

# In-situ Flushing of Petroleum Contaminated Soil

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## Acknowledgements

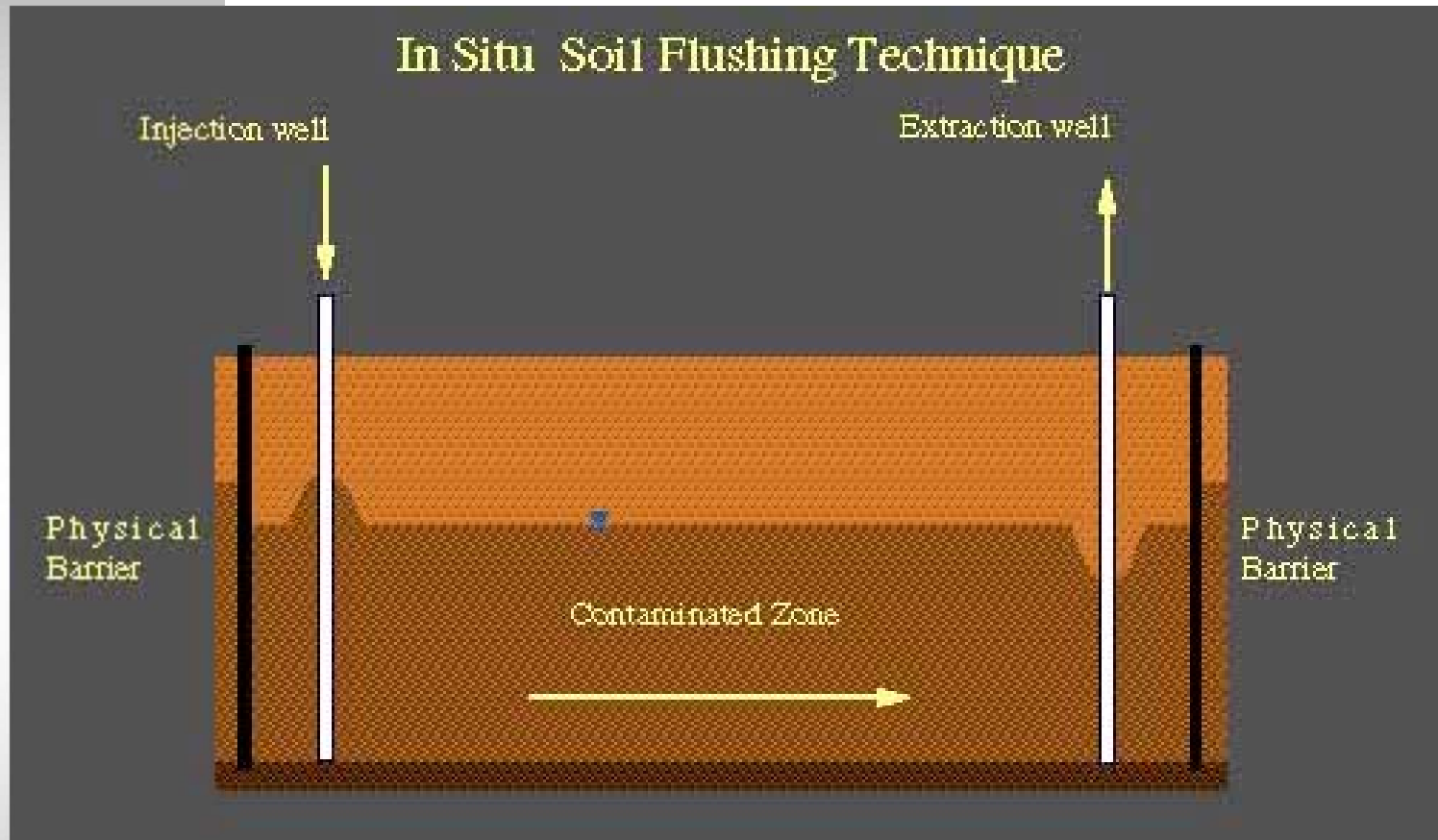
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- Environment Canada
- The City of Calgary

## Scope of the presentation

- ✓ Principle of in-situ flushing
- ✓ Lignin derivatives as prospective flushing agents
- ✓ Work objectives
- ✓ Experimental setup
- ✓ Results and discussions
  - Petroleum hydrocarbons
  - Heavy metals
- ✓ Conclusions and recommendations

# Principle of the process



Source: P.Benchimol, M.Pandit "Remediation of Contaminated Soils..."  
Civil Eng. Dept., Virginia Tech, 1998

## Main features of in-situ flushing

- ✓ Lower costs associated with in-situ treatment (no soil excavation/ building demolition required)
- ✓ Minimal interruption of commercial/industrial activities at the site
- ✓ Treatment rates are generally slower than for ex-situ treatment
- ✓ “Open” treatment system

## Lignosulfonates as prospective flushing agents

- By-product of pulp and paper industry (~3,000,000 t/year worldwide).
- Uses: vanillin, industrial surfactants, polymer fillers, etc.
- Chemical properties: phenolic, carboxylic, aldehyde groups.
- Act as mild surfactants
- Bind metal ions.

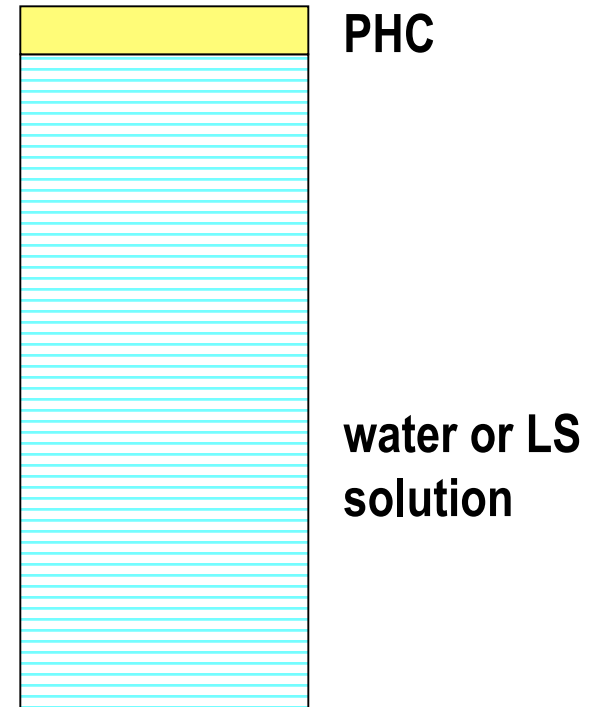
## Work objectives

- ✓ Bench-scale study
  - To evaluate commercially available lignosulfonates as flushing agents in in-situ treatment of petroleum contaminated soils
- ✓ In case of a successful treatment
  - Provide recommendations for for a pilot-scale study



## Experimental setup

- ✓ Phase I – aqueous solubility tests
- ✓ Oil added to 100 ml of water or LS solution
- ✓ Parameters under evaluation: type and concentration of LS, pH, and contact time





## Experimental setup

- ✓ Phase II – slurry leaching
- ✓ 100-200 g samples of contaminated soil (spiked and actual)
- ✓ Parameters under evaluation: type and concentration of LS, pH, and contact time



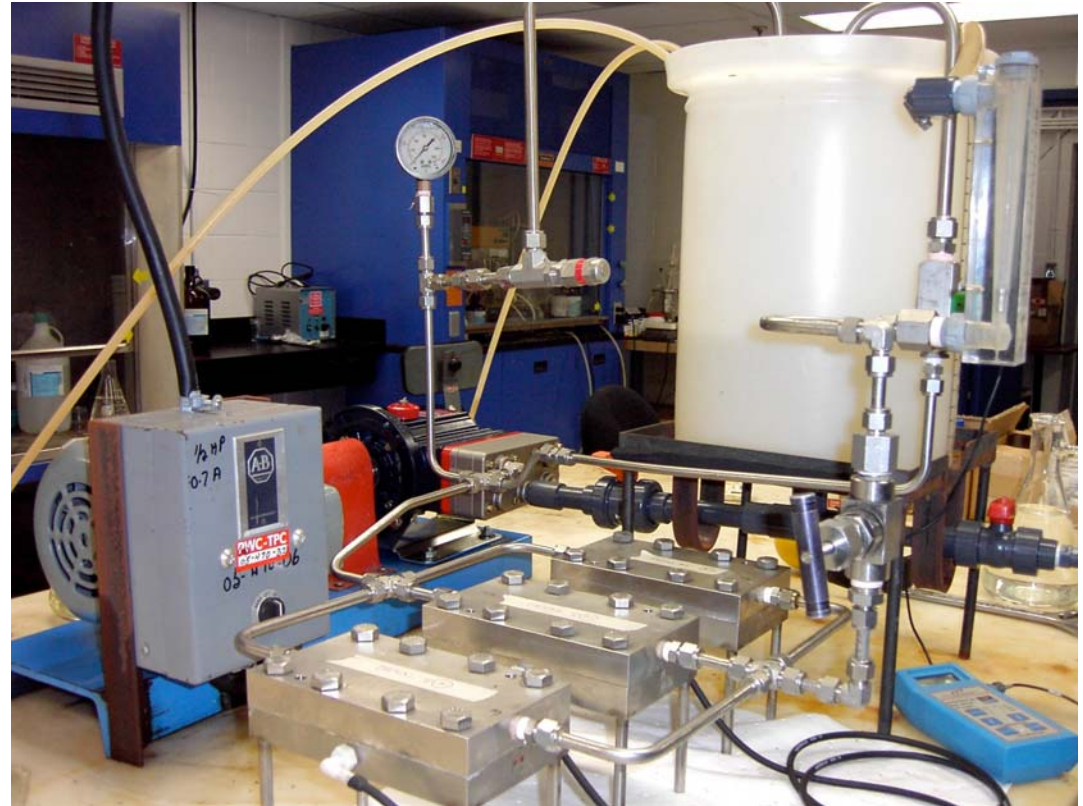
## Experimental setup

- ✓ Phase III – column leaching
- ✓ 1,000-1,500 g samples of contaminated soil
- ✓ Parameters under evaluation: volume and concentration of flushing solution, pressure, and contact time

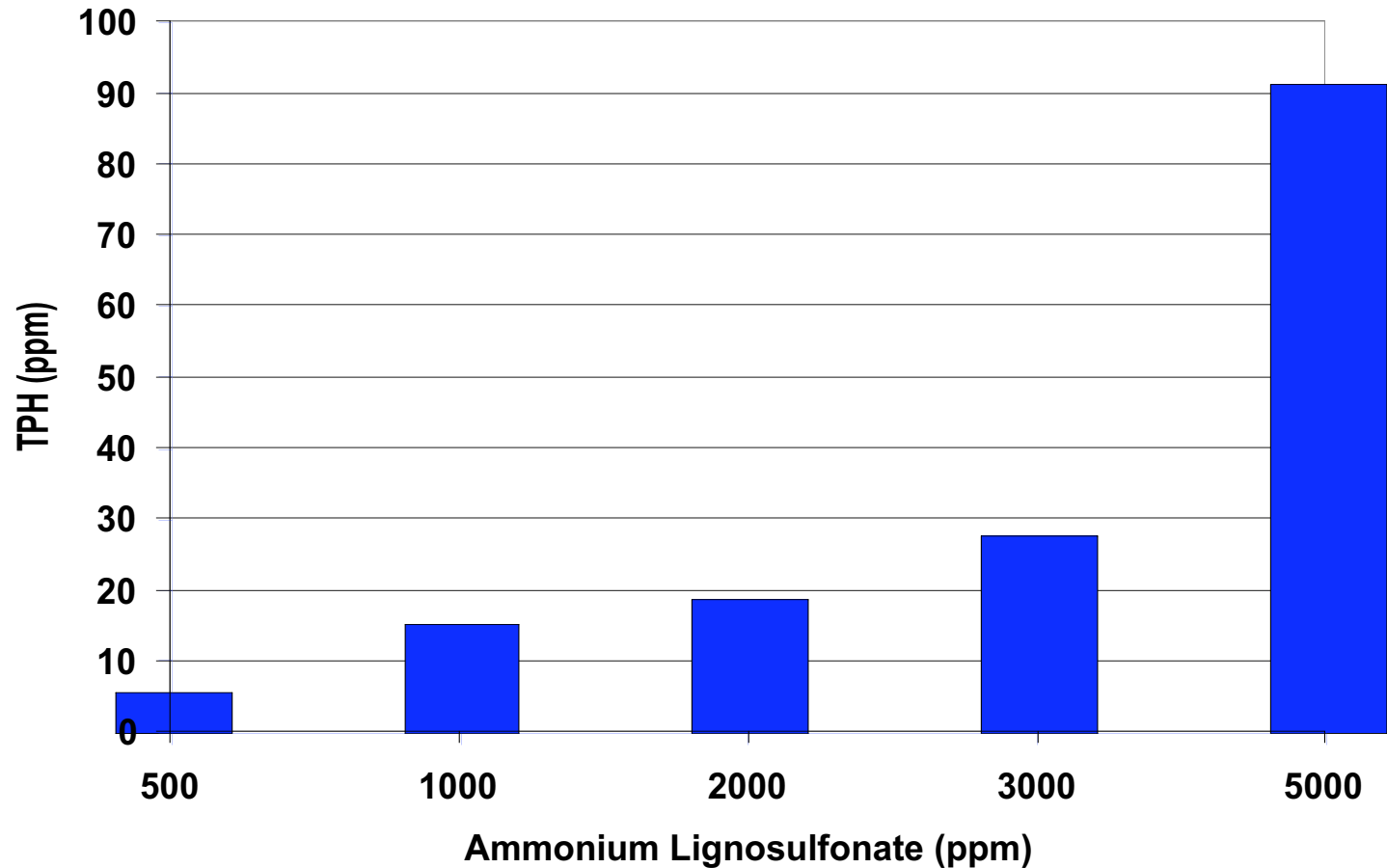


## Experimental setup

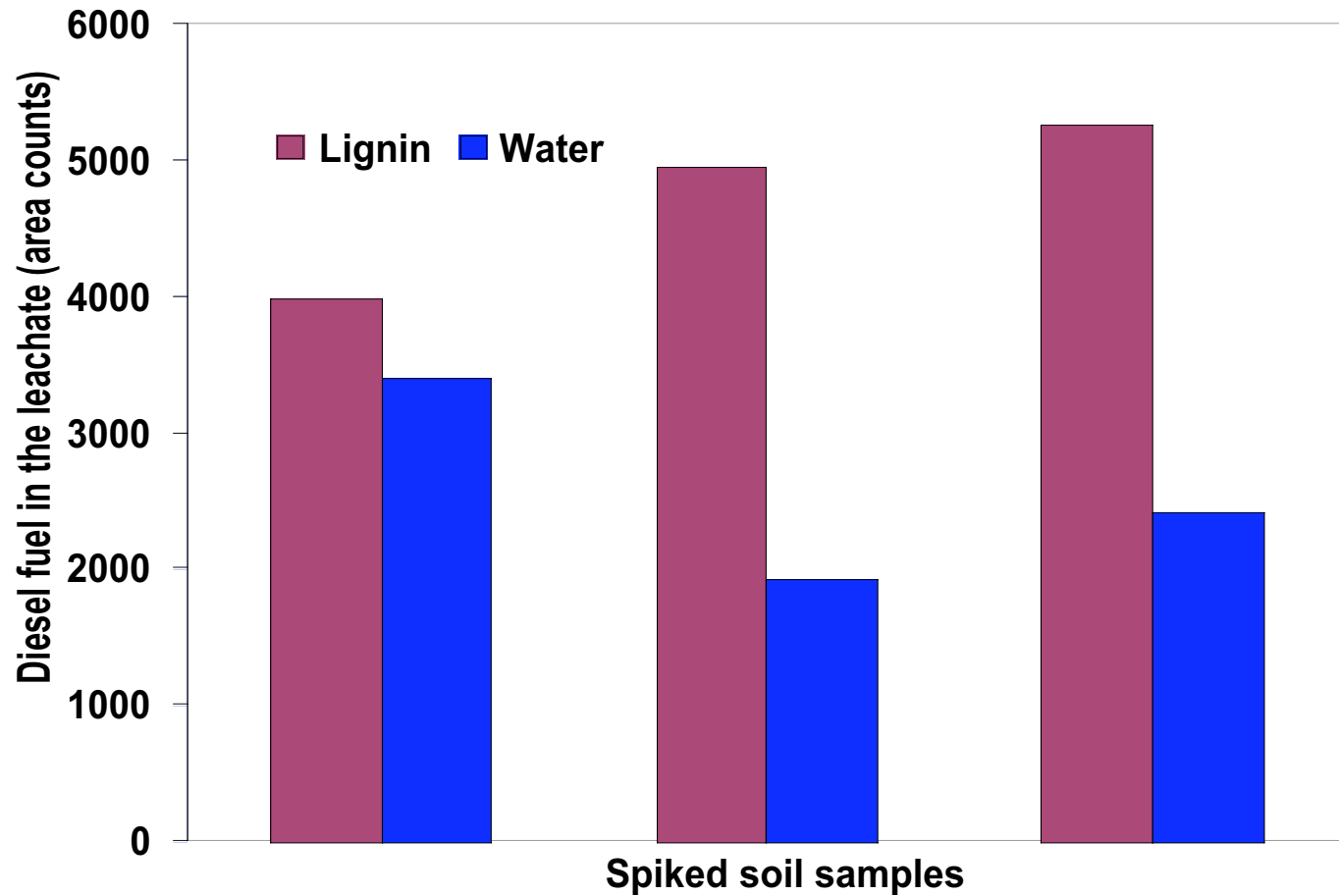
- ✓ Phase IV – leachate treatment
- ✓ Membrane filtration used to concentrate contaminants and reduce the leachate volume



# Test results: The effect of lignosulfonate concentration on the solubility of petroleum hydrocarbons

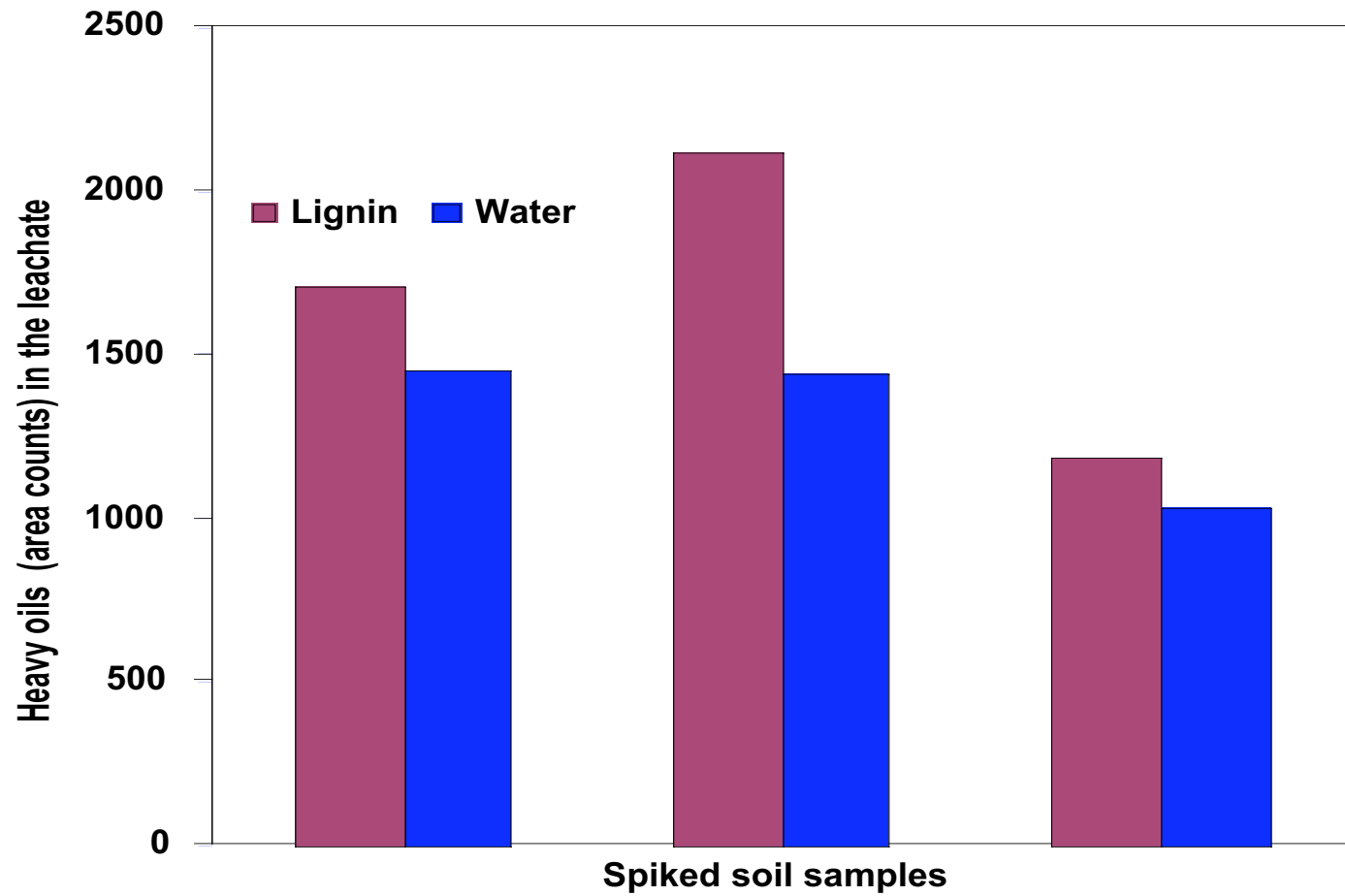


## Results of slurry leaching: Diesel fuel

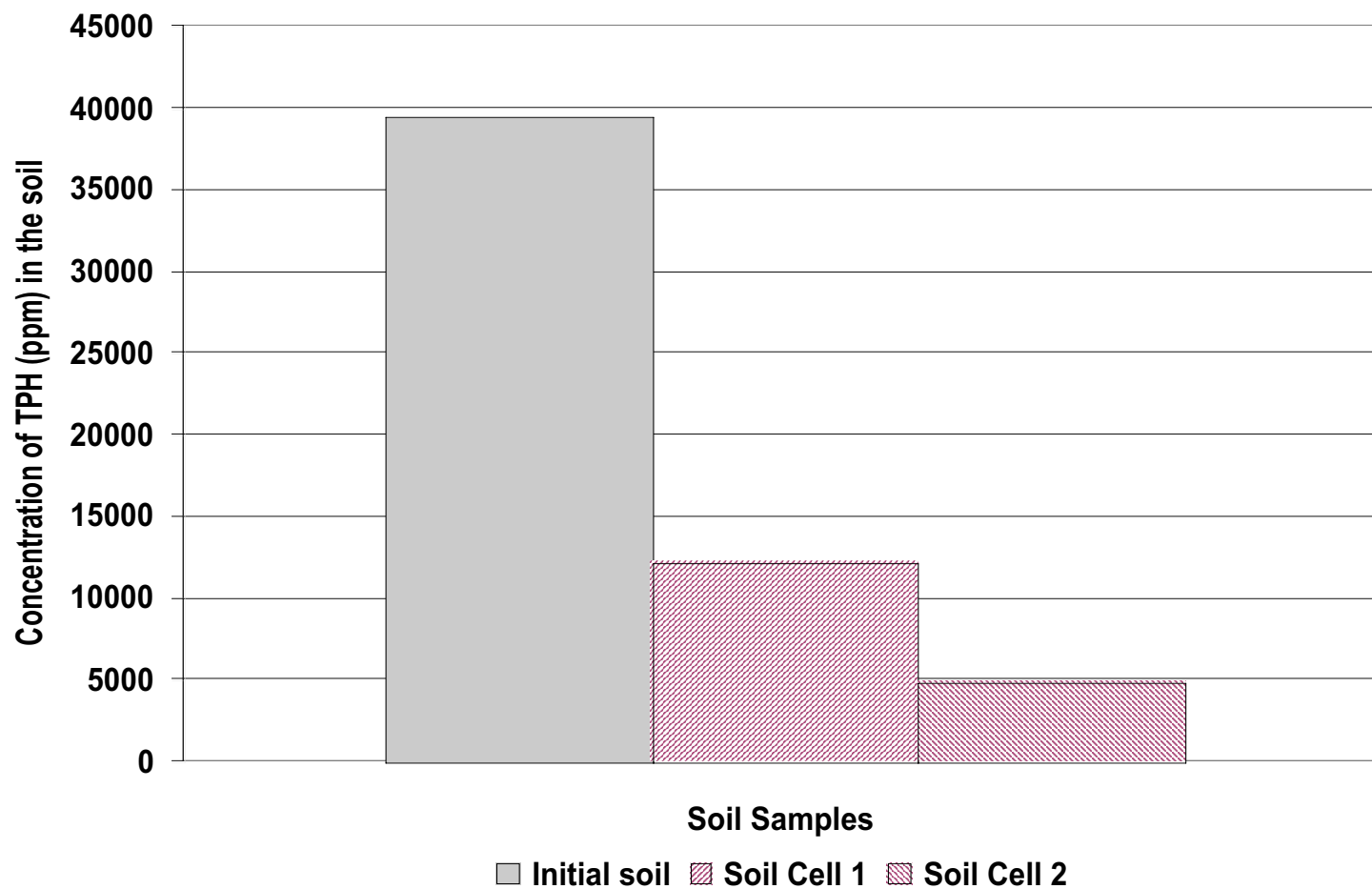




## Results of slurry leaching: Heavy oils



## Results of column leaching: Total petroleum hydrocarbons





## Test results: Leachate treatment



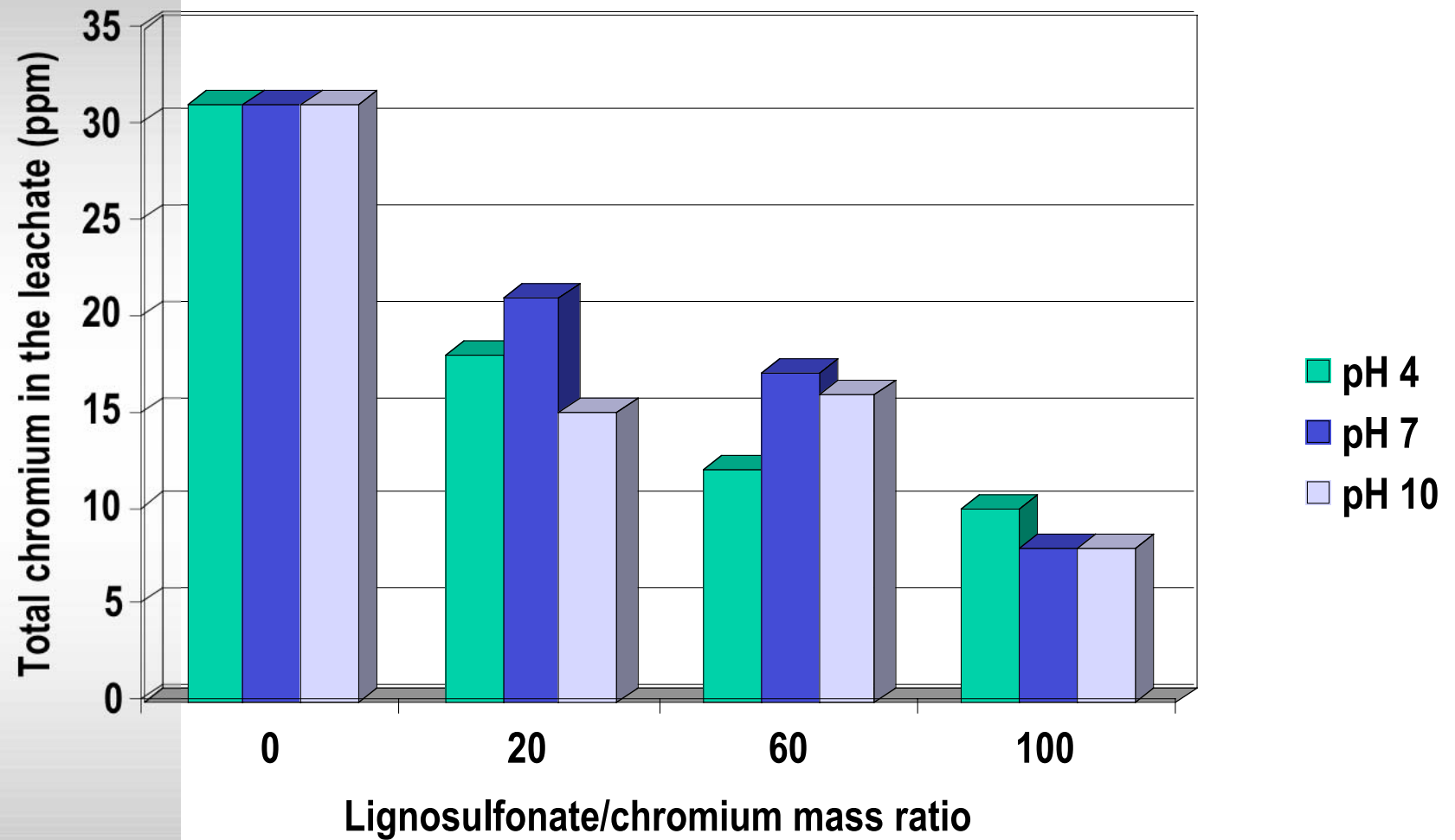
- ✓ Permeates generated using semi-permeable membranes Desal-5, G50 and G20 (all of GE Osmonics)
- ✓ Concentrate is biodegradable

## Removal of heavy metals

| Contaminants* | Percentage removal     |                         |
|---------------|------------------------|-------------------------|
|               | “conventional” process | with lignin derivatives |
| Hg            | 0%                     | 15%                     |
| U             | 0%                     | 42%                     |
| Cd            | 26%                    | 70%                     |
| Cr (III)      | 0%                     | 24%                     |
| Pb            | 29%                    | 75%                     |

\* Initial concentrations: 500 mg/L

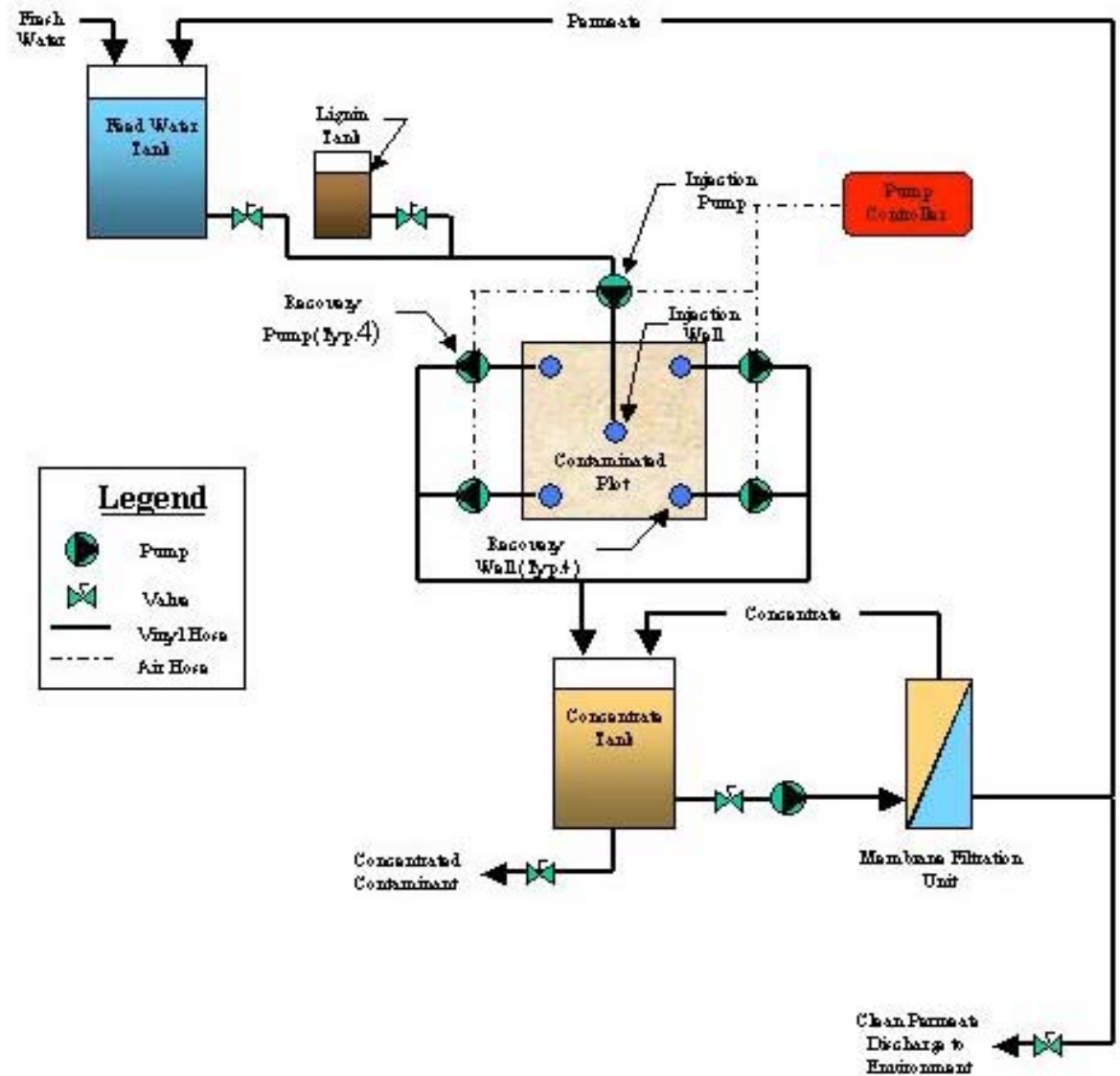
# Stabilization of hexavalent chromium



## Conclusions and recommendations

- ✓ Lignosulfonates enhance the removal of petroleum hydrocarbons from soil in in-situ flushing
- ✓ Ammonium lignosulfonate is the most effective agent
- ✓ Leachate can be concentrated using membrane filtration
- ✓ Heavy metal removal observed. Possibility for the treatment of mixed contaminated soil
- ✓ Stabilization of hexavalent chromium in the soil observed
- ✓ Pilot-scale trial is recommended

# Flow chart of the proposed pilot test system



# Cross section of the treatment zone

