

# **An investigation into the treatment of co-mingled saline-hydrocarbon affected soil and groundwater**

by:

**Gordon H. Bures, M.Eng., P.Eng.**  
**Frac Rite Environmental Ltd.**

**Dr. Edwin Liem, Ph.D.,**  
**Black Earth Humates Ltd.**



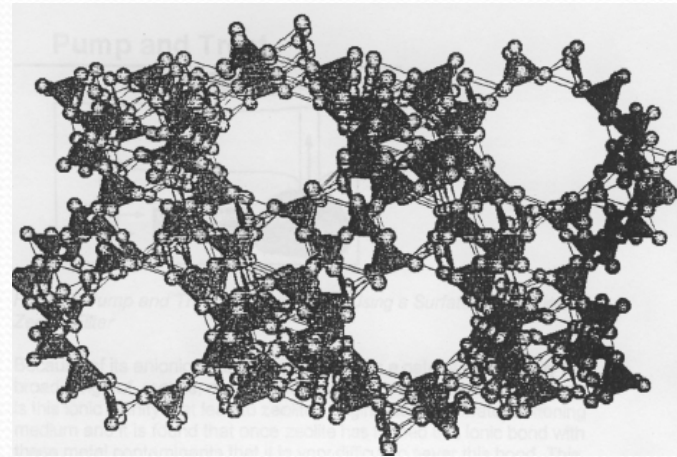
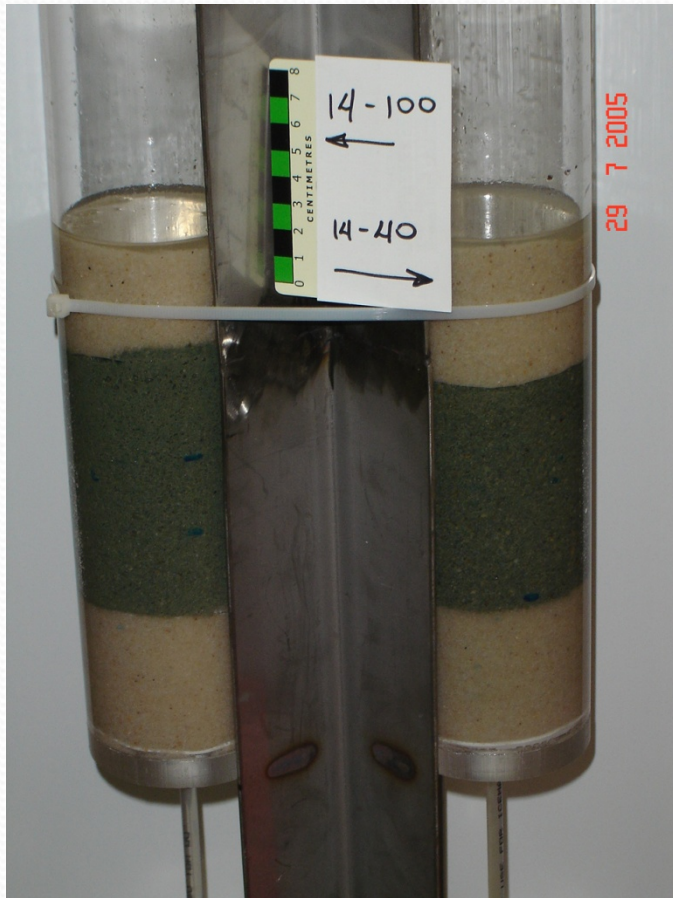
FRAC RITE ENVIRONMENTAL LTD.

# Outline

- What are Zeolites and Humates?
- Existing environmental applications
- Rationale for column testing
- Column testing and results
- Conceptual field pilot application



# Zeolites



- Naturally occurring, crystalline, porous aluminosilicate structure
- Anionic framework, high CEC
- Open structure, molecular “sieve” - traps contaminants

# Zeolite Column Test Results: 2005

TABLE 3 –  
Column Test Salinity Parameters

PORE VOLUMES:	COLUMN 1: DISCHARGE WATER SALINITY (-14 + 100 MESH ZEOLITE)									
	Cl (mg/L)	EC (uS/cm)	SO <sub>4</sub> (mg/L)	TDS (mg/L)	pH	Ca (mg/L)	K (mg/L)	Mg (mg/L)	Na (mg/L)	SAR
PRE-TEST	9830	19500	439	13600	7.0	2430	66.0	579	2410	11.4
0-2	8920	17500	378	12200	6.9	3230	280.0	496	1080	4.7
20-22	9560	19000	405	13300	7.2	2430	232.0	565	2310	11.0
40-42	9660	19300	395	13500	7.2	2380	168.0	565	2350	11.2
60-62	9770	19900	401	14000	7.3	2430	129.0	571	2400	11.4
80-82	9670	19400	373	13600	7.4	2320	90.4	547	2280	11.1
100-102	9840	19500	477	13600	7.5	2460	95.6	578	2440	11.5
120-122	9910	19700	412	13800	7.6	2580	96.0	602	2540	11.7
PORE VOLUMES:	COLUMN 2: DISCHARGE WATER SALINITY (-14 + 40 MESH ZEOLITE)									
	Cl (mg/L)	EC (uS/cm)	SO <sub>4</sub> (mg/L)	TDS (mg/L)	pH	Ca (mg/L)	K (mg/L)	Mg (mg/L)	Na (mg/L)	SAR
PRE-TEST	9830	19500	439	13600	7.0	2430	66.0	579	2410	11.4
0-2	8200	17000	321	11900	7.0	2860	239.0	436	1200	5.5
20-22	9460	19300	386	13500	7.2	2490	218.0	578	2340	11.0
40-42	9610	19500	383	13600	7.3	2450	156.0	570	2370	11.2
60-62	9580	19300	386	13500	7.5	2250	129.0	533	2230	11.0
80-82	9640	19500	410	13600	7.4	2350	122.0	562	2310	11.2
100-102	9570	19800	374	13800	7.5	2390	101.0	560	2360	11.3
120-122	9680	19400	400	13600	7.5	2400	98.5	562	2360	11.3
140-142	9790	19500	394	13600	7.6	2570	114.0	609	2550	11.7
POST TEST ZEOLITE:	ZEOLITE SALINITY AFTER TESTING									
	Cl (mg/L)	EC (dS/m)	SO <sub>4</sub> (mg/L)	SPECIFIC GRAVITY	pH	Ca (mg/L)	K (mg/L)	Mg (mg/L)	Na (mg/L)	SAR
US MESH:										
-14 + 100	6070	17.9	334	1.49	7.5	1920	71	176	1980	11.6
-14 + 40	4070	13.0	226	1.25	7.7	1120	60	96	1490	11.5

Note: Cl – chloride; EC – Electrical Conductivity; SO<sub>4</sub> – sulphate; TDS – Total Dissolved Solids; Ca – calcium; K – potassium; Mg – magnesium; Na – sodium; SAR – Sodium Adsorption Ratio.

TABLE 1 –  
Column Test Specifications

Parameter	Specifications (actual)	
	-14 + 100 zeolite test column	- 14 + 40 zeolite test column
Dry weight	324 g	300 g
Wet weight	330 g	330 g
Pore volume	249 cm <sup>3</sup>	209 cm <sup>3</sup>
Sample dimensions (saturated)	101 mm Height x 67.5 mm Diameter	88 mm Height x 67.5 mm Diameter
Consolidation Pressure	60 kPa	60 kPa
Water feed rate	10 ml/min ± 15%	10 ml/min ± 15%
Column water head	1 m ± 10%	1 m ± 10%
Sampling frequency	7.60 hrs	6.25 hrs
No. of samples taken	7	8
Total test volume	30.4 litres	29.7 litres
Test duration	53.2 hrs	50 hrs

- High permeability:  
 $K = 10^{-4}$  m/s
- Na reduced 60%
- Cl reduced 15%
- SAR reduced from 11.4 to 4.7



# Humates



**HUMATES:** Organic material formed from the decay of plant and animal residues. Large organic molecules with many functional groups.

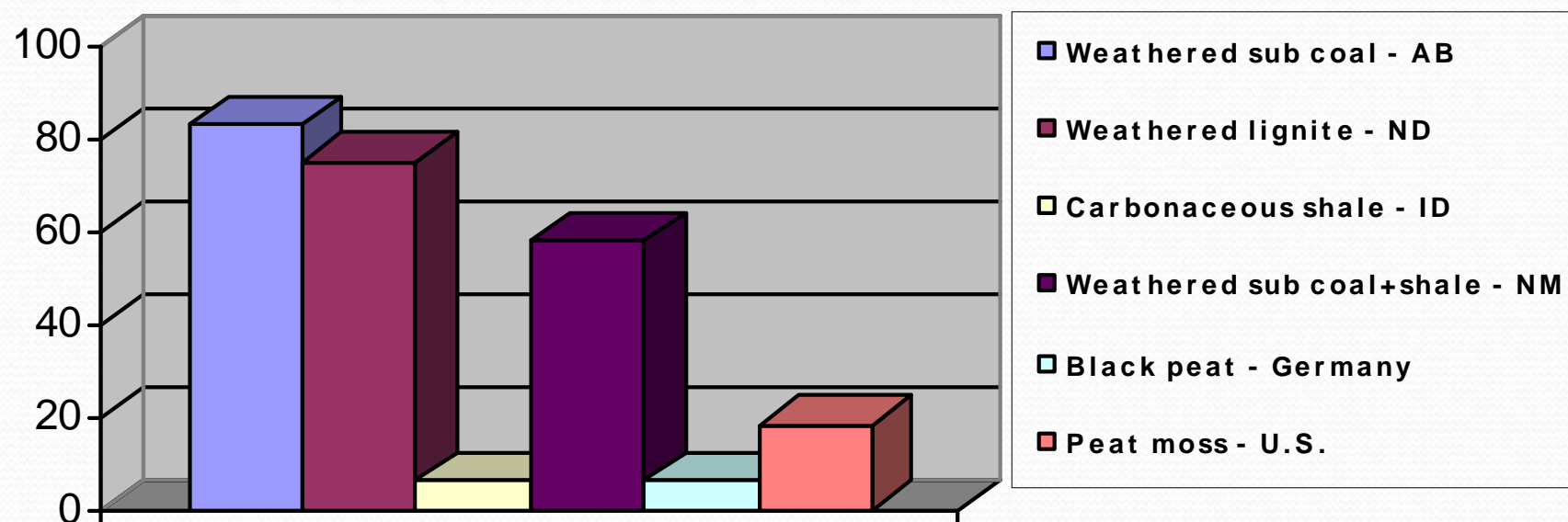
- Comprised of 90% C and O
- 10% H, N, S and trace elements
- Two main components of Humates are HUMIC and FULVIC acid

- **ATTRIBUTES:**

- Chelator to fixate sodium
- Increases solubility of PHCs
- Provides carbon for Microbes to convert PHCs to acids and sugars
- Cation and water retention

# HUMATE COMPARISONS

## % Total Humic Acids - Dry Matter



----. 1998, 2000, 2004, 2006, 2008, 2011. Lab Reports. A&L Western Laboratories, Modesto, CA. 6 pages.

Hoffman, G. K. et al. 1994. Overview of Humate Production in North America. (In) Proceedings 30th Forum on the Geology of Industrial Minerals. 54 – 70.





# Present Remedial Applications

- Mine tailings remediation for metals reduction (humates)
- Treatment of co-produced saline formation waters in oil and gas industry (zeolites)
- Phytoremediation (humates and zeolites)
- Treatment of saline-hydrocarbon drilling mud returns (humates and zeolites)
- Treatment of sodic soils in dry land farming (humates)



# Rationale for Treatability Testing

Zeolites/humates have been used in reclamation to remediate surficial soils, but ...



There isn't yet a commonly used *in situ* approach to cost-effectively treat deeper impacted subsoils.

- Co-mingled saline-hydrocarbon impacts reside at many existing and former facilities for which a passive and cost-effective *in situ* remedial approach is needed, especially in deeper subsoils.
- This approach needs to address both soil and groundwater impacts



# Humate and Zeolite Column Testing

- Concept evolved from 2005 zeolite column study on sodic/saline impacted groundwater to include treatment of co-mingled petroleum hydrocarbons
- IRAP supported column study was prepared in 2009-2010 to study potential for using combination of humates and zeolites to remediate petroleum hydrocarbons concurrently with salts through physical attenuation mechanisms
- Humates also facilitate biodegradation of PHCs, which was not examined in column study

# Test Objectives

The specific objectives of the laboratory testing program were to evaluate the performance of the combination of amendments with respect to:

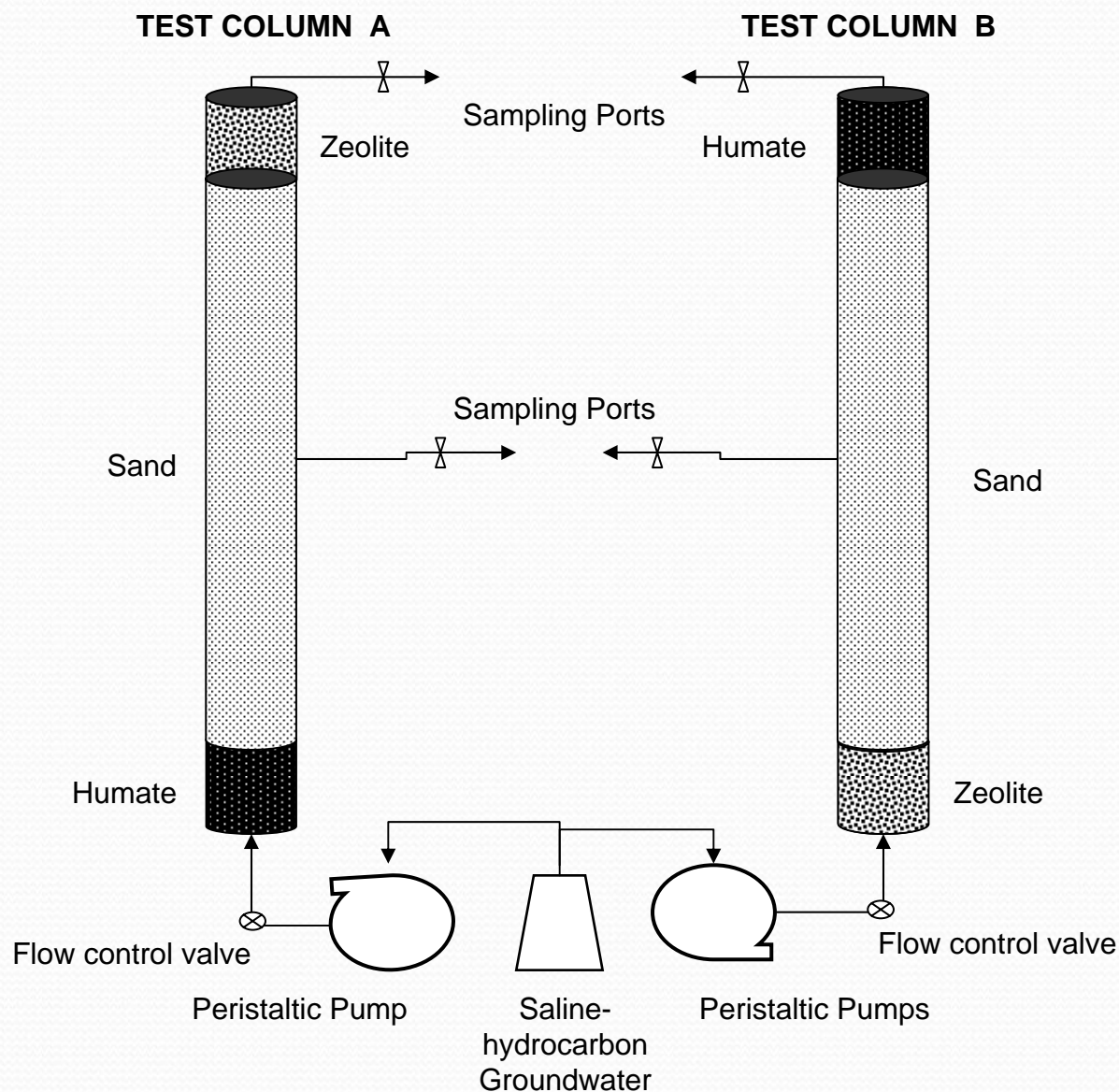
- Reduction in salinity indicator parameters;
- Reduction in sodium and chloride concentrations;
- Reduction petroleum hydrocarbon constituents (i.e. BTEX, F1, F2);
- Remediation effectiveness relative to the *sequence of treatment* in the test columns;
- Sodium and salinity retention as measured by Sodium Adsorption Ratio (SAR) and Electrical Conductivity, respectively; and,
- Efficacy for implementation of zeolites and humates in the field for passive treatment of saline-hydrocarbon impacted soil and groundwater.



# Configuration of Test Columns

**Column Test Specifications**

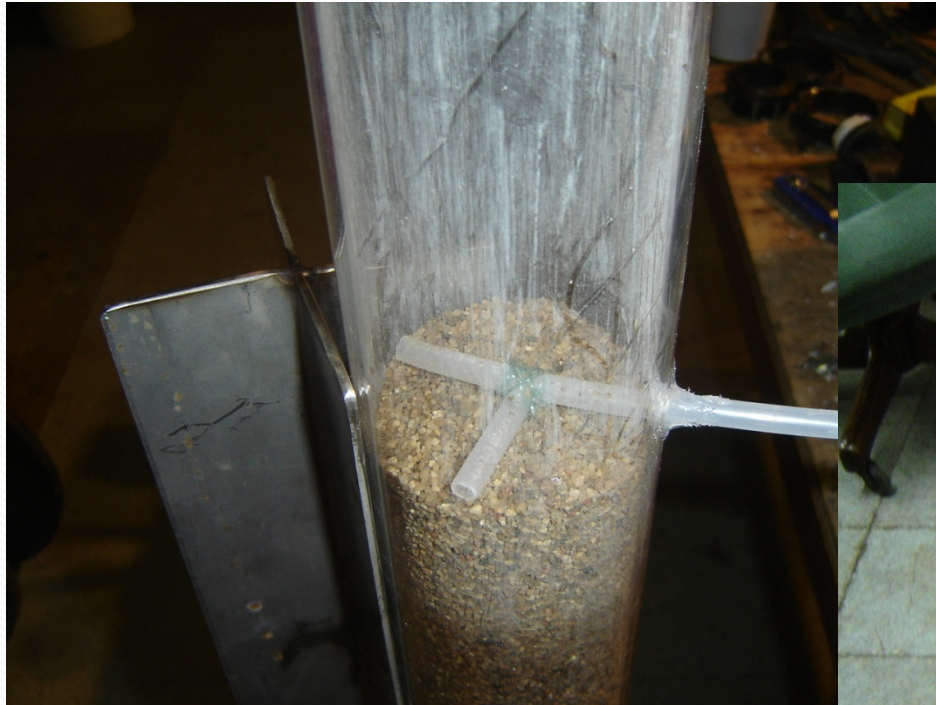
Parameter	Specifications (actual)	
	Column A	Column B
Sand weight	3,654.9 g	3,674.5 g
Humates weight	959.0 g	1,015.2 g
Zeolites weight	1,212.9 g	1,137.1 g
Configuration bottom-top	Humates - Zeolites	Zeolites - Humates
Humate Dimensions	300 mm Height x 68.5 mm Diameter	320 mm Height x 68.5 mm Diameter
Zeolite Dimensions	320 mm Height x 68.5 mm Diameter	300 mm Height x 68.5 mm Diameter
Total column pore volume	2,000 cm <sup>3</sup>	2,000 cm <sup>3</sup>
Flow rate	50 ml/min	Variable: 5-59ml/min
Sampling frequency	40min	variable
No. of samples taken	11 water, 2 "soil"	8 water, 2 "soil"
Total test volume	13 litres	14 litres
Test duration	4:41	16:47



FRAC RITE ENVIRONMENTAL LTD.



# Column Construction and Apparatus



**Mid – column sampling port construction**

**Peristaltic flow pumps and saline groundwater source**





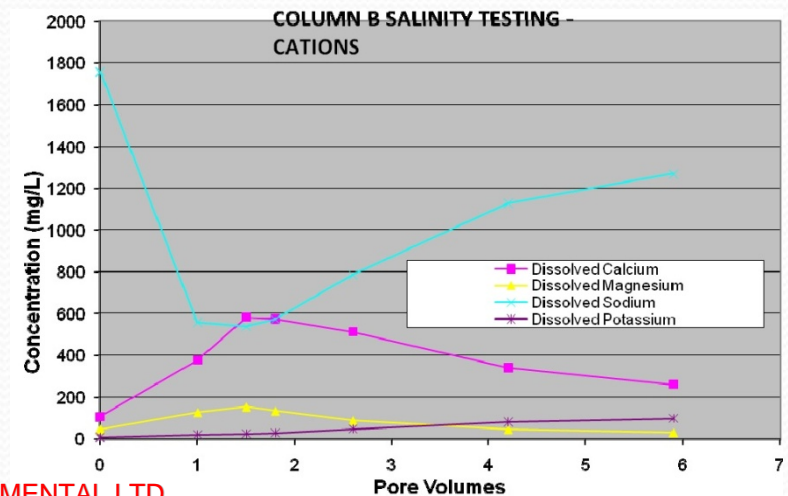
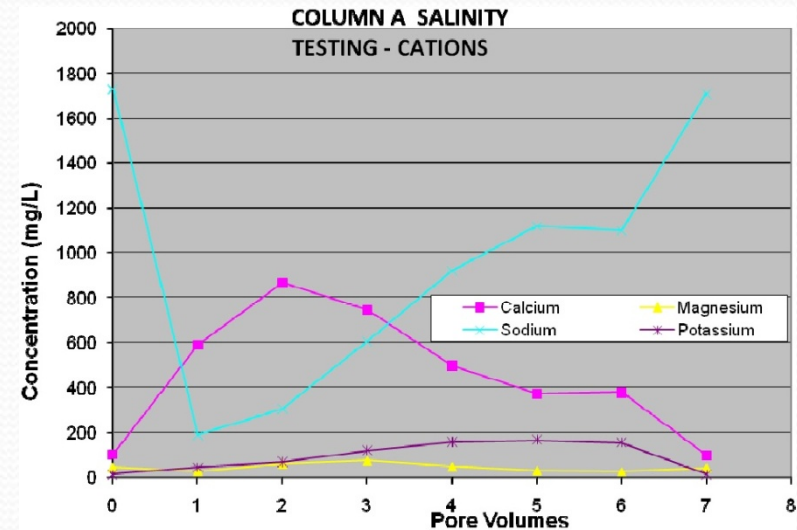
# Geotechnical Properties

<i>Test Parameter</i>	<i>Geotechnical Properties</i>		
	<b>Humate</b> -10+40 mesh “leonardite”	<b>Zeolite</b> -14+40 mesh “clinoptilolite”	<b>Silica Sand</b> -16+40 mesh
Specific gravity	1.3	2.3	2.6
Porosity (%)	29	58	36

# Results – Sodium Reduction

Maximum reduction in sodium concentration after 2-3 pore volumes:

- Cation exchange manifested as substitution of calcium for sodium in zeolite lattice structure
- Column A (Humate-Zeolite) treatment sequence had greatest sodium removal (89%)

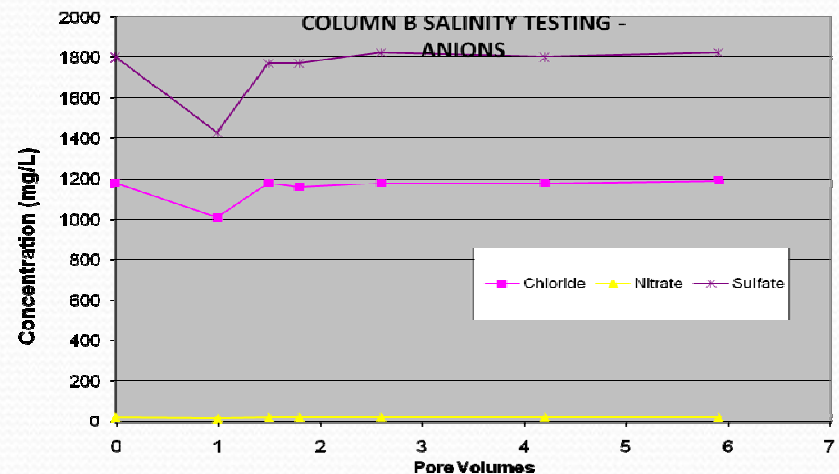
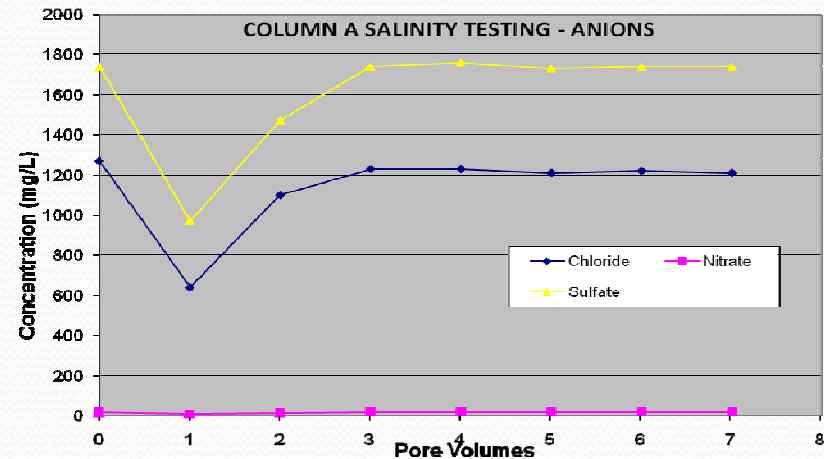




# Results – Chloride Reduction

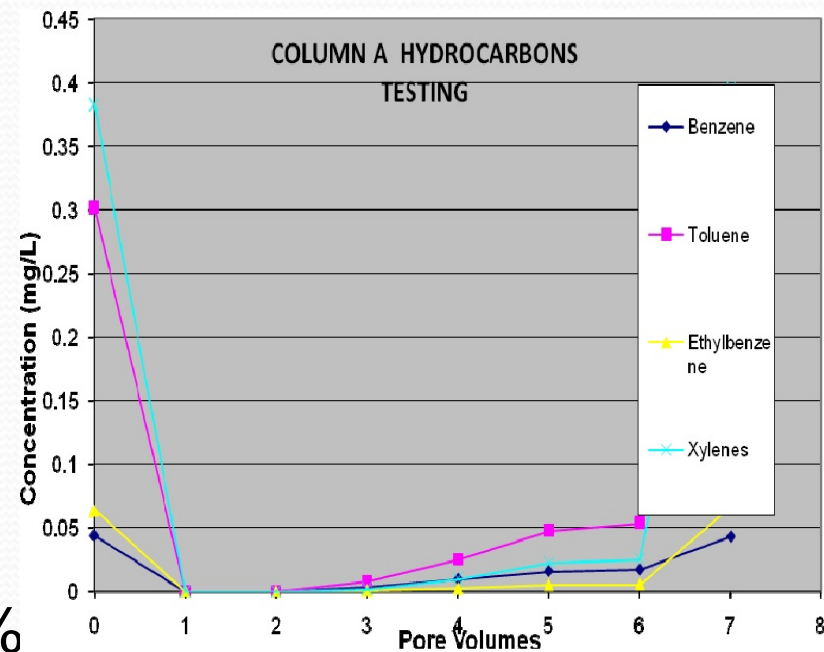
Maximum reduction in chloride concentration after 1 pore volume:

- Column A (Humate-Zeolite) treatment sequence had greatest chloride removal (50%)
- 35% of chloride attenuation attributable to humates, and 15% to zeolites



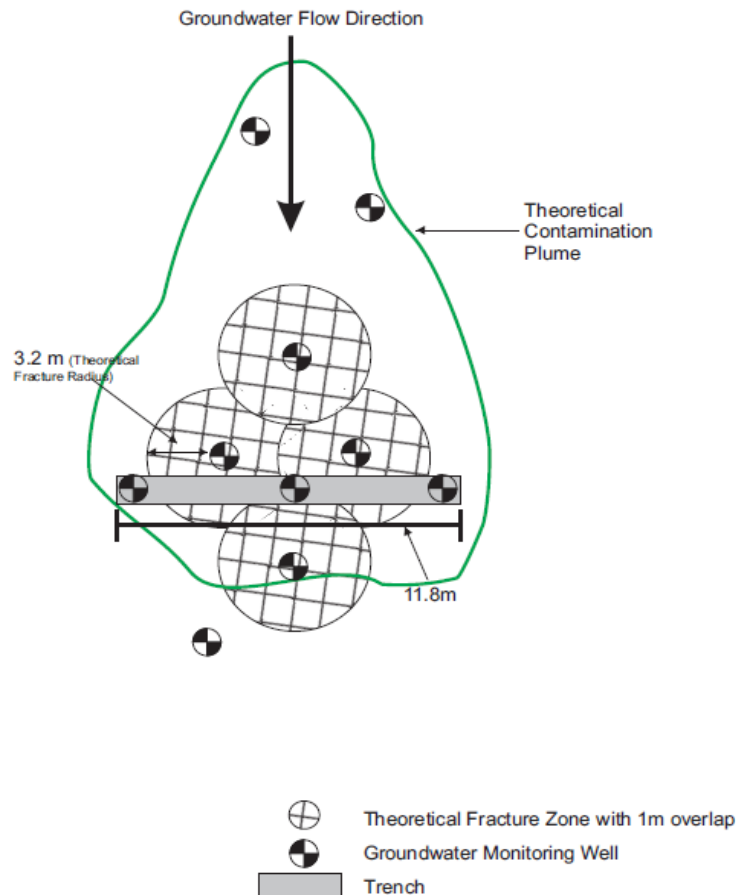
# Results – BTEX Reduction

- Significant reduction in BTEX concentration after 2-6 pore volumes (Column A)
- Concentrations of BTEX in effluent stream reduced to ND concentrations at 1 – 2 pore volumes
- Humates responsible for  $\geq 87\%$  of BTEX adsorption





# Conceptual Salt – Hydrocarbon Treatment Barrier



- Hydraulic Fractures create a network of interconnected sand drainage pathways to direct flow in a low permeability soil towards a zeolite/humate treatment trench.
- System can be easily scaled up to include captive deionizations or reverse osmosis.

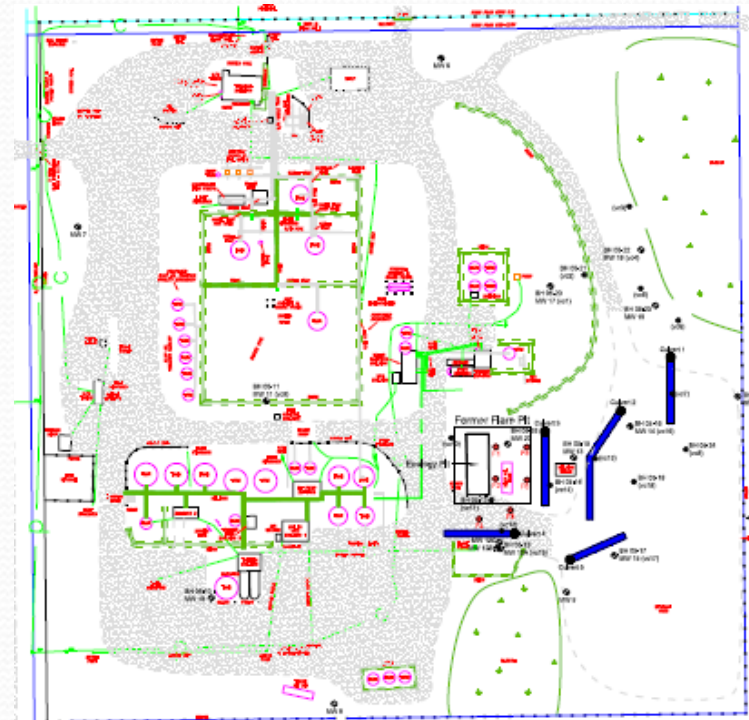
(2012 IRAP Phase 2 Field Pilot)

# Salt Remediation using Interceptor Trenches



## Objective:

Create a network of sand drainage fractures to connect with interceptor trenches for enhancing salt capture and treatment using reverse osmosis.





# Conclusions

- Most effective treatment results for salinity and petroleum hydrocarbon constituents were in Column A, i.e. humate treatment followed by zeolite treatment.
- Chloride reduction also noted in this configuration, possibly attributable to attenuation of chloride compounds in solution.
- Column test results are conservative in that they don't consider bioremediation processes
- Column test results can be used in design of treatment trench to conduct a field pilot