







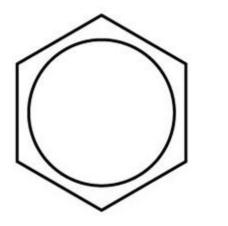
Authors

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Benzene in the environment

- BTEX family
- Benzene CAS#: 71-43-2
- Colorless liquid
- Sweet odor, smell at 60 ppm
- Dissolves in water (limited)
- Evaporates easily
- Highly flammable
- Benzene comes from industries and nature
- It breaks slowly in water/soil

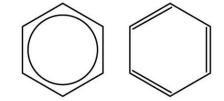






Reactions - Types of Substitutions

• Nitration: $C_6H_6 + HNO_3 \longrightarrow C_6H_6NO_2 + H_2O$



- Halogenation: Carbon Replaced by Halogen
- Sulfonation: Carbon Replaced by Sulfer
- Friedel-crafts: Acyl group (RCO-) Attaches to Ring to Make a Keytone

Physical Properties

	Formula	Molecular Weight (g/mol)	Vapor Pressure mm Hg at 25°C	Odor Threshold	Log K _{ow} (Octanol /Water)
Benzene	C ₆ H ₆	78.11	95.1	1.5 ppm (5 mg/m3)	2.13

Sources of Benzene

- Origins
 - Old time: 1800s from coal tar
 - Modern: Petroleum (top 20 USA production)
- Industries: Styrofoam, resin, nylon, rubber, lubricant, dyes, detergent, drug & pesticides
- Nature: gas emission, volcano, forest fires



Health and Environmental Hazard

- Exposure by inhalation and skin
- Exposure from products (glue, tobacco, etc.)
- Emissions of car and industries: 20% total
- Permissible level in water 0-5 ppb
- 10ppb/water 0.4ppb/air, 1 extra cancer in 100,000



Legal Facts

- July 1, 1999, Gasoline supply less than 1%
- Test Methods: D2163 14e1
- Benzene is classified as carcinogenic to humans (i.e., as a non-threshold toxicant - a substance for which there is considered to be some probability of harm for the critical effect at any level of exposure)
- It is considered to be "toxic" as defined under Paragraph 11(c) of the Canadian Environmental Protection Act.
- On the basis of available data, benzene is not considered to be "toxic" as defined under Paragraphs 11(a) and 11(b) of CEPA (Canadian Council of Ministers of the Environment).



Benzene Derivatives (properties)

	Formula	Molecular Weight (g/mol)	Vapor Pressure mm Hg at 25°C	Odor Threshold	Log K _{ow} (Octanol /Water)	
Benzene	C ₆ H ₆	78.11	95.1	1.5 ppm (5 mg/m3)	2.13	
Toluene	C ₇ H ₈	92.15	28.4	2.9	2.69	
Ethylbenzene	C ₈ H ₁₀	106.16	9.53	2.3	3.13	сн ₃ сн ₃
Xylene	C ₈ H ₁₀	106.16	6.728	1.1 (as per m-xylene)	3.12	



Toxicity of Toluene, Ethylbenzene & Xylenes

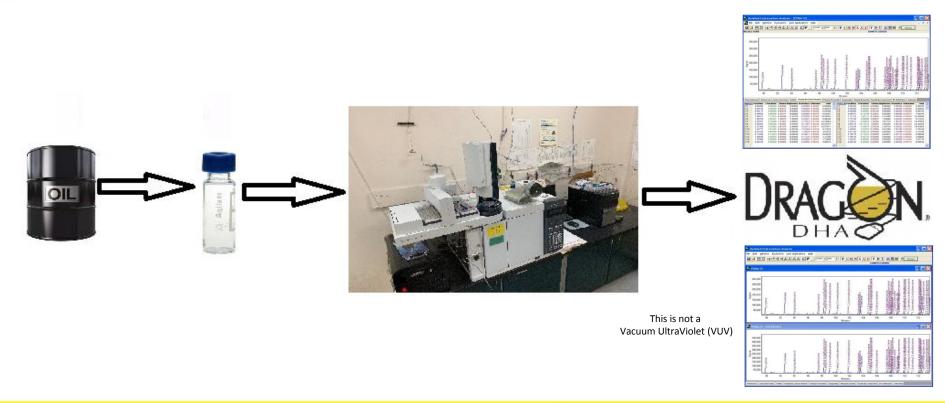
- Toxic to Vertebrate, Invertebrate, Plants, Earthworm, wheatgrass, etc.
- Toxic as low as 1 mg/L
- Cause acute and chronic effects (CNS)
- Acute: CNS dysfunction, fatigue, sleepiness, headaches, and nausea, swollen liver, congestion and hemorrhage of the lungs, tubular kidney necrosis, and impaired respiratory function.
- Chronic: drowsiness, ataxia, tremors, cerebral atrophy, nystagmus (involuntary eye movements), and impaired speech, hearing, and vision



Research Question

 Are there more of similar benzene compounds in sources such as oil?

Methodology - Method



Results

- A total of 48 benzene compounds
- BTEX contain 6 in total
- There are 42 more benzene like compounds

Identified Benzene Species (1)

	Component Name	Mol %	Molecular Weight	Density	Structure
1	benzene	0.3124	78.114	0.8789	
2	Ethylbenzene	0.4932	106.168	0.867	CH ₃
3	1,3-dimethylbenzene (m-Xylene)	0.3615	106.168	0.8642	CH ₃
4	1,4-dimethylbenzene (p-Xylene)	0.2728	106.168	0.861	CH ₃
5	1,2-dimethylbenzene	0.2997	106.168	0.8802	CH ₃
6	i-propylbenzene (Cumene)	0.0838	120.195	0.8618	CH ₃
7	n-propylbenzene (phenylpropane)	0.1736	120.195	0.862	
8	1,3-methylethylbenzene	0.2783	120.195	0.8645	

Identified Benzene Species (2)

	Component Name	Mol %	Molecular Weight	Density	Structure
9	1,4-methylethylbenzene	0.1407	120.195	0.8612	H ₃ C
10	1,3,5-trimethylbenzene	0.1128	120.195	0.8652	CH ₃
11	1,2-methylethylbenzene	0.2246	120.195	0.8807	
12	1,2,4-trimethylbenzene	0.2884	120.195	0.8758	H ₃ C CH ₃
13	i-butylbenzene	0.0129	134.222	0.8532	
14	sec-butylbenzene	0.0917	134.222	0.862	

Identified Benzene Species (3)

	Component Name	Mol %	Molecular Weight	Density	Structure
15	1,2,3-trimethylbenzene	0.12	120.195	0.8944	<u></u>
16	1,3-methyl-i-propylbenzene	0.0703	134.222	0.861	JJ
17	1,4-methyl-i-propylbenzene	0.0669	134.222	0.8573	
18	1,2-methyl-i-propylbenzene	0.2116	134.222	0.8766	
19	1,3-diethylbenzene	0.0832	134.222	0.8639	H H M



Identified Benzene Species (4)

	Component Name	Mol %	Molecular	Density	Structure
20	1,3-methyl-n-propylbenzene	0.1035	Weight 134.222	0.8609	CH ₃
					H ₃ C CH ₃
21	1,4-diethylbenzene	0.0366	134.222	0.862	
22	1,4-methyl-n-propylbenzene	0.0559	134.222	0.8584	CH ₃
					п,
23	n-butylbenzene	0.0582	134.222	0.861	
24	1,3-dimethyl-5-ethylbenzene	0.0613	134.222	0.88	H H

Identified Benzene Species (5)

	Component Name	Mol %	Molecular Weight	Density	Structure
25	1,2-diethylbenzene	0.0846	134.222	0.8799	H H H H
26	1,2-methyl-n-propylbenzene	0.2263	134.222	0.8736	+
27	1,3-dimethyl-2-ethylbenzene	0.024	134.222	0.8904	
28	1,4-methyl-t-butylbenzene	0.0435	148.24	0.85	CH ₃ CH ₃

Identified Benzene Species (6)

		Component Name	Mol %	Molecular Weight	Density	Structure
2	29	1,2-dimethyl-3-ethylbenzene	0.1057	134.222	0.8921	H H H H H
3	30	1,2-ethyl-i-propylbenzene	0.0161	148.24	0.89	CH ₃
3	31	1,2,4,5-tetramethylbenzene	0.0614	134.222	0.8875	
3	32	1,2-methyl-n-butylbenzene	0.0163	148.24	0.89	

Identified Benzene Species (7)

	CN	NA - LO/	NA - I I	D 14	Characteria
	Component Name	Mol %	Molecular	Density	Structure
			Weight	ı	
33	1,2,3,5-tetramethylbenzene	0.0961	134.222	0.8903	
34	1,2-methyl-t-butylbenzene	0.0054	148.24	0.89	
35	1,2-ethyl-n-propylbenzene	0.1794	148.24	0.89	
36	1,3-methyl-n-butylbenzene	0.0584	148.24	0.89	

Identified Benzene Species (8)

	Component Name	Mol %	Molecular Weight	Density	Structure
37	1,3-di-i-propylbenzene	0.0616	162.272	0.89	
38	s-pentylbenzene	0.0516	148.24	0.89	
39	n-pentylbenzene	0.1	148.24	0.89	
40	1,2-di-i-propylbenzene	0.0857	162.272	0.89	

Identified Benzene Species (9)

	Component Name	Mol %	Molecular Weight	Density	Structure
41	1,4-di-i-propylbenzene	0.2529	162.272	0.89	
42	1,4-ethyl-t-butylbenzene	0.1247	162.272	0.89	
43	1,3-di-n-propylbenzene	0.0519	162.272	0.89	

Identified Benzene Species (10)

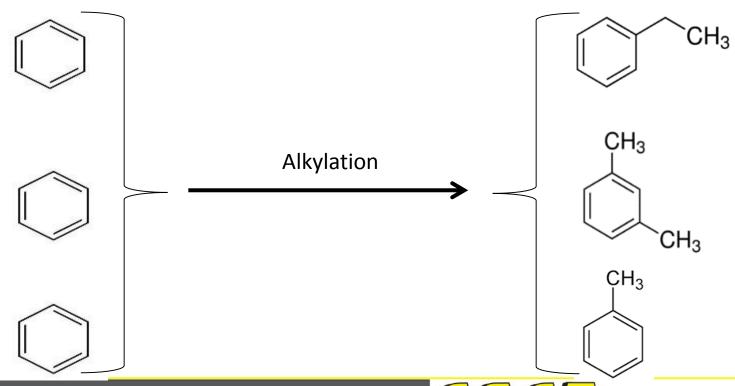
	Component Name	Mol %	Molecular	Density	Structure
			Weight		
44	1,3,5-triethylbenzene	0.0354	162.272	0.8897	H H H H H H H H H H H H H H H H H H H
45	1,2,4-triethylbenzene	0.0771	162.272	0.8897	H H H H H H H H H H H H H H H H H H H
46	1,4-methyl-n-pentylbenzene	0.0688	162.272	0.8897	
47	n-hexylbenzene	0.1058	162.272	0.8897	
48	1,2,3,4,5-pentamethylbenzene	0.1097	148.24	1	



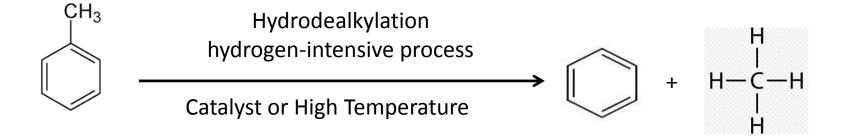
Concentrations

- BTEX: 1.7396 Mol%
- Other Alkylated Benzenes: 4.3167 Mol%

Alkylation of Benzene



Hydrodealkylation



Schedule 1 (July 11, 2018)

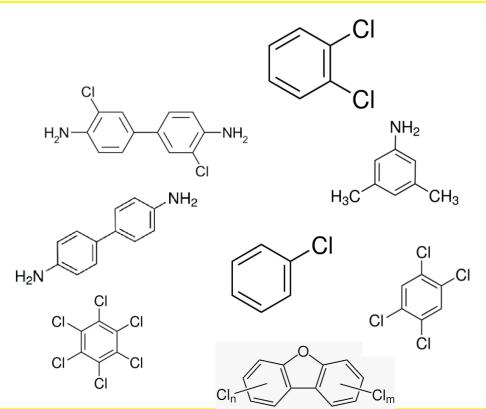
- Benzene
- Hexachlorobenzene
- Tetrachlorobenzenes
- Pentachlorobenzene
- 1,2-Benzenediol
- 1,4-Benzenediol

- Benzene, (chloromethyl)-,
- Benzene, 1-methyl-2nitro-
- Benzene, 1,2dimethoxy-4-(2propenyl)-



Priority Substances List 1 (PSL1), 1989

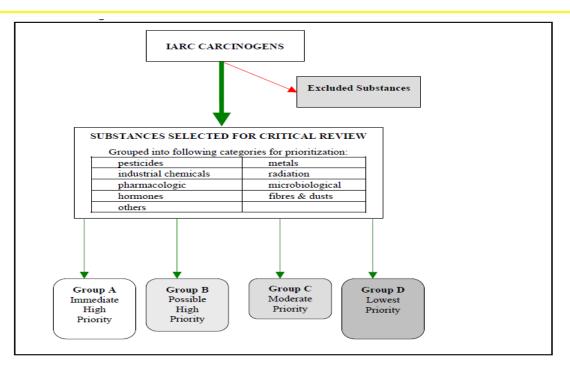
- 1,2-Dichlorobenzene
- 1,4-Dichlorobenzene
- 3,5-Dimethylaniline
- Benzidine
- Chlorobenzene
- Hexachlorobenzene
- Polychlorinated Dibenzofurans
- Tetra/Trichlorobenzenes
- PAHs, Toluene, Benzene, Xylenes



Priority Substances List 2 (PSL2) (CEPA) December, 1995

- Phenol
- Nonylphenol

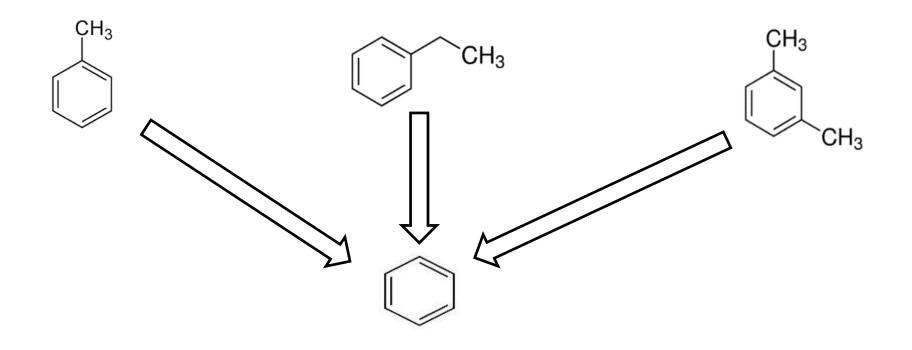
CAREX Canada – Prioritization Process Occupational



CAREX Canada is a national surveillance project that estimates the number of Canadians exposed to substances associated with cancer in workplace and community environments.



Does A Reverse Process Exist?



An Example of Dealkylation

Remember...

- BTEX: 1.7396 Mol%
- Other Alkylated Benzenes: 4.3167 Mol%

Conclusion and Suggestions

- Alkylated benzene compounds deemed to be toxic due to their structural similarity with BTEX compounds.
- Toxicity and carcinogenicity of all benzene compounds need to be assessed.
- Any petroleum product must be assessed for new benzene compounds.
- Pyrolysis derivatives of benzene contain products can be measured for potential hazard.
- PIONA GC-FID is an easy to operate instrument and cost effective for alkylated benzene content analysis.





Laboratory solutions

Environmental,

Energy, Mining,

Agri-food, Industrial

Transportation

Life Sciences

Thank You

Questions and Discussion



