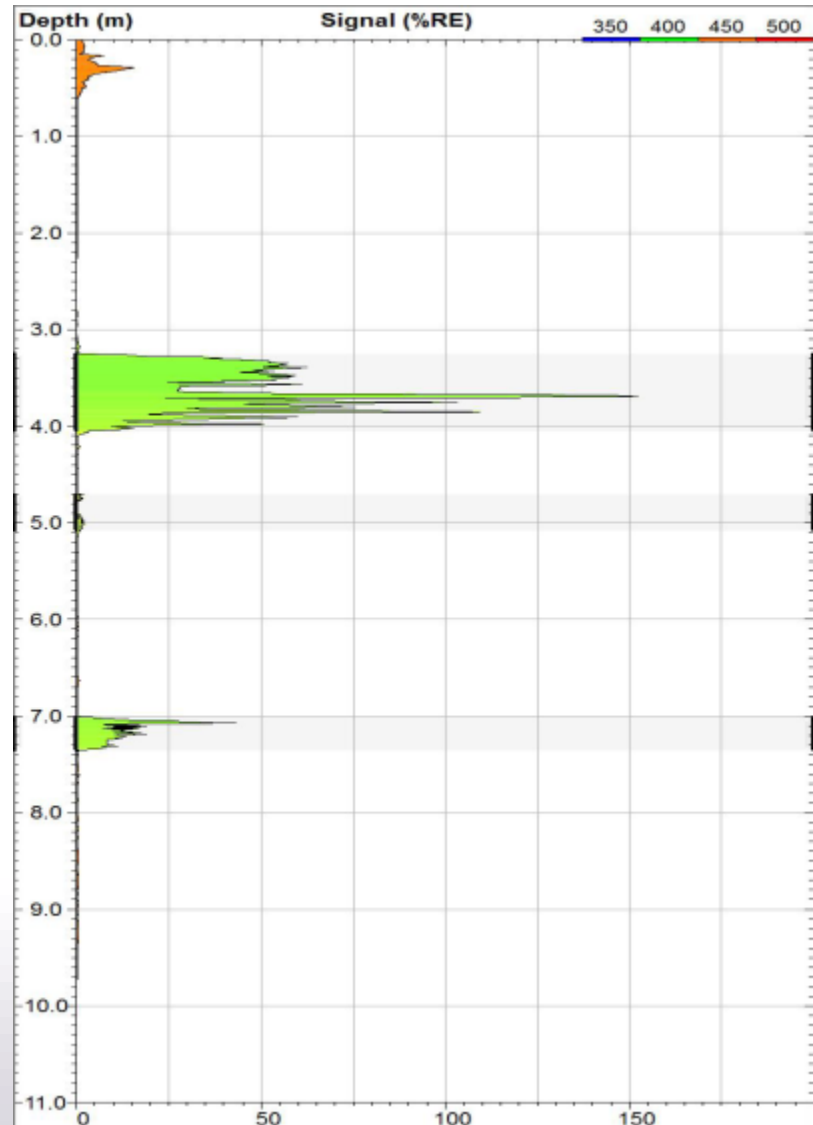
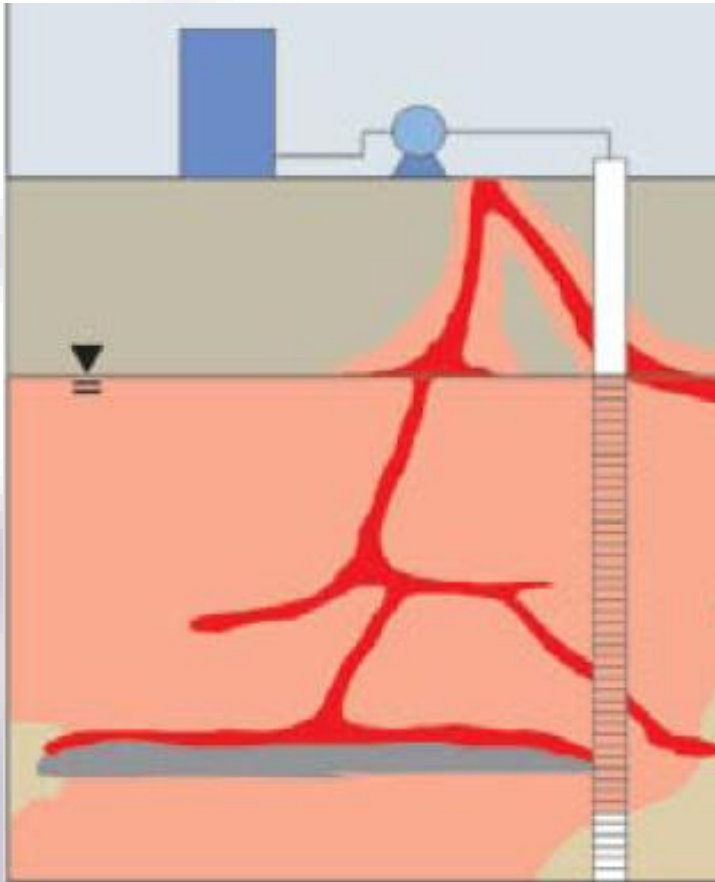
1,100 ppm

10,000 ppm

84,000 ppm

Gasoline

Laser Induced Fluorescence

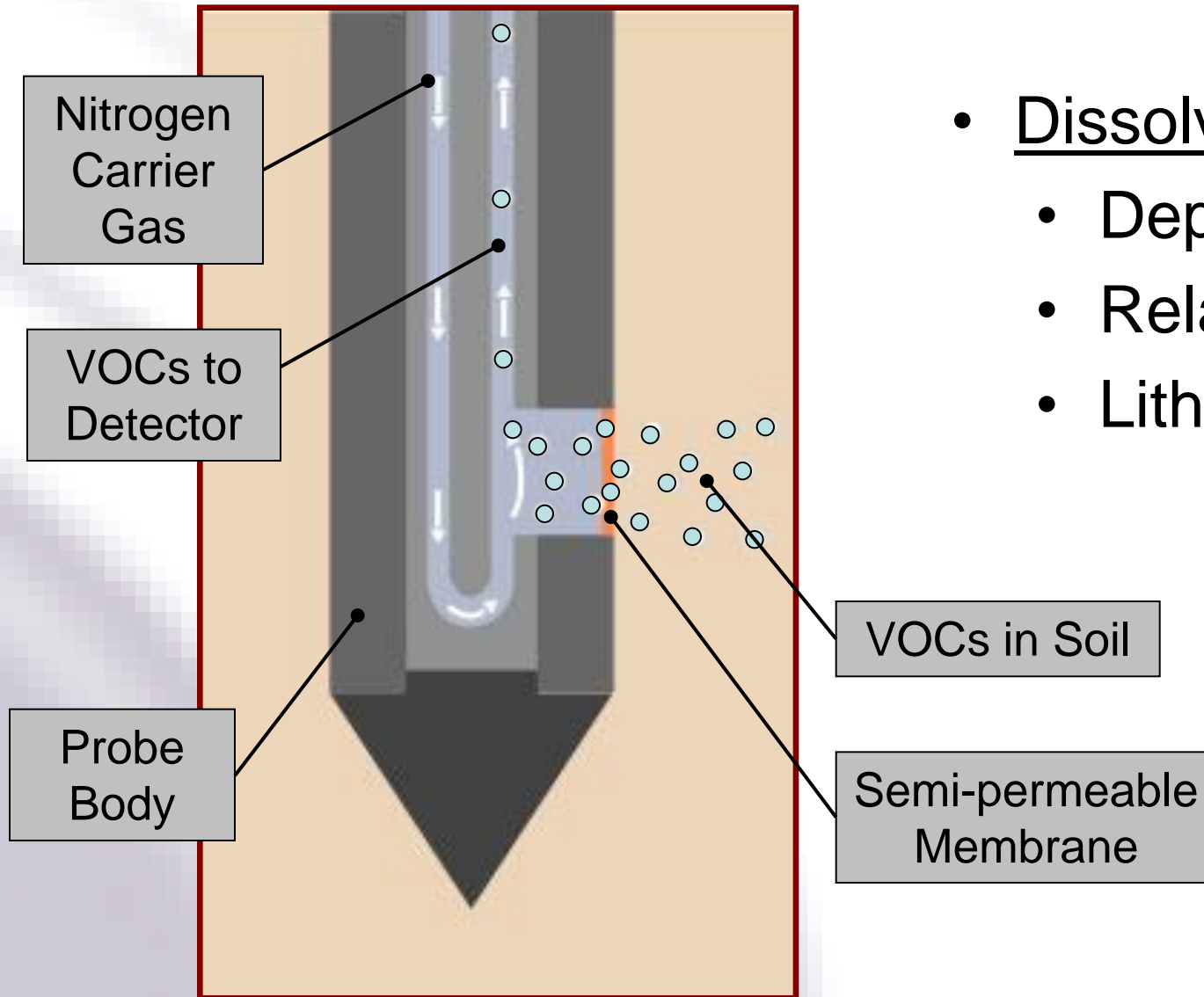


Contaminant Characterization

- Membrane Interface Probe



Membrane Interface Probe

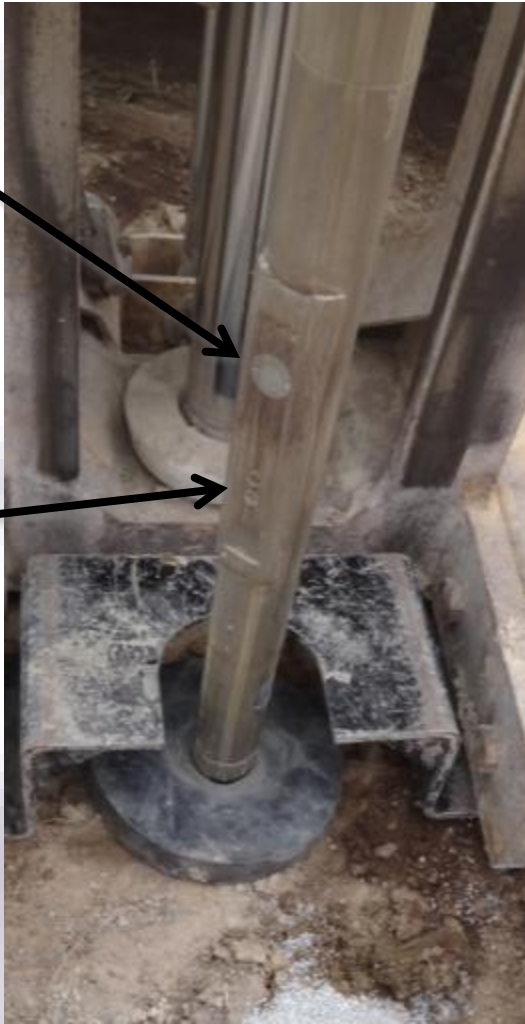


- Dissolved phase:
 - Depth
 - Relative conc.
 - Lithology

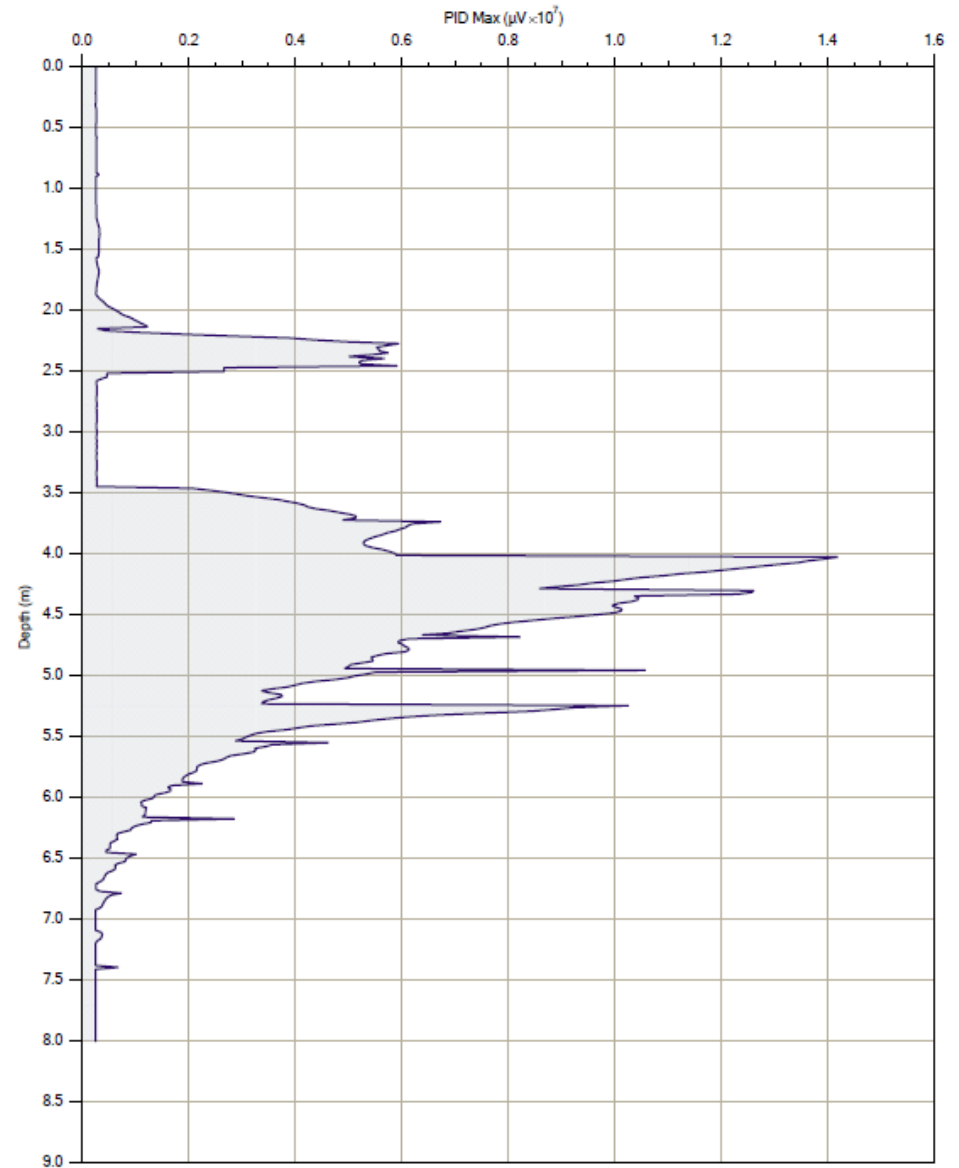
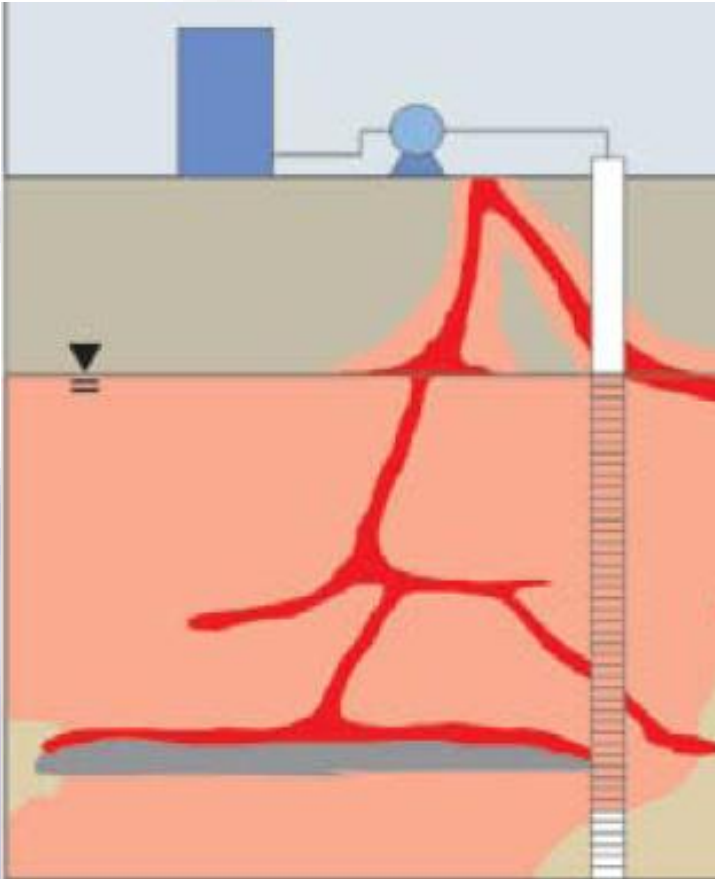
Membrane Interface Probe

Membrane

Heater
Block



MIP Example



Characterization – In Sights

Standard Characterization

- Monitoring Wells and Sampling (3 m)

Advanced Characterization Tools

- Laser Induced Fluorescence (0.01 m)
 - Free Product (LNAPL)
- Membrane Interface Probe (0.3 m)
 - Dissolved Phase (Plume)
- 3-D view: superior understanding of aquifer
- Remedial design: Target discrete zones



Oxidant Delivery



- Oxidant Delivery Method
 - Injection Wells
 - Injection Points
- Delivery Experiment
 - Injection Well
 - Injection Points



Oxidant Delivery Method

- Article published in fall 2010
 - Ground Water Monitoring & Remediation
- 5 authors (Krembs et al. 2010):
 - Created a database
 - Examined 242 in-situ projects ('96 to '07)
 - All oxidation projects:
Permanganate, catalyzed hydrogen peroxide (H_2O_2), ozone, persulfate, peroxone (ozone and H_2O_2), percarbonate
 - Oxidants = liquids or gases



Delivery Method

Gas
Injection

Oxidant Delivery Method	Percent of Sites
Injection wells	47
Direct push	23
Sparge points	14
Trenches, Galleries	10
Recirculation	7
Fracturing	6
Mechanical mixing	2
Horizontal wells	1

Percentages greater than 100% because multiple delivery techniques were used at some sites.

Delivery Method

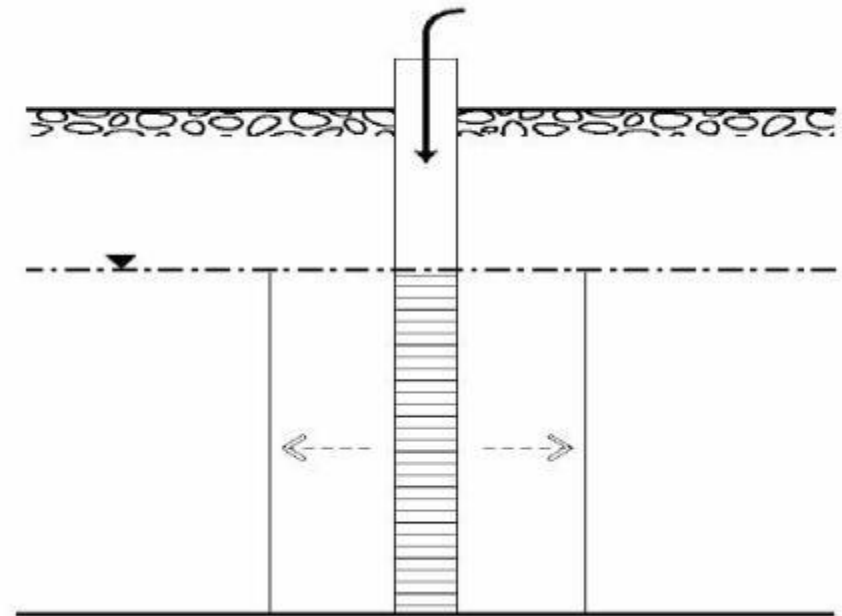
Liquid
Injection
only

Oxidant Delivery Method	Percent of Sites
Injection wells	54
Direct push	26
Sparge points	
Trenches, Galleries	11
Recirculation	8
Fracturing	7
Mechanical mixing	2
Horizontal wells	1

Percentages greater than 100% because multiple delivery techniques were used at some sites.

Delivery Method

Injection Well



Source: ITRC, 2005

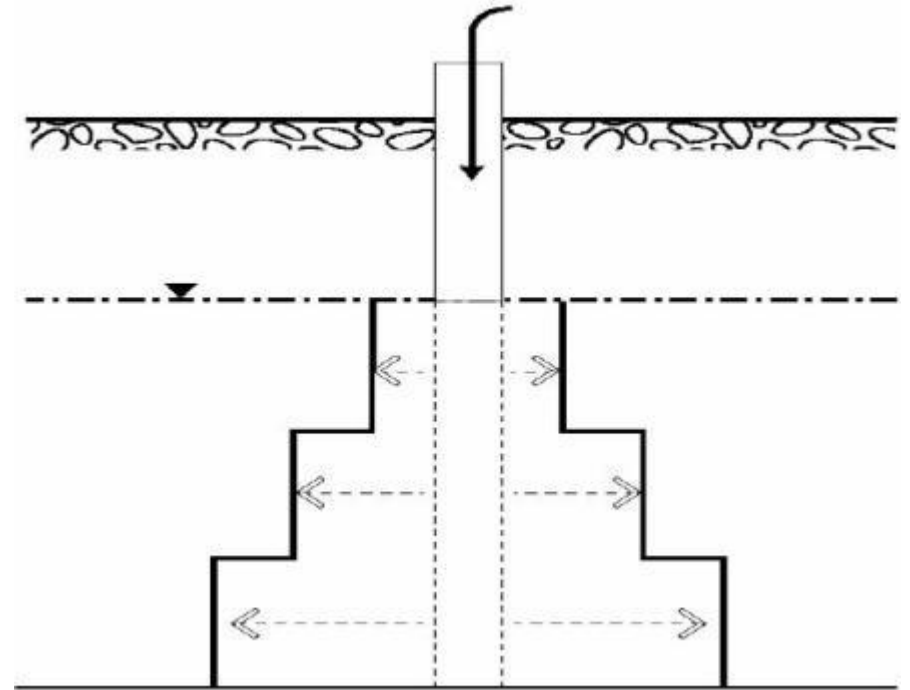


Delivery Methods

Direct Push



Bottom Up

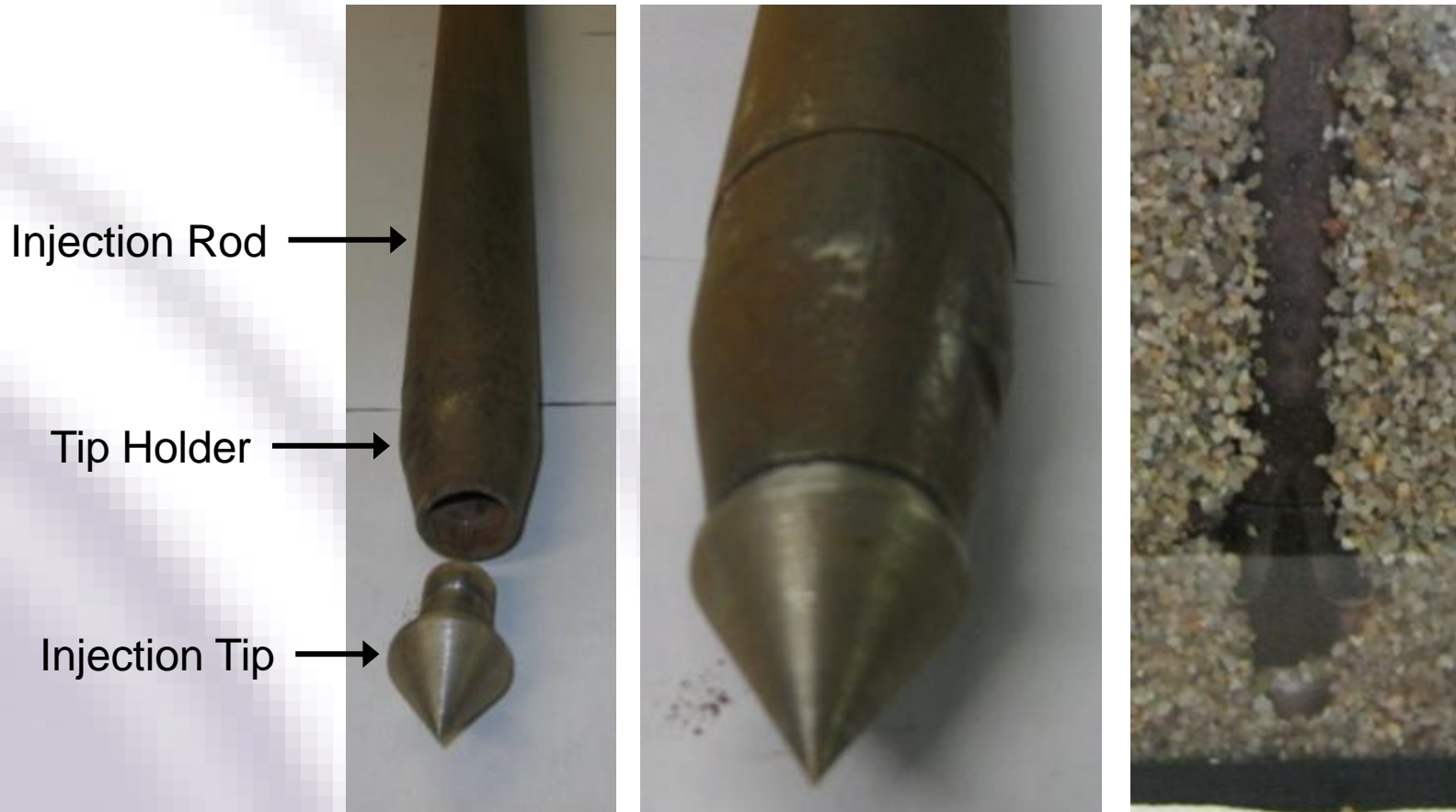


Source: ITRC, 2005



Delivery Methods

Injection Point



Delivery Laboratory Experiment

Overview

- Purpose:
 - Assessment of Dye Distribution
- Experiments:
 - Distribution from Injection Well
 - Coarse Sand
 - Distribution from Injection Point
 - Coarse Sand
 - Sand with Clay Layer



Delivery Laboratory Experiment

Glass Container

- 51 by 26 by 31 cm
- 41 L

Coarse Sand

- Filter Sand (#3)
- Bentonite Clay

Water and Dye

- Food grade dye

Injection

- Gravity Feed (0 psi)
- Hydraulic Head=1.5 m
- Flow = 1.0 to 4.5 lpm



Delivery Laboratory Experiment

Experiment #1 Injection into Injection Well



Delivery Laboratory Experiment



Injection Well, Experiment #1
2" diameter PVC Slot 10 Well, screened to bottom of tank
Flow = 4.5 lpm

Delivery Laboratory Experiment

Experiment #2

Injection into Temporary Point



Delivery Laboratory Experiment



Pulling Injection Rod to remove Injection Tip

Delivery Laboratory Experiment



Injection into Injection Point, Experiment #1
Injection rod located 9 cm from bottom of tank
Flow = 4.0 lpm



Delivery Laboratory Experiment



Injection into Injection Point, Experiment #2
Injection rod located 9 cm from bottom of tank
Flow = 2.8 lpm



Delivery Laboratory Experiment

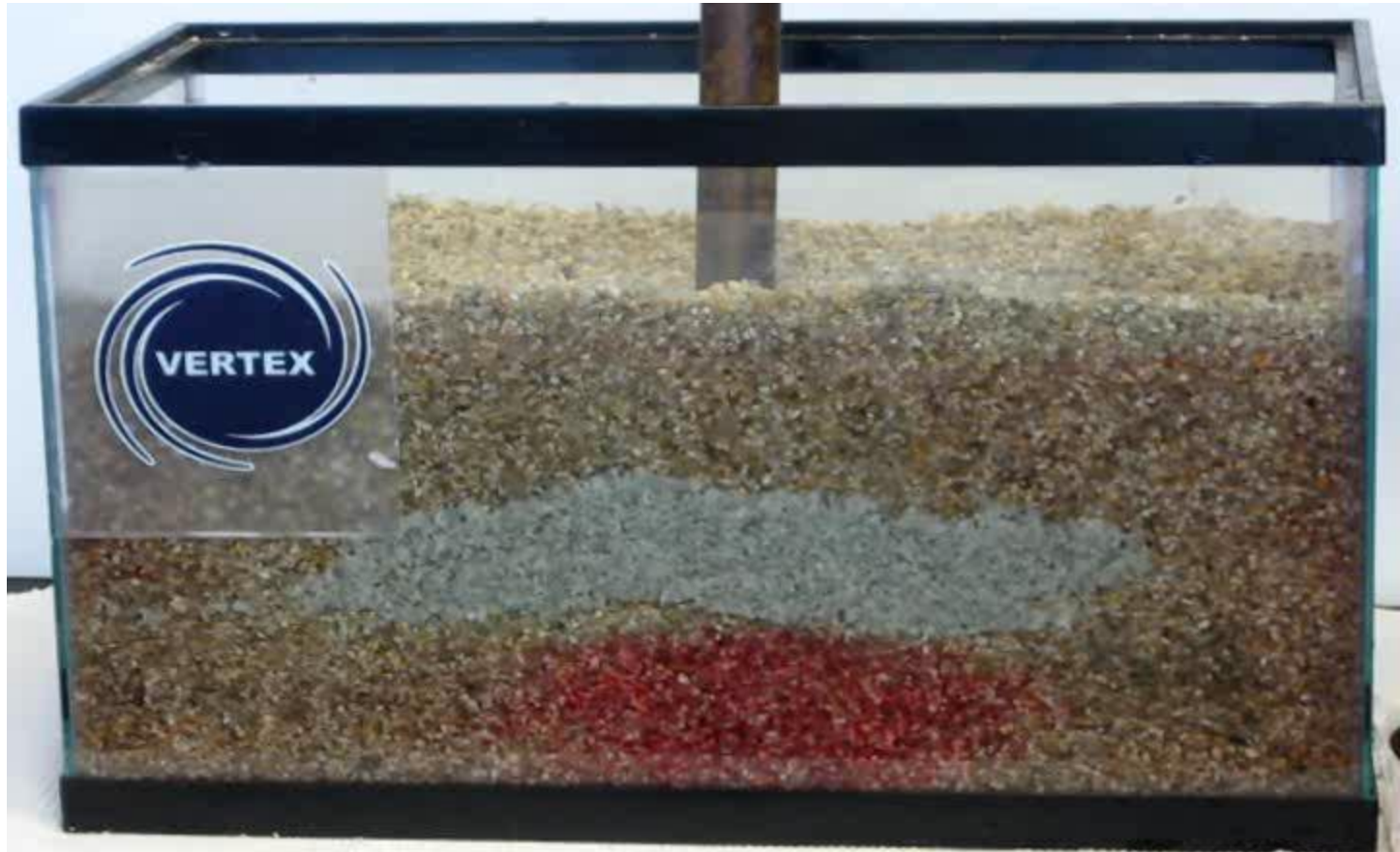
Experiment #3 Injection into Temporary Point Clay Aquitard



Delivery Laboratory Experiment



Delivery Laboratory Experiment



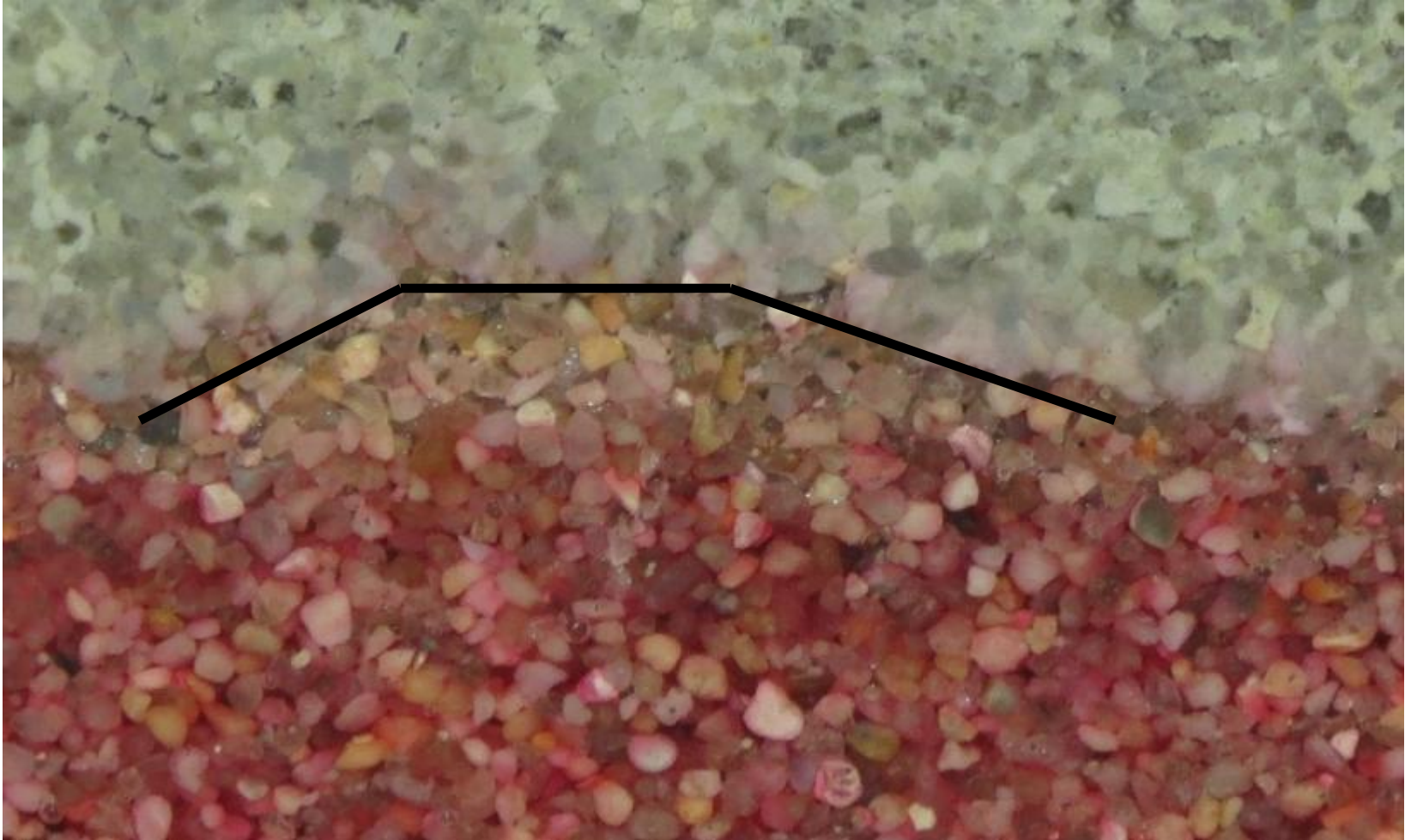
Injection into Injection Point, Experiment #3
Injection rod located 5 cm from bottom of tank
Flow = 1.0 lpm



Delivery Laboratory Experiment



Delivery Laboratory Experiment



Delivery Laboratory Experiment



Injection into Injection Point, Experiment #3
Time = 0 min

Delivery Laboratory Experiment



Injection into Injection Point, Experiment #3
Time = 2 min

Delivery Laboratory Experiment



Injection into Injection Point, Experiment #3
Time = 4 min

Delivery Laboratory Experiment



Injection into Injection Point, Experiment #3
Time = 5 min

Delivery Laboratory Experiment



Injection into Injection Point, Experiment #3
Time = 15 min

Delivery – In Sights

- Homogenous sand:
 - Well = Horizontal distribution
 - Point = Radial distribution
- Clay aquitard:
 - Significant influence on dye distribution
 - Tight soil will control distribution of liquid
 - No dye migration into clay
 - Idea: Injection point could be placed into clay to deliver directly to the clay unit



Conclusions / In Sights

- Oxidant chemistry works
- Better oxidant delivery = better remediation
- Two important factors:
 - Understanding contaminant distribution
 - Designing delivery system
- Contaminant Distribution
 - Monitoring well (300 cm resolution)
 - LIF – free phase (1 cm resolution)
 - MIP – dissolved phase (30 cm resolution)
 - Detailed data – produce 3D plots, and design remediation approach accordingly



Conclusions / In Sights

- Oxidant Delivery System
 - 2 most common methods: Wells and Points
 - Wells:
 - Good for homogeneous aquifers and evenly distributed contamination
 - If multiple injections, lower cost option
 - Limited ability to deliver into tighter soils, due to preferential flow into coarse soils
 - Major problem, considering most historical contamination is located within tight soils



Conclusions / In Sights

- Injection Points:
 - Pin point delivery – flow from end of injection rod allows accurate vertical placement of oxidant
 - Vertical distribution – rods can be advanced to specific depths to target impacted seams
 - Targeting tight soil layer = distribution into clays
 - Higher injection pressure with steel injection rod construction
 - Multiple injections, different locations
- Overall: When designing an in-situ remediation program, consider advanced characterization techniques and delivery method





Questions?

Thank You for Your Time

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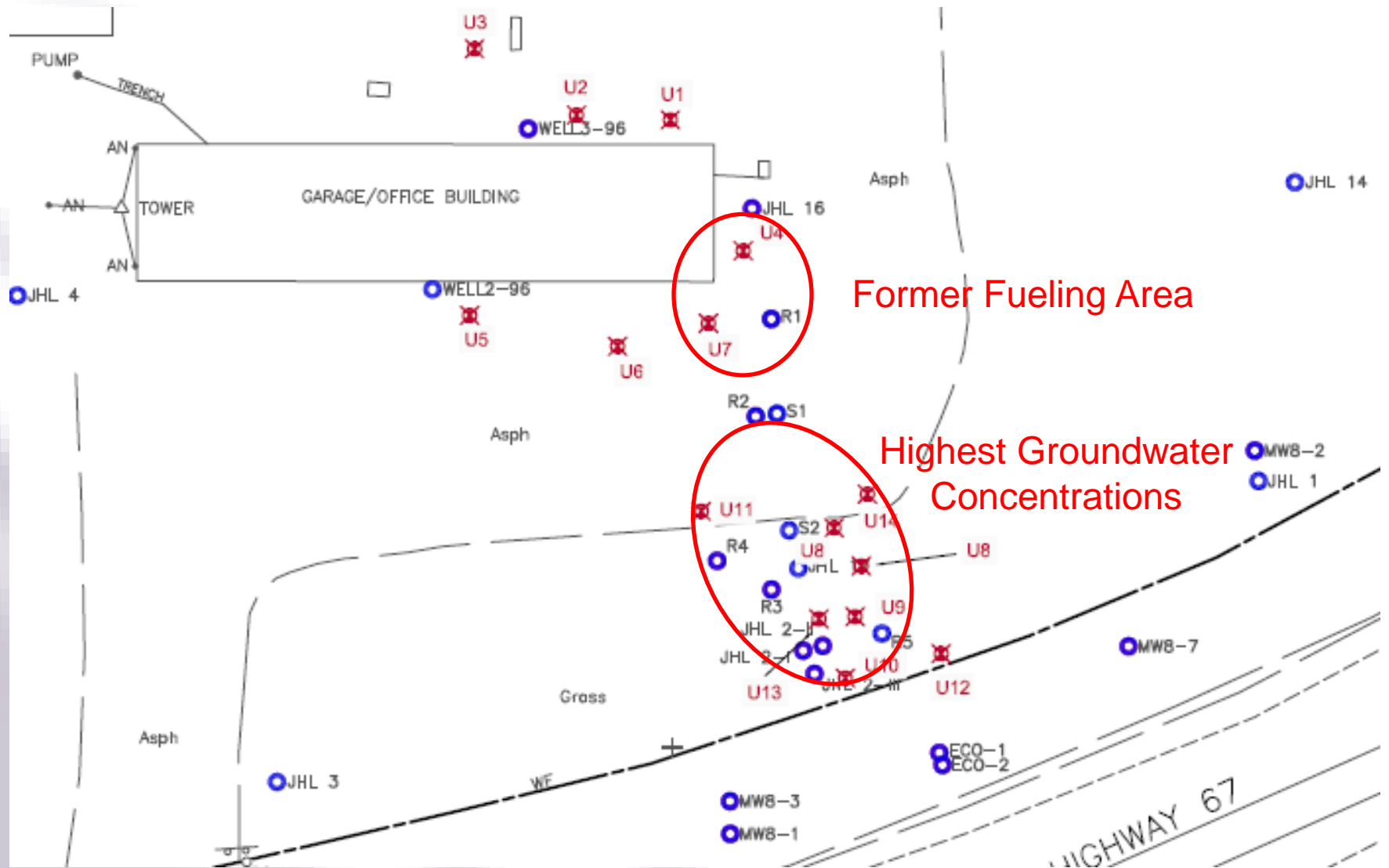
Case Study

Advanced Characterization and Oxidant Delivery

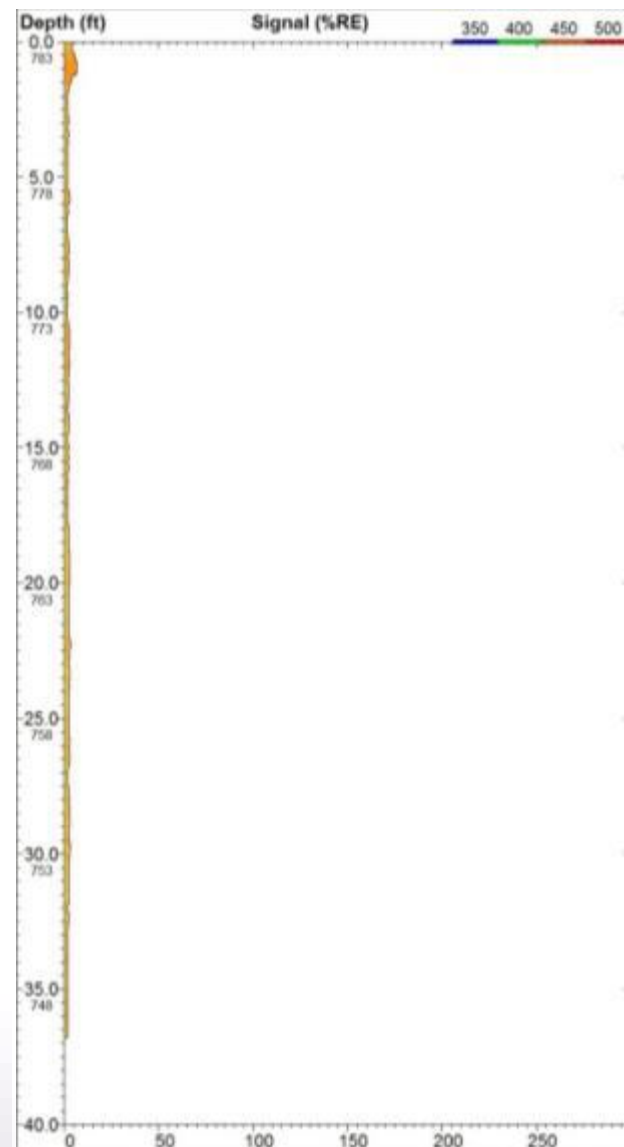
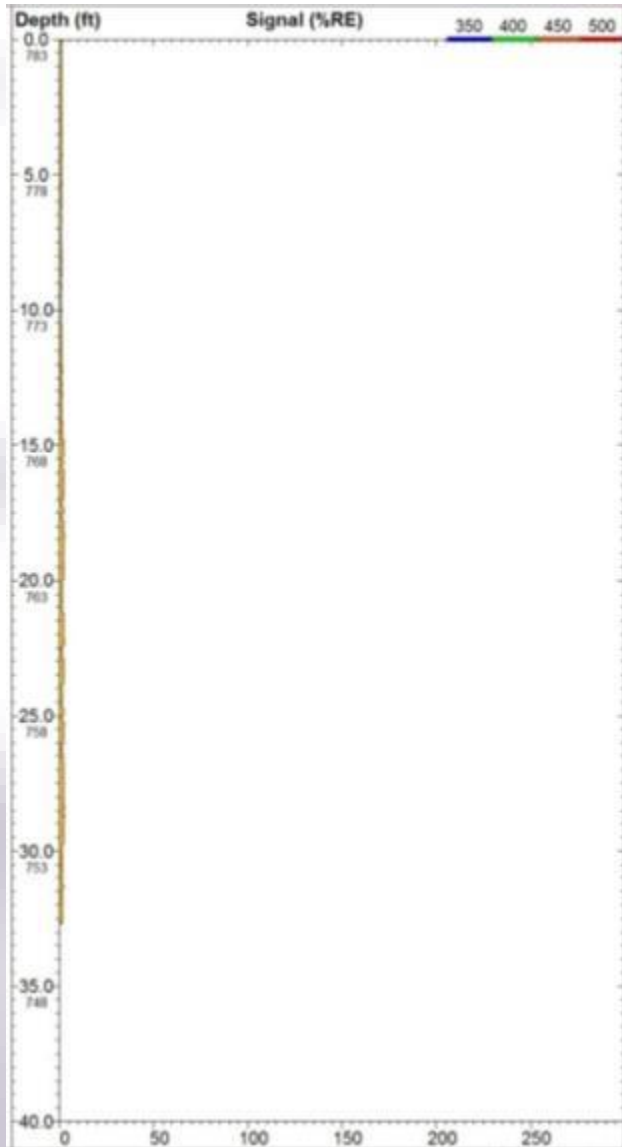
- Laser Induce Fluorescence
 - Free Product (LNAPL)
- Membrane Interface Probe
 - Dissolved Phase (Plume)
- Injection Points
 - Oxidant Delivery



Case Study



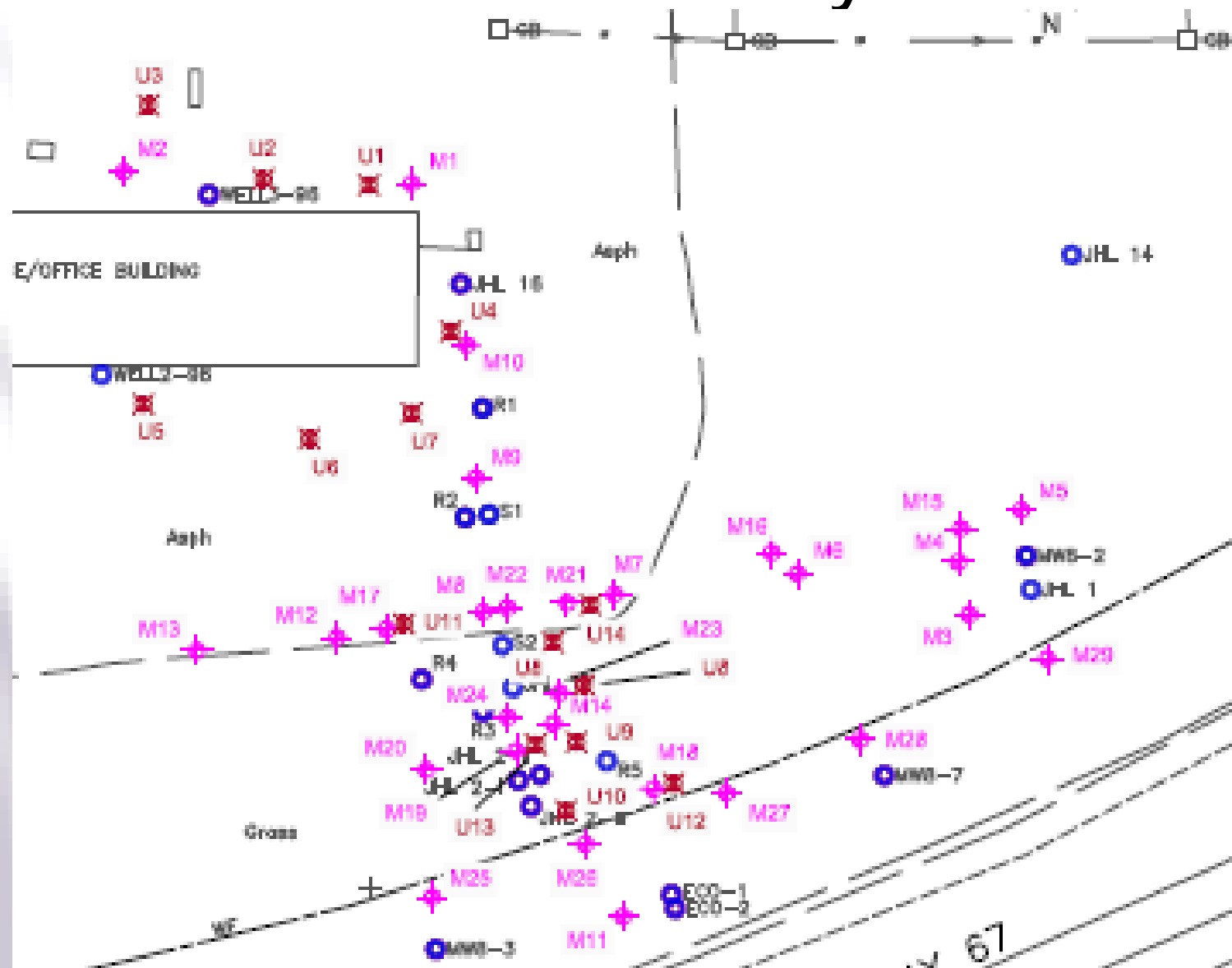
Case Study – LIF Results



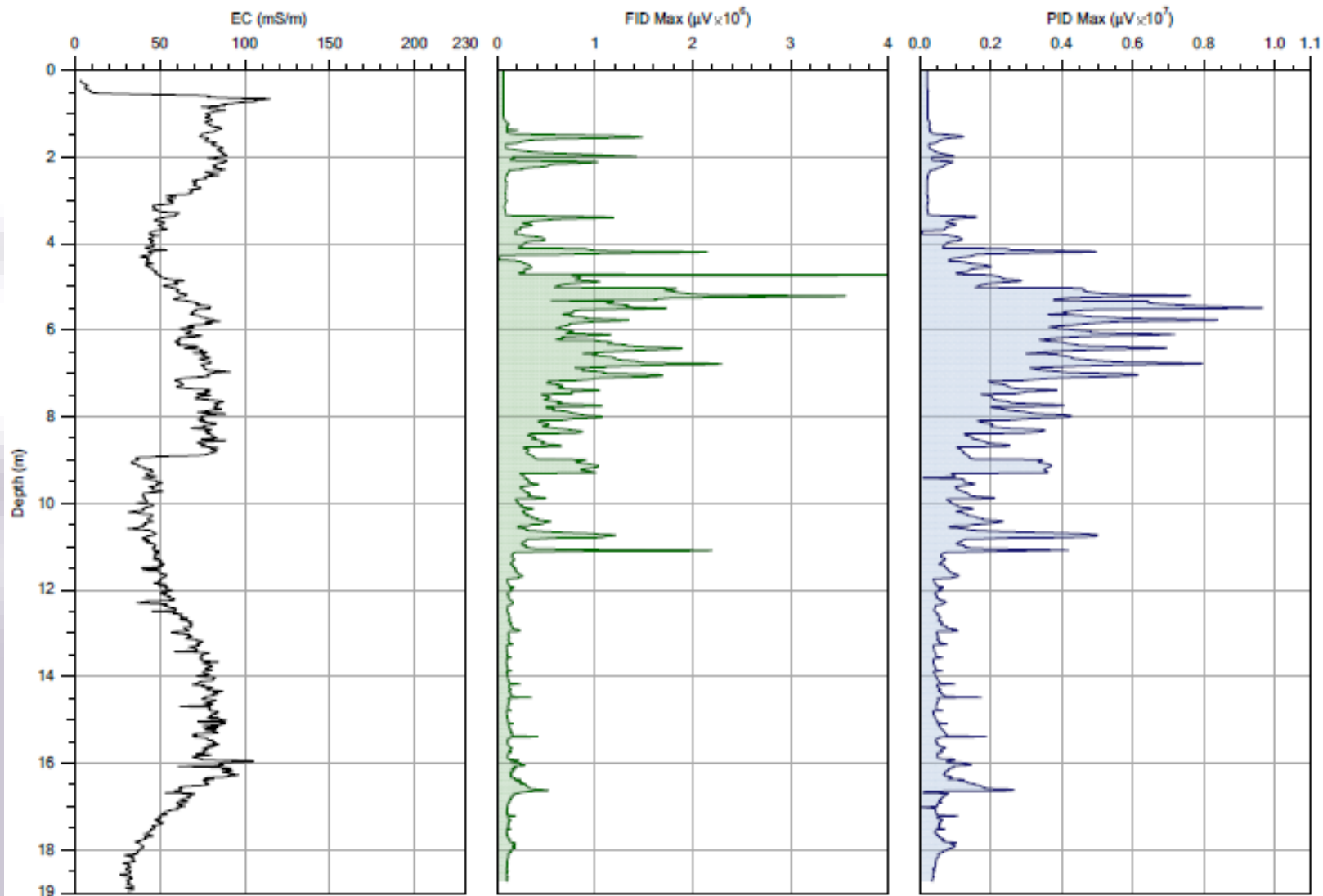
No NAPL
detected



Case Study



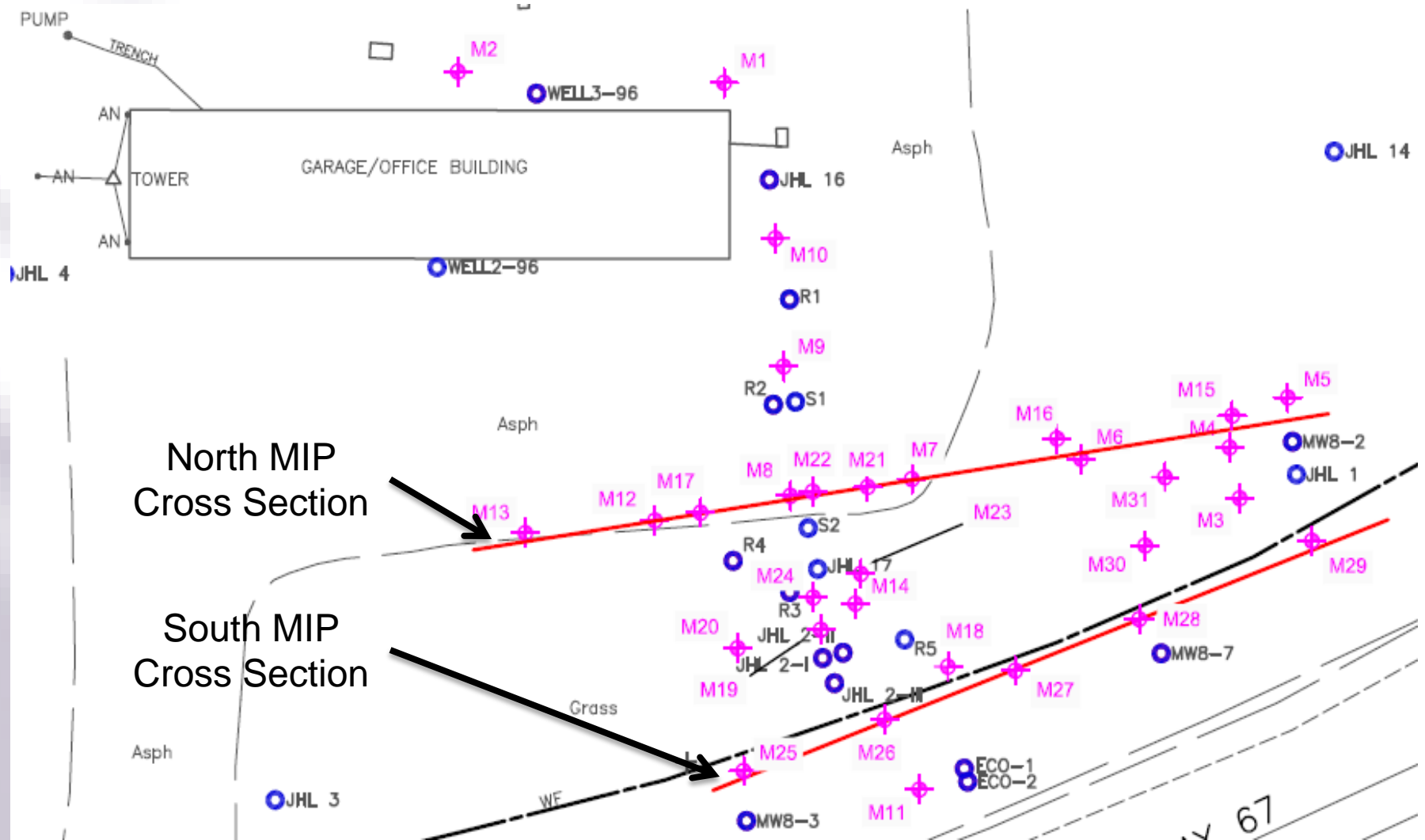
Case Site – MIP Results



↑
Dissolved
phase plume
2 m to 11 m
below
ground
↓

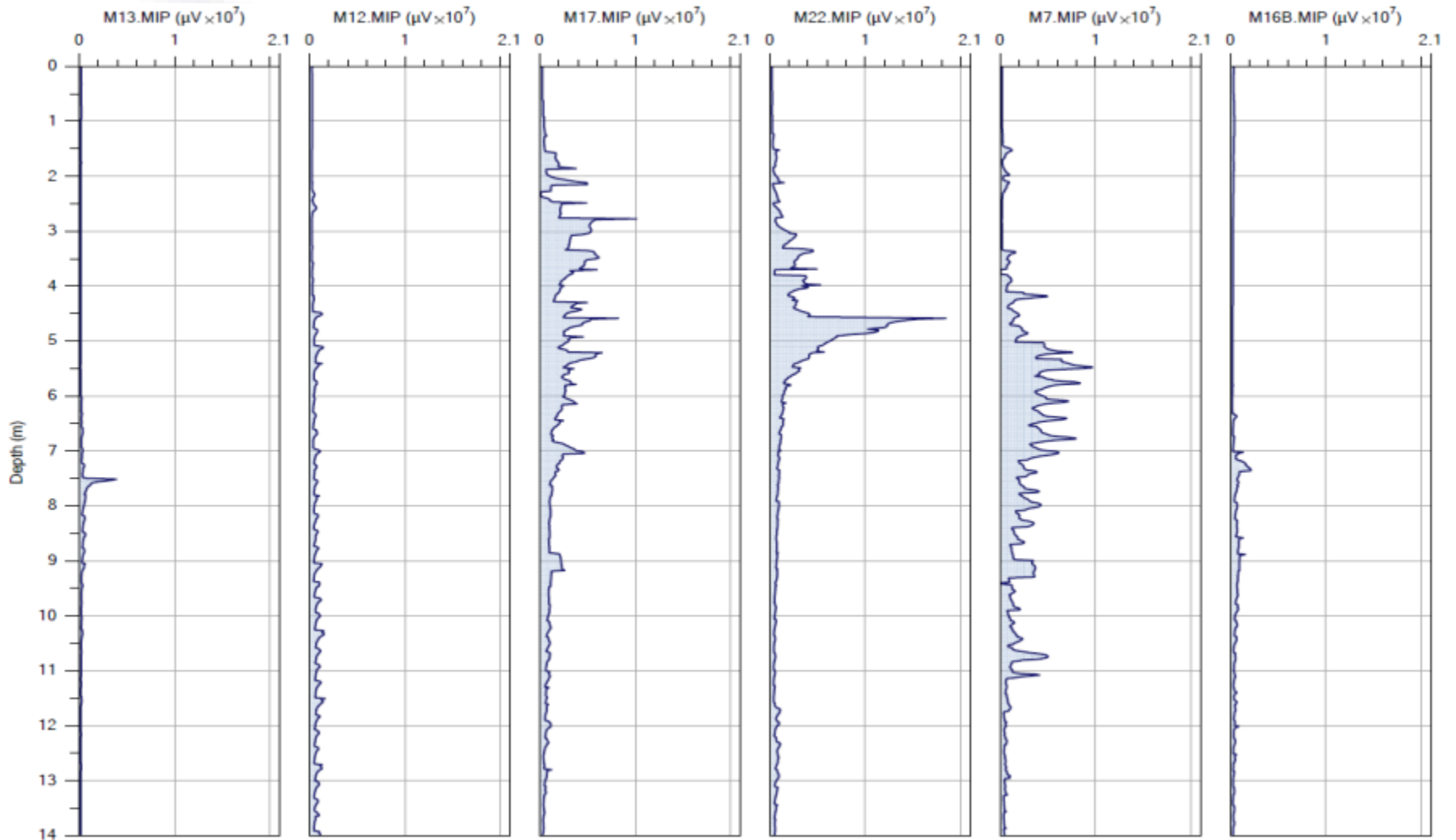


Case Site – Results



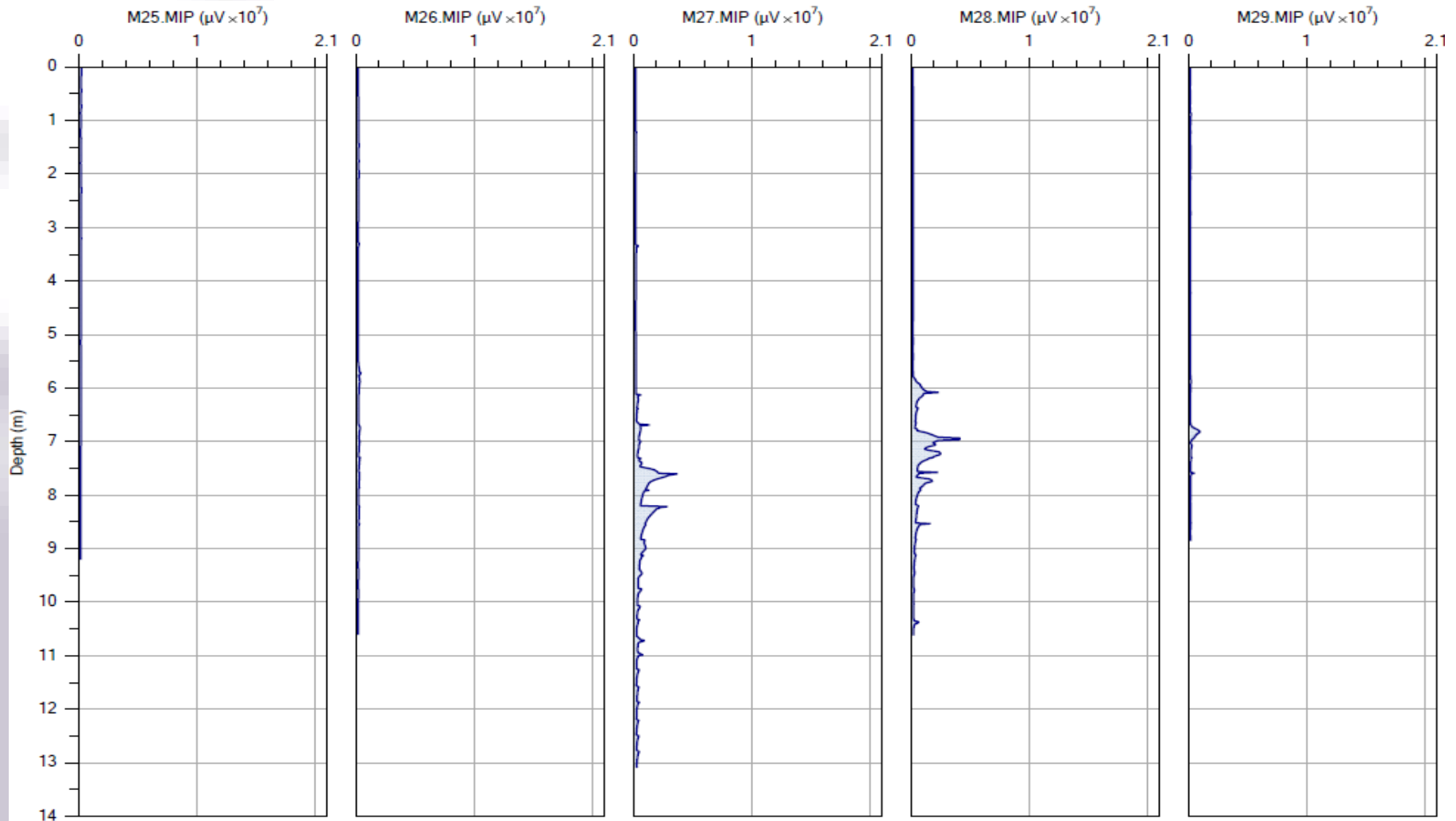
Case Site – Results

North MIP Cross Section



Case Site – Results

South MIP Cross Section



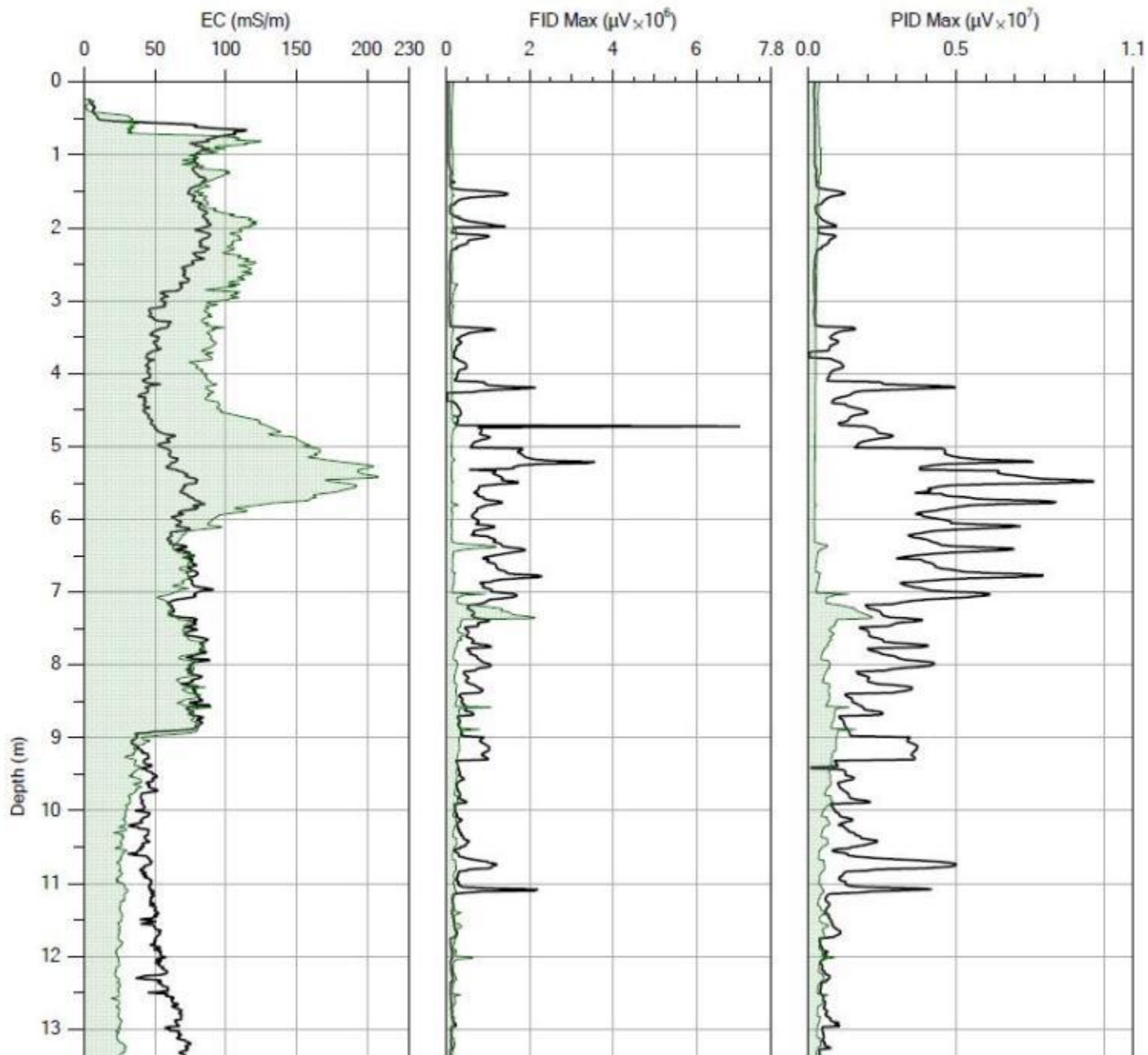
MIP Results

Before ISCO

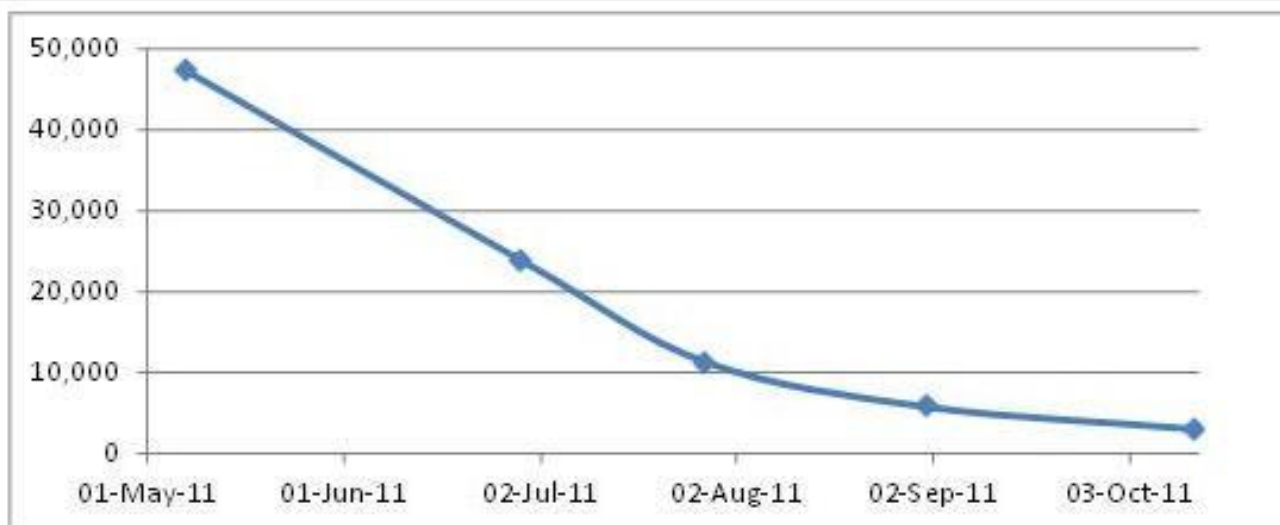
Black Line
May 13/2011

After ISCO

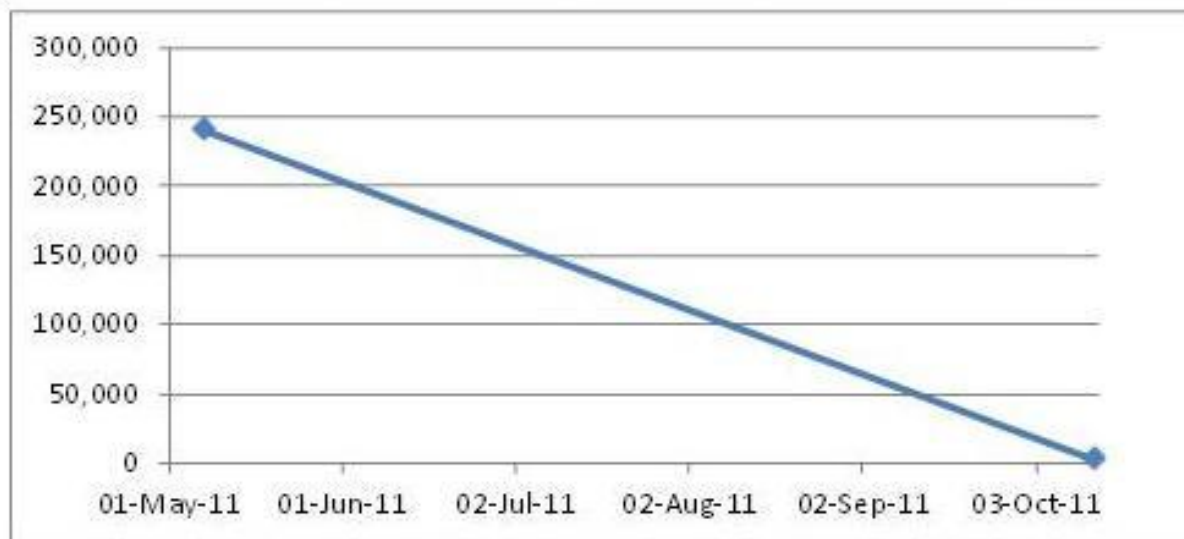
Green Line
Sept 27/2011



PARAMETER	MOE Groundwater Standards (ug/L)	JHL2-II (11 m, 35 ft)					% Reduction
		MMM	Vertex	Vertex	Vertex	MMM	
		07-May-11	29-Jun-11	28-Jul-11	01-Sep-11	13-Oct-11	
Benzene	5.0	9,500	4,400	1,100	800	1,100	88
Toluene	24	9,100	4,500	2,200	540	780	91
Ethylbenzene	2.4	690	160	85	-	46	93
Xylenes (total)	300	2,900	2,740	2,000	1,100	1,100	62
PH(F1-C6 to C10)	750	25,000	12,000	5,900	3,300	ND	99.7
PH(F2-C10 to C16)	150	ND	-	-	ND	ND	-
PH(F3-C16 to C34)	500	ND	-	-	ND	ND	-
PH(F4-C34 to C50)	500	ND	-	-	ND	ND	-
Total BTEX/PHCs	-	47,190	23,800	11,285	5,740	3,026	94



PARAMETER	MOE Groundwater Standards (ug/L)	R3		
		MMM	MMM	% Reduction
		07-May-11	13-Oct-11	
Benzene	5.0	2,100	170	92
Toluene	24	15,000	370	98
Ethylbenzene	2.4	2,300	ND	99.98
Xylenes (total)	300	15,000	1,100	93
PH(F1-C6 to C10)	750	140,000	1,100	99.2
PH(F2-C10 to C16)	150	61,000	160	99.7
PH(F3-C16 to C34)	500	4,300	ND	98.8
PH(F4-C34 to C50)	500	790	ND	94
Total BTEX/PHCs	-	240,490	2,900	98.8



Case Study

- Average PHC groundwater reduction across Site = 96%
- LIF: safe commencement of ISCO
- MIP: ISCO re-design
- Zones of high impacts targeted, resulting in excellent treatment

