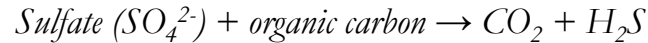


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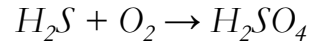
*West Geomatics and Environmental Services: helping clients address water, wastewater, reclamation, remediation, closure and abandonment challenges by applying leading-edge tools, innovation and expertise combined with an uncompromising commitment to safety and sustainability.*

# $H_2S$ Occurs Commonly

- *Within hydrocarbon reservoirs through microbial and thermochemical sulfate reduction*
- *By-product of microbial digestion within anaerobic (oxygen-depleted <0.1 mg/L) environments by sulfate-reducing bacteria:*



- *In sewers and wastewater piping, sulfuric acid is generated when  $H_2S$  is absorbed into a film (slime) on the crown of the pipe and sulfur-oxidizing bacteria react:*



- *$H_2S$  can be found in sewers, silos, sumps, pits, ponds, mine adits, landfills, wastewater treatment plants, manure piles, hot asphalt paving, food processing, pulp and paper processing, tanning and textile manufacturing*

## *H2S Fatalities Occur Over Wide Spectrum of Industries*

- *Agricultural operation pump house alarm investigation*
- *Grease trap vault cleaning*
- *Bulk milk tank cleaning*
- *Poultry blood wastewater tank cleaning*
- *Waste heat boiler valve replacement in sulfur plant*
- *Water well confined space*
- *Paper mill sulfuric acid spill reaction with wastewater with gas escaping through manhole cover*
- *Utility construction trench*
- *Sewer trench installation*
- *Sewer plug removal*
- *Gas plant*
- *Pump replacement machine shop*
- *Removal of old sewer pipes*
- *Sump pump replacement in wet well*
- *Asphalt mix tank chemical addition*
- *Oil battery tank gauging*
- *Heater Treater Repair*
- *Wastewater Treatment Plant leak repair*
- *Fish factory wastewater slurry pipe installation*
- *Sewer pipe cleaning*
- *Restaurant sewer pump repair*
- *Hog farm utility corridor*
- *Gas well abandonment*
- *Dry well repair*
- *Sewer lift station inspection*
- *Shipyards tank cleaning*
- *Sewer lift station repair*
- *Sewer pipe replacement*
- *Cement truck cleaning*
- *Tannery tank filling*

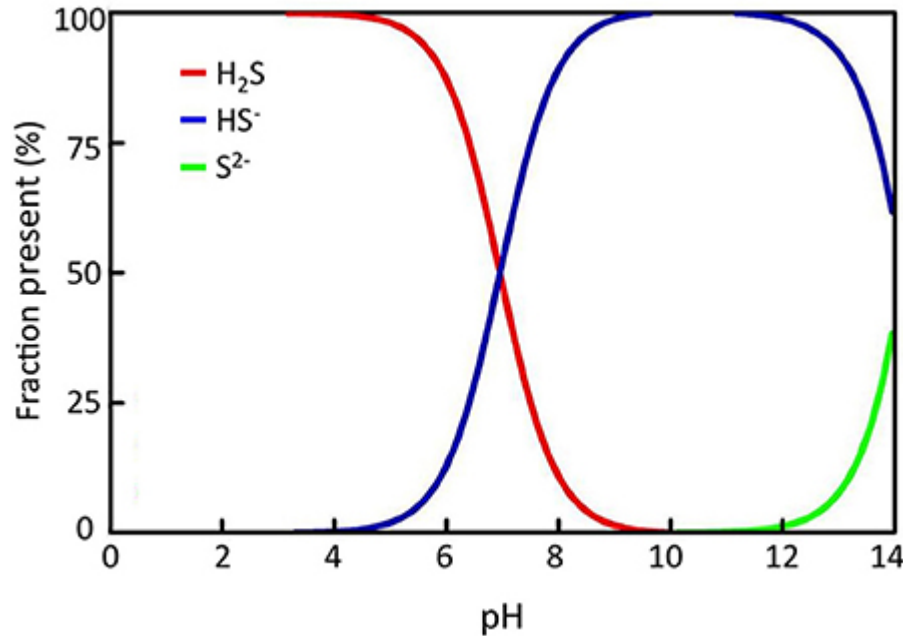
## *Higher risk industries*

- *Hendrickson et al (2004) published analysis of 7 years of data from the 90's with important findings*
- *50% of fatalities occur within 1<sup>st</sup> year of employment*
- *Fatalities were most common in waste management (24%) and petroleum/natural gas processing (18%)*
- *Poisonings were most common in sewerage (33%) and petroleum manufacturing (23%)*
- *35% of fatalities were in companies with greater than 100 employees with 24% in companies less than 10 workers*

# *Physical and Chemical Properties*

- *Colourless, flammable and highly toxic gas*
- *Denser than air = 1.2 times*
- *Slightly water soluble (0.4% at 20 °C)*
- *Highly flammable and explosive between 4% and 45% (concentration in air)*
- *Solubility changes with water temperature*

*Dissolved  $H_2S$  Gas vs Dissolved  $HS^-$  Ions? It depends...*

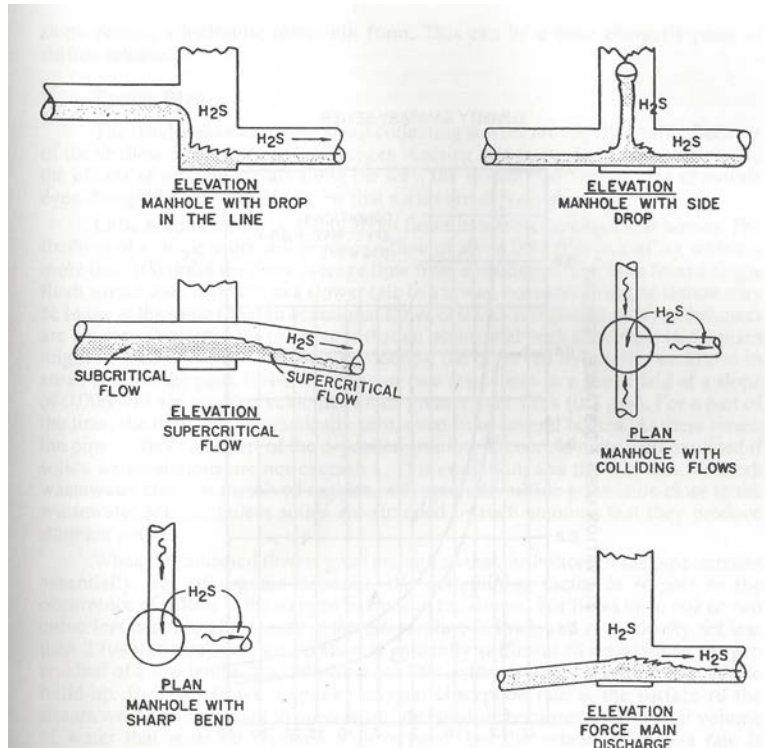


## *H<sub>2</sub>S Release Affected by pH, Temperature and Turbulence*

- *Lowering pH increases the amount of dissolved gas that can evolve especially in turbulent conditions*
- *Lowering temperature increases H<sub>2</sub>S solubility in sewage: 5160 mg/L @ 10 °C vs 3925 mg/L @ 20 °C*
- *Increased turbulence improves H<sub>2</sub>S release into headspace*



# Examples of turbulent flows releasing $H_2S$



Source: ASCE Manual 69

# Hierarchy of Action Levels

<i>Criteria</i>	<i>Description</i>	<i>Threshold</i>
<i>Olfactory detection</i>	<i>Offensive smell</i>	<i>as low as 0.5 ppb</i>
<i>Sewer bylaws</i>	<i>Prohibited substance</i>	<i>0</i>
<i>Sewer bylaws</i>	<i>Restricted substance</i>	<i>Dissolved sulphide &lt;1 ppm in effluent</i>
<i>ACGIH (CUPE)</i>	<i>8-hour Threshold Limit Value (TLV)</i>	<i>1 ppm</i>
<i>ACGIH (CUPE)</i>	<i>15-minute Short-Term Exposure Limit (STEL)</i>	<i>5 ppm</i>
<i>Alberta OH&amp;S Regulation 87-2009</i>	<i>8-hour Occupational Exposure Limit (OEL)</i>	<i>10 ppm</i>
<i>Alberta OH&amp;S Regulation 87-2009</i>	<i>15-minute ceiling (OEL)</i>	<i>15 ppm</i>
<i>Respiratory Distress</i>	<i>Irritation nose, throat, lungs</i>	<i>50 ppm</i>
<i>National Institute for Occupational Safety and Health (NIOSH)</i>	<i>Immediate Danger to Life of Health (IDLH)</i>	<i>100 ppm</i>
<i>Lethal</i>	<i>Cardiac arrest</i>	<i>1000 ppm</i>

# *Key Considerations for Managing Safety*

- *Identify potential sources of H<sub>2</sub>S especially anaerobic digestion of organic matter*
- *Identify downstream conditions that can affect releases especially chemical addition, turbulence, change in pH, change in temperature*
- *Address upset conditions that disturb equilibrium of system*
- *Provide adequate detection: both process and personal*
- *Implement engineering controls to prevent generation*
- *Adopt operating protocols to react immediately*

# *H<sub>2</sub>S Treatment Options*

- *Technology Selection Depends Upon Desired Outcome:*
  - *meeting effluent discharge criteria*
  - *odour elimination*
  - *occupational health and safety*
  - *reduce damage to infrastructure*
- *Technology types:*
  - *physical-mechanical*
  - *chemical*
  - *biological*
  - *combinations*

# $H_2S$ Treatment Technologies: High Water Volume

<i>Technology</i>	<i>Description</i>
<i>Modular Stripper-Polisher</i>	<i>5-stage process (1) inlet separator (2) natural gas mixer (3) flash vessel (4) polishing chemical (5) off-gas to recovery or flare</i>
<i>Conventional Stripper (Debolt)</i>	<i>5-stage process (1) inlet separator (2) degassing tower (3) stripping tower (4) polishing chemical (5) off-gas to incinerator</i>
<i>Oxidation: Hydrogen peroxide <math>H_2O_2</math> (Permian)</i>	<i>5 stage process (1) oxidant addition (2) reactor (3) filtration (<math>5\ \mu\text{m}</math>) (4) residual scavenger (5) elemental sulfur disposal</i>
<i>Oxidation: Chlorine dioxide <math>ClO_2</math> (Permian)</i>	<i>5-stage process (1) oxidant addition (2) reactor (3) filtration (<math>5\ \mu\text{m}</math>) (4) residual scavenger (5) elemental sulfur disposal</i>
<i>Oxidation: ozone (Permian)</i>	<i>4-stage process (1) hydrodynamic cavitation (2) ozonation (3) acoustic cavitation (4) electrochemical oxidation</i>
<i>Scavenger (Permian)</i>	<i>4-stage process (1) gas scrubber (2) scavenger (3) tank scrubber (4) sweet water disposal/recycle</i>
<i>Triazine (Common)</i>	<i>Direct injection or contactor tower.</i>

# $H_2S$ Treatment Technologies: Low Volume

Technology	Description
Chlorine dioxide $ClO_2$	Prevent formation of $H_2S$ (water). Applied at high doses to remove biofilm layer without forming colloidal sulfur.
Sodium Nitrate $NaNO_3$	Prevent formation of $H_2S$ (water). Applied to retard septicity by providing alternative source of oxygen to anaerobic microorganisms rather than sulfate. $BOD + NO_3^- \rightarrow N_2 + CO_2 + \text{biomass}$
Sodium Chlorite $NaClO_2$	Remove $H_2S$ (water). Selectively oxidizes $H_2S$ without reacting with ammonia or forming chlorinated organics. Used for remote sites for long duration control. $2H_2S + NaClO_2 \rightarrow 2H_2O + NaCl$
Hydrogen Peroxide $H_2O_2$	Remove $H_2S$ (water). Oxidizes $H_2S$ and decomposes into $O_2$ and $H_2O$ . $H_2S + H_2O_2 \rightarrow S^0 + 2H_2O$ (pH < 8.5) $H_2S + 4H_2O \rightarrow SO_4^{2-} + 2H_2O$ (pH > 8.5)
Nitrate $NO_3$	Removal (water). Bacteria utilize nitrate for biooxidation of $H_2S$ . $8NO_3^- + 5H_2S \rightarrow 5SO_4^{2-} + 4N_2$
Iron Salts: Ferrous $Fe^{2+}$ Ferric $Fe^{3+}$	Removal (water). React with dissolved $H_2S$ to form precipitate. $Fe^{2+} + HS^- \rightarrow FeS + H^+$ $2Fe^{3+} + 3HS^- \rightarrow Fe_2S_3 + 3H^+$
Iron Sponge	Removal (gas). React with $H_2S$ to produce water and ferric sulfide. $FeO_3 + 3H_2S \rightarrow Fe_2S_3 + 3H_2O$
Carbon Absorption	Impregnated with sodium hydroxide (NaOH) or potassium hydroxide (KOH) to increase affinity for $H_2S$
Scrubbing	Liquid packed bed or misting system in which foul air is mixed with pH-controlled water or vapour

# Variety of Treatment Applications



*Beverly Hills WTP*



*Horn River Flowback*



*Produced Water and Flowback*

# *Detection Methods*

- *Gas detection tubes: colorimetric test in which H<sub>2</sub>S reacts with lead acetate coated beads (Gastec, Matheson-Kitagawa)*
- *Passive sampler badges (Morphix)*
- *Electrochemical handheld (Odalog, Jerome)*
- *Electrochemical fixed (Cti)*
- *Tunable diode laser fixed (Galvanic)*
- *Tape-based fixed (Galvanic)*
- *Open path laser fixed (Boreal)*



# *Due Diligence Factors*

- *H<sub>2</sub>S Alive Training*
- *Engineering Controls*
- *Detection Equipment*
- *Personal Protective Equipment*

# Lessons-Learned

- *Identify conditions that generate H<sub>2</sub>S*
- *Understand how changes in water chemistry alter availability: pH, T and turbulence*
- *Understand how changes in air flow distribute H<sub>2</sub>S*
- *Address root causes of potential problems with engineering controls*
- *Ensure sufficient treatment capacity is in place to deal with upsets*
- *Ensure personal monitors in place where there is still a risk of exposure*
- *Evacuate first: ask questions later*

# *Questions or Suggestions*

[bill.berzins@westx.com](mailto:bill.berzins@westx.com)

*Bill Berzins, M.A.Sc, P.Eng. (BC,AB,SK)*

*403.807.2782*