



CREATING AND DELIVERING BETTER SOLUTIONS

Sulfolane Impacted Soil and Groundwater Treatability Study

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Outline

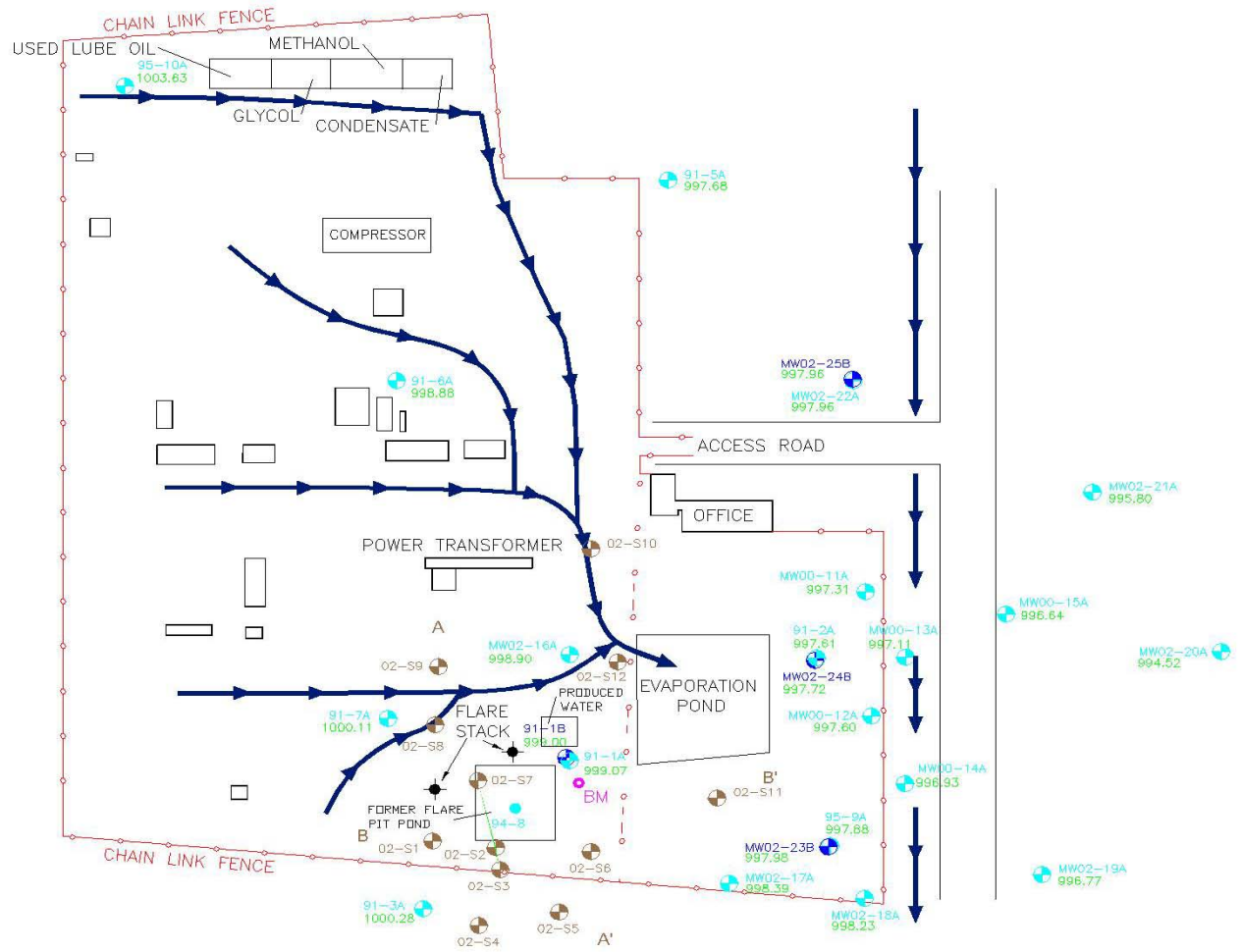
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2. Soil and Groundwater Quality Monitoring Results
3. Sulfolane – Sulfinol[®]
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Introduction

- Active Sour Gas Plants
- Gas Sweetening
- Gas Dehydration
- Treatment

Products : Condensate and Natural Gas

Site Plan



Physical Settings

Topography and Drainage:

- Porcupine Hills (1,600 masl).

Geology:

- Fill/gravel 0.5 m.
- Silty Clay Till – 2.5 m to 9.2 m, sand lenses.

Hydrogeology:

- Shallow water table at 1.7 m to 3.4 m.
- Groundwater flows towards east.
- Bulk hydraulic conductivity 1.02 E-5 cm/s to 1.41 E-7 cm/s .

Soil and Groundwater Quality

Under the Alberta Environment Approval to Operate:

- Soil Management Program – Alberta Environment Soil Monitoring Directive (1996) – every five years.
- Groundwater Quality Monitoring Program initiated in 1991 – annually.

Soil

- Elevated concentrations of petroleum hydrocarbons (PHCs) detected – benzene, toluene, ethylbenzene, and xylenes (BTEX) and F₁, F₂, and F₃.
- Elevated chloride concentrations detected.

Groundwater

Historically elevated concentrations of:

- PHC – BTEX and F₁ and F₂.
- Dissolved Organic Carbon (DOC).
- Chlorides.
- Sulphate (naturally occurring).
- Nitrites and Nitrates.
- Total Dissolved Solids (TDS).

The DOC and PHC concentrations do not correlate.

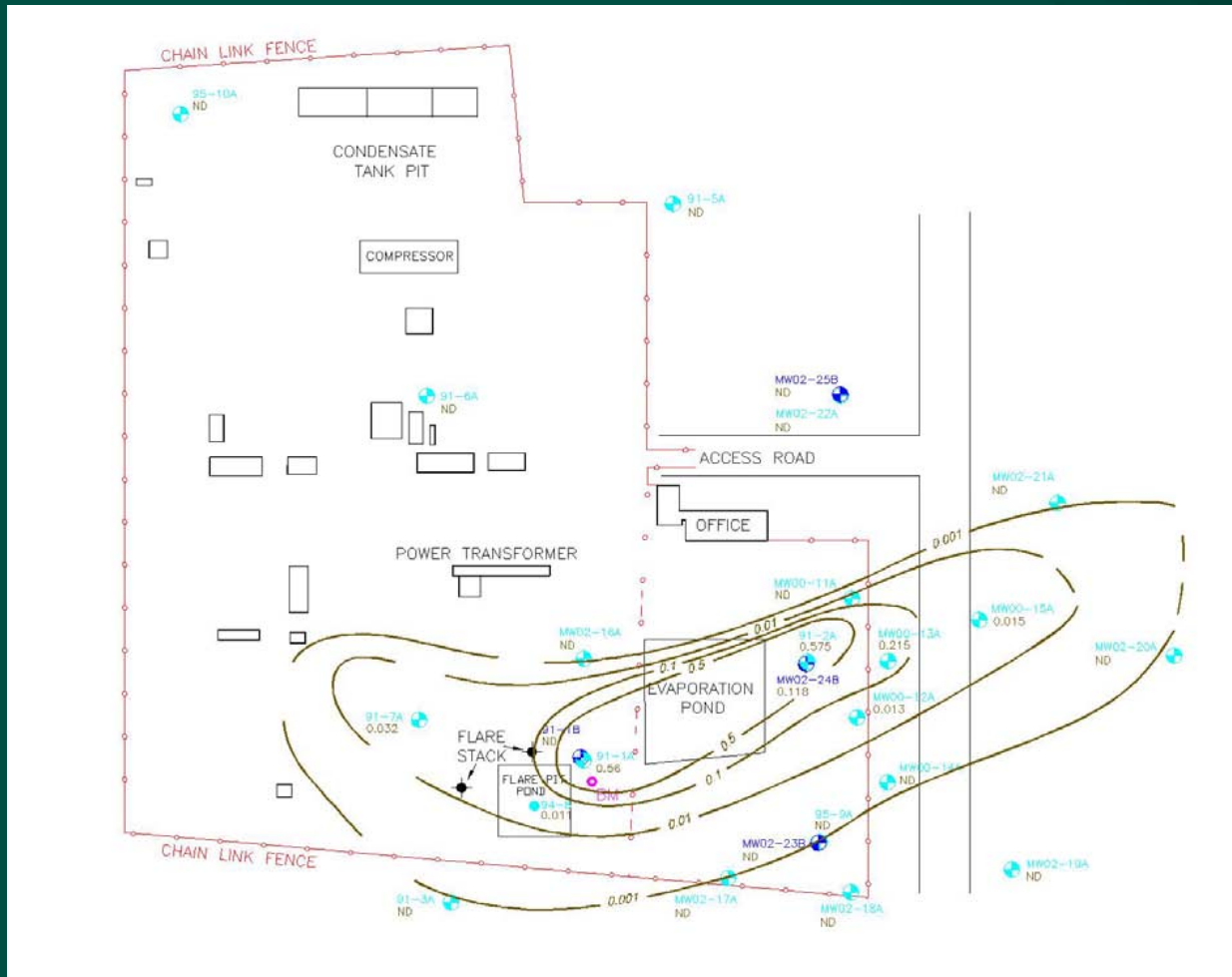
Assessment

- Resolution of elevated concentrations of DOC initiated the plant site product inventory review.
- Product review detected that, in 1992, existing amine process replaced Sulfinol process.
- Sulfinol process was mixture of Sulfinol[®] and DIPA.
- In 2000, Sulfolane analyses were added to the regular annual groundwater quality monitoring.

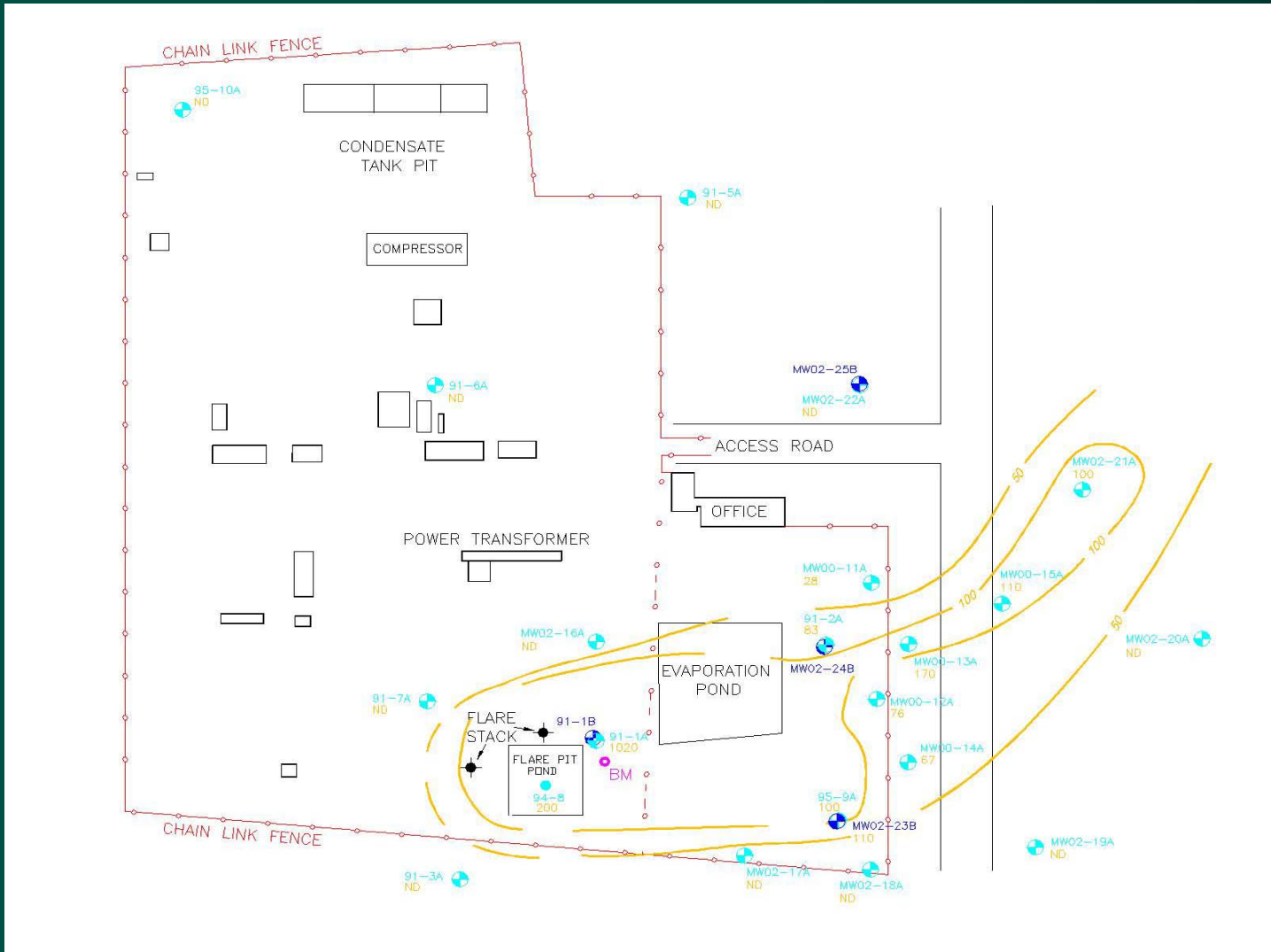
Assessment continued...

- Half the monitoring wells returned concentrations of sulfolane greater than background concentrations.
- In 2002, soil and groundwater with elevated concentrations of PHC and Sulfolane were partially delineated in the vicinity of the former flare pit and east of the Evaporation Pond.
- Sulfolane was detected in all collected soil samples in the vicinity of former flare pit.

Benzene Concentration Distribution Map 2002



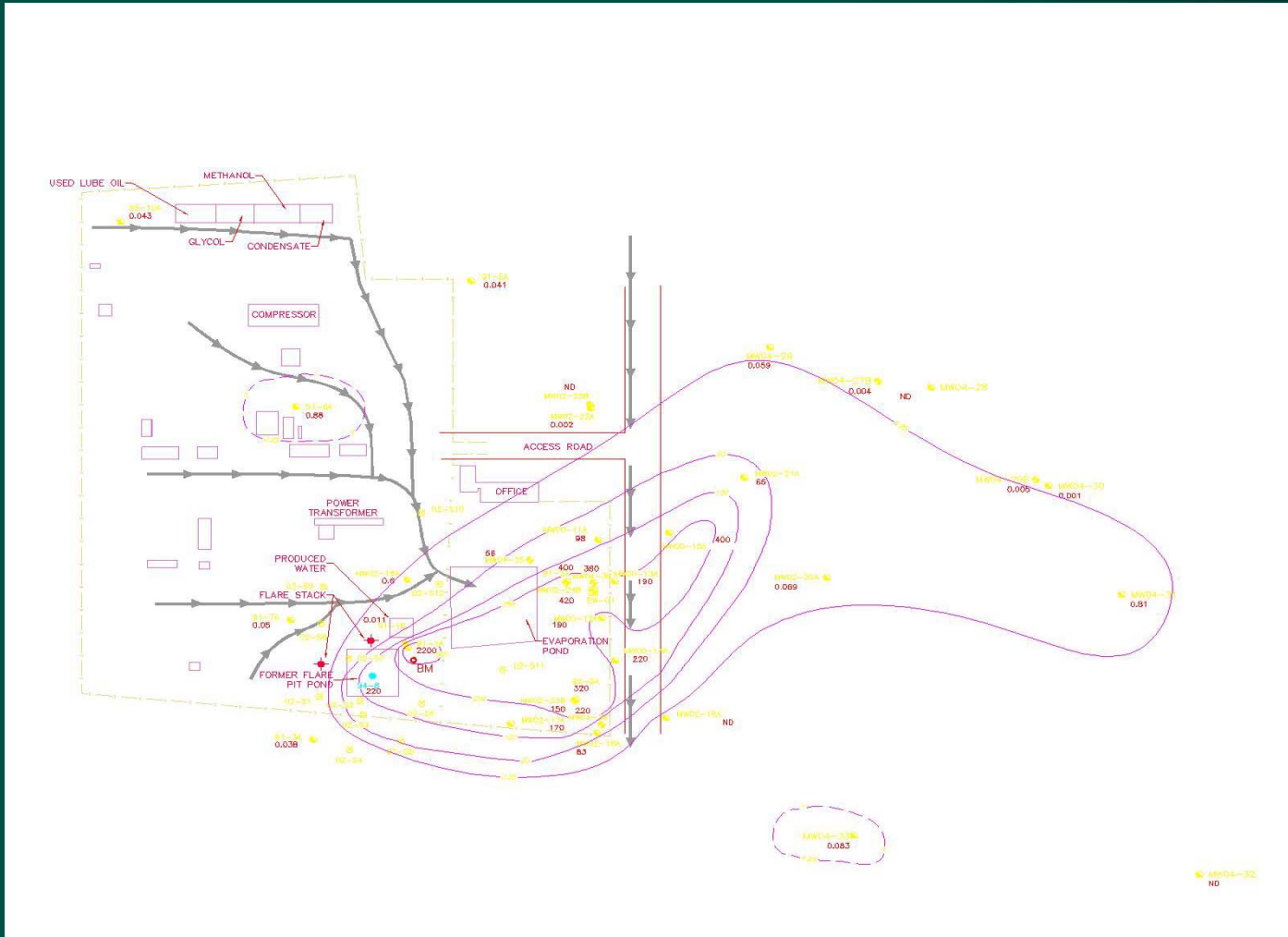
Sulfolane Concentration Distribution Map 2002



Assessment continued...

- In 2004, additional groundwater quality monitoring wells were installed to further delineate the groundwater with concentrations of sulfolane greater than background concentrations.
- One groundwater extraction well and one observation well were installed to determine hydrogeological properties of a shallow water-bearing zone.

Sulfolane Concentration Distribution Map 2004



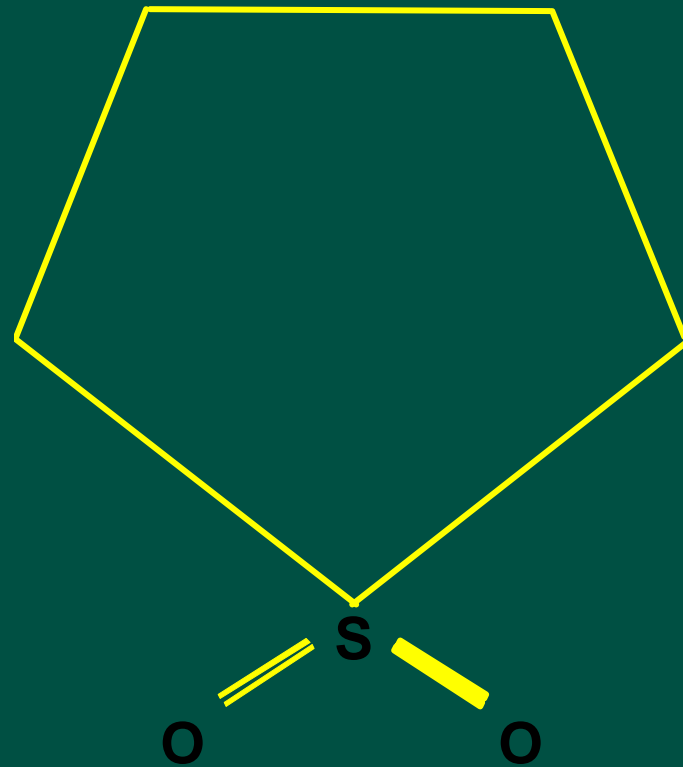
Assessment continued...

- In 2005, the Canadian Council of Ministers of the Environment (CCME) and Alberta Environment adopted the Canadian Association of Petroleum Producers (CAPP) guidelines for sulfolane: in soil (2.3 mg/kg) and groundwater (0.26 mg/L).

Sulfolane

- Sulfolane (C₄H₈O₂S) – common trade name for an organic chemical tetrahydrothiophene 1,1-dioxide, colourless, very polar, highly soluble in water, and extremely stable.
- Sulfinol[®] – solvent developed by Shell in early 1960s for extracting aromatics from hydrocarbons; second major application is in the process of 'sweetening' natural gas.
- Sulfinol[®] is slightly heavier than water (1,060 g/L).

Sulfolane



Laboratory Trials

Soil:

- Bio-treatability.

Groundwater:

- Bio-treatability.
- Chemical Oxidation (Mineralization).

Soil Trial

- Five soil samples collected.
- Sulfolane concentrations ranged from 350 mg/kg to 3,400 mg/kg.
- Two samples with highest concentrations of sulfolane were homogenized – 1,400 mg/kg.
- Samples for analytical analyses collected on days 0, 15, 30, 45, 63, and 78.

Soil Trial

7 Bio-reactors:

- Control – no additives.
- Sterile.
- Condensate impacted.
- Ammonia nitrogen (83 mg/kg).
- Ammonia nitrogen (232 mg/kg).
- Ammonia phosphate (83 mg/kg).
- Ammonia phosphate (232 mg/kg).

Oxygen concentrations maintained >10%.

Groundwater Trials

Bio-treatment:

- Aeration (GLR micro-bubbles[®]) and nutrient amendment.

Chemical Oxidation:

- H_2O_2 .
- UV radiation.
- Combination of both of the methods.

Groundwater Trials

Aeration Trial:

- 24-hour trial.
- Continuous aeration - $O_2 > 7.7$ mg/L.
- Nutrient - Ammonia phosphate.
- Samples collected at 0, 8, and 24 hours.

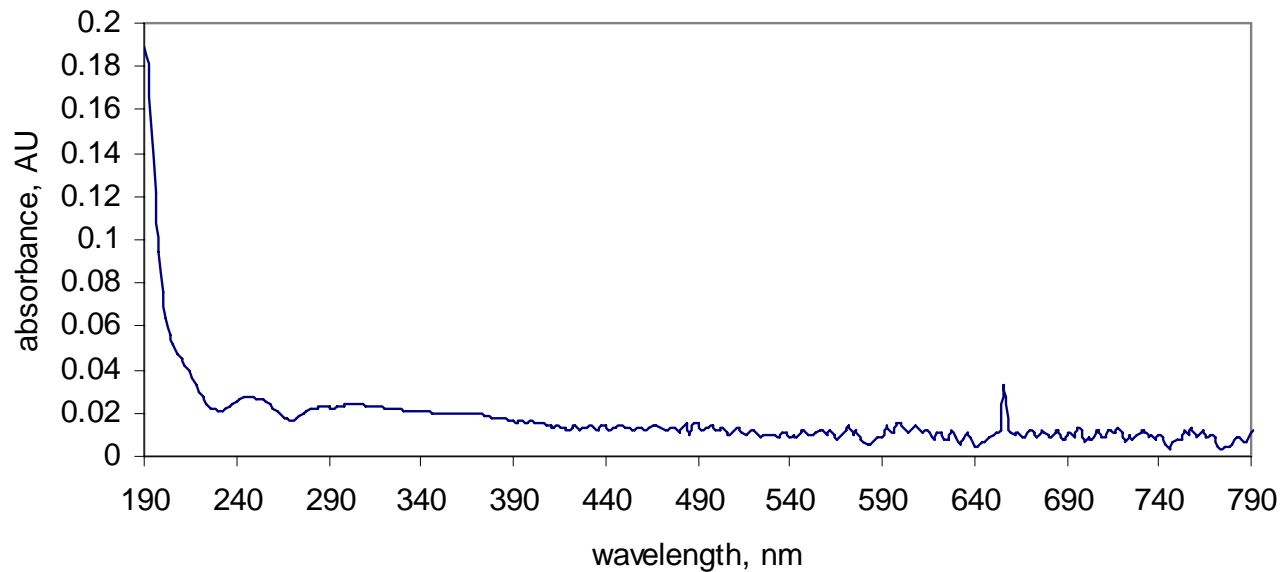
Groundwater Trials

Chemical Oxidation:

- 7-day trial.
- H_2O_2 concentration = 10 g/200 ml.
- UV radiation – 7 W UV bulb.
- $\text{C}_4\text{H}_8\text{O}_2\text{S} + 13\text{H}_2\text{O}_2 = 4\text{CO}_2 + \text{H}_2\text{SO}_2 + 16\text{H}_2\text{O}.$

Groundwater Trials

Spectrophotometric Evaluation of Light Absorbance by
Sulfolane ([c]=600 mg/L)



Results - Soil

			Ammonia	Nitrate	Ammonia	Phosphate
Days	Control	Sterile	83 mg/kg	232 mg/kg	83 mg/kg	232 mg/kg
0	1,300	1,400	1,700	1,700	1,400	1,400
15	1,700	1,600	1,500	1,500	1,500	1,400
78	1,100	1,200	860	760	2.1	2.9

Results - Groundwater

Sample Description	Sulfolane (mg/L)	Removal Ratio (%)	Time (hours)
Blank	1,200 (1,800)		
Chemical oxidation	950	79	
UV Irradiated sample	1,000	83	
Chem. Oxidation + UV	13	99	168
Aeration + Nutrients	490	73	24

Results - Groundwater

Energy consumed by the samples was calculated using the following formula:

- $P = E_{\text{photon}} * I/S = 0.2 \text{ m J/sec cm}^2$

Where:

- E_{photon} – energy of a single photon
- I – intensity of the photon flux in the system, and
- S – surface of irradiated vessel

Results - Groundwater

[Using the sun as source of UV radiation]

Based on commonly used potassium ferrioxalate system irradiation value, the estimated solar energy is:

$$E_{\text{solar}} = 3.5 \text{ mW/cm}^2$$

Therefore, a week of UV radiation in the lab will be equal to **10 hours** of sun exposure.

Conclusions

Soil:

- Sulfolane biodegradation is possible (bacteria require longer time to adjust to environment).
- After 78 days of incubation with ammonia phosphate fertilizer, the soil sample becomes non-toxic to Microtox[®].
- Full sulfolane biodegradation occurred with an ammonium phosphate fertilizer.
- No obvious production of toxic by-products.

Conclusions

Groundwater:

- The sulfolane degradation occurred under the chemo-physical and biodegradation (aeration) processes.
- The chemical oxidation and UV irradiation trial achieved 95% removal of sulfolane after one week of treatment.
- The biodegradation treatment achieved 73% of sulfolane removal after 24 hours.

Conclusions

Chemical Oxidation:

- If field scale treatment utilizes sunlight, 10 hours of the daily light will be required.

**Note: UV reactor should be considered as more effective method*

Biodegradation:

- Field scale treatment would require a water treatment and storage facility.

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